



Stratospheric ozone depletion inside the volcanic plume shortly after the 2022 Hunga Tonga eruption

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Abstract. Near-term in-plume ozone depletion was observed for about 10 d by the Aura Microwave Limb Sounder (MLS) right after the January 2022 Hunga Tonga–Hunga Ha’apai (HTHH) eruption. This work analyzes the dynamic and chemical causes of this ozone depletion. The results show that the large water injection (~ 150 Tg) from the HTHH eruption, with ~ 0.0013 Tg injection of ClO (or ~ 0.0009 Tg of HCl), causes ozone loss due to strongly enhanced HO_x and ClO_x cycles and their interactions. Aside from the gas-phase chemistry, the heterogeneous reaction rate for $\text{HOCl} + \text{HCl} \rightarrow \text{Cl}_2 + \text{H}_2\text{O}$ increases to $10^4 \text{ cm}^{-3} \text{ s}^{-1}$ and is a major cause of chlorine activation, making this event unique compared with the springtime polar ozone depletion where $\text{HCl} + \text{ClONO}_2$ is more important. The large water injection causes relative humidity over ice to increase to 70 %–100 %, decreases the $\text{H}_2\text{SO}_4 / \text{H}_2\text{O}$ binary solution weight percent to 35 % compared with the 70 % ambient value, and decreases the plume temperature by 2–6 K. These changes lead to high heterogeneous reaction rates. Plume lofting of ozone-poor air is evident during the first 2 d after the eruption, but ozone concentrations quickly recover because its chemical lifetime is short at 20 hPa. With such a large seawater injection, we expect that ~ 5 Tg Cl was lifted into the stratosphere by the HTHH eruption in the form of NaCl, but only ~ 0.02 % of that remained as active chlorine in the stratosphere. Lightning NO_x changes are probably not the reason for the HTHH initial in-plume O_3 loss.

1 Introduction

Stratospheric ozone concentrations change after volcanic eruptions for a variety of reasons. Enhanced polar ozone depletion occurs after large or medium volcanic eruptions (Hofmann and Oltmans, 1993; Portmann et al., 1996; Solomon et al., 2016) because heterogeneous reactions on volcanically enhanced sulfate aerosols result in amplified anthropogenic ClO_x - and BrO_x -induced ozone loss. Tie and Brasseur (1995) demonstrated that mid- and high-latitude O_3 changes after a volcanic eruption largely depend on chlorine loading. For the preindustrial era and in the absence of anthropogenic halogens in the stratosphere, O_3 would slightly increase in the middle atmosphere after a large volcanic eruption due to the suppression of NO_x -catalyzed destruction by the heterogeneous creation of HNO_3 on volcanic aerosols. After the 1991 Pinatubo eruption, the radiative heating caused by volcanic aerosols perturbed the local temperature and circulation, which lifted the ozone layer and caused equatorial ozone depletion (Kinnison et al., 1994). Wang et al. (2022) reported that, in the case of the Hunga-Tonga eruption, mid-latitude ozone reduction was primarily caused by anomalous upwelling. Enhanced water can also change O_3 . In the lowermost stratosphere, H_2O injection through deep convection or tropopause cirrus clouds could change the catalytic chlorine and/or bromine free-radical chemistry and shift the total available inorganic chlorine towards the catalytically active free-radical form, ClO (Solomon et al., 1997; Anderson et al., 2012).

Evan et al. (2023) report observations of decreased O_3 and HCl and increased ClO in the first week following the Hunga Tonga–Hunga Ha’apai (HTHH) eruption at 20 hPa; these observations are related to the injected H_2O exceeding the normal range of the stratospheric variability. Here, we use the Whole Atmosphere Community Climate Model version 6 (WACCM6; Zhu et al., 2022) to analyze the dynamic and chemical contributors to this initial in-plume ozone depletion and to understand the climate model performance. A lofting plume can bring ozone-poor tropospheric air into the stratosphere and cause in-plume low ozone values compared with the surrounding stratospheric air (Yu et al., 2019). For a submarine volcanic eruption, the in-plume air composition is not only impacted by tropospheric air but also by the seawater and volcanic gases (including H_2O , CO_2 , SO_2 , HCl , HF , H_2S , S_2 , H_2 , CO , and SiF_4 .) and volcanic minerals. For the HTHH initial plume, besides high H_2O and high SO_2 , Microwave Limb Sounder (MLS) observations indicate that the in-plume air carried high CO (Fig. A1), relatively low ozone, and high ClO compared with the surrounding air. We constrain the initial plume chemical compounds based on observational data from the MLS and then analyze how stratospheric chemistry changes the plume composition. We will answer the following scientific questions:

1. What are the initial conditions in the volcanic plume?
2. What are the main causes of in-plume ozone depletion?
3. How do volcanic injections impact heterogeneous reactions that cause chlorine activation in the plume?

2 Observational data description and model setup

The MLS instrument aboard the Earth Observing System (EOS) Aura satellite was launched into a near-polar Sun-synchronous orbit in 2004. This work uses the MLS version 4 product for O_3 , ClO , temperature, and CO data during the first 10 d after the eruption, as recommended by Millán et al. (2022). The vertical resolution of these MLS products is typically around 3–5 km in the stratosphere. All data used here were screened using the methodology indicated in Livesey et al. (2022). We use the MLS H_2O data to identify the plume location and define it as regions with water vapor concentrations higher than 10 ppmv.

Vömel et al. (2022a) provide water vapor radiosonde measurements during the first three global circumnavigations of the plume. Here, we calculate the relative humidity with respect to ice (RH_i) and compare the observed values with the simulated values.

We use the 70-layer WACCM model as described in Zhu et al. (2022), injecting SO_2 (0.42 Tg) and H_2O (150 Tg). The model has a horizontal resolution of $0.9^\circ \times 1.25^\circ$ (latitude \times longitude). The injection plume in the model includes about 40 grid points. The model’s vertical resolution is about 1 km in the stratosphere. The model atmosphere is nudged to the Goddard Earth Observing System version 5 (GEOS-5) Modern-Era Retrospective analysis for Research and Applications (MERRA) meteorological analysis (Rienecker et al., 2008) until 14 January, 1 d before the eruption day. After 15 January, we run the model freely with a fully interactive atmosphere and ocean for 10 d.

We constrain the simulated volcanic aerosol, H_2O , and chlorine via comparison to observations during the first 10 d after the eruption. Zhu et al. (2022) show that the simulated aerosol backscatter coefficient agrees with observations from the Cloud-Aerosol Lidar and Infrared Pathfinder Satellite Observations (CALIPSO) satellite on 17 January. The simulated H_2O agrees with the MLS (Millán et al., 2022; Zhu et al., 2022) from 1 February to 1 April 2022. Here, we compare the simulated H_2O with the radiosonde observations of humidity (Vömel et al., 2022a) during the first week. Figure 1 shows the RH_i on 18 and 19 January observed by the radiosonde and from nearby simulated model output. Both the observations and simulations show a relative humidity of between 70 % and 100 %. The radiosonde observations have a much higher vertical resolution than the model. Therefore, they show multiple layers of water enhancement, whereas the model only shows one.

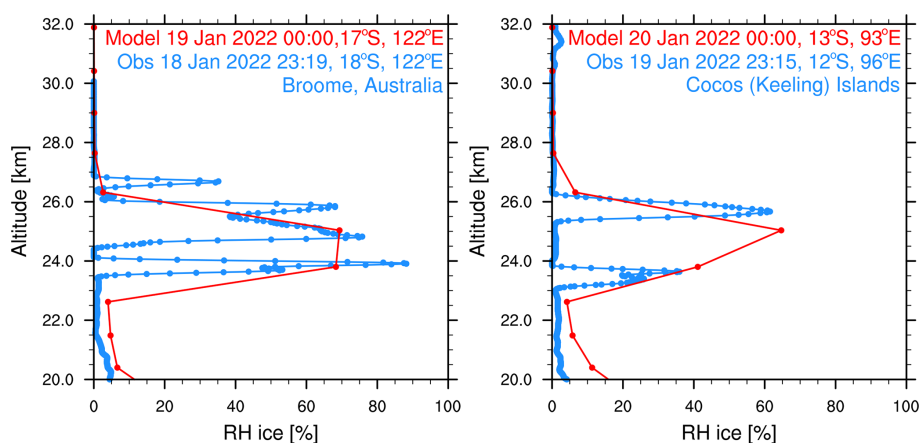


Figure 1. Relative humidity with respect to ice saturation vapor pressure from radiosondes (blue) (Vömel et al., 2022) and simulation (red). The profiles are picked at nearby locations. Note that the observations are about 45 min earlier in time than the simulations, which places them on a different day.

We constrain the chlorine injection using MLS ClO observations at 20 hPa. Figure 2a shows ClO from the MLS observations and the model simulations at 20 hPa from 18 to 24 January. The MLS values are selected from locations where the water vapor mixing ratio is higher than 10 ppmv, indicating that these values are inside the volcanic plume. Figure 2b and c show the simulated daytime ClO for one plume location for each day. The dates are marked next to each plume. MLS observations show elevated ClO, about 5 to 10 times higher than the ambient values (Fig. 2a). If we only inject SO₂ and H₂O (the H₂O_SO₂ case defined in Table 1), we get a ClO amount about twice as large as the background (Fig. 2b), which is much lower than observed. The change in ClO indicates that H₂O alters the Cl_y partitioning. To match the observed values, we need to inject 0.0013 Tg of ClO (Fig. 2c). This is equivalent to injecting ~ 0.0009 Tg of HCl (Fig. A2). In our simulations, injecting ClO and HCl does not lead to different HOCl (Fig. A3), ClO, or O₃ levels after 15 January, indicating that the balancing of ClO and HCl inside the HTHH plume happens very quickly. Unfortunately, the HOCl retrieval from the MLS is not suitable for scientific use at this pressure level, so we cannot validate it. We choose the ClO injection case in our following analysis. Note that the MLS ClO vertical resolution is ~ 2 km near 20 hPa, which is coarser than the model vertical resolution (~ 1 km at 20 hPa).

To investigate the O₃ decrease and its related chemical evolution during the first 10 d, we conduct several simulations, as described in Table 1.

3 Results

Evan et al. (2023) show that the HTHH in-plume ozone depletion at 20 hPa lasts at least 10 d after the HTHH eruption, which they attribute to the heterogeneous chlorine activation

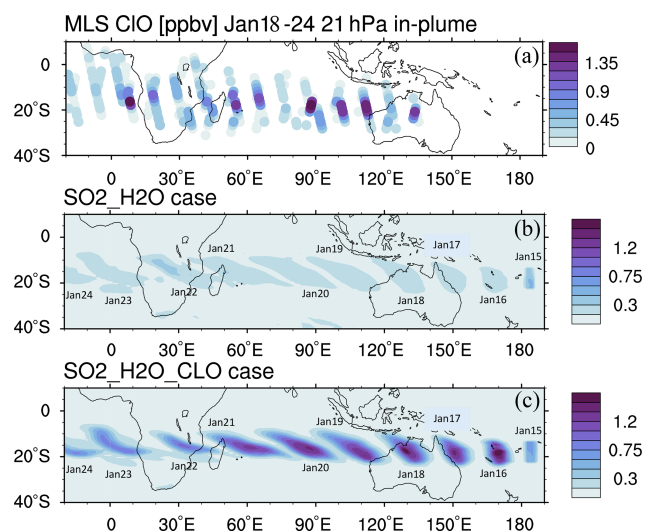


Figure 2. (a) MLS in-plume ClO observations from 18 to 24 January. “In-plume” is defined as the area in which the water vapor mixing ratio is larger than 10 ppmv. MLS in-plume ClO data are not recommended for scientific use until 18 January 2022. The simulated 10 d evolution of in-plume ClO in the SO₂_H₂O (b) and SO₂_H₂O_CLO (c) cases is also shown. The modeled ClO concentrations are only taken during daytime each day (at either 06:00 or 12:00 UTC).

on humidified volcanic aerosols. Here, we analyze the contributions to this initial in-plume O₃ depletion considering three processes: (1) increasing H₂O injection may enhance the HO_x catalytic cycle and HO_x–ClO_x interactions; (2) increasing ClO during the injection phase may deplete ozone due to both heterogeneous reactions and gas-phase reactions; (3) the rising plume from the troposphere may carry ozone-poor tropospheric air into the stratosphere.

Table 1. Model cases descriptions.

Name	Description
Nonvolc	There is no injection of volcanic H ₂ O or SO ₂ .
H2O_SO2	The H ₂ O and SO ₂ injection profiles follow Zhu et al. (2022).
H2O_SO2_CIO	Besides H ₂ O and SO ₂ , 0.00013 Tg of ClO is injected. The ClO injection profile is proportional to H ₂ O injection.
H2O_SO2_CIO_nohet	Same setting as H2O_SO2_CIO but the heterogeneous chemical reactions for HCl+HOCl, ClONO ₂ + H ₂ O, and ClONO ₂ + HCl are turned off.
SO2_CIO	The SO ₂ injection profile follows Zhu et al. (2022). No water is injected. A total of 0.00013 Tg of ClO is injected using the same profile as H2O_SO2_CIO.
lowO3	The O ₃ is reduced to 75 % of its original value at 20 hPa.
H2O_SO2_lowO3	H ₂ O and SO ₂ are injected and O ₃ is reduced to 75 % of its original value.
H2O_SO2_CIO_lowO3	H ₂ O, SO ₂ , and ClO are injected and O ₃ is reduced to 75 % of its original value.
H2O_SO2_NO	A total of 0.003 Tg of NO is injected in addition to H ₂ O and SO ₂ .

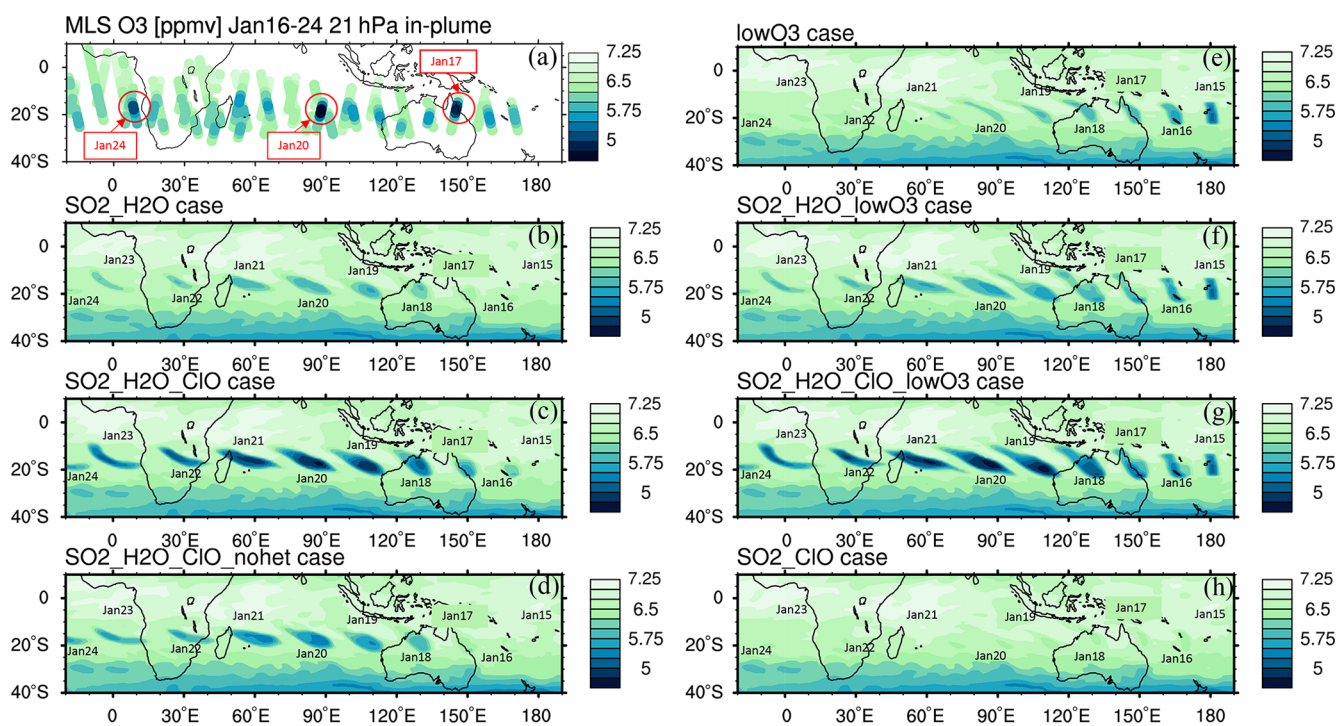


Figure 3. (a) The MLS in-plume O₃ observation from 16 to 24 January. “In-plume” is defined as in Fig. 2. Note that MLS ozone retrievals were unaffected by the plume, leading to the addition of 2 extra days of data for this figure. The locations and days with low O₃ values used in Fig. 6 are marked with circles. (b–h) The simulated 10 d evolution of in-plume O₃ in seven model cases with various injections of SO₂, H₂O, ClO, and low initial O₃ is also shown. Panel (d) uses the same injection as panel (c) but with heterogeneous reactions (i.e., HCl + HOCl, ClONO₂ + H₂O, and ClONO₂ + HCl) turned off. The simulated O₃ in the H₂O_SO₂ case uses one model time step each day that occurs near local noon.

The MLS observed in-plume low ozone concentrations at 20 hPa (Fig. 3a), especially during the following 3 d: ozone concentrations of 4.8 ppmv on 17 January, 4.6 ppmv on 20 January, and 5.1 ppmv on 24 January. These are ozone anomalies of about 1.7, 1.9, and 1.4 ppmv, respectively. The anomalies are calculated using the background average values in this area (6.5 ppmv) subtracting the low ozone values. Note that any interpretation of these O₃ anomalies needs to consider the coarse MLS vertical resolution (~3 km). Because the plume is spatially small during the initial days, the MLS tracks do not capture the maximum plume perturbation every day. The simulation with water injection (Fig. 3b) accelerates the HO_x catalytic cycle and shows evident O₃ reduction, although less than observed. Once we inject ClO on top of the massive water injection (Fig. 3c), O₃ loss is significantly enhanced and is close to the observations after 18 January. The difference between Fig. 3d and c is caused by heterogeneous reactions, which usually only happen in the stratospheric polar springtime where they cause the Antarctic ozone hole and Arctic ozone depletion. Heterogeneous reactions become important, despite the high non-polar temperatures, because of the massive quantity of water injected. The heterogeneous reaction rate is strongly related to the relative humidity (Shi et al., 2001). Usually, during the polar night, the relative humidity is higher (RH_i 60%–100%) than in the non-polar stratosphere because of the low temperature (< 195 K). Here, the water injection increases the relative humidity (Fig. 4c). Enhanced water causes the weight percent of H₂SO₄ of the sulfuric acid aerosol to decrease from 70% to 35% (Fig. 4b). The massive water injection also causes the in-plume temperature to drop by about 2–6 K (Fig. 4f) (Solomon et al., 2016). All of these factors (temperature decrease, relative humidity increase, and particle H₂SO₄ dilution) can increase the three heterogeneous reaction probabilities (HCl + HOCl, ClONO₂ + H₂O, and ClONO₂ + HCl). As shown in Fig. 5, when the water vapor amount is near the climatological value of 6 ppmv, the heterogeneous reaction probability reaches 10⁻² to 10⁻¹ when the temperature is ~190 K. Meanwhile, the reaction probability is similar for temperatures of 215 K when the water vapor is ~600 ppmv in the simulations, as was the case for the HTHH plume during the week following the eruption. Constellation Observing System for Meteorology, Ionosphere, and Climate-2 (COSMIC-2) radio occultation observations showed even higher water vapor mixing ratio during the first week: the maximum values over 20–22 January are ~1000–2000 ppmv (Randel et al., 2023). Also, because the in-plume and out-of-plume chemical concentrations are different, we apply both conditions (solid and dashed lines) to show how the different HCl, HOCl, and ClONO₂ conditions alter the HCl + HOCl and ClONO₂ + HCl reaction probabilities by 1 order of magnitude. Volcanic sulfur injection also increases the sulfate surface area density (Fig. 4a) that provides extra surfaces for heterogeneous reactions.

Comparing Fig. 3b and c with MLS observations, we can see that the chemical reactions do not explain the O₃ loss during the first 3 d of the eruption (15–17 January, low O₃ near 160° E in the MLS observation). This discrepancy suggests that the plume contains some ozone-poor tropospheric air after the injection into the stratosphere. We ran three cases with initial low ozone. For the low-O₃ case (Fig. 3e), we inject only ozone-poor air without volcanic H₂O or SO₂. It shows low O₃ as observed during the first couple of days, but ozone recovers quickly because the O₃ chemical lifetime is short at 20 hPa inside the plume (Fig. A4). The H₂O_SO₂_lowO₃ case (Fig. 3f) shows ozone loss similar to the observations in the first 6 or 7 d. By adding the ClO and initial ozone-poor air (Fig. 3g), we obtain persistent low O₃ values that agree with the lowest observed values better than the other cases (Fig. 6a). Compared with Fig. 3b, Fig. 3d has slightly more ozone depletion, indicating that the extra chlorine injection impacts O₃ even without heterogeneous chemistry. However, without including the high amounts of injected water, the additional ClO alone cannot deplete ozone much (Fig. 3h).

Figure 6 shows the O₃ anomaly evolution from several model cases (panel a) and the percentage contributions to the total ozone loss (panels b and c). The model case with all injections (initial low O₃, high H₂O, and high ClO) agrees well with the MLS observations on the 3 d with the lowest O₃ values (Fig. 6a). In Fig. 6b and c, the black bars represent the contribution from the low-O₃ injection, which is significant during the first couple of days but diminishes quickly. From these percentage values, we conclude that the lofting of low O₃ carried in the plume cannot be the reason for the low O₃ values after 3 d. Chemistry is the main reason that this O₃ depletion lasts so long.

There are two ways to look at the chemical contributors to ozone loss based on our model runs. The first is to separate the contributors due to various injections (Fig. 6c): H₂O injection accounts for about 30%–40% of the ozone loss most of the time (blue) and ClO injection accounts for 50% of the ozone loss most of the time (red). However, we cannot simply attribute the largest contribution to the ClO injection, because if we only inject ClO, it does not produce much ozone depletion (Fig. 6a, magenta). It is the ClO_x–HO_x interactions that accelerate O₃ depletion.

A second way to look at the causes for ozone loss is to separate the contributions from the gas-phase chemistry and the heterogeneous chemistry (Fig. 6b). The model run with the H₂O and ClO injections but without the heterogeneous chemistry shows that the gas-phase chemistry (yellow bars) account for more than 47% of the ozone loss from 18 to 24 January. Heterogeneous chemistry (green bars) destroys about 30% of the ozone. Hence, both heterogeneous chemistry and gas-phase chemistry are important for O₃ depletion. Once we turn off the heterogeneous chemistry, the partitioning between active chlorine and chlorine in the reservoirs is changed. The order in which the processes are accounted for can affect the resulting breakdown. Thus, we cannot sim-

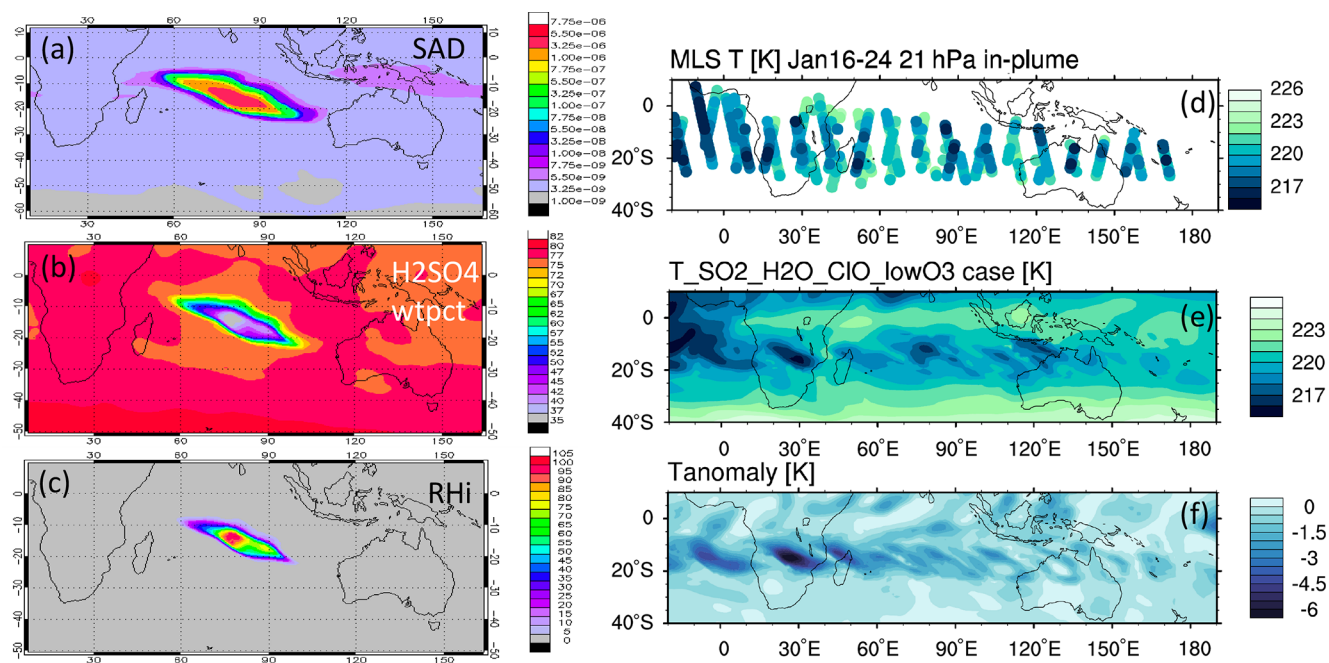


Figure 4. Panels (a), (b), and (c) present the simulated surface area density, simulated $\text{H}_2\text{SO}_4/\text{H}_2\text{O}$ weight percent, and relative humidity on 20 January at 20 hPa, respectively. (d) The temperature evolution during the first 10 d at 20 hPa from the MLS. (e) simulated temperature evolution in the $\text{SO}_2\text{-H}_2\text{O-CIO-lowO}_3$ case, and (f) temperature difference between the $\text{SO}_2\text{-H}_2\text{O-CIO-lowO}_3$ case and the Nonvolc case are also shown.

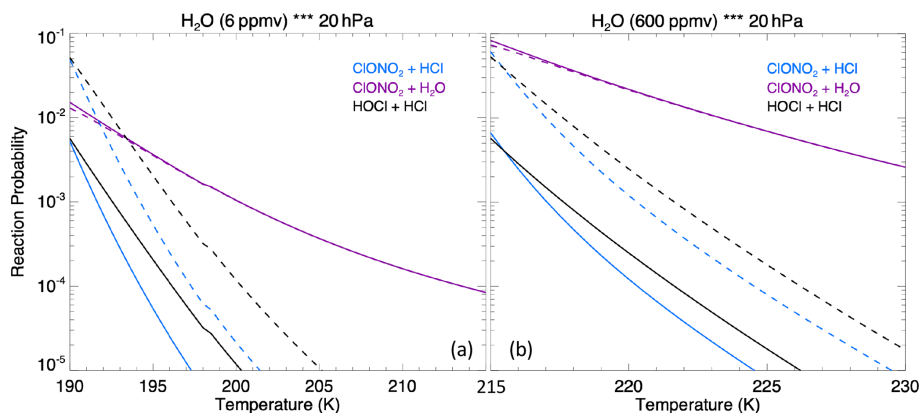


Figure 5. The heterogeneous reaction probability for three reactions on sulfate surfaces ($\text{ClONO}_2 + \text{HCl}$, $\text{ClONO}_2 + \text{H}_2\text{O}$, and $\text{HOCl} + \text{HCl}$) as a function of water vapor assuming a $0.4\ \mu\text{m}$ particle size at 20 hPa. Panel (a) assumes 6 ppmv of ambient water vapor and panel (b) assumes 600 ppmv of ambient water vapor. The solid lines use the out-of-plume chemical concentration on 20 January: 1.0 ppbv of HCl, 0.03 ppbv of HOCl, and 0.5 ppbv of ClONO_2 ; the dashed lines use the in-plume chemical concentration: 0.1 ppbv of HCl, 1.0 ppbv of HOCl, and 0.05 ppbv of ClONO_2 . These values are based on the simulation output.

ply say that gas-phase chemistry contributions are larger than heterogeneous chemistry. Both are clearly significant.

To better understand which reactions are critical in the HTHH plume, we investigate the simulated reaction rates related to HO_x and chlorine compounds (Fig. 7). These reactions reflect how the water and ClO injections strengthen the in-plume $\text{HO}_x\text{-ClO}_x$ interactions, chlorine activation, and the relative importance of each heterogeneous reaction

rate. The WACCM model uses the methods developed by Shi et al. (2001) for heterogeneous reaction rate calculations. Figure 7a shows the HO_x cycle inside and outside the water plume during daytime on 20 January at 20 hPa. The $\text{HO}_2 + \text{O}_3$ reaction rate increases by a factor of 4 (from 5×10^4 to $2 \times 10^5\ \text{cm}^{-3}\ \text{s}^{-1}$), $\text{OH} + \text{O}$ increases by a factor of \sim four (from 2×10^4 to $7.5 \times 10^4\ \text{cm}^{-3}\ \text{s}^{-1}$), and $\text{HO}_2 + \text{O}$ increases by a factor of 4 (from 2×10^4 to $8 \times 10^4\ \text{cm}^{-3}\ \text{s}^{-1}$). In

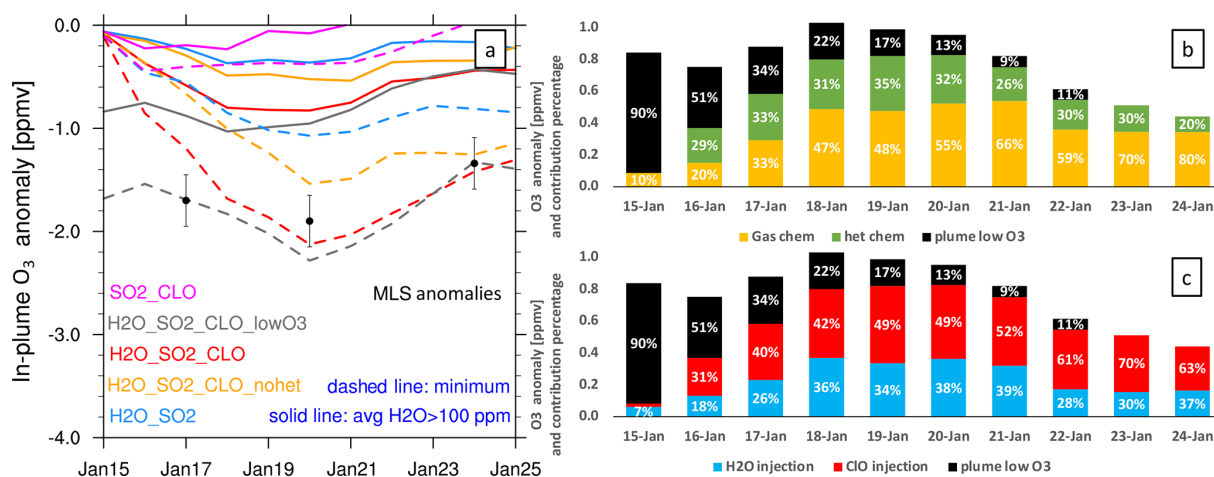


Figure 6. Panel (a) presents the O₃ anomaly in different model cases. The solid lines are the average O₃ anomaly at 20 hPa on each day near local noon where water vapor is larger than 100 ppmv. Here, 100 ppmv is suggested by Evan et al. (2023), who found that O₃ anomalies are not significant for a 10 ppmv threshold but are significant for a 100 ppmv threshold. The dashed lines are the simulated maximum O₃ anomaly on each day at 20 hPa. The black dots show the 3 d during which the MLS measures the lowest O₃ values (explained in Fig. 3a). Panel (b) presents the percentage contributions to ozone loss from gas-phase chemistry (orange, H₂O_SO₂_CLO_nohet), heterogeneous chemistry (green, H₂O_SO₂_CLO minus H₂O_SO₂_CLO_nohet), and low-O₃ air carried into the stratosphere (black, H₂O_SO₂_CLO_lowO3 minus H₂O_SO₂_CLO). Panel (c) shows the percentage contributions to ozone loss from H₂O injection (blue, H₂O_SO₂ minus Nonvolc), ClO injection (red, H₂O_SO₂_CLO minus H₂O_SO₂), and low-O₃ air carried into the stratosphere (black, H₂O_SO₂_CLO_lowO3 minus H₂O_SO₂_CLO).

addition, the extra HO_x plays a large role in chlorine activation. Figure 7b shows the chlorine compound reactions inside the HTHH initial plume. The HOCl photolysis rate increases by a factor of ~ 30 inside the plume (from $6 \times 10^3 \text{ cm}^{-3} \text{ s}^{-1}$ outside the plume to $2 \times 10^5 \text{ cm}^{-3} \text{ s}^{-1}$) due to the high HOCl mixing ratio, which is the dominant process causing the increase in chlorine activation to Cl. The HOCl concentration remains high due to the enhanced ClO_x–HO_x interaction (i.e., ClO + HO₂ → HOCl + O₂ reaction) as well as the increase in the heterogeneous reaction rate of ClONO₂ + H₂O by 5 orders of magnitude (from 0.3 to $1 \times 10^4 \text{ cm}^{-3} \text{ s}^{-1}$). The large amounts of HOCl also make the heterogeneous reaction of HOCl + HCl faster than the ClONO₂ + HCl reaction; the latter reaction is known as the major reaction contributing to the chlorine activation that has a hand in polar ozone depletion. Figure A5 shows the uptake coefficient for the three heterogeneous reactions HCl + HOCl, ClONO₂ + H₂O, and ClONO₂ + HCl on 20 January. The reaction probability of ClONO₂ + HCl is increased by 8 orders of magnitude (from the background value of 10^{-10} to 10^{-2}). This value is even higher than that suggested by Evan et al. (2023), who estimate that enhanced water vapor increases the uptake coefficient of ClONO₂ + HCl to $10^{-4} \text{ cm}^{-3} \text{ s}^{-1}$. The reaction probability of HCl + HOCl and ClONO₂ + H₂O increases to 10^{-2} . Furthermore, inside the plume, the reactions that convert Cl back to HCl are slower than their activation rate.

Besides the ozone loss reactions, ozone production reactions are also significantly altered by the water plume (Fig. 7c). HO₂ + NO is usually not an important process for

O₃ production in the stratosphere (more important in the troposphere). The reaction rate doubles inside the plume (from 1×10^5 to $2 \times 10^5 \text{ cm}^{-3} \text{ s}^{-1}$). Note that we do not inject lightning NO_x in this case, a possible scenario during the eruption phase, that can also further increase the O₃ production (detailed in Sect. 4).

Comparing the partitioning of Cl_y (Cl + ClO + 2Cl₂ + 2Cl₂O₂ + OCIO + HOCl + ClONO₂ + HCl + BrCl) reveals the in-plume chlorine activation processes (Fig. 8). Outside the plume, HCl and ClONO₂ are dominant, indicating that most of the Cl is in reservoirs. While inside the water plume, both the H₂O_SO₂ and H₂O_SO₂_CLO cases show strong depletion of the reservoirs HCl and ClONO₂, and most of the Cl_y is either in the form of HOCl (a short-lived reservoir) or is activated in the form of ClO. Unlike the chlorine activation process in the polar winter, HOCl is the highest in the HTHH plume because heterogeneous chemistry is not fast enough to destroy HOCl to produce ClO. In the case without heterogeneous chemistry, HCl and ClONO₂ are dominant in the plume, indicating that heterogeneous chemistry is the main process of converting HCl to active chlorine. Comparing total Cl_y and ClO in all panels, ClO does not exceed a quarter of the Cl_y, indicating that adding 0.00013 Tg of ClO through injection is one way to produce the observed ClO. There is a possibility that ClO is converted from other Cl_y species through chemical reactions that we are not aware of, as this was a very unusual eruption.

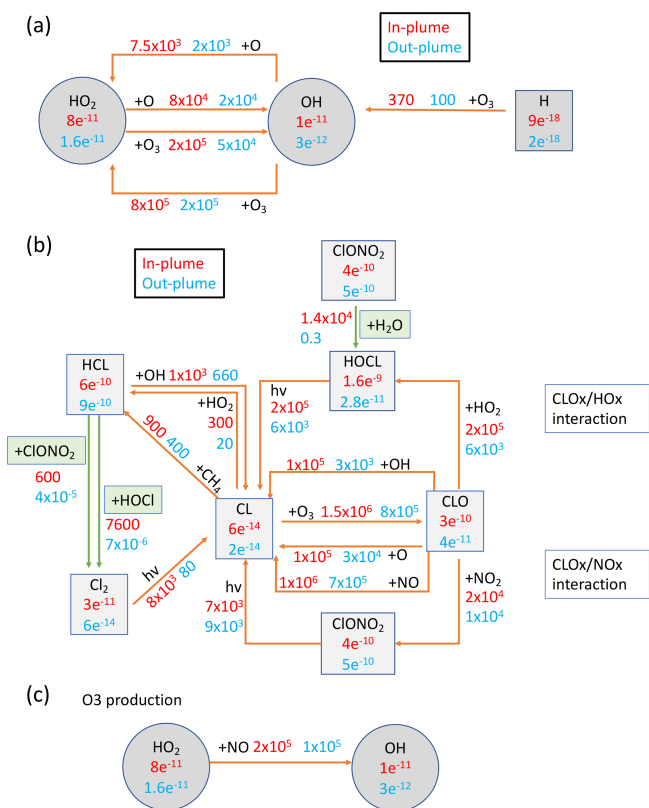


Figure 7. Reactions inside and outside the plume (in $\text{cm}^{-3} \text{s}^{-1}$) and compound concentrations (in mol mol^{-1}). Red numbers represent values outside the plume and blue numbers represent values inside the plume and orange numbers represent values inside the plume. Panel (a) presents the HO_x balance and its interaction with O_x during daytime at 20 hPa on 20 January 2022. Panel (b) shows chlorine compound reactions in the H₂O–SO₂–ClO case. Panel (c) gives the HO_x cycle impact on O₃ production. Green arrows represent the heterogeneous reactions for chlorine activation. H₂O is ~ 600 ppm inside the plume and ~ 5.5 ppm outside the plume. Cl_y is ~ 4.2 ppbv inside the plume and 1.5 ppbv outside the plume.

4 Discussion

The ozone loss inside the HTHH plume during the first 10 d provides a unique opportunity to study stratospheric chemistry and to understand the performance of the state-of-the-art WACCM climate model, because the HTHH-injected ClO and H₂O exceed the normal range of stratospheric variability. These volcanic injections strongly altered the ClO_x–HO_x interactions and heterogeneous reaction rates, producing different chemical pathways for chlorine activation and ozone depletion compared with what occurs in the Antarctic ozone hole or Arctic ozone depletion in the polar stratospheric winter and spring. HOCl is identified as playing a large role in the in-plume chlorine balance and heterogeneous processes. The high HOCl concentrations are a result of the very high in-plume water vapor content, which makes this event different

from chemistry in the Antarctic ozone hole, where ClONO₂ is more important.

This study also raises an interesting question regarding where the Cl comes from in the volcanic injection. Seawater contains 3.5 % sea salt, which implies that about 5 Tg of NaCl could have been injected, assuming that the injected 150 Tg of H₂O came from seawater. However, we only need to inject 0.00013 Tg of ClO to match the MLS ClO observations during the first few days after the eruption. We also conducted a test injecting an equivalent amount of HCl (0.0009 Tg), which resulted in a similar HOCl, ClO, and O₃ pattern (Figs. A2, A3). If we inject more HCl or ClO, ClO would exceed the observed concentration, causing the depletion of OH and slowing down SO₂ oxidation. Evidently, if the water came from seawater, most NaCl was not converted to HCl but stayed in the stratosphere as particles. Vernier et al. (2023) sampled NaCl particles 8 months after the eruption near Brazil. Based on their sampled NaCl concentration, we estimate that 0.5–1 Tg of NaCl may have been injected and stayed in the atmosphere. There are several possibilities as to why this event did not inject 5 Tg of NaCl into the stratosphere: remote-sensing particle size estimations (Khaykin et al., 2022) and in situ measurements (Asher et al., 2023) indicate that the particles were submicron sized. However, sea salt particles injected into the lower troposphere by wind are mainly particles larger than 10 μm. Hence, if the volcanic injection had similar sized NaCl particles, most of them may have quickly fallen out of the stratosphere. In addition, the majority of NaCl might have been washed out during the first couple of hours of plume injection by acting as nuclei for ice particles. It is also possible that the reactions that might release Cl from NaCl may not efficiently lead to reactive Cl. For example, HNO₃ can heterogeneously react on sea salt very quickly in the troposphere to release HCl (De Haan and Finlayson-Pitts, 1997; Guimbaud et al., 2002; Murphy et al., 2019). This reaction may be accelerated by HTHH high humidity, even if the temperature is low in the stratosphere. HCl could be removed by condensing in supercooled water, which would reduce HCl vapor concentrations by up to 4 orders of magnitude, preventing substantial stratospheric chlorine injection (Tabazadeh and Turco, 1993). Finally, it may be that the water injected came from magmatic water or from seawater that percolated into the volcano and was released as steam. Such water would not be rich in NaCl. In that case, Cl observed by Vernier et al. (2023) may have been bound up in minerals of the volcanic ash. Other halogen species, such as bromine and iodine, are often observed after volcanic eruptions (large amounts of BrO were observed after HTHH in the troposphere; Li et al., 2023). However, they can lead to much stronger ozone depletion if they persist in the stratosphere. As the elevated Cl in the model can explain the O₃ depletion well, the impact of bromine and iodine on stratospheric O₃ is minimal for this eruption.

In addition, NO_x can be produced by lightning inside or around the volcanic plume. Observations show that there

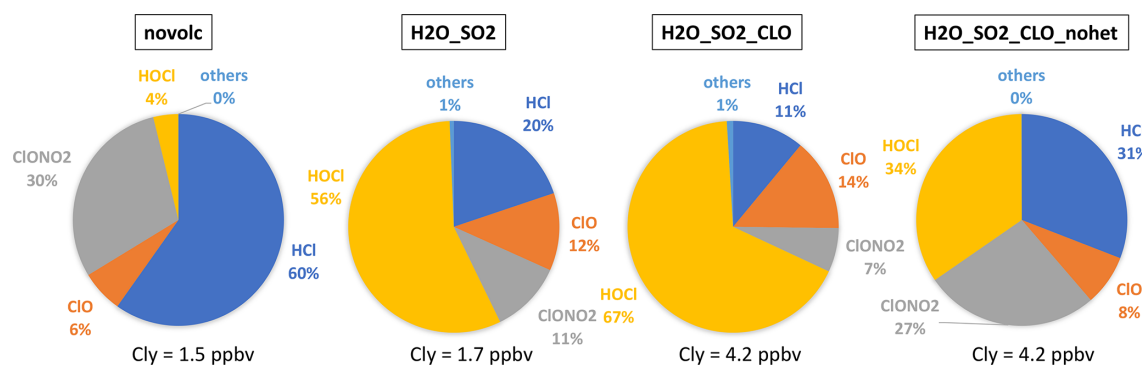


Figure 8. The percentage of each inorganic chlorine compound ($\text{Cl}_y = \text{Cl} + \text{ClO} + 2\text{Cl}_2 + 2\text{Cl}_2\text{O}_2 + \text{ClOCl} + \text{HOCl} + \text{ClONO}_2 + \text{HCl} + \text{BrCl}$) inside and outside the plume. The slight difference between Nonvolc Cl_y and $\text{H}_2\text{O_SO}_2$ Cl_y is because H_2O injection changes the plume dynamics in the free-running simulations.

was a record number of lightning events in this volcanic plume. Almost 400 000 flashes were observed by the Global Lightning Detection Network (GLD360) network over the 6 h of the most active eruption period (~ 590 000 total flashes were observed) (Global Volcanism Program, 2022). Considering that tropospheric global models use a lightning source of 5 Tg(N) yr^{-1} and an average flash the Optical Transient Detector and Lightning Imaging Sensor (OTD/LIS) satellite sensors produced an average global flash rate of 44 ± 5 flashes per second, an injection of ~ 0.001 – 0.003 Tg N (0.002 – 0.006 Tg NO) would be expected for the HTHH eruption. We conducted a model run with H_2O , SO_2 , and an injection of 0.003 Tg NO (the $\text{H}_2\text{O_SO}_2\text{_NO}$ case), showing that this additional NO has little impact on the O_3 loss and ClO levels during the first 10 d (Fig. A6). Compared with the $\text{H}_2\text{O_SO}_2$ case, the simulated O_3 loss in the $\text{H}_2\text{O_SO}_2\text{_NO}$ case increased by $\sim 5 \times 10^5 \text{ molecules cm}^{-3} \text{ s}^{-1}$; however, at the same time, the O_3 production rate increased by $\sim 5 \times 10^5 \text{ molecules cm}^{-3} \text{ s}^{-1}$. The $\text{NO} + \text{HO}_2$ reaction rate in the $\text{H}_2\text{O_SO}_2\text{_NO}$ case increased by 5 times compared with the $\text{H}_2\text{O_SO}_2$ case. Therefore, lightning NO_x probably does not contribute to the HTHH initial in-plume O_3 loss. Because of the high water vapor concentration, NO would convert to HNO_3 in the first couple of days. Unfortunately, we lack observations of HNO_3 , NO, or NO_2 right after the eruption. The MLS observations in February (Fig. A7) and the model simulations with H_2O injection or $\text{H}_2\text{O} + \text{NO}$ injections show elevated HNO_3 compared with the background.

Appendix A

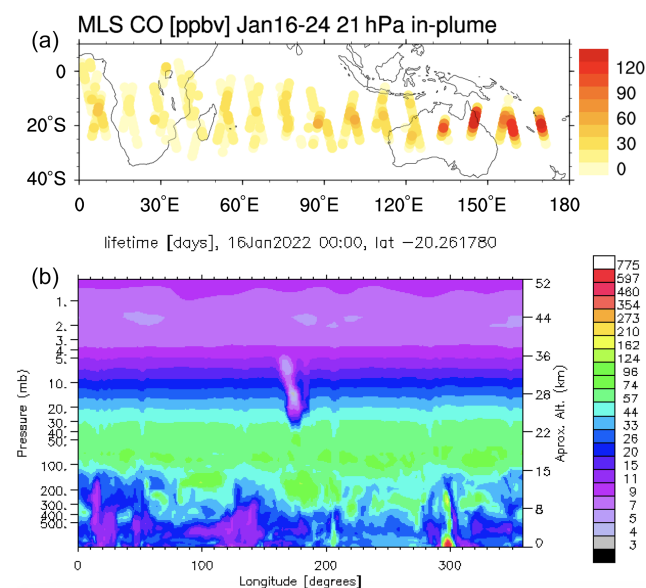


Figure A1. Panel (a) shows the MLS in-plume CO observation during the first 10 d after the eruption. Panel (b) shows that the CO lifetime on 16 January at 20°S is shortened from a month to a few days because of the volcanic water plume. The observed CO mixing ratios of around 120 ppmv seem incompatible with typical CO levels over oceanic regions, indicating the production of CO within the magma chamber or in the hot plume itself.

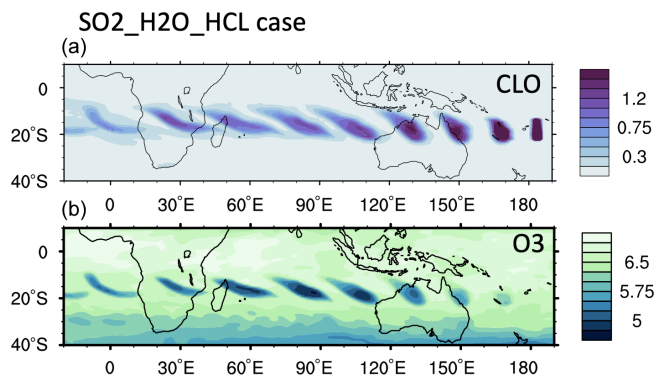


Figure A2. The (a) CLO and (b) O₃ evolution from the model case with an HCl injection of 0.000092 Tg (equivalent to 0.00013 Tg of CLO injection).

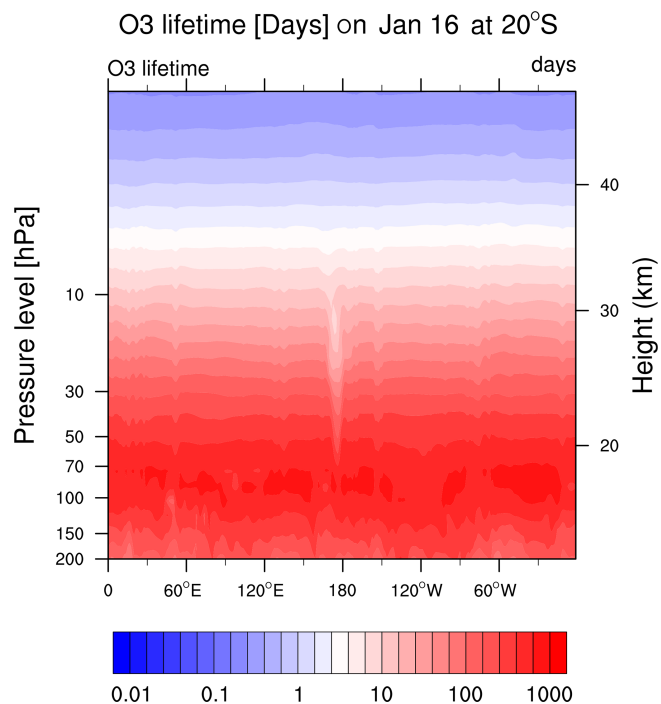


Figure A4. The O₃ chemical lifetime is about 1–2 months at 20 hPa and is reduced to 10 d at the HTHH location.

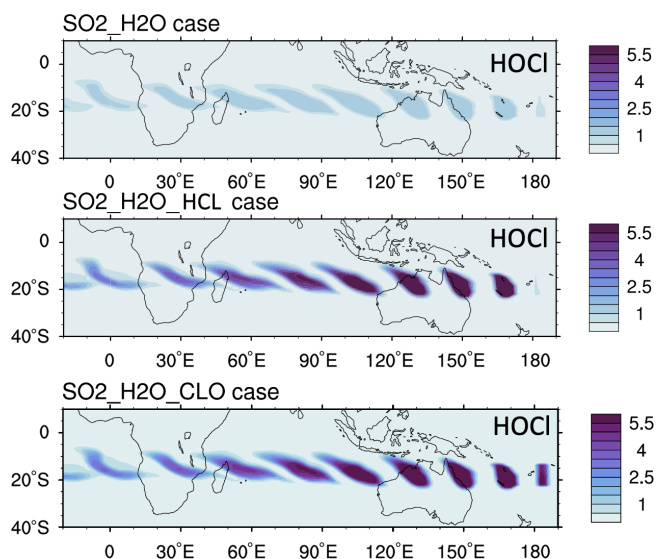


Figure A3. The HOCl evolution from the three model cases.

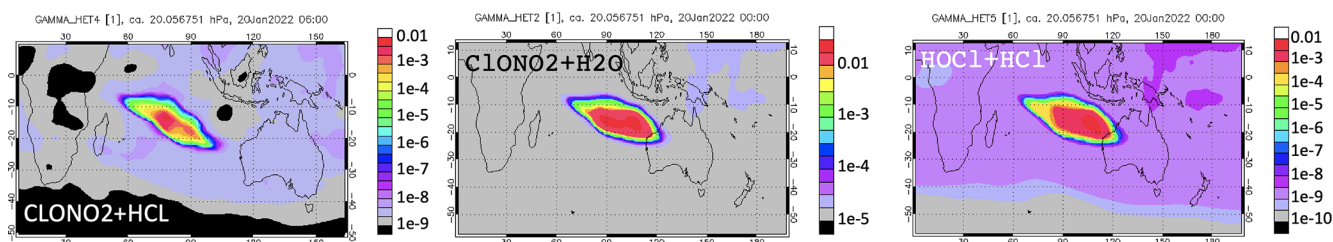


Figure A5. Heterogeneous reaction probabilities for the three heterogeneous reactions on 20 January at 20 hPa.

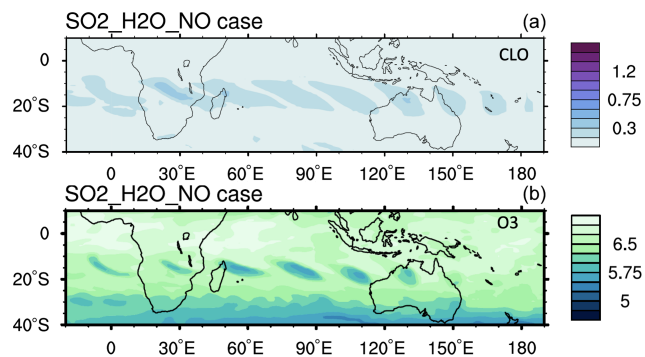


Figure A6. The (a) CIO and (b) O₃ evolution from the model case with an NO injection of 0.003 Tg, which is identical to the SO₂_H₂O case. The CIO and O₃ enhancements are due to the H₂O injection.

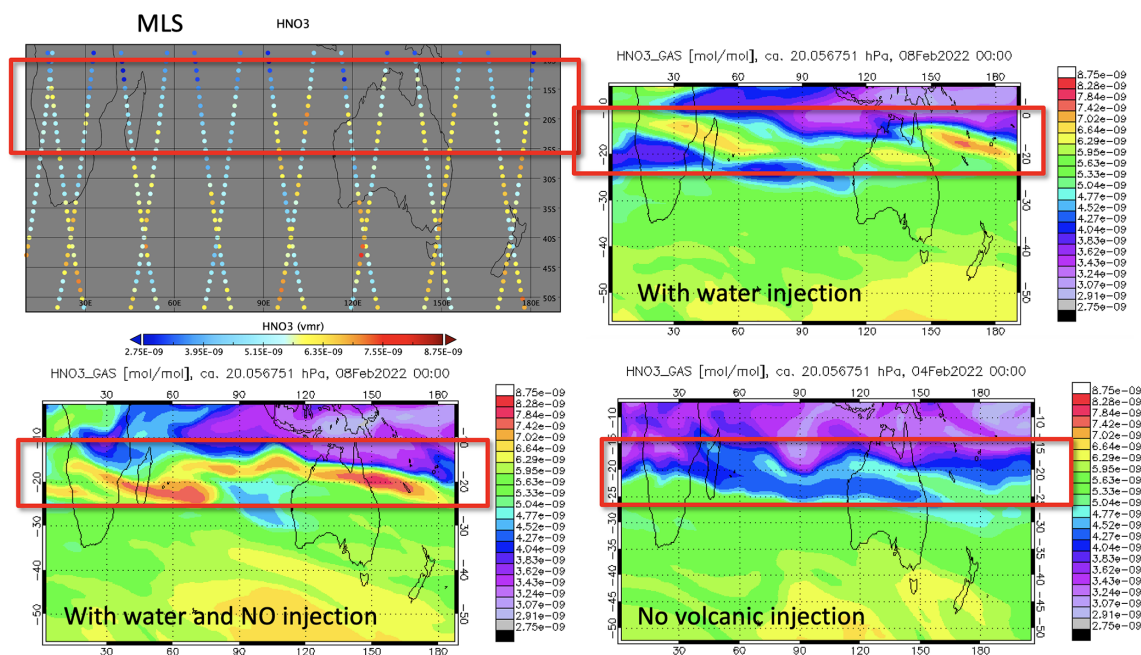


Figure A7. HNO₃ observed by the MLS on 8 February 2022 compared to the model simulation with water and NO injection as well as the no volcanic injection case. The MLS shows similar elevated HNO₃ to the simulation case with H₂O injection or with H₂O / NO injection.

Code availability. The Community Earth System Model 2 (CESM2) is available on the CESM trunk to any registered user: <https://www.cesm.ucar.edu/models/cesm2/download> (NCAR, 2023).

Data availability. The main simulation data generated during this study are available at <https://doi.org/10.17605/OSF.IO/F69NS> (Zhu, 2023). Aura MLS v4 data are available at <https://doi.org/10.5067/Aura/MLS/DATA2009> (Lambert et al., 2015). Water vapor radiosonde data are available at <https://doi.org/10.5065/p328-z959> (Vömel et al., 2022b).

Author contributions. YZ, RWP, DK, and KHR designed the experiments. YZ performed the simulations and prepared the manuscript with contributions from all co-authors. DK examined the sensitivity of the stratospheric H₂O abundance to the reaction probability (Fig. 5). LM, HV, and SE provided observational data and analysis. RWP, DK, OBT, JZ, ST, CGB, XW, WJR, and KHR participated in the modeling data analysis.

Competing interests. At least one of the (co-)authors is a member of the editorial board of *Atmospheric Chemistry and Physics*. The peer-review process was guided by an independent editor, and the authors also have no other competing interests to declare.

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