

**LOW INCIDENCE OF MICROPLASTIC CONTAMINANTS IN PACIFIC OYSTERS  
FROM THE SALISH SEA, USA**

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15    **HIGHLIGHTS**

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17    - Microplastics in oysters were found in 5 of 10 sites in the Salish Sea, Washington.

18    - Only ~2% of the microparticles were identified as microplastics by RMS and FTIR.

19    - Sorbitan derivatives, polyamide resins, cellulose, and minerals were also present.

20    - Microfibers were observed but were not confirmed as plastic or polymeric with RMS.

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## ABSTRACT

Plastic pollution is a threat to marine life with long term impacts to ecosystems and organisms in the sea. In this study, we quantified the presence of microparticles in wild populations of Pacific oysters (*Crassostrea gigas*) from the Salish Sea, Washington State. Examination under a dissecting microscope revealed 63% of oysters contained microparticles (~1.75 microparticles per oyster) and microfibers were the dominant type of particles. Using Raman microspectroscopy (RMS) and Fourier Transform Infrared microspectroscopy ( $\mu$ -FTIR) we found that only ~2 % of these microparticles were synthetic and included polymers such as polystyrene, polyethylene, polypropylene, poly(bisphenol A carbonate), rayon, and polyacrylate. It is important to note that of the 447 microparticles analyzed with RMS, 41% showed fluorescence interference, impeding the determination of their identification. The remaining microparticles were cellulose derivatives, shell fragments, biological or proteinaceous material, salts, minerals, and gypsum. Fourier Transform Infrared spectroscopy equipped with a diamond Attenuated Total Reflectance accessory (ATR-FTIR) showed the presence of sorbitan derivatives in all samples examined (n = 213). These findings provide the first baseline for microplastic and other particles in oysters from the west coast of the United States integrating results from ATR-FTIR,  $\mu$ -FTIR, and RMS, in addition to visual sorting. These results suggest there is low retention of plastic particles in Pacific oysters from the Salish Sea, but further research is needed to determine the composition of microparticles with fluorescence interference.

[229 words]

## INTRODUCTION

Plastic pollution is a threat to marine life around the globe with long term impacts to coastal ecosystems and the industries that depend on them. Small plastic particles or microplastics are synthetic solid particles or polymeric matrices, that have regular or irregular shape and range in size from 1µm to 5mm (Arthur et al. 2009, Frias & Nash 2019). Microplastics are insoluble in water and can have either primary or secondary manufacturing origin. Microplastics are considered to be primary in origin when they are manufactured to be small, or secondary when they are a result of a slower, long-term breakdown from larger plastics. Primary and secondary microplastics can reach the ocean from wastewater treatment plant runoff, or form when plastic debris is fragmented by UV radiation and wave abrasion (e.g. Weinstein et al. 2016). Currently, microplastics have been reported along shorelines around the world on all continents (Moore 2008, Browne et al. 2011, Cole et al. 2011, Hirai et al. 2011, Waller et al. 2017) and have been found in sediments, throughout the water column, and in the digestive systems, respiratory structures, and tissues of marine organisms worldwide (Andrady 2011, Browne et al. 2011, Cole et al. 2011, Depledge et al. 2013, Ling et al. 2017, Munari et al. 2017). Because of their small size, microplastics can be easily ingested by marine organisms, particularly by species that have limited ability to select particles during feeding (Ward et al. 1994, Ward & Shumway 2004). The accumulation of microplastics in organisms may have detrimental effects on feeding rates, energy storage, reproduction and overall fitness (Paul-Pont et al. 2016, Sussarellu et al. 2016, Welden & Cowie 2016, Harris and Carrington 2019). Furthermore, microplastics do not degrade quickly and may be transferred through trophic interactions (Farrell & Nelson 2013, Au et al. 2017), creating a pathway to ingestion of microplastics by animals at higher trophic levels, including fish, predatory mammals, and humans (Wang et al. 2019).

Microplastics are usually transported in the water, either by marine currents or in freshwater systems by rivers (see review by Cole et al. 2011 for marine environments, Siegfried et al. 2017, Horton & Dixon 2018). In regions with high population density, it is likely that more microplastics will be released to the waterways (e.g. Miller et al. 2017). Similarly, waters that are stagnant, or have longer residence times (e.g. in a bay, inlet, or fjord), may accumulate more microparticles at depth and in sediments (Wessel et al. 2016). Therefore, marine organisms living under different oceanographic conditions could be differentially exposed to microplastic contaminants. Previous research suggests that microfibers from clothing are the most common type of secondary microplastic worldwide (Carr 2017, Cesa et al. 2017). When clothes are washed, these small fibers are released into the water, and their small size allows them to pass through filters, into grey water and eventually local waterways. Other secondary microplastic sources include the breakdown of plastic bags and containers that are transported into waterways, and the breakdown of fishing or aquaculture gear. Primary sources can be microbeads used in cleaning and beauty products that are also transported by waterways. Distinguishing primary vs. secondary plastics or synthetic vs. natural particles in the marine environment is a difficult task. In our study we only use the term microplastic (regardless of origin) for particles that were identified as synthetic polymers by Raman microspectroscopy (RMS), Fourier Transform Infrared Spectroscopy equipped with a diamond Attenuated Total Reflectance accessory (ATR-FTIR), or FTIR microspectroscopy ( $\mu$ -FTIR). Other particles that were similar in size and visual characteristics to microplastics but with unknown composition were referred to as microparticles, sometimes also referred to as “suspected microplastics”.

Microplastics have been found in ecologically and commercially important bivalves from

Europe, Asia, Brazil, Canada, and the United States (Van Cauwenberghe & Janssen 2014, Mathalon & Hill 2014, Rochman et al. 2015, Li et al. 2016), however, there is a lack of research in the eastern North Pacific, one of the most productive regions in the world and with high potential for marine aquaculture (Baechler et al. 2019, Granek et al. 2020). On the west coast of the United States, two studies have assessed and quantified the presence of microplastics in commercially important bivalves (Rochman et al. 2015, Baechler et al. 2019). One study found that 33% of the Pacific oysters sampled in California markets (n = 12) contained 0-2 microparticles per individual (Rochman et al. 2015). These particles were identified as microfibers using visual sorting but no further analyses were performed to assess their chemical identity. The other study carried out in the Oregon coast found that microplastics were present in Pacific oysters and razor clams, but only 26 of the 2428 microfibers were analyzed with FTIR. In Canada, visual sorting also revealed the presence of microparticles in wild and cultured Manila clams (8-11 particles/clam; Davidson & Dudas 2016, Murphy 2018), oysters, and mussels (5.6-6.57 per gram; Mathalon & Hill 2014, Murphy 2018). Murphy (2018) was the only other study in addition to Baechler et al. (2019) that examined the chemical identity of microparticles. Using FTIR analysis, this study showed that only half of the microparticles extracted were synthetic polymers (including plastic). In this context, it is fundamental that microplastic determinations are based on chemically identified polymers, otherwise overestimation of microplastics and misinterpretation of results may occur (Shumway et al. 2018).

In this study, we examine the presence and distribution of microplastic pollution in the Pacific oyster, *Crassostrea gigas*, an ecologically and economically important shellfish species in Washington State, USA (Fig. 1). As ecosystem engineers, their presence creates a hard-bottom substrate that provides habitat and protection for other organisms. Furthermore,

oysters are important food sources for aquatic animals and can filter high volumes of water removing organic and inorganic particles from the water column, resulting in cleaner water (Coen et al. 2007). In the Salish Sea and Puget Sound in particular, there are multiple self-sustaining *C. gigas* populations in bays and inlets where the oysters grow either individually or in dense mats on rocks and soft-bottom substrates. Economically, the hardiness and rapid growth (10-15 cm in 2 years) of *C. gigas* are major advantages for the aquaculture industry. Pacific oysters currently lead the production output of the aquaculture industry in Washington State, the top producer of shellfish in the US with around 200 million dollars in sales in 2014 (Washington Sea Grant 2015). Understanding the abundance and diversity of microplastics in *C. gigas* will help determine if the species is vulnerable to these contaminants and to what degree. Specifically, our goals are to: (i) determine the abundance and type of microplastics in naturally occurring populations of *C. gigas* in Washington State, (ii) identify the material of these microplastics, and (iii) identify potential areas that are ‘hotspots’ for microplastic accumulation in Washington State. We hypothesize that microfibers will be the dominant type of microplastics in oysters, and that regions with longer water residence times will have oysters with higher concentration of microplastics due to longer exposure to these contaminants over time. Considering that marine plastic waste is expected to increase (Andrady 2011, C3zar et al. 2014, Lebreton et al. 2018), this study will contribute to establishing a baseline for the eastern North Pacific, and provide evidence to inform and direct mitigation strategies aimed at aiding in the detection of microplastic sources and the production of healthy shellfish.

## **METHODS**

### **Location and species**

The study was performed in Puget Sound in Washington State, the leading producer of farmed bivalves in the United States (Fig. 1). Puget Sound is located in the southern end of the Salish Sea, is the second largest estuary in the country, and is a complex glacial system with seven sub-basins that have distinct characteristics due to the degree of water mixing, freshwater influence, and wind fetch. The sampling sites are located throughout Puget Sound to capture the variability in environmental conditions between basins along the estuary (Fig. 1, Table 1). Sites in Northern Puget Sound often experience shorter water residence times (2 months) than the average for Puget Sound due to their proximity to the Pacific Ocean (Encyclopedia of Puget Sound). In contrast, South Puget Sound has longer water residence times (2-4 months) due to limited water circulation and mixing (Finlayson 2006). Puget Sound estuary is within close proximity to some of the largest cities in Washington State (Seattle, Tacoma and Olympia, Fig. 1) with a population of approximately 3,867,000 people, making it the 14<sup>th</sup> largest metropolitan area in the United States (United States Census Bureau 2017). Previous studies in the Puget Sound region have found that microplastics are present in sediment and water samples (Masura et al. 2015, Eshom-Arzadon 2017). However, there are no published studies examining accumulation of microplastics in shellfish that naturally grow in the area.

### **Sample collection and processing**

We collected 30 *C. gigas* individuals during the low tide at ten sites in Puget Sound estuary between January and April 2018 (Table 1, Fig. 1). All collection sites are State Parks or public access beaches with self-sustaining Pacific oyster populations. Within 48 hs of collection, we dissected oysters and measured the right and left valves to the nearest mm with a digital caliper. We then removed the soft tissue from the shells, weighed it to the nearest mg



with an XS304 Mettler Toledo digital scale, and stored it at -20 °C until the microplastic extraction protocol was performed.

### **Tissue digestion and microplastics extraction**

We used a standardized extraction protocol developed by Li et al. (2015) to extract potential microplastics from oysters. We placed each oyster tissue in a 1 L glass beaker and digested it with 250 ml of 30% hydrogen peroxide in an oscillator (3500 VWR Advanced) at 65 °C for 24 hs. In some cases, larger oysters took up to 48 hs to be fully digested. During digestion we covered the beakers with aluminum foil to prevent dehydration and airborne contamination of the samples. Each digestion cycle, we processed three samples and a blank extraction control. This number was determined by the maximum number of beakers that fit into the oscillator in the incubator (4 beakers total). We then added 750 ml of 25% saline solution to the digested solution. The water for the saline solution was filtered with a 1.0 µm pore size nuclepore hydrophilic membrane (VWR Whatman). In the saline solution, positively buoyant microparticles floated to the surface while heavier, biotic (undigested tissue), and abiotic (sediment) sank to the bottom. After 24 hs, we filtered the solution with 5 µm pore size Whatman nitrate cellulose membrane using a vacuum pump system under a fume hood. Finally, we placed each filter membrane in a labeled and sealed Petri dish and left it to dry for 48 hs.

Given that airborne microplastic contamination is very common (Foekema et al. 2013, Torre et al. 2016, Granek et al. 2020), we adopted a series of measures to prevent it. In addition to the control blanks throughout the protocol, we rinsed the glass beakers and other lab material with deionized filtered water three times prior to use (Li et al. 2015). To prevent contamination from our own clothing we used a 100% cotton laboratory coat during all

experimental procedures and analyses of the filters (Van Cauwenberghe & Janssen 2014).

## **Observation and quantification of microplastics and microparticles**

Determination of microplastics was performed using the definition proposed by Frias & Nash (2019). Only synthetic polymers identified by RMS, ATR-FTIR, or  $\mu$ -FTIR were considered microplastics. Other particles similar in size and visual characteristics to microplastics but with unknown composition are referred as microparticles throughout the manuscript. Once the filters were dry, we visually sorted and enumerated microparticles under a dissecting microscope. Visual sorting is one of the most common ways to quantify and identify potential microplastics, and differentiate morphotypes such as fibers (slender and elongated), spheres (round, similar to a ball in shape), flakes (small and very thin layer of larger plastic debris), and fragments (isolated or incomplete part of larger plastic debris that did not fit any of the previous categories; Hidalgo-Ruz et al. 2012, Li et al. 2016). The dimensions and coloration of the microparticles on the filters were obtained by taking pictures (Nikon Eclipse Ni camera with a 4x objective attached to a dissecting scope) and estimating their length using the open software ImageJ.

We used a d'Agostino test to test for normality in our data. The null hypothesis for this test is that data are normally distributed (not skewed) and the alternative hypothesis is that data are skewed. In our case the test indicated the data are skewed (skew = 2.38,  $z = 9.21$ ,  $p < 2.2e-16$ ). As a result, we used a Kruskal-Wallis rank sum tests were used to determine if there were significant differences in oyster weight per site and microparticle abundance between sites. A Spearman correlation test was also used to determine if there was a significant positive association between number of microparticles and oyster weight per site. All

analyses were carried out in R Statistical Computing software, version 3.5.3 (R Core Team 2019).

To account for potential sources of contamination we visually inspected the filters of blank extraction controls, and compared the microparticles found in them with those retrieved in the filters with sample after digestion. If there was an agreement between the type or color of microparticle in the samples, with the ones found in the controls, these microparticles were not considered towards the total count as they were considered to come from contamination.

### **Validation of microplastic identity through polymer chemical analyses**

After categorizing and measuring microparticles retained in filters we employed RMS and ATR-FTIR to determine the identity of the polymers that make up those microparticles. These non-destructive techniques record the signals absorbed or produced from individual microparticles, and are commonly used for microplastic identification due to their high accuracy in polymer recognition (e.g. Enders et al. 2015, Loder et al. 2015, Tagg et al. 2015, Imhof et al. 2016). RMS provides information about the structure of a molecule depending on its polarizability, while FTIR identifies the presence of certain functional groups in an organic molecule depending on the molecules' change in dipole moment (Hind et al. 2001).

In recent studies, FTIR has been more frequently used for marine microplastic identification than RMS (Lenz et al. 2015). However, RMS can provide advantages over FTIR such as lower minimum microparticle size identification (Lenz et al. 2015, Kappler et al. 2016). In addition, RMS measurements can be done on thicker or strongly absorbing microparticles because measurements do not depend on the transmission of excited light through the sample material (Lenz et al. 2015). Both RMS and FTIR have limitations in the identification of

244 molecules because of the different approaches used to determine the polymer identity.

245 However, when used in combination they become a powerful tool for polymer

246 characterization.

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248 Across ten locations, we randomly selected 69 oyster samples and 23 controls to analyze by

249 RMS. For each oyster (filter sample) we selected representative microparticles ( $n = 3-10$

250 depending on microparticle density on the filter) of various sizes and shapes and analyzed

251 them using a Renishaw inVia Raman microspectrometer with 785 nm and 514 nm lasers.

252 Two different objectives (5x and 10x) were used to optimize the analytical laser spot for

253 spectral analysis, and the laser power and acquisition times were varied depending on each

254 microparticle's sensitivity to thermal damage. Then, we individually and manually matched

255 each microparticle's Raman spectrum to a known plastic Raman spectrum in the Renishaw

256 Raman Database of Polymers library, containing 267 polymer entries (Renishaw). Each

257 spectrum was subjected to data processing depending on the signal-to-noise ratio and

258 fluorescent interference (which produces a curved, sloped baseline). Baseline correction

259 allows for the removal of a distorted spectrum before comparing to a known reference

260 spectrum. While some spectra received baseline correction (linear or polynomial baseline

261 correction available in the Renishaw Windows-based Raman environment software) not

262 every spectrum required baseline correction before library searching. The reference spectrum

263 from the library database matched when all the reference peaks (in wavenumbers) appeared

264 in the sample spectrum. Additional peaks appear in the sample spectrum because of

265 proteinaceous or biological contaminants or effects from environmental degradation. It is

266 important to note that all the spectra present in the polymer library correspond to virgin

267 polymers which have not been subject to any environmental degradation. Comparing spectra

268 between virgin polymers and non-virgin polymers exposed the marine environment can  
269 hinder the ability to detect and identify microplastics.  
270

271 In our samples, physical degradation (such as cracks, color change, or embrittlement) of the  
272 microplastics was not observed or quantified visually. However, it is possible that  
273 fragmentation, due to physical degradation, occurred as the microplastics identified were all  
274 less than 50  $\mu\text{m}$  in size. For the positively identified microplastics in our samples, the Raman  
275 spectrum and all of the major peaks matched the reference spectrum exactly indicating that  
276 degradation did not hinder our identification analysis via RMS (Lenz et al. 2015).  
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278 We also used a Bruker Vertex 70 ATR-FTIR to determine polymer identification of 70 filters  
279 with samples and 24 filters used as controls. Since the diamond ATR crystal has an area of 4  
280  $\text{mm}^2$  and the filters have an area of  $\sim 1662 \text{ mm}^2$ , each filter was analyzed in two different  
281 random areas to confirm whether or not the samples or controls were uniform. We first  
282 analyzed the control filters to obtain a background spectrum and capture background sources  
283 such as light or other residual signals from the atmosphere. Next, we obtained the samples'  
284 spectrum and subtracted the background spectrum from the sample's spectrum to obtain the  
285 FTIR spectrum which we then matched to an FTIR spectral library. Similar to RMS, the  
286 FTIR library contained only virgin polymer spectra.  
287

288  $\mu$ -FTIR was performed on selected individual microparticles from three filters that had  
289 confirmed microplastics from RMS to determine if both techniques detected the same  
290 contaminants.  $\mu$ -FTIR was performed on 5  $\mu\text{m}$  - 50  $\mu\text{m}$  microparticles with a Thermo  
291 Scientific Nicolet iNTM10 FT-IR microscope with a fixed focal length 15x objective. The

microparticles were isolated from the filters and, depending on the sample, were analyzed with either micro-ATR or micro-reflection techniques.

## RESULTS

### Observation and measurement of microparticles

Using a dissecting scope, we counted and measured microparticles from 213 of the 300 oysters collected throughout ten sites in Puget Sound (Table 2). This difference is due to the fact that several filters had plenty of digested organic matter and we could not process them. Of the 213 filters we were able to process (one filter per oyster), 63% ( $n = 134$ ) had at least one microparticle visible under the dissecting scope, and in 96% of the cases ( $n = 129$ ) these microparticles were microfibers. The rest of the microparticles (4% of the total) were classified as flakes or fragments. For 37% of the filters ( $n = 79$ ) we did not observe microparticles resembling plastic under the dissecting scope. Across all sites, mean microfiber length was  $621.93 \mu\text{m}$  ( $n = 276$ ), median fiber length was  $485.14 \mu\text{m}$  ( $n = 276$ ), with minimum and maximum values ranging between  $102.45 \mu\text{m}$  and  $2885.49 \mu\text{m}$  respectively ( $n = 276$ ).

When grouped per site, the mean number of microparticles per oyster ranged between 0.69 and 3 (Table 2, Fig. 2A) and there were significant differences between sites (Kruskal-Wallis rank sum test,  $\chi^2 = 32.84$ ,  $p = 0.0001$ ), where Oakland Bay and Samish Bay had significantly more microparticles than the other sites (Wilcoxon rank sum test,  $W = 322$ ,  $p = 0.0099$ , Fig. 2A ). Mean oyster weight per site ranged between 12.23 gr to 59.85 gr (Table 2, Fig. 2B). There were significant differences in oyster weights between sites (Kruskal-Wallis rank sum test,  $\chi^2 = 103.07$ ,  $p < 2.2 \cdot 10^{-16}$ ), where Oakland Bay oysters were significantly smaller (Wilcoxon rank sum test,  $W = 101.5$ ,  $p = 0.0006$ , Fig. 2B) and Samish Bay oysters were

significantly larger than the other sites (Wilcoxon rank sum test,  $W = 415$ ,  $p = 1.41^{-10}$ , Fig. 2B). However, overall, the number of microparticles was not associated with the biomass (size) of the oyster (Spearman's correlation,  $S = 15983$ ,  $p = 0.912$ , Fig. 2C).

In terms of coloration, the majority of the microparticles identified under the dissecting scope were dark (either blue or black), light (yellow, white, silver), clear, red, purple and green (Fig. 3A).

### **Validation of microparticles through polymer identification analyses**

We analyzed a total of 447 microparticles from oyster and control filters using RMS (Supplementary Table 1). The number of microparticles analyzed per site differed as it depended on the density of microparticles found on each filter. The shapes of the microparticles varied from fiber, shard, irregular, and spheroid with microparticles sizes ranging from 20  $\mu\text{m}$  to 13000  $\mu\text{m}$ . Out of the 447 microparticles examined, only eight (~2% of the total number of microparticles analyzed) were identified as microplastics (Fig. 4, Fig. 5, Fig. 6, Table 3). Seven of those microplastics were found in filters with oyster samples and only one microplastic (polystyrene) was found in one control filter. In the oyster samples, two microparticles were identified as polystyrene (PS), three particles as polypropylene (PP), and two as polyethylene (PE). All of these microplastic samples ranged in size from 50 - 150  $\mu\text{m}$  and the shapes were irregular, spheroid, or shard. In terms of coloration, the majority of the microparticles identified using RMS were clear, light (mostly opaque white), dark (either blue or black), and pink or purple (Fig. 3B).

RMS results showed that 10% ( $n = 46$ ) of the microparticles that were colorless or opaque, with an irregular or fiber-like shape, were identified as "cellulose", or a mixture of cellulose-

derived polymers (Fig. 4). These cellulose-derived microfibers were found on both control samples and oyster samples, but more predominantly on the control samples and had the same composition of the filter papers used. RMS also identified 18% (n = 81) of microparticles as either poly(ethylene glycol) monooleate or polyamide resin (Table 3, Fig. 4). Differentiation of these particles were not possible because of a combination of high fluorescence interference and because the spectra for the two polymers are similar. RMS results showed that 29% (n = 127) of the microparticles surfaces were biological, salt, or mineral in nature. These microparticles were predominantly < 75  $\mu$ m and of various shapes and are labeled as 'others' (Fig. 4). The remaining proportion of the microparticles analyzed, 41% (n = 185), contained high fluorescence interference, possibly from a layer of biofilm, and weak Raman signals which prohibited our ability to identify them despite organic matter removal efforts. Based on their RMS spectra, we suspect that some of these microparticles are biological materials (such as protein and carbohydrates), silicates, minerals, or shell fragments, similar to those found by Wagner et al. (2017).

Examinations using ATR-FTIR revealed that 100% of the filters examined with oyster samples from all sites (n = 71) contained a mixture of sorbitan derivatives such as sorbitan monopalmitate, sorbitan trioleate, sorbitan monolaurate, sorbitan monooleate, or polysorbate (a representative FTIR spectrum of sorbitan monopalmitate is shown in Fig. 7). Sorbitan derivatives were found consistently across the entire filter for all samples.  $\mu$ -FTIR was performed on selected particles in a small subset of our samples that had microplastics present.  $\mu$ -FTIR confirmed our RMS results such that the majority of the microparticles were identified as proteinaceous, fatty acid esters, and additionally identified the presence of polyester, rayon, poly(t-butyl acrylate), and poly(bisphenol A carbonate) (Table 3).



Due to the small number of microplastics found in this study we were not able to establish any comparisons between sites with different oceanographic conditions. Similarly, the small number of microplastics per site (Table 2) also limited our ability to determine any hotspots for microplastic contamination in the different basins of Puget Sound estuary.

## **DISCUSSION**

The global demand for plastics has consistently increased over recent years (Andrady 2011, Browne et al. 2011), leading to alarming rates of pollution worldwide (Gallo et al. 2018). Marine plastics have been estimated to exceed 5 trillion pieces (Eriksen et al. 2014), posing challenges for the health of marine ecosystems and the industries that depend on them. This first baseline of microplastic presence in Pacific oysters from Puget Sound revealed five major findings: (1) ~2% of the microparticles were identified as microplastics using RMS or  $\mu$ -FTIR, (2) microplastics in oysters were present in five out of ten sites examined, (3) sorbitan derivatives were present in all oyster samples, (4) 41% of microparticles showed fluorescence interference and could not be identified with RMS, and (5) when observed under a dissecting scope, fibers were the most common type of microparticle found, but most of these fibers did not have a plastic composition as indicated by RMS and  $\mu$ -FTIR.

### **Particles identified under the dissecting microscope**

Oysters from all ten sites had microparticles that were identified mostly as microfibers (96% of microparticles) under a dissecting microscope. Relative abundances of these microparticles varied between sites but mean values ranged between 1 and 4 microparticles per oyster per site (Fig. 3). These findings agree with what has been reported for bivalves from Europe (Van Cauwenberghe & Janssen 2014, Li et al. 2018), China (Li et al. 2015, Li et al. 2016, Qu et al.

2018), Brazil (Santana et al. 2016), Canada (Davidson & Dudas 2016), and the United States (Rochman et al. 2015), among many others. Because visual sorting has limitations for accurate polymer identification, we complemented our observations with three different techniques that allow for polymer identification: RMS, ATR-FTIR, and  $\mu$ -FTIR.

### **Plastic particles identified using RMS, ATR-FTIR, and $\mu$ -FTIR**

The majority of microparticles in our study were analyzed using RMS, and this technique was complemented with ATR-FTIR and  $\mu$ -FTIR for a smaller set of samples. Out of 447 microparticles examined, only eight particles were identified as microplastics by RMS and their spectra matched with polypropylene (PP), polyethylene (PE) or polystyrene (PS) (Fig. 6). These three polymers have been identified in surface waters and sediments in the Atlantic (Moret-Ferguson et al. 2010, Enders et al. 2015, Woodall et al. 2014, Courteney-Jones et al. 2017) and North Pacific Oceans (Rios et al. 2007), and also in bivalves from China and Europe (Li et al. 2016, Li et al. 2018, Li et al. 2018). PE along with PP are the most commonly used hydrocarbon polymers (Da Costa et al. 2018), and there are many different sources and a wide range of domestic and industrial applications for them (Andrady 2011, Zhao et al. 2018). PP for example, is used for rope, bottle caps and nettings (Smith et al. 2018). High and low density PE are used for milk and juice jugs, plastic bags, six pack rings, netting and drinking straws (Smith et al. 2018), and PS is used for plastic utensils and food containers (Smith et al. 2018).

RMS showed that 81 microparticles were either poly(ethylene glycol) monooleate and/or polyamide resin. The similar Raman spectra of these two compounds made differentiation of

the exact chemical identification challenging even with samples that have not been exposed to environmental factors. This challenge coupled with high fluorescence interference and weak Raman signal rendered differentiation impossible for the microparticles analyzed. However, poly(ethylene glycol) monooleate is water soluble and it is likely that this compound was washed away during the sample processing. If this was the case, then the compound in the oyster samples is most likely a polyamide resin. Given that polyamide resins are not ubiquitous or widely used, it is surprising they are found in high abundance in the oyster. If these resins are synthetic in origin they should be considered microplastics (Frias & Nash 2019), but the resins could also have a natural origin if they are the result of the oxidation of oyster protein by hydrogen peroxide. To test these hypotheses we ran a simple test to determine if the polyamide resins could be natural in origin by doing the same digestion protocol we used for the oyster samples on an egg. Eggs are also high in protein and are a closed system that should be naturally free of microplastics. The filter resulting from the egg processing was scanned with RMS and the Raman spectra matched the one observed for all the polyamide resin particles found in our samples (Supplementary Image 1). While this test is not conclusive, it provides strong support to the idea that these polyamide resins present in every filter can come from oysters instead of synthetic sources. Further, investigations using ATR-FTIR revealed that 100% of the filters with oyster samples from all sites contained sorbitan derivatives. While ATR-FTIR has a penetration depth of  $\sim 2\text{ }\mu\text{m}$ , RMS has a penetration depth of  $12\text{ }\mu\text{m}$  at  $785\text{ nm}$ . Thus, it is possible the microparticles are poly(ethylene glycol) monooleate or polyamide resins coated with sorbitan derivatives.

Both sorbitan derivatives and poly(ethylene glycol) monooleate are most often used as

437 emulsifiers and surfactants (Bobin et al. 1999). Sorbitan derivatives are not always polymeric  
438 (polysorbate being a polymeric sorbitan derivative) and are considered synthetic waxes  
439 derived from the dehydration of sorbitol. Sorbitan derivatives such as sorbitan monolaureate,  
440 monooleate, and monostearate are FDA-approved for oral administration up to 25 mg per kg  
441 body weight as food additives (Nielloud and Marti-Mestres 2000). In 2018, the European  
442 Food Safety Authority re-evaluated the use of sorbitan derivatives as food additives and  
443 found that the acute toxicity of these compounds is low, however, more data is needed to  
444 decrease uncertainty in exposure assessment (European Food Safety Authority 2015).

445

446 Other ubiquitous polymers present in the samples were cellulose-derived. It is likely that  
447 these polymers originated during the extraction protocol. We used filters that are made up of  
448 nitrate-cellulose and it is possible that some degradation occurred during the peroxide  
449 treatment that led to cellulose particles in the filters. Future studies could benefit from using  
450 other types of filters such as glass filters to solve this. Other cellulose-derived polymers that  
451 were identified by RMS included cellulose acetate which are used in cigarette filters, hygiene  
452 products, and clothing (Woodall et al. 2014, Andrady 2015). This contaminant is usually  
453 introduced in marine environments via sewage discharge (Browne et al. 2011).

454

455 Lastly, four additional microparticles were identified as plastics/synthetic polymers with  $\mu$ -  
456 FTIR. These microparticles came from three of the same filters also used for RMS and were  
457 identified as polyester, rayon, poly(*t*-butyl acrylate), and poly(bisphenol A carbonate).  
458 Polyester is a synthetic polymer that is petroleum based and mostly used as fibers used to  
459 manufacture fabrics that make up blankets, fleece and other clothing items. Rayon in contrast

is considered a semi-synthetic fiber derived from wood pulp. Rayon is mostly used for clothing, and when it is bleached with dioxins to make white clothes it can become toxic. Poly(*t*-butyl acrylate) is a polymer mostly used in paints, coatings, adhesives, fuel and textiles but there is not enough information available about its toxicity to marine animals. Poly(bisphenol A carbonate) or polycarbonate (PC) is a polymer formed by monomers of bisphenol A carbonate (BPA) and is one of the most widely used thermoplastics (Quaranta et al. 2017). PC is used in a wide range of industrial applications, such as automotive and transportation, building and construction, packaging, medical, data storage, and interactive software media (Siddiqui et al. 2018). Traditional methods of PC disposal (such as landfilling and incineration) are related to the leaching of BPA and PBA products (Siddiqui et al. 2018). Previous research has also shown that BPA can be released when polycarbonates are biodegraded by marine microorganisms (Artham & Doble 2012). PC found in one of the oyster filters warrants some concern as degradation of this polymer to BPA may induce differential effects in the gonads of male and female oysters (Luo et al. 2017). BPA has received attention because it is mildly estrogenic (Im & Loffler 2016) and can be an endocrine disruptor toxic to marine organisms (Artham & Doble 2012).

#### **Methods to identify plastic polymers and associated caveats**

Microplastics research is a field that is growing at an exponential pace but there are still several challenges associated with the chemical identification of microparticles. The identification of polymers is challenging in part due to the large number of mixtures of polymers that are produced by manufacturers today. Therefore, no single analytical approach is perfect to determine the composition of microparticles (Elert et al. 2017), and studies

aiming to chemically identify microparticles can benefit from using different methods to analyze microplastic composition. This is hard to achieve however, as techniques such as RMS, FTIR, and  $\mu$ -FTIR require training and expertise to use the equipment appropriately (Granek et al. 2020). These methods are also costly and time consuming, particularly in the case of RMS.

Of the three, ATR-FTIR is the least costly and is most suitable to analyze particles that are larger than 100  $\mu\text{m}$ . The caveat of ATR-FTIR is that it can detect materials that are coated on the surface of a sample, but may not identify the underlying material as it only has a penetration depth of  $\sim 2\text{ }\mu\text{m}$ . Another disadvantage with ATR-FTIR is that brittle samples can fracture upon contact with the diamond ATR (Wagner et al. 2017). ATR-FTIR and  $\mu$ -FTIR use mid-IR light while RMS uses sub-micron light (usually a laser source) to probe microparticle identities. Therefore, while IR can only identify particles larger than 10-20  $\mu\text{m}$ , RMS has the ability to resolve samples larger than 1  $\mu\text{m}$  (Silva et al. 2018). Both  $\mu$ -FTIR and RMS require low amounts of sample, can identify particles with minimal sample preparation, and distinguish between plastics and natural particles from marine organisms or soil (Silva et al. 2018). Compared to Raman, FTIR has been around longer and the technology has evolved with the polymer industry. There is a greater abundance of historical IR spectroscopic data for polymer analysis and thus more available reference spectra for microplastic identification (Araujo et al. 2018).

There are also different caveats that need to be taken into account for each technique, such as the issue of fluorescence interference with RMS (Araujo et al. 2018). In our study, almost 41% of the microparticles analyzed showed high fluorescence and this limited our ability to

detect the RMS signal and identify the composition of the particles. Thus, it is likely we may have underestimated the amount of microplastics present in the samples. The sources of fluorescence in microparticles can be many. Coloring agents, biological material, or degradation products from the environment can attach/adsorb to the plastic particles and influence their spectra. Many colorants such as pigments and dyes strongly fluoresce in visible light and can also hinder the acquisition of spectra, while inorganic pigments are more likely to change the pattern of peaks that are displayed (Frederiks 2012). Bacteria present in biofilm (Araya et al. 2003, Rummel et al. 2017), and algal phaeopigments (Carlson & Shapiro 1981, Mitchell & Kiefer 1988) can also be major contributors to fluorescence. To reduce fluorescence interference in some of our samples, it was necessary to bleach or illuminate them with the laser before spectral acquisition. This treatment, however, occasionally led to microparticles degradation.

Another caveat are additive compounds included to the basic polymer matrix when commercial plastics are manufactured (Lenz et al. 2015). These added compounds can alter the reading of the plastic polymers by RMS. Similarly, plastic microparticles obtained from the marine environment can have altered chemical fingerprints as a result of weathering (Lenz et al. 2015). These alterations can sometimes lead to a falsified polymer type signal (Lenz et al. 2015) and misidentification. This issue has been mentioned as one of the main drawbacks of matching spectra to a library because reference libraries cannot cover the whole variety of particles present in marine samples, both in terms of additives and/or degradation (Lenz et al. 2015). As this challenge begins to be recognized, several authors have recommended that spectra from degraded polymers are also included in polymer libraries in

order to increase the possibilities of identification of microparticles that have been exposed to environmental conditions. Lenz et al. (2015) showed for example, that PVC can be degraded to a point where successful identification is not possible if using a virgin PVC reference. Similarly, when exposed to artificial seawater, PE pellets undergo considerable structural modifications as a result of the formation of new functional groups in the polymer that ultimately yield polymers with a distinct composition (Da Costa et al. 2018).

## **Conclusions and implications of this research**

The findings presented here indicate that visual sorting should not be assumed to be a reliable method for microplastic identification. Over 60% of the oysters analyzed had a microparticles that looked like a microplastics however only 2% of the microparticles observed were confirmed to have a plastic composition. Our research further shows that Pacific oysters from Puget Sound in Washington State, USA, are not accumulating large amounts of microplastics, on the contrary, we have no evidence to support the idea that there is more than one microplastic per oyster. As pointed out previously, 41% of the samples were covered with fluorescent material and the methods have caveats to consider, but based on the 60% of the samples that were identifiable, the presence of microplastics is very low. This does not undermine in any way the seriousness and emergency of plastic contamination in marine environments and its effects on marine biota. This research instead provides the first baseline value for microplastics in Pacific oysters from Washington State based on three different techniques - in addition to visual sorting. Information from this first baseline study will give ecosystem managers, the aquaculture industry, and the general public an important indicator of bivalve health. This pioneer contribution can be considerably improved by analyzing more of the oyster samples with  $\mu$ -FTIR, by managing fluorescence interference



under RMS, and using additional chemical identification techniques. Having a more comprehensive dataset of results as well polymer libraries with spectra of polymers that were exposed to the environmental degradation, will allow scientists to identify with more accuracy the types of microplastics, resins, additives, and other contaminants present in marine organisms.

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## FIGURE CAPTIONS

**Figure 1.** Location of Puget Sound and its basins in Washington State, USA. The five basins indicated are Juan de Fuca (JF) Strait in purple, Hood Canal in green, South Puget Sound (PS) in yellow, Central Puget Sound in red and Whidbey Basin in blue. The image on the right shows the study species, the Pacific oyster, *Crassostrea gigas*.

**Figure 2.** Panel showing (A) boxplots with the number of microparticles per oyster that were identified using visual sorting under the dissecting microscope. Site names are organized alphabetically. Thick horizontal lines represent median values per site, boxes enclose the 25<sup>th</sup>-75<sup>th</sup> percentiles and whiskers indicate the minimum and maximum values, (B) boxplots showing mean oyster weight per site, (C) scatterplot of the relationship between mean oyster weight and number of microparticles per oyster for all pooled sites. R-squared value for regression: 0.003.

**Figure 3.** Pie charts showing (A) the percentage of different colors observed in microparticles identified as microfibers under the dissecting scope (size ranges: 102 to 2885  $\mu\text{m}$ ) and (B) the percentage of different colors observed in microparticles under RMS (size ranges between 20 to 50  $\mu\text{m}$ ). In a clockwise direction, ‘clear’ indicates transparent particles, ‘red’ indicated red and pink particles, ‘purple’ indicates violet and purple particles, light indicates white, yellow, silver and amber particles, ‘green’ indicates greenish particles and ‘dark’ indicates all particles that were blue, black or dark in coloration.

**Figure 4.** Map showing the location of sampling sites (1-10) in Puget Sound, Washington. Site names are organized N-S as follows: (1) Samish Bay, (2) Sequim Bay, (3) Mystery Bay, (4) Jacoby State Park, (5) Heritage County Park, (6) Illahee State Park, (7) North Bay, (8)

Kopachuck State Park, (9) Penrose Point, (10) Oakland Bay. The barplot to the right shows the proportion of different microparticles identified by RMS per site. It is important to note that we did not process 100% of the particles per filter therefore the graph does not represent all the particles that were present, only the ones that were randomly sampled. It is also of note that the number of particles sampled varied per site as it depended on the density of particles per filter (number of microparticles analyzed per site is shown in Table 2). ‘Cellulose’ are mainly cellulose fibers, ‘fluorescence’ are unidentified particles due to high fluorescence interference, ‘others’ are shell particles, grains of sand, and gypsum, ‘resins’ are polymers such as sorbitan monopalmitate, and ‘plastics’ are particles that were identified as polyethylene (PE), polypropylene (PP) and polystyrene (PS).

**Figure 5.** Size distribution and identification of microparticles from oyster samples using RMS. Most microparticles ranged between 20 to 50  $\mu\text{m}$  in size. The identity of the particle is color coded according to the key in the top right-hand corner of the image. The darker blue color is an overlap between the blue and pink colors. Polystyrene (PS), polypropylene (PP), and polyethylene (PE) microparticles (in red) were all less than 150  $\mu\text{m}$ . One 13000- $\mu\text{m}$  microfiber particle and particles categorized as others (such as gypsum, salt, minerals, or shells) were omitted from the histogram for clarity.

**Figure 6.** Spectra of microplastic particles identified using RMS. Red spectra are the individual plastic RMS spectra and the blue spectra are the reference library spectra. (a) North Bay Oyster 3a is a 75- $\mu\text{m}$  polystyrene shard, (b) North Bay Oyster 3e is a 100- $\mu\text{m}$  polypropylene irregular-shaped particle, (c) North Bay Oyster 3h is 50- $\mu\text{m}$  polystyrene shard, (d) North Bay Oyster 13d is a 150- $\mu\text{m}$  polypropylene spheroid, (e) Oakland Oyster 4d is a 100- $\mu\text{m}$  polyethylene shard, (f) Samish Control 4-6a is a 100- $\mu\text{m}$  polystyrene irregular-

shaped particle, (g) Sequim Oyster 25d is a 150- $\mu\text{m}$  polypropylene shard, (h) Jacoby Oyster 31b is a 125- $\mu\text{m}$  polyethylene irregular-shaped particle.

**Figure 7.** FTIR spectrum of North Bay oyster 1 (in red) and FTIR reference spectrum of sorbitan monopalmitate (in blue). This spectrum is representative of the large majority of the spectra observed with FTIR.

**Figure 8.** Microplastic particles identified from three filter samples using  $\mu$ -FTIR. Red spectra are the individual microplastic  $\mu$ -FTIR spectra and the purple or black spectra are the library reference spectra. (a) Oakland Oyster 4a is a polyester fiber, (b) Oakland Oyster 4b is a rayon fiber, (c) Oakland Oyster 4c is a poly(t-butyl acrylate) irregular-fragment, (d) Jacoby Oyster 31a is a poly(bisphenol A carbonate) particle.

## TABLE CAPTIONS

**Table 1.** Sampling sites organized N-S according to the Puget Sound basin where they are located (see Fig. 1). Site coordinates and characteristics of the basins are also included. ‘SP’ indicates State Park and ‘CP’ indicates County Park.

**Table 2.** Number of oysters processed under the dissecting scope and RMS per site. For microparticles identified under dissecting scope we present the mean value of the ones that were visually sorted as microplastics, and the mean value per gram of oyster. For microparticles identified under RMS we present the number of particles analyzed per site and the percentage of those that were confirmed as plastic.

**Table 3.** Number and type of pollutants found per site. The main source of each pollutant is indicated, as well as the method of detection in oyster tissue samples from Washington state.

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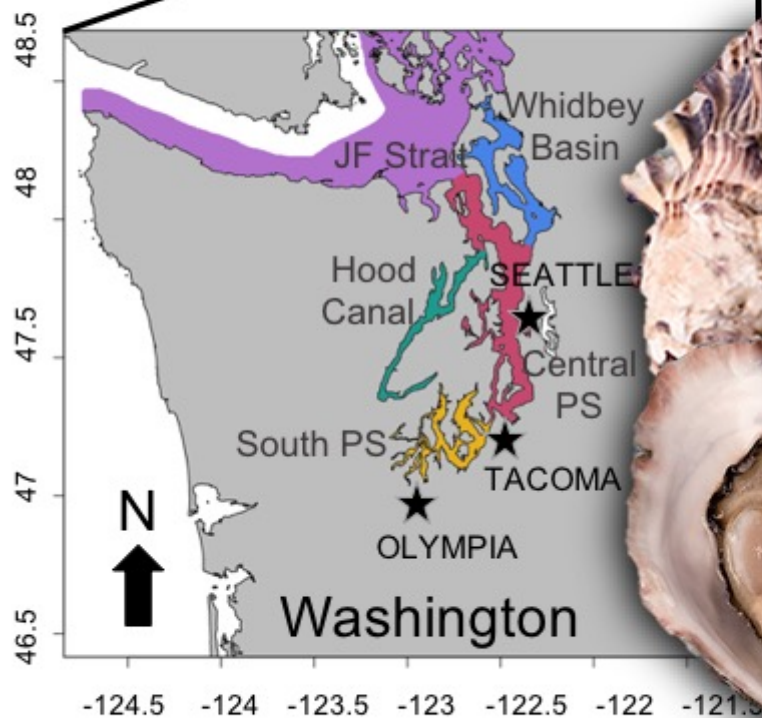
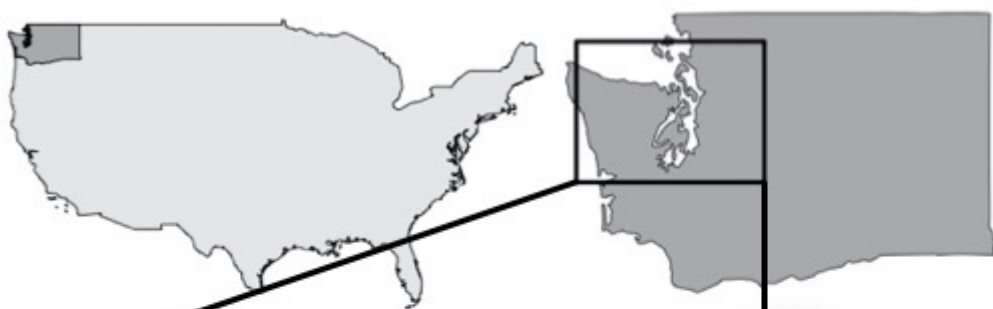
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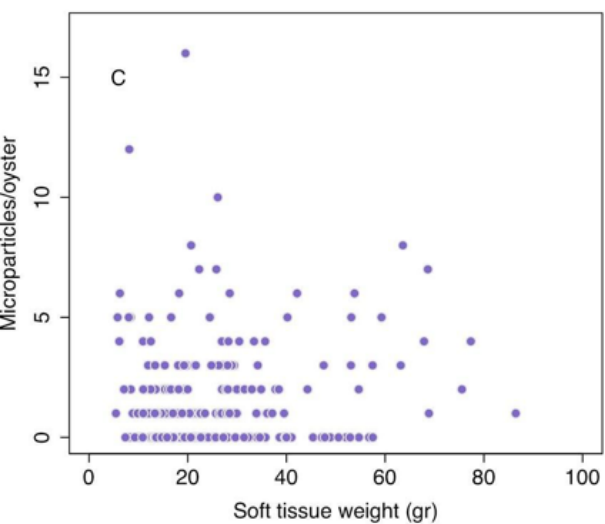
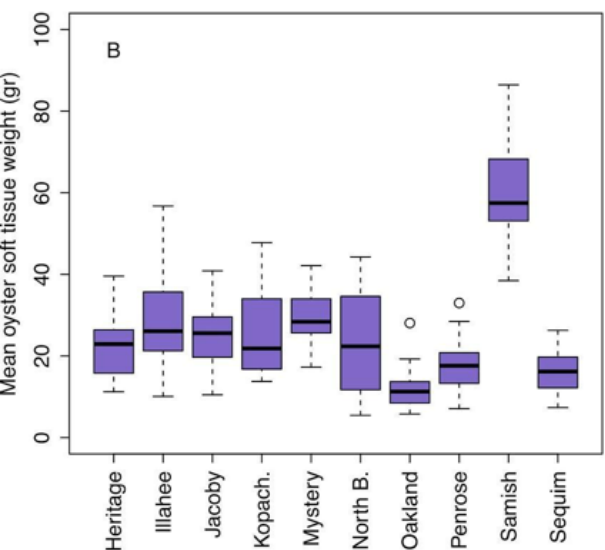
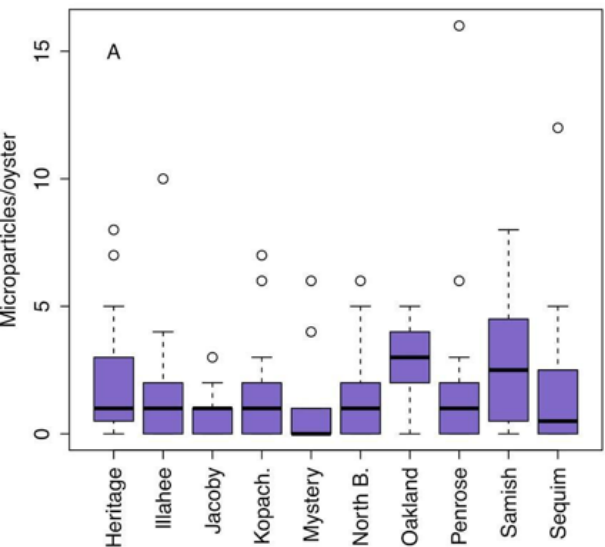
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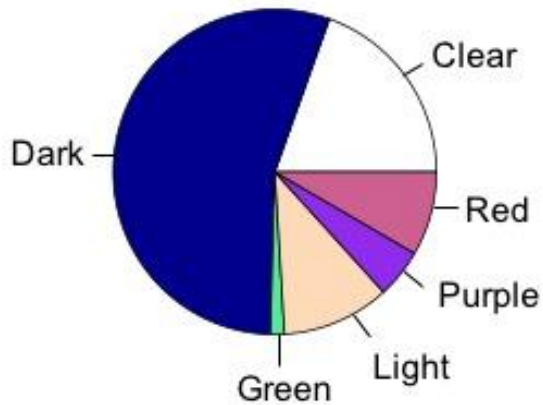
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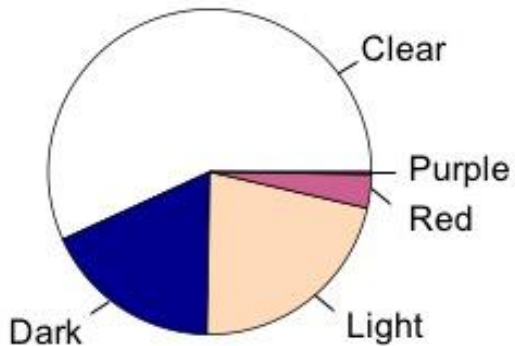




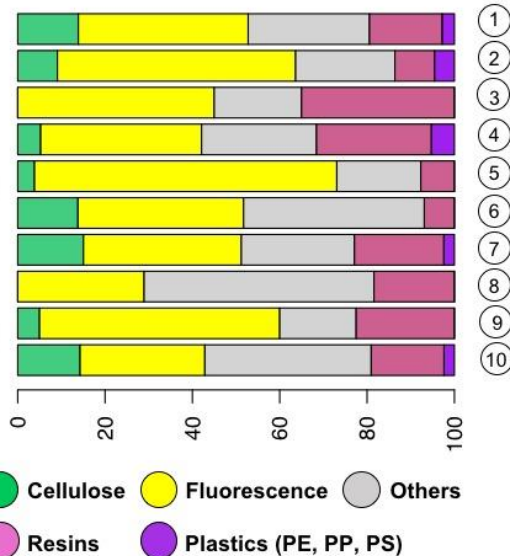
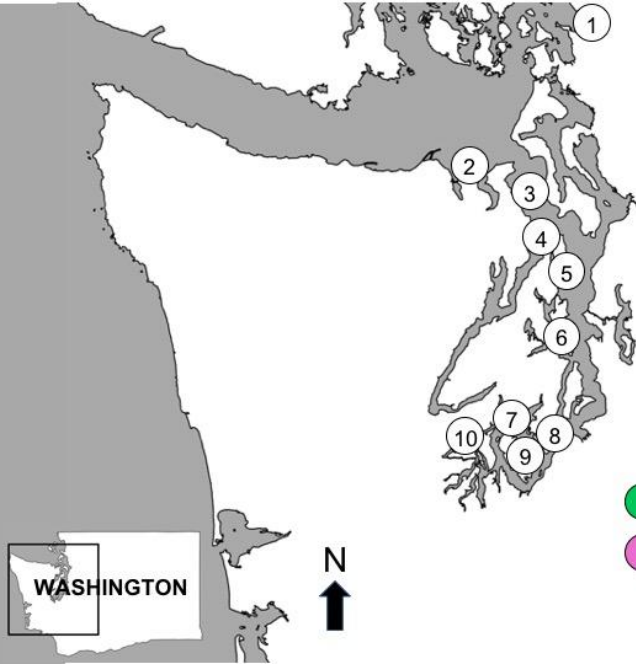
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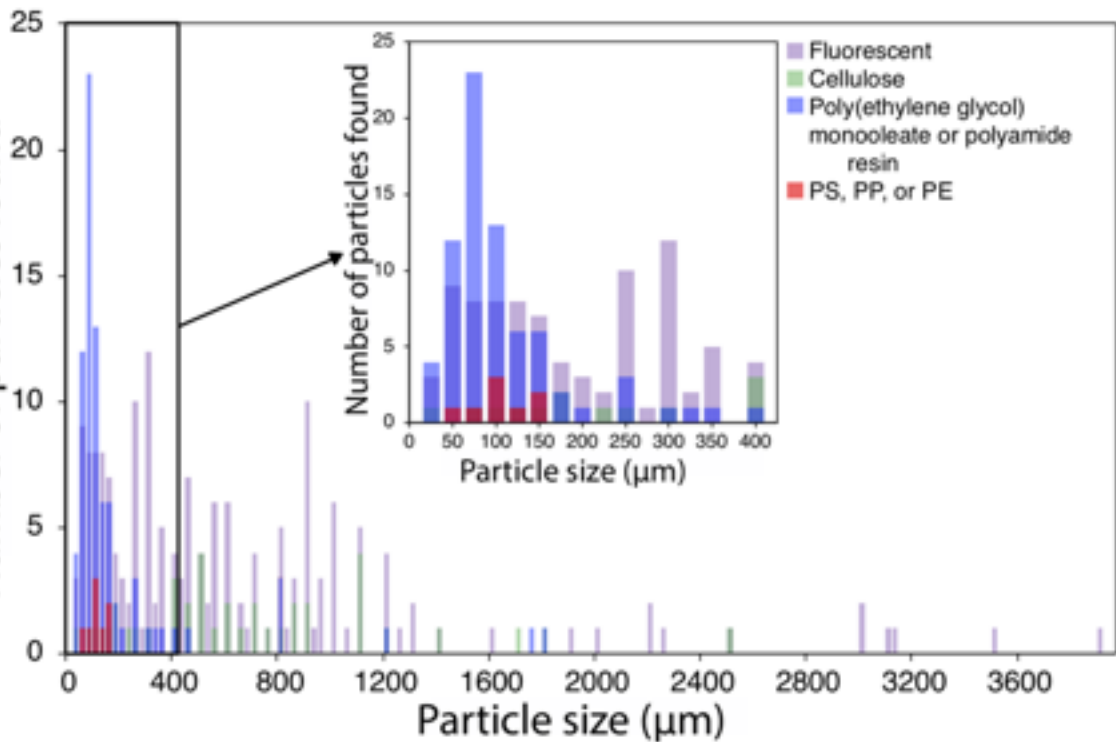
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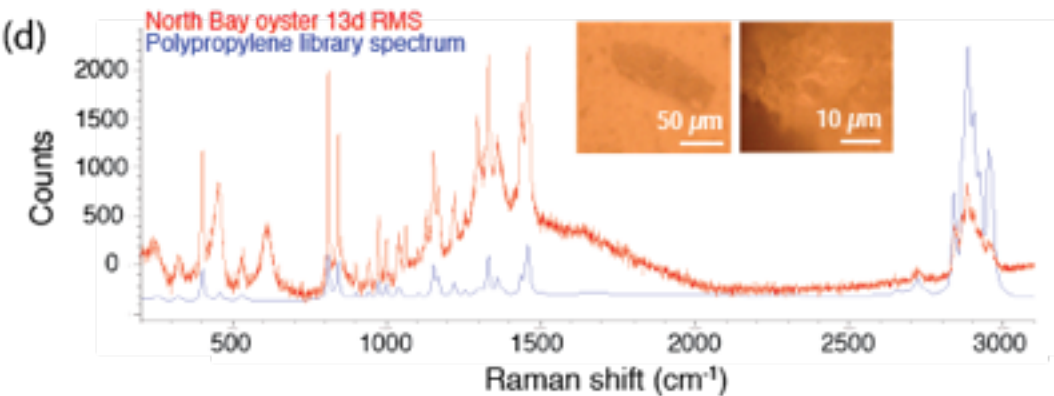
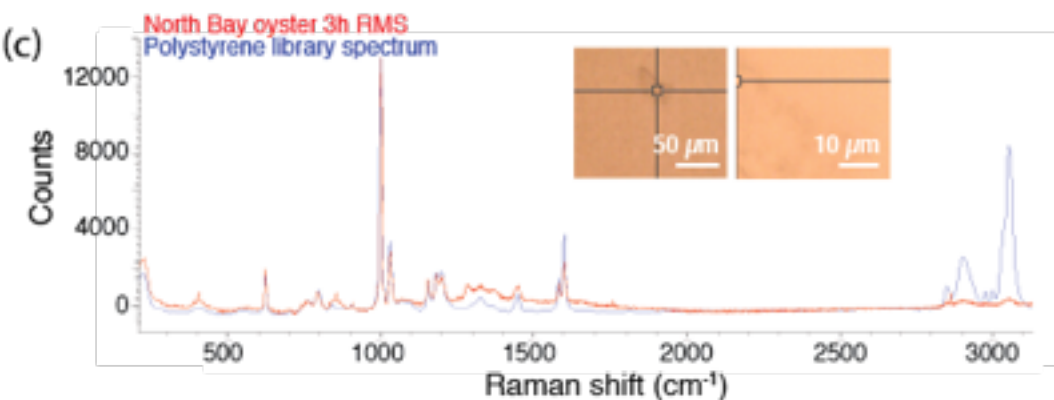
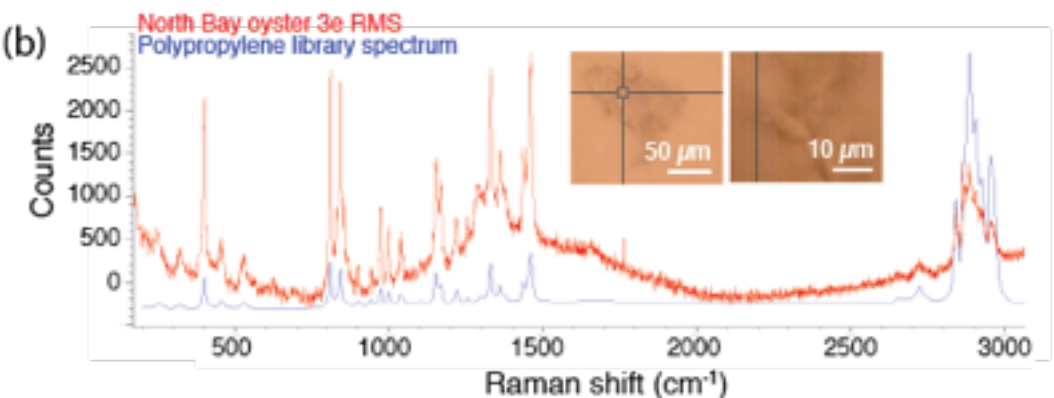
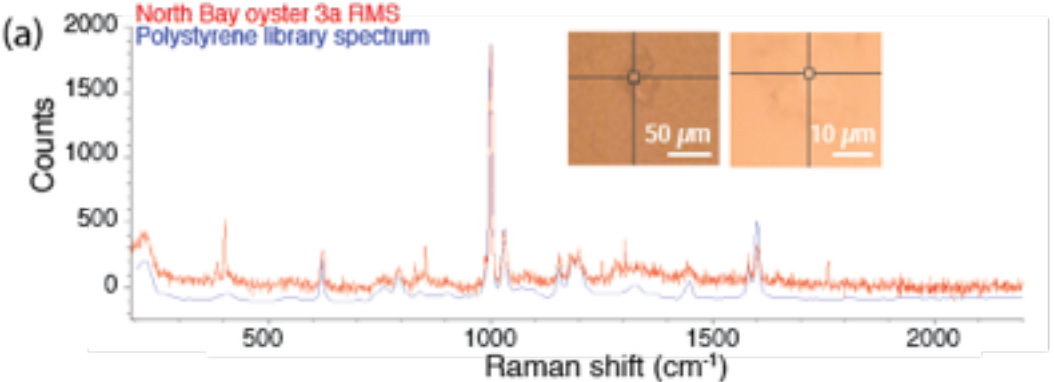


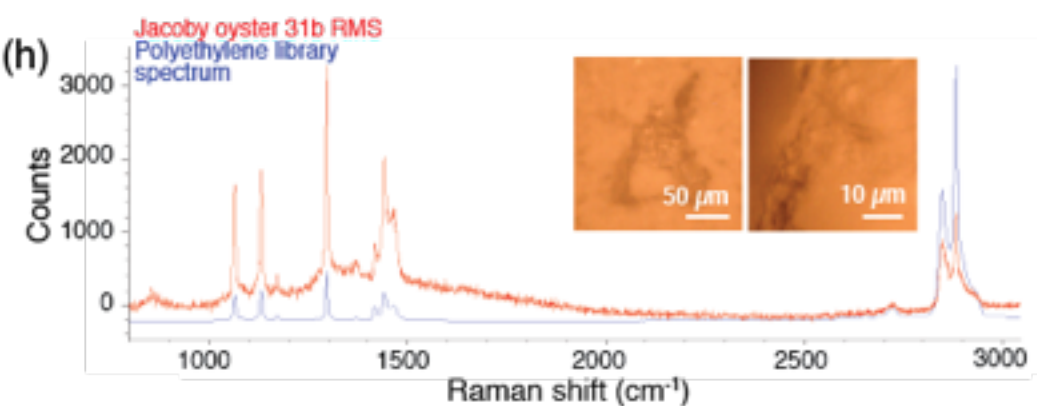
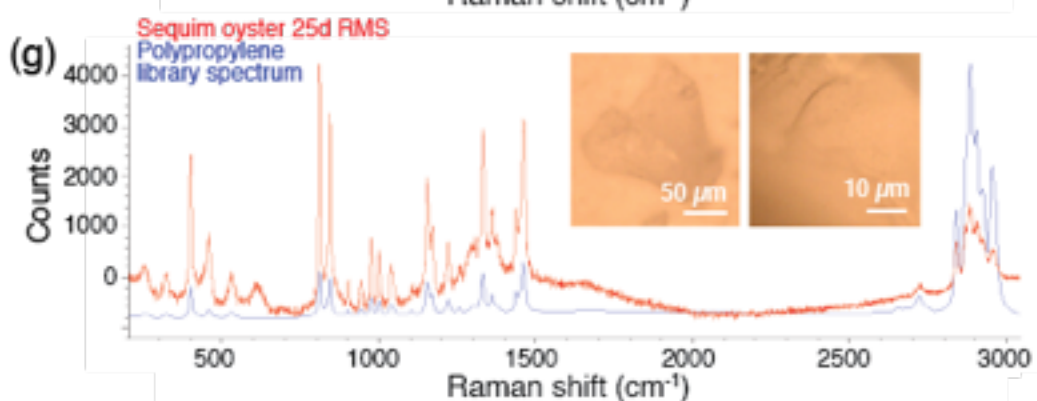
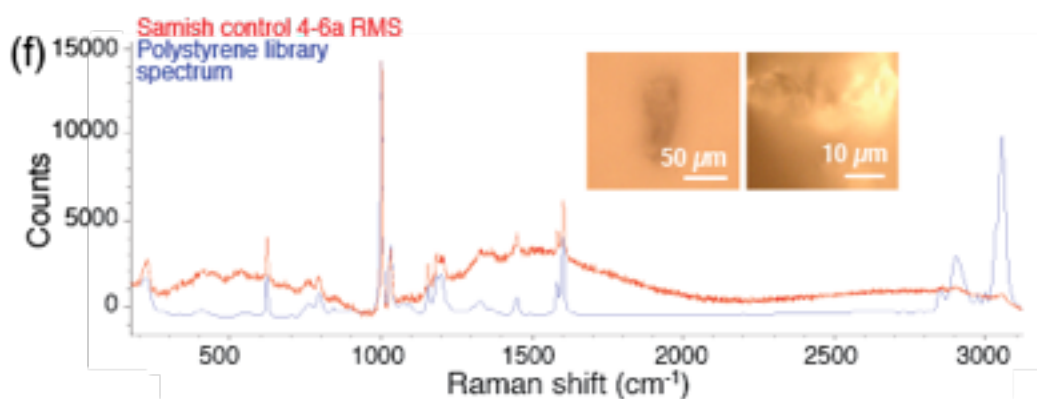
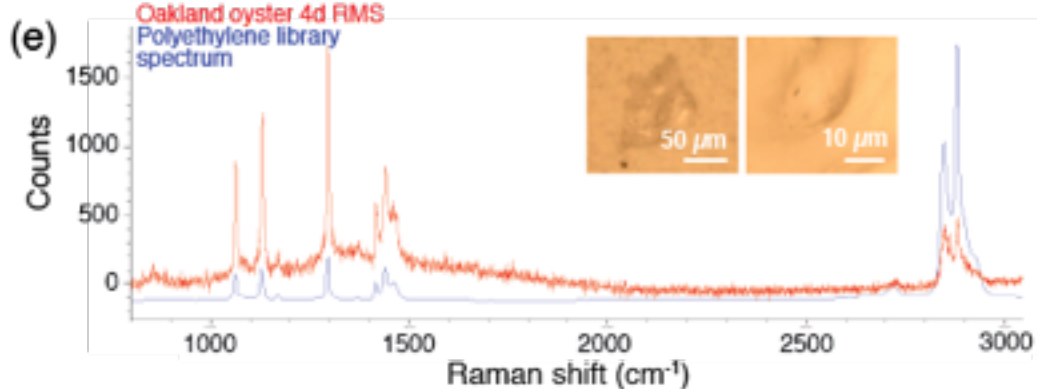


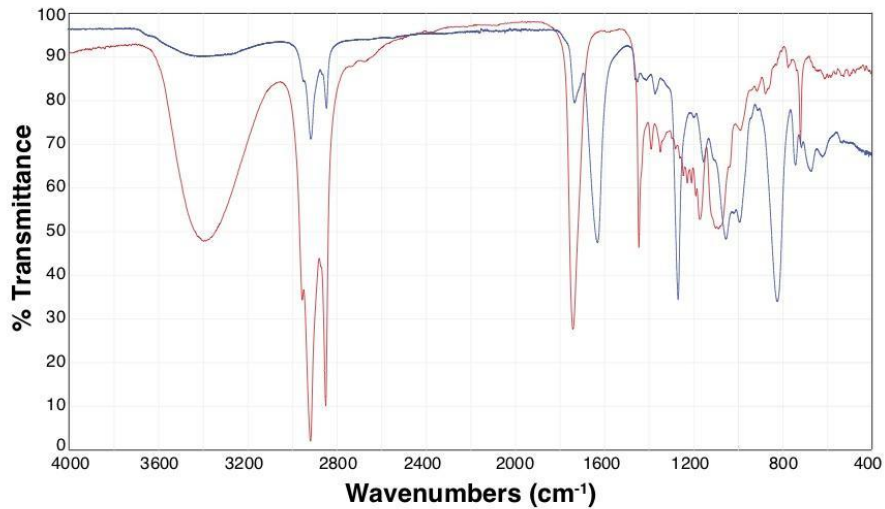


Number of particles found









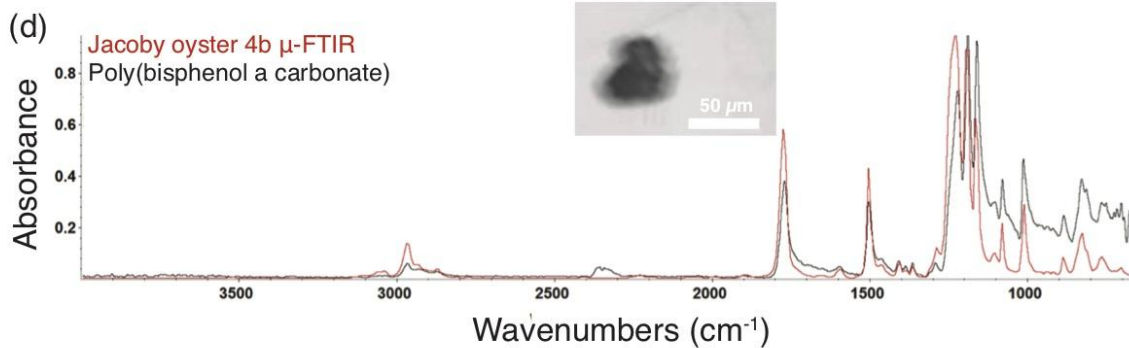
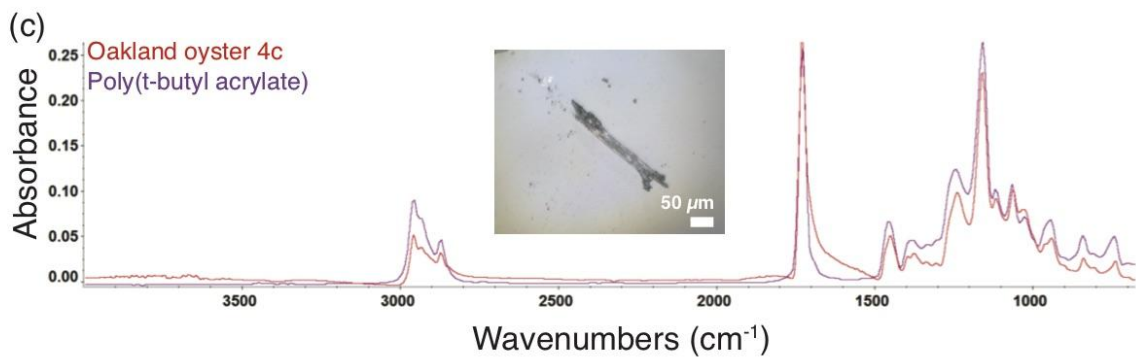
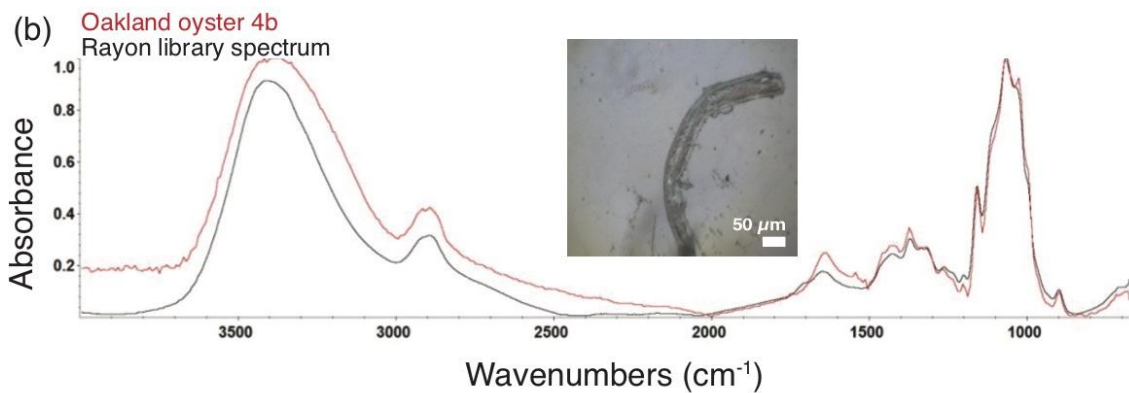
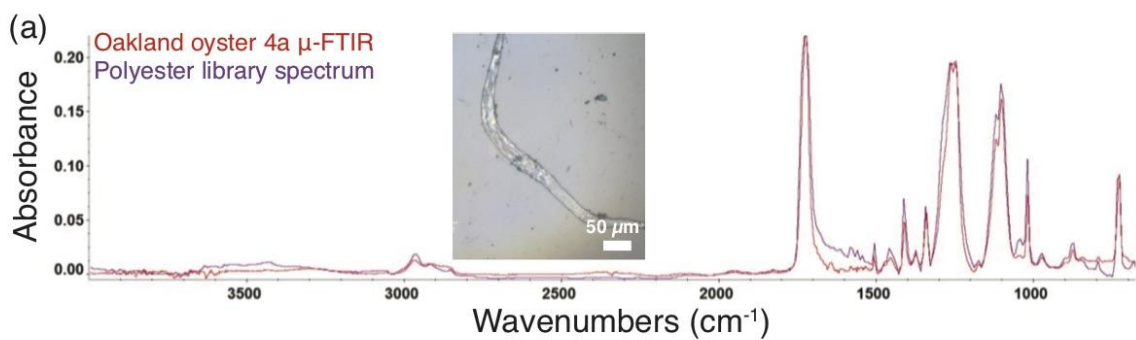


TABLE 1

Basin	Site	Coordinates	Characteristics
Juan de Fuca Strait	Sequim Bay	48° 2' 26" N, 123° 2' 6" W	Open basin connected to the Pacific Ocean shorter water residence times than average for Puget Sound (approximately 1 month). Sequim Bay is however narrow and north-facing, factors that increase water residence times relative to the rest of the basin.
Whidbey Basin	Samish Bay	48° 35' 23" N, 122° 30' 7" W	Close to Juan de Fuca Strait, shorter water residence times than average for Puget Sound (approximately 1 month)
Central Puget Sound	Heritage CP	47° 50' 24" N, 122° 35' 16" W	This basin includes areas draining into small creeks that flow directly into Puget Sound from Seattle and off Vashon Island. Water residence time approximately 2 months.
	Illahee SP	47° 35' 48" N, 122° 35' 57" W	
	Jacoby SP	47° 52' 7" N, 122° 38' 12" W	
	Mystery Bay SP	48° 3' 34" N, 122° 41' 50" W	
South Puget Sound	Kopachuck SP	47° 18' 31" N, 122° 41' 16" W	Long, narrow bays, islands and small inlets. Water residence time approximately 2 months.
	North Bay	47° 23' 25" N, 122° 48' 51" W	
	Oakland Bay	47° 13' 29" N, 123° 4' 8" W	
	Penrose Point	47° 15' 32" N, 122° 44' 54" W	

TABLE 2

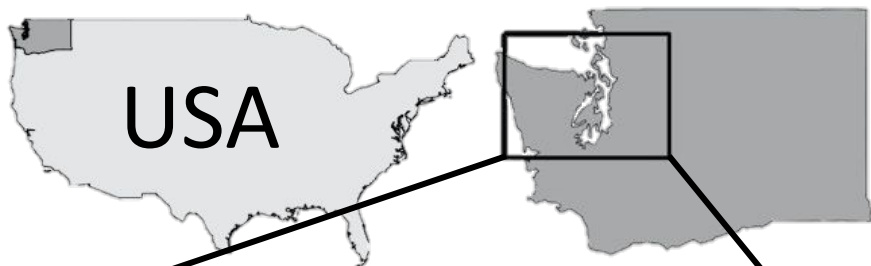
Site	Number of oysters processed	Mean oyster weight	Mean number of particles per oyster (using a dissecting scope)	Mean number of particles per gram (using a dissecting scope)	Particles examined using RMS	Percentage of RMS plastic particles (%)
Heritage CP	20	22.41	2.05	0.09	35	0
Illahee SP	21	28.98	1.71	0.07	28	0
Jacoby SP	23	25.17	0.69	0.03	19	5.3
Kopachuck SP	22	26.66	1.36	0.05	38	0
Mystery Bay SP	21	29.48	0.76	0.02	21	0
North Bay	21	23.48	1.62	0.11	167	2.4
Oakland Bay	21	12.23	3	0.3	42	2.4
Penrose Point SP	24	18.18	2	0.11	40	0
Samish Bay	20	59.85	2.8	0.05	37	2.7
Sequim Bay	20	16.23	1.7	0.14	22	4.5



TABLE 3

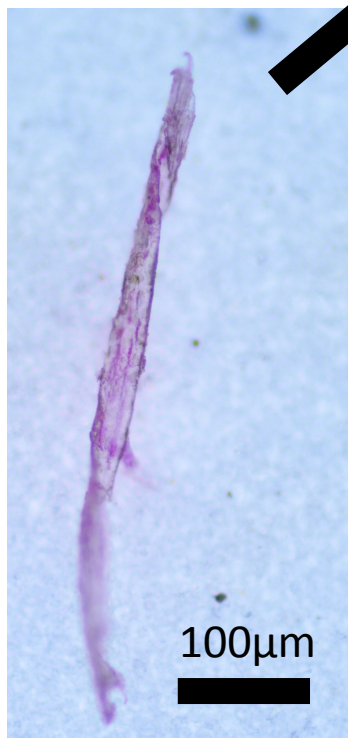
Pollutant	Possible Sources	No. particles found	Sites	Method used
Polyethylene (PE)	Milk and juice jugs, plastic bags, six pack rings, drinking straws	2	Jacoby SP, Oakland Bay	RMS
Polystyrene (PS)	Plastic utensils, food containers	3	Samish Bay, North Bay	RMS
Polypropylene (PP)	Rope, bottle caps, nettings	3	Sequim Bay, North Bay	RMS
Polyester	Fabrics, fibers and outerwear	1	Oakland Bay	$\mu$ -FTIR
Rayon	Semi-synthetic fiber derived from wood pulp	1	Oakland Bay	$\mu$ -FTIR
Poly (t-butyl acrylate)	Paints, coatings, adhesives, textiles	1	Oakland Bay	$\mu$ -FTIR
Poly (bisphenol A carbonate)	Automotive and transportation, building and construction, packaging, medical, storage devices, interactive software media	1	Jacoby SP	$\mu$ -FTIR
Poly(ethylene glycol) monooleate or polyamide resins	Emulsifiers, surfactants, printing inks	81	All sites	RMS

Sorbitan derivatives	Emulsifiers, surfactants	Non-solid polymers (uncountable)	All sites	ATR-FTIR
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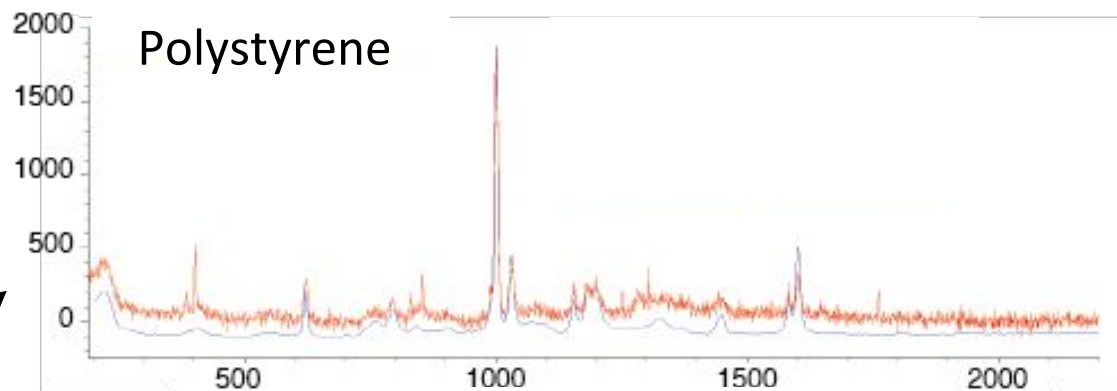
300 Pacific oysters

447 microparticles processed

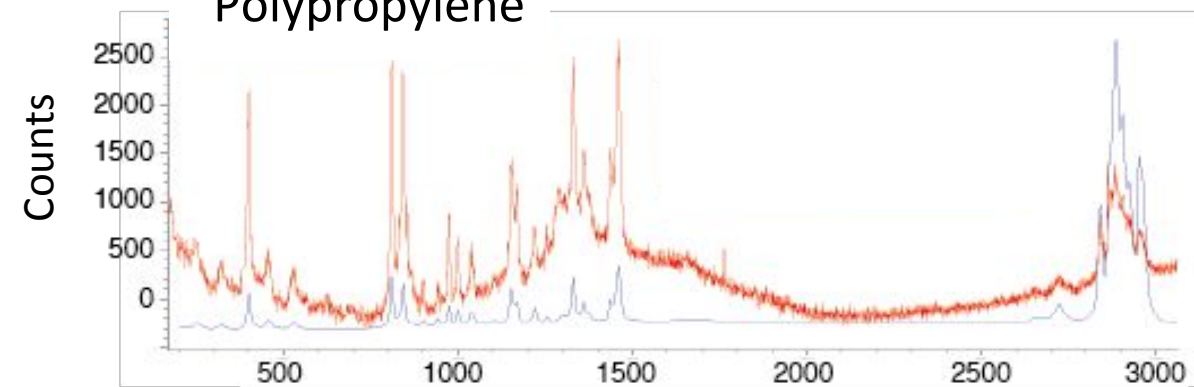


Counts

Polystyrene



Polypropylene



Polyethylene

