

1 **Evaluating ammonia (NH₃) predictions in the NOAA National Air Quality**
2 **Forecast Capability (NAQFC) using *in-situ* aircraft and satellite**
3 **measurements from the CalNex2010 campaign**
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34 **Abstract:**

35 Atmospheric ammonia (NH₃) is not only a major precursor gas for fine particulate matter
36 (PM_{2.5}), but it also negatively impacts the environment through eutrophication and acidification.
37 As the need for agriculture, the largest contributing source of NH₃, increases, NH₃ emissions will
38 also increase. Therefore, it is crucial to accurately predict ammonia concentrations. The objective
39 of this study is to determine how well the U.S. National Oceanic and Atmospheric
40 Administration (NOAA) National Air Quality Forecast Capability (NAQFC) system predicts
41 ammonia concentrations using their Community Multiscale Air Quality (CMAQ) model (v4.6).
42 Model predictions of atmospheric ammonia are compared against measurements taken during the
43 NOAA California Nexus (CalNex) field campaign that took place between May and July of
44 2010. Additionally, the model predictions were also compared against ammonia measurements
45 obtained from the Tropospheric Emission Spectrometer (TES) on the Aura satellite. The results
46 of this study showed that the CMAQ model tended to under predict concentrations of NH₃.
47 When comparing the CMAQ model with the CalNex measurements, the model under predicted
48 NH₃ by a factor of 2.4 (NMB = -58%). However, the ratio of the median measured NH₃
49 concentration to the median of the modeled NH₃ concentration was 0.8. When compared with
50 the TES measurements, the model under predicted concentrations of NH₃ by a factor of 4.5
51 (NMB = -77%), with a ratio of the median retrieved NH₃ concentration to the median of the
52 modeled NH₃ concentration of 3.1. Because the model was the least accurate over agricultural
53 regions, it is likely that the major source of error lies within the agricultural emissions in the
54 National Emissions Inventory. In addition to this, the lack of the use of bidirectional exchange of
55 NH₃ in the model could also contribute to the observed bias.

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1. Introduction and Background

Ammonia (NH_3) is an important gas in the atmosphere. Major sources of NH_3 include livestock, fertilizer, soil, biomass burning, industry, vehicles, the ocean, humans, and waste disposal/recycling activities, with agricultural emissions accounting for about 90% of NH_3 emissions into the atmosphere (Anderson *et al.*, 2003; Aneja *et al.*, 2009). As the world's population continues to increase, the fertilizer and agricultural (both crop and animal) industries will also increase, thus leading to increasing NH_3 emissions into the atmosphere (Heald *et al.*, 2012), which could cause a number of impacts to both human health and the environment.

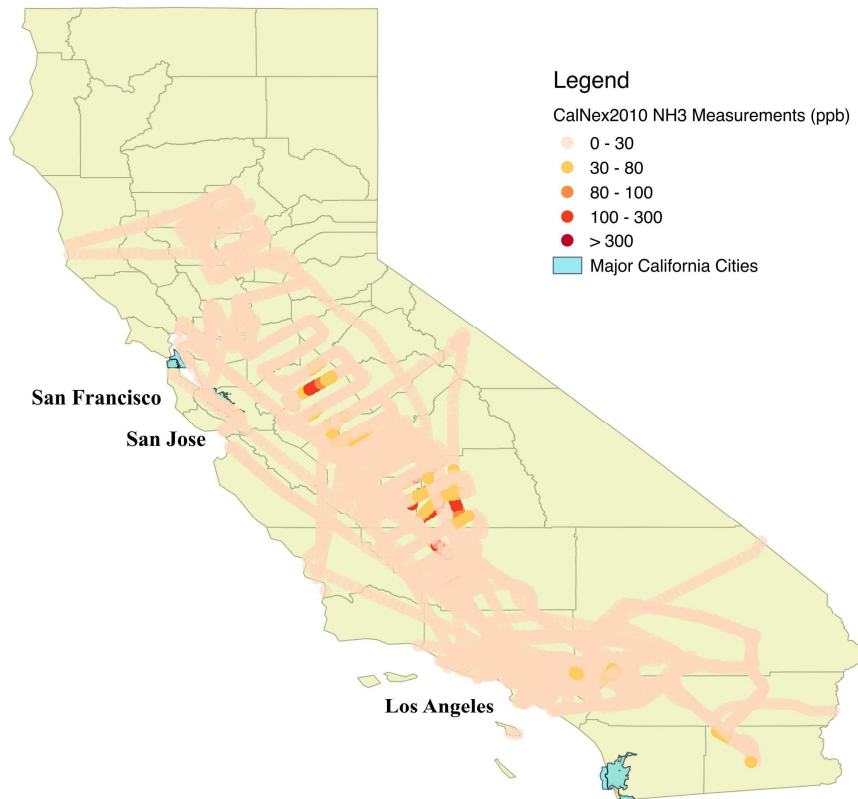
NH_3 reacts with sulfuric, nitric and hydrochloric acids to form ammonium sulfate, ammonium bisulfate, ammonium nitrate and ammonium chloride aerosols, all of which contribute to the formation of fine particulate matter ($\text{PM}_{2.5}$) (Robarge *et al.*, 2002; Baek and Aneja, 2004; Baek *et al.*, 2004; Renner and Wolke, 2010; Wang *et al.*, 2012; Kwok *et al.*, 2013). Exposure to elevated $\text{PM}_{2.5}$ concentrations is a major concern for human health and welfare due to the particles' ability to penetrate deep into the respiratory tract. There are many adverse health effects associated with elevated concentrations of fine particulate matter, such as cardiovascular and respiratory issues and even death (Anderson *et al.*, 2003; Pope *et al.*, 2009; Behera and Sharma, 2010a, b). Fine particulate matter is also associated with a number of environmental impacts, such as reducing visibility and changing the earth's radiational balance (Behera and Sharma, 2010a, b; Fan *et al.*, 2005; Heald *et al.*, 2012; Wang *et al.*, 2012).

In addition, NH_3 is also important in the environment due to its role in acid deposition and the nitrogen cycle, which is one of the most important nutrient cycles for living organisms. NH_3 and ammonium (NH_4^+) in the atmosphere are deposited to the surface via wet and dry deposition, thus increasing the amount of reduced nitrogen (Robarge *et al.*, 2002). This could lead to a number of negative impacts on the environment, such as soil acidification, eutrophication, as well as decreasing the resistance of vegetation to drought and frost damage (Robarge *et al.*, 2002). NH_3 in agricultural soil also plays a significant role in the formation of nitrous oxide (N_2O), a major greenhouse gas. The oxidation of NH_3 during the nitrification process can produce N_2O in a number of different pathways, such as through the denitrification process.

Due to the importance of atmospheric NH_3 , it is necessary that the air quality models are able to accurately predict concentrations of NH_3 . The purpose of this research is to determine how accurate the National Oceanic and Atmospheric Administration's (NOAA) National Air Quality Forecast Capability (NAQFC), which uses the Community Multiscale Air Quality (CMAQ) model (v4.6), predicts NH_3 and ammonium concentrations during the CalNex2010 (California Nexus) field campaign. During the CalNex2010 field campaign, *in-situ* measurements of pollutants were obtained via aircraft between May and July, 2010, across much of California (Ryerson *et al.*, 2013). Figure 1 shows the flight paths taken during the field campaign.

Model predictions of NH_3 and NH_4^+ are compared with measurements taken via aircraft as well as with satellite measurements obtained from the Tropospheric Emission Spectrometer (TES) aboard NASA's Aura satellite in order to determine the accuracy of NOAA's CMAQ model. Because agricultural emissions of NH_3 are difficult to quantify, there is much uncertainty in the emission inventory used by the CMAQ model. The objective of this research is to not only determine the accuracy of NOAA's CMAQ model, but also to

111 identify potential ways to improve the NH₃ emissions inventory used in the CMAQ model
112 for California.
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114 **Figure 1.** The flight paths taken during the CalNex field campaign, with the colored dots representing locations
115 where NH₃ measurements were made. The blue polygons represent major California cities for reference.
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118 2. Methodology

119 2.1. Air Quality Model

120 Version 4.6 of the CMAQ model, using the CB05-AERO5 chemical mechanism, was
121 used to predict the concentrations of NH₃ and NH₄⁺, at 12 km grid resolution, from May to
122 July, 2010. The meteorological predictions used within the CMAQ model were generated by
123 the North American Mesoscale Forecast System (NAM). The NH₃ and ammonium
124 emissions used in the model were obtained via the 2005 US Environmental Protection
125 Agency's (EPA) National Emissions Inventory (NEI). The emission data used in NOAA's
126 NAQFC system was based on the 2005 US Environmental Protection Agency's (EPA) 2005
127 National Emissions Inventory (NEI), with an update using the Cross-State Air Pollution
128 Rule (Pan *et al.*, 2015; Tong *et al.*, 2015; Canty *et al.*, 2015; Duncan *et al.*, 2015).
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132 2.2. Aircraft Measurements

133 Daytime measurements of atmospheric NH₃ and NH₄⁺ from NOAA's WP-3D aircraft
134 from May 4th to June 20th, 2010, taken over California are used here. Gaseous NH₃ was
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136 measured using chemical ionization mass spectrometry (CIMS) at 1 Hz (~100 m spatial
137 resolution) with typical inaccuracies $\pm 30\% \pm 0.2$ ppbv and a 1σ uncertainties of 0.08 ppbv
138 (Nowak *et al.*, 2010; Nowak *et al.*, 2012). NH_4^+ concentrations were measured using a
139 compact time-of-flight mass spectrometer with 2σ uncertainties of $\pm 34\% \pm 0.06 \mu\text{g m}^{-3}$
140 (Bahreini *et al.*, 2009; Bahreini *et al.*, 2012; Nowak *et al.*, 2012). NH_4^+ was measured every
141 10 seconds and then averaged over one minute, while NH_3 was measured every second and
142 then averaged over 1 minute (Nowak *et al.*, 2010; Nowak *et al.*, 2012). The one minute
143 averages were compared with the model estimates. Measurements of NH_4^+ were taken in μg
144 standard m^{-3} , where standard signifies that these measurements were taken at standard
145 temperature and pressure, while measurements of NH_3 were taken in $\mu\text{g m}^{-3}$. Therefore, for
146 this comparison, it was necessary to convert the $\mu\text{g standard m}^{-3}$ to $\mu\text{g m}^{-3}$ using
147 measurements of the ambient atmosphere that were taken during the flight. Meteorological
148 parameters including temperature, dew point temperature, potential temperature, relative
149 humidity, wind speed and wind direction were measured (Ryerson *et al.*, 2013). In addition
150 to this, the aircraft's navigation system and global positioning system (GPS) measured the
151 location, altitude, speed, bearing and the angle of descent were recorded. The CMAQ model
152 predictions of NH_3 and NH_4^+ (in $\mu\text{g m}^{-3}$) were used for comparison against each 1-minute
153 aircraft measurement. The CMAQ prediction for each measurement location and time was
154 computed by 4-dimensional interpolation across space and time, using the model grid
155 centroids surrounding the measurement point for the hours before and after the
156 measurement.

157 158 2.3. Satellite Measurements

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160 Predictions from the CMAQ model were also compared against satellite NH_3
161 concentrations retrieved from infrared spectra gathered by the Tropospheric Emission
162 Spectrometer (TES) on the National Aeronautics and Space Administration's (NASA) Aura
163 satellite. TES is a high spectral resolution infrared Fourier Transform spectrometer (FTS)
164 (Beer, 2006) that covers the spectral range 650–3050 cm^{-1} (Bowman *et al.*, 2006). TES has a
165 spatial resolution of 5.3 x 8.5 km nadir and 37 x 23 km limb and has a spectral resolution of
166 0.5 x 5 km nadir and 2.3 x 23 km limb (Beer *et al.*, 2001; Zhu *et al.*, 2013). TES measures
167 the Earth's infrared light energy and follows a sun-synchronous orbit, making observations
168 on a 16-day cycle, with roughly 1 pass during the day and 1 pass during the night over each
169 region every other day (Clarisse *et al.*, 2010; Zhu *et al.*, 2013). Atmospheric ammonia
170 concentrations are derived from TES by observing changes in the infrared radiation intensity
171 between 940 cm^{-1} and 970 cm^{-1} . The TES ammonia retrievals use a forward radiative transfer
172 model (RTM) to compute the expected intensity of radiation at the top of the atmosphere for
173 an estimated ammonia concentration. The assumed concentration of ammonia is varied to
174 minimize the error between the spectrum predicted by the RTM and the spectrum actually
175 measured by the satellite. This results in an estimate of the vertical profile of the ammonia
176 concentration for the region sensed by the satellite. (Bowman *et al.*, 2006; Shephard *et al.*,
177 2011; Shephard and Cady-Pereira, 2015). Finally, the meteorological conditions
178 (temperature, relative humidity, etc.) are used with the a priori NH_3 profile to estimate the
179 atmospheric NH_3 concentration (Herman and Osterman, 2014).

180 TES performed 6 transect measurements over the CalNex study domain between May 7,
181 2010, and June 3, 2010. In order to compare with the CMAQ model predictions, the NH_3

concentration for the aircraft sampling height was extracted from the total column data based on the assumed a priori profile. This study used only measurements that met TES Species Retrieval Quality criteria and for which the degrees of freedom for signal (DOFS) exceeded 0.5. It is important to note that there are some uncertainties associated with this data. For example, the satellite retrieval of ammonia concentration may be biased toward the a priori assumption. In addition to this, the estimated vertical distribution of ammonia is also impacted by the a priori assumption made.

2.4. Model to Measurement Comparisons

Similar to the work of Battye *et al.* (2016), the normalized mean bias (NMB) was calculated using the following equation:

$$NMB = \frac{1}{N} \frac{\sum_{i=1}^N [C_{mod}(i) - C_{obs}(i)]}{\sum_{i=1}^N C_{obs}(i)},$$

in order to determine the accuracy of the NAQFC CMAQ model. The ratio of the average measured concentration to the average model prediction ($R_{o/m}$) was calculated using:

$$R_{o/m} = \frac{\sum_{i=1}^N C_{obs}}{\sum_{i=1}^N C_{mod}},$$

where $C_{mod}(i)$ is the model prediction, $C_{obs}(i)$ is the observed concentration at a given location and time, and N is the number of observations. The relationship between NMB and $R_{o/m}$ is as follows:

$$NMB = \frac{1}{R_{o/m}} - 1.$$

3. Results and Discussion

3.1. Aircraft Measurements Compared with Model Predictions

Table 1 compares the aircraft measurements of NH_3 and NH_4^+ taken during the field campaign with the model predictions predicted by NOAA's CMAQ model and the calculated NH_x (NH_3 (g) + NH_4^+ (p)) concentrations. The average concentration of the 1-minute averaged NH_3 observations in the CalNex field campaign was 4.1 ± 14.8 ppbv ($2.7 \pm 9.9 \mu\text{g m}^{-3}$), with a maximum 1-minute average concentration of 380.1 ppbv ($254.7 \mu\text{g m}^{-3}$) and a maximum 1-second concentration of 963 ppbv ($669 \mu\text{g m}^{-3}$). In contrast to this, the model predicted an average NH_3 concentration of 1.7 ± 2.4 ppbv ($1.1 \pm 1.4 \mu\text{g m}^{-3}$), with the maximum predicted NH_3 concentration at 17.3 ppbv ($11.3 \mu\text{g m}^{-3}$). Thus, the measured concentration of NH_3 was a factor of 2.4 higher than what was modeled by the CMAQ forecasting model, with a normalized mean bias of -58%.

Table 1. Comparison of in situ aircraft measurements with model predictions for NH₃, NH₄⁺, and NH_x.

	NH ₃ (ppbv)	NH ₃ (μg m ⁻³)	NH ₄ ⁺ (μg m ⁻³)	NH _x (μg m ⁻³)
Measured concentrations				
Average	4.1	2.7	0.4	3.1*
Standard deviation	14.8	9.9	0.7	10.6*
Maximum	380.1	254.7	6.7	254*
Median	0.8	0.5	0.2	1.0
Model predictions				
Average	1.7	1.1	0.6	1.7*
Standard deviation	2.4	1.4	0.8	2.2*
Maximum	17.3	11.3	7.3	11.5*
Median	0.9	0.6	0.3	1.2*
Comparison statistics				
Normalized mean bias		-58%	43%	-44%
Ratio of average measured value to average modeled value		2.4	0.7	1.8
Ratio of median measured value to median modeled value		0.8	0.7	0.8
Correlation coefficient (r)		0.34	0.62	0.30
Coefficient of determination (r ²)		0.12	0.38	0.09
Number of observations		8,181	4,605	4,605

* Denotes calculated value, not measured

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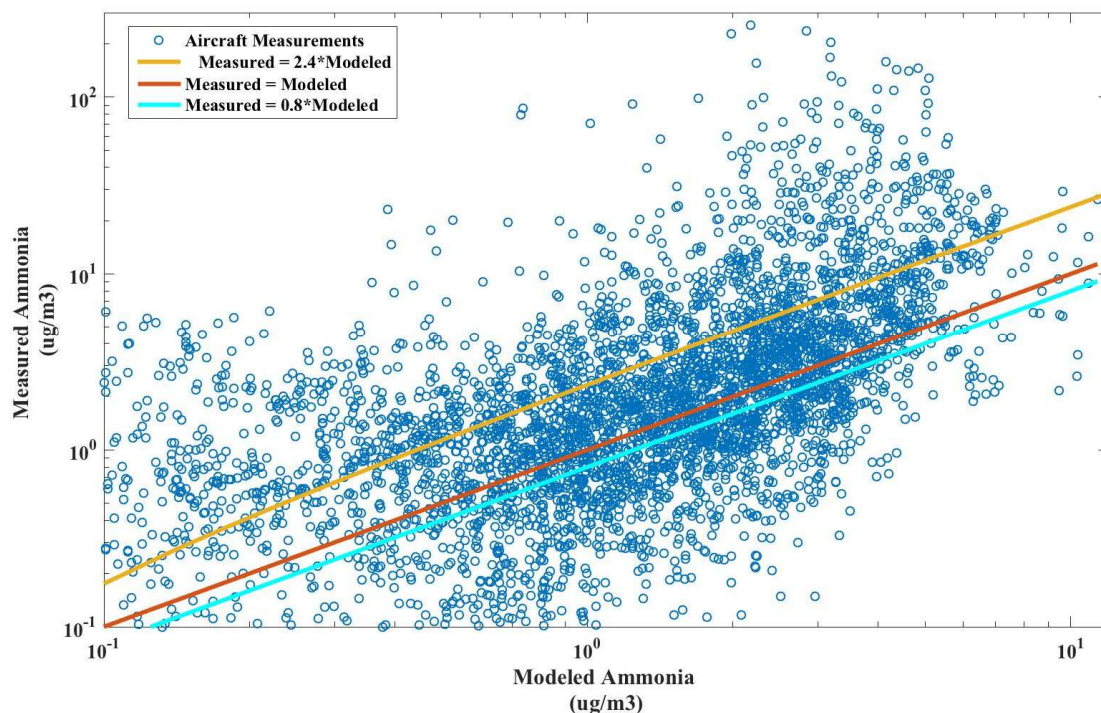
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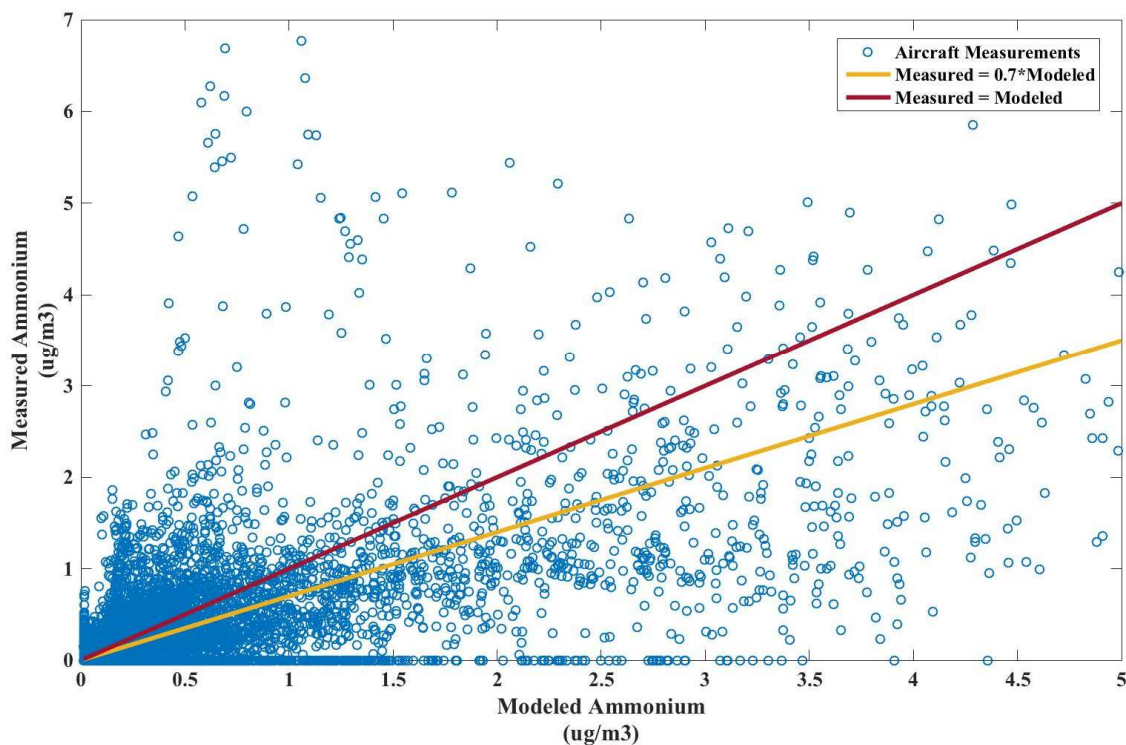
In order to take out the influence of the outlier data points, the median was also calculated for the modeled and measured NH₃ concentration. The median NH₃ concentration measured in the field campaign was 0.8 ppbv (0.5 μg m⁻³), while the median modeled concentration was 0.9 ppbv (0.6 μg m⁻³). The ratio of the measured median concentration and the modeled median concentration of ammonia is 0.8, which suggests that the model is fairly accurate without the influence of the elevated outlier NH₃ concentrations observed in the field campaign.

Figure 2 shows the measured concentrations of NH₃ (y-axis) compared with the model predictions of NH₃ (x-axis) on a log-log scale plot, with the plotted measurements represented as the blue dots, the gold line representing the actual measured trend line, the cyan-green line representing the bias line given by the ratio of the medians and the red line representing where the measured points would have fallen if the model correctly predicted the measurements. The log-log plot was chosen for this figure due to the large range observed in measured ammonia concentrations. When comparing the actual trend line with the modeled one-to-one line (i.e. when the measured = modeled), it appears that the model under predicted concentrations of NH₃ by a factor of 2.4. However, when comparing the one-to-one line with the bias line given by the median ratio, it appears that the model only under predicted NH₃ concentrations by a factor of 0.8.



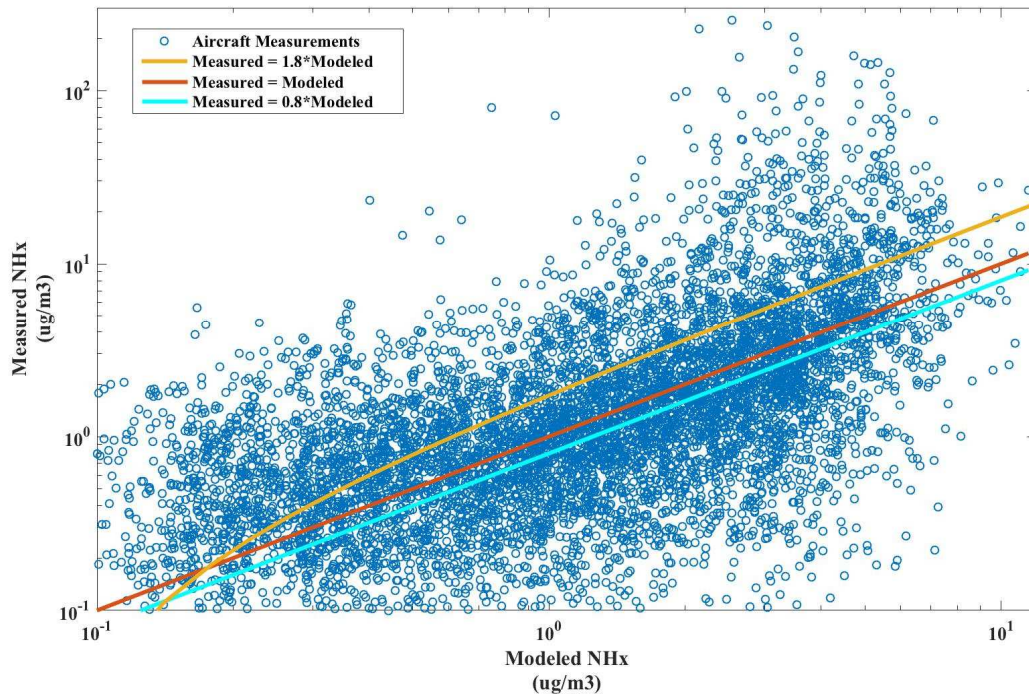
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 250 **Figure 2.** Aircraft *in-situ* measurements of NH₃ (blue dots) plotted against model predictions on a log-log scale plot.
 251 The red line shows where the measured points should have fallen if the model predictions were exactly correct and
 252 the gold line shows the actual measured trend line. The actual trend line (gold line) is plotted above the one-to-one
 253 line (red line), while the bias line given by the median ratio is given by the cyan-green line.
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255 The average NH₄⁺ concentration for the CalNex study area was found to be 0.4 ± 0.7
 256 $\mu\text{g}/\text{m}^3$, with a maximum concentration of $6.7 \mu\text{g m}^{-3}$. In comparison, the CMAQ model
 257 predicted an average NH₄⁺ concentration of $0.6 \pm 0.8 \mu\text{g m}^{-3}$, with a maximum predicted
 258 concentration of $7.3 \mu\text{g m}^{-3}$. The concentration of NH₄⁺ for the study area was found to be a
 259 factor of 0.7 lower than the prediction made by the CMAQ model and the calculated
 260 normalized mean bias was found to be 43%. When comparing the median values of the
 261 measured and modeled data, the measured NH₄⁺ median was $0.2 \mu\text{g m}^{-3}$ and the modeled
 262 NH₄⁺ median was $0.3 \mu\text{g m}^{-3}$. This corresponds to a ratio of the measured to modeled
 263 median concentration of 0.7, which is equivalent to the ratio of the modeled to measured
 264 concentration. Figure 3 shows the comparison of the measured concentration of NH₄⁺
 265 compared with the modeled concentrations, plotted with the observed trend line (gold) and
 266 the one-to-one line (red). It is necessary to note that the bias line given by the median ratio is
 267 not shown because it is equivalent to the observed trend line. Because both the modeled and
 268 measured ammonium concentrations were less than $10 \mu\text{g m}^{-3}$, this figure was plotted on a
 269 linear scale as oppose to a log-log scale. Unlike the comparison with NH₃ concentrations,
 270 the modeled NH₄⁺ concentrations were fairly close to the measurements made, with the
 271 model slightly over predicting, particularly at higher concentrations of NH₄⁺. This suggests
 272 that the model had a fairly good handle on the conversion of gaseous NH₃ to particulate
 273 ammonium and thus the conversion was likely not limited by the concentration of gaseous
 274 NH₃.
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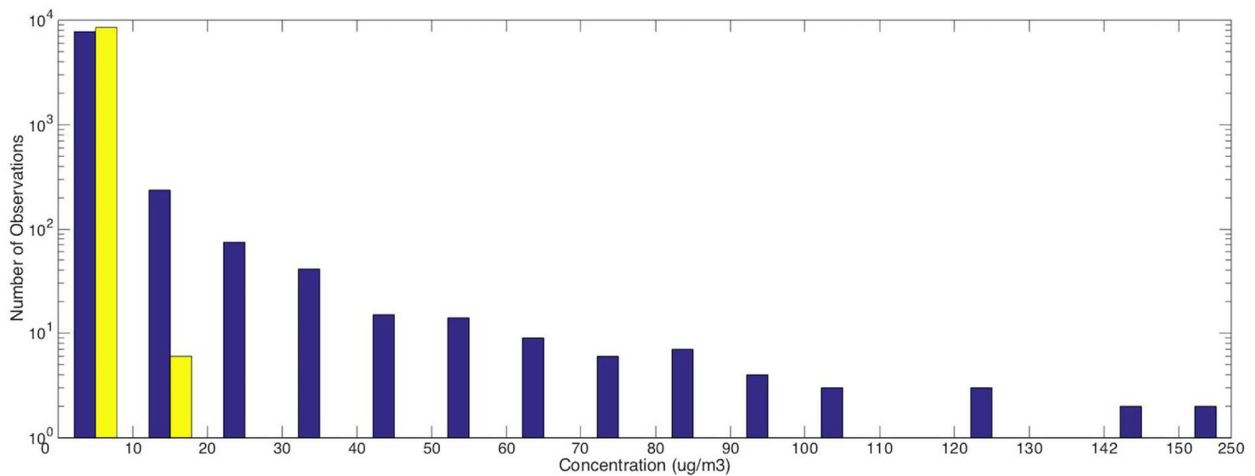
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 277 **Figure 3.** Aircraft *in-situ* measurements of NH_4^+ (blue dots) plotted against model predictions on a linear scale plot.
 278 The red line shows where the measured points should have fallen if the model predictions were exactly correct, the
 279 gold line shows the actual measured trend line. In this case, the bias line is equivalent to the actual trend line and is
 280 therefore not plotted in Figure 3.

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 282 The average concentration of NH_x measured during the field campaign was $3.1 \pm 10.6 \mu\text{g m}^{-3}$, with a maximum concentration of $254 \mu\text{g m}^{-3}$. In comparison to this, the CMAQ model
 283 predicted an average concentration of $1.7 \pm 2.2 \mu\text{g m}^{-3}$. The maximum predicted NH_x
 284 concentration was $11.5 \mu\text{g m}^{-3}$. The measured concentration of NH_x was found to be a factor
 285 of 1.9 higher at very low concentrations ($<10^{-0.5}$) of NH_x than what was predicted by the
 286 CMAQ model and a factor of 1.9 lower than what was predicted by the CMAQ model at
 287 higher concentrations. The average measured to modeled ratio of 1.9 corresponds to a
 288 normalized mean bias of -44%. However, comparing the medians of the measured and
 289 modeled NH_x concentration, the medians were found to be $1 \mu\text{g m}^{-3}$ and $1.2 \mu\text{g m}^{-3}$,
 290 respectively. This corresponds to a median ratio of 0.8, which is lower than the ratio that
 291 was observed when comparing the average modeled NH_x concentrations with the average
 292 measured NH_x concentrations. Figure 4 shows the comparison of the measured concentration
 293 of NH_4^+ compared with the modeled concentrations on a log-log scale plot. Similar to Figure
 294 2, this was also plotted on a log-log due to the large range in measured NH_x concentrations.
 295 As described above, this figure shows the model tends to under predict concentrations of
 296 NH_x at higher concentrations of NH_x and the model tends to over predict concentrations of
 297 NH_x at lower concentrations. However, when comparing the bias line of the median ratio
 298 (cyan-green line) with the one-to-one line (red line) it appears that the measured values are
 299 fairly close to the modeled projections.
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Figure 4. Calculations based off of aircraft *in-situ* measurements (referred to as measured), of NH_x (blue dots) plotted against calculated model predictions (referred to as modeled) on a log-log scale plot. The red line shows where the measured points should have fallen if the model predictions were exactly correct, the yellow line shows the actual measured trend line, and the cyan-green line shows the bias line given by the median ratio.



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Figure 5. Histogram comparing the modeled (yellow) versus the measured (purple-blue) NH_3 concentration with respect to the number of observations. Both the modeled and the measured concentration of NH_3 occurred most frequently at lower concentrations. However, the range of observations at different concentrations is much larger for measured values. This suggests that there was a greater variation in the concentration measured as opposed to the modeled concentration, which occurred primarily at concentrations less than $30 \mu\text{g m}^{-3}$.

Figure 5 shows a histogram comparing the modeled and measured NH_3 concentrations. The extreme values measured during the field campaign are significantly higher than the

317 extreme values predicted by the model. While both the modeled and measured
318 concentrations show a similar negative pattern, where the number of observations is highest
319 at the lowest concentration and then rapidly decreases thereafter, the slope of the decrease is
320 significantly different. The number of observations from the field campaign gradually
321 decreases with increasing ammonia concentrations while the number of modeled
322 observations drop exponentially with increasing ammonia concentrations, such that there are
323 no observations above $30 \mu\text{g m}^{-3}$. The 98th percentile of the measured NH_3 values was found
324 to be $23.2 \mu\text{g m}^{-3}$ while the 98th percentile of the modeled NH_3 values was $5.1 \mu\text{g m}^{-3}$,
325 showing the vast under estimation of NH_3 concentrations by the model.

326 Spatial patterns in the model prediction error were identified by comparing the model
327 bias (model concentration – measured concentration) in relation to NH_3 point sources and
328 agricultural sources. Figure 6 plots both the agricultural and point source emissions
329 (obtained from the US EPA’s National Emissions Inventory) with relation to the model bias.
330 Figure 6A shows all the calculated model bias for the period while Figure 6B only shows
331 model bias over 50 ppbv.

332 The majority of the high model biases occur over large agricultural regions and the
333 highest model bias occurs over point sources. This suggests that the 2005 NEI under predicts
334 NH_3 concentrations within this region of California. When comparing the 2005 NEI NH_3
335 emissions with the 2011 NH_3 NEI emissions for the study domain, it is found that there is a
336 25% increase in the NH_3 emissions, which would certainly account for some of the observed
337 bias. However, the 2014 NEI NH_3 emissions, which uses an entirely different methodology
338 for their agricultural emissions, are 62% higher than the 2011 NEI NH_3 emissions and 85%
339 higher than the 2005 NEI NH_3 emissions (EPA, 2005; EPA, 2011; EPA, 2014). According
340 to the US EPA 2014 National Emissions Inventory (version 1) Technical Supporting
341 Document, this version of the NEI has updated the agricultural livestock ammonia
342 methodology in order to incorporate both new observational data as well as new process
343 based methods. In addition to this, the methodologies used to develop emissions from
344 fertilizer application have been entirely changed. For example, in this inventory, ammonia
345 emissions from agricultural soils are estimated using the bidirectional version of the CMAQ
346 model (v5.0.2) coupled with the Fertilizer Emissions Scenario Tool for CMAQ FES-
347 C(v1.2) (EPA, 2016). Based on this, it is likely that the agricultural emissions used in the
348 model contributed to much of the biases observed in this study.

349 Another potential cause for the discrepancies between the model and the measured values
350 could be the NEI’s handle on the diurnal and seasonal representation of ammonia emissions
351 in this region. However, it is important to note that the diurnal representation of ammonia
352 emissions has been updated within the Sparse Matrix Operator Kernel Emissions (SMOKE)
353 model and this has been released in the newer (2011, 2014) versions of the NEI (Zhu et al.,
354 2015).

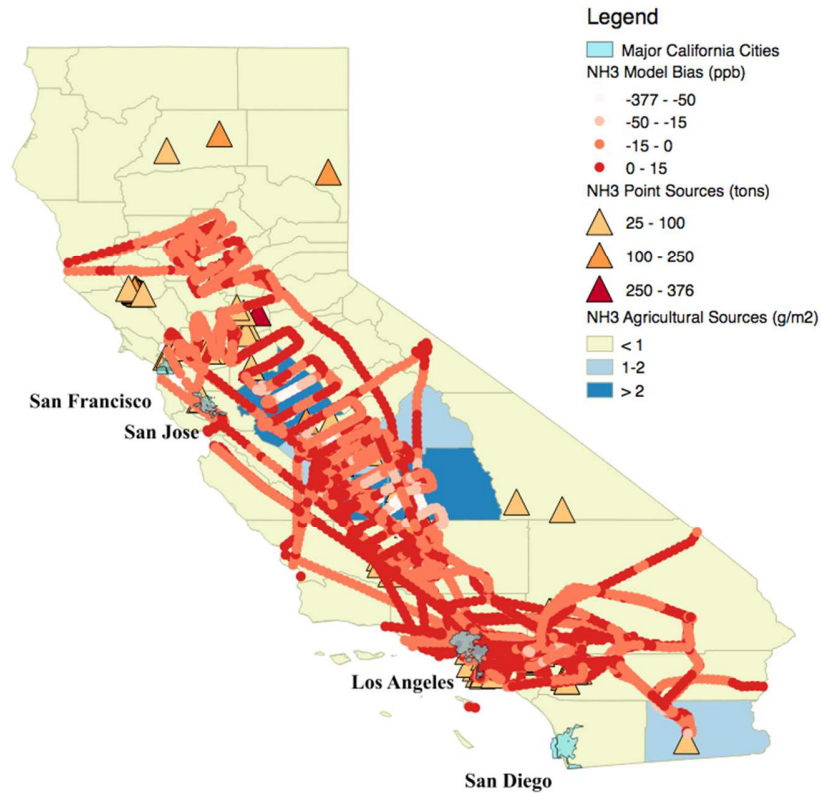
355 356 *3.2. Model Predictions Compared with Satellite Retrievals*

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358 The CMAQ model prediction of NH_3 was compared with the TES satellite retrieval of
359 concentrations measured at the CalNex aircraft measurement heights (Table 2, Figure 7a).
360 The average TES NH_3 concentration was $14.8 \pm 11.8 \mu\text{g m}^{-3}$, with a maximum retrieved
361 concentration of $40.5 \mu\text{g m}^{-3}$ and a median concentration of $10.4 \mu\text{g m}^{-3}$, while the
362 associated average CMAQ model concentration was found to be $3.3 \pm 1.0 \mu\text{g m}^{-3}$, with a
363 maximum predicted concentration of $4.9 \mu\text{g m}^{-3}$ and a median concentration of $3.4 \mu\text{g m}^{-3}$.

364 As Figure 7a shows, the majority of the NH₃ retrieval measurements fall above the one to
365 one line (where the modeled NH₃ = the TES NH₃ retrieval), which suggests that the TES
366 retrieval measurements are higher than the concentrations predicted by the model. This
367 normalized mean bias of the TES retrieval was found to be -77%, which corresponds to a
368 ratio of the average measured NH₃ value to the average NH₃ TES retrieval of 4.5. Similarly,
369 the ratio of the median measured value to median modeled value was 3.1. The average total
370 column loading (mg m⁻²) measured by TES was 7.2 ± 6.7 mg m⁻², with a maximum total
371 column loading of 40 mg m⁻² and a median value of 5.1 mg m⁻². In contrast to this, the
372 average total column loading predicted by the CMAQ model was 0.002 ± 0.002 mg m⁻²,
373 with a maximum total column loading of 0.008 mg m⁻² and a median of 0.002 mg m⁻² (Table
374 2, Figure 7b). This corresponds with a NMB of -99% and a ratio of the average measured
375 NH₃ value to the average NH₃ TES retrieval 3600. The ratio of the median measured value
376 to median modeled value was 2550. Because the satellite samples a larger volume of air than
377 the CIMS, it is expected that the retrieved range of concentrations of ammonia would be
378 narrower than those observed by the aircraft due to the fact that there is much more
379 variability on a smaller spatial scale. In addition to this, it would follow that the satellite
380 retrievals would also be lower than those observed by the aircraft (assuming the aircraft was
381 targeting emission sources), due to the fact that the concentration of atmospheric ammonia
382 drops exponentially with increasing distance from the source, thus returning values that are
383 similar to the volume modeled by CMAQ. While the authors tried to pair the
384 CMAQ/aircraft measurement data with the TES retrievals such that they were as close as
385 possible, a potential reason for the discrepancies between the TES NH₃ retrieval and the
386 CMAQ model estimates, could be due to the fact that the measurements did not align 100%
387 in time and space.

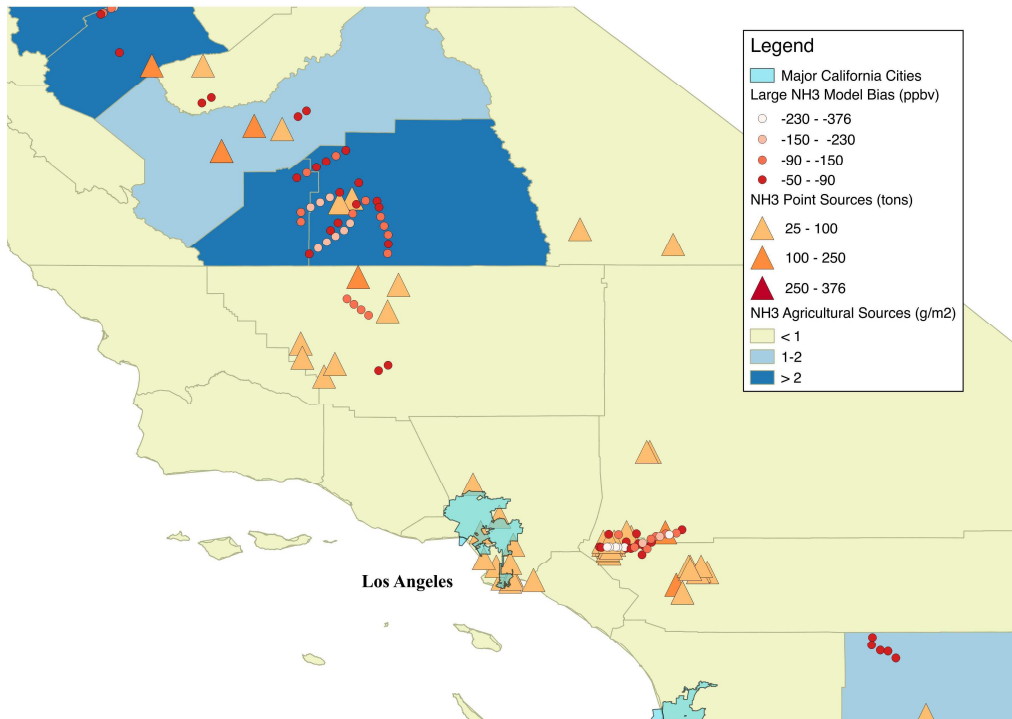
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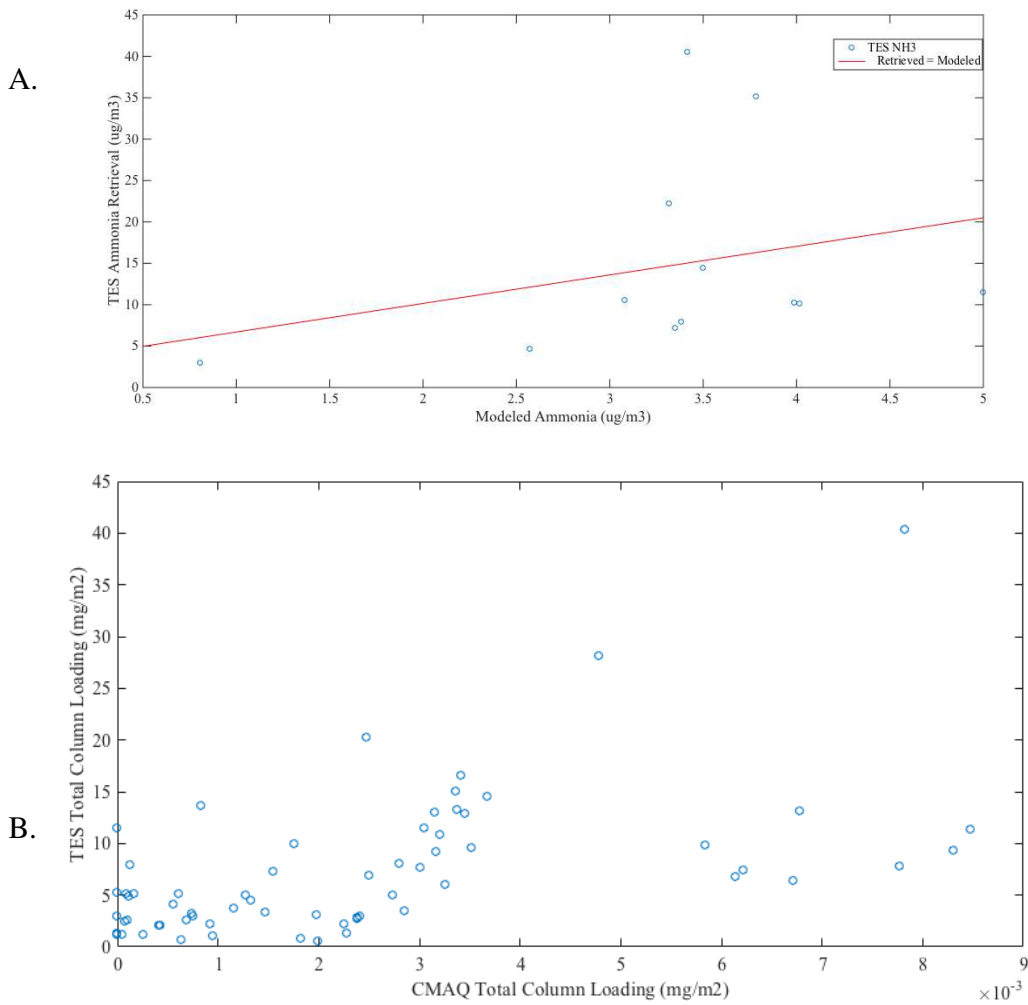
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Figure 6. The agricultural and point source emissions (obtained from the US EPA's National Emissions Inventory) plotted with relation to the model bias. Figure 6A shows all the model biases obtained from the CalNex study while Figure 6B shows only the largest model bias. Note that major under estimates occurred in close proximity to both NH₃ point sources as well as agricultural sources of NH₃.

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Figure 7. The CMAQ model prediction of NH₃ compared with the TES satellite retrieval at the aircraft measurement level plotted on a log-log scale plot (A) and the model the total column loading compared with the TES total column loading retrieval (B). The red line in Figure 7A shows where the TES retrieval points should have fallen if the model predictions were exactly correct. Looking at the order of magnitude, it is evident that the modeled ammonia concentrations were much lower than the retrieved ammonia concentrations at the aircraft height. This was also found when comparing the modeled total column loading of ammonia with the total column loading of ammonia retrieved by TES.

431 **Table 2.** Comparison of CMAQ model predictions at the aircraft level and the total column loading with the
 432 corresponding TES retrievals for NH₃.

	NH ₃ At aircraft level ($\mu\text{g m}^{-3}$)	NH ₃ Total column loading (mg m^{-2})
TES retrievals		
Average	14.8	7.2
Standard deviation	11.8	6.7
Maximum	40.5	40.4
Median	10.4	5.1
Model predictions		
Average	3.3	0.002
Standard deviation	1.0	0.002
Maximum	4.9	0.008
Median	3.4	0.002
Comparison statistics		
Normalized mean bias of TES retrieval	-77%	-99%
Ratio of average TES retrieval value to average modeled	4.5	3600
Ratio of median measured value to median modeled value	3.1	2550
Correlation coefficient (r)	0.28	0.54
Coefficient of determination (r ²)	0.08	0.30
Number of observations	12	66

433
 434 **3.3. Satellite Retrievals Compared with In-Situ Aircraft Measurements**
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436 In this study, only 12 satellite retrievals with adequate data quality lined up with the
 437 aircraft measurements taken during the CalNex study (Figure 8), therefore limiting the
 438 results. Table 3 and Figure 9 compare the *in-situ* aircraft measurements with the satellite
 439 observations obtained from TES. The average corresponding NH₃ concentration measured
 440 from the CalNex campaign was found to be $33.1 \pm 35.5 \mu\text{g m}^{-3}$, with a maximum NH₃
 441 concentration of $108.8 \mu\text{g m}^{-3}$ and a median value of $10.2 \mu\text{g m}^{-3}$. In comparison to this, the
 442 average TES NH₃ concentration was $14.8 \pm 11.8 \mu\text{g m}^{-3}$, with a maximum measured
 443 concentration of $40.5 \mu\text{g m}^{-3}$. However, the median value observed for the TES NH₃
 444 concentration was $10.4 \mu\text{g m}^{-3}$, which is similar to that of the NH₃ aircraft measurements
 445 median. Thus, the ratio of the median measured value to the median modeled value is 0.98.



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447 **Figure 8.** The 12 locations where the CalNex *in-situ* measurements could be compared with the TES satellite
 448 retrieval of NH₃. The pink colored triangles represent the locations where the *in-situ* and satellite measurements
 449 were compared while the blue polygons represent the major cities in California.

450

451 As Figure 9 shows, the majority of the NH₃ measurements fall above the one to one line.
 452 This normalized mean bias of the TES retrieval was found to be -55%, which corresponds to
 453 a ratio of the average measured NH₃ value to the average NH₃ TES retrieval of 2.2. The
 454 median of the *in-situ* aircraft measurements was 10.2 μg m⁻³, while the median of the
 455 CMAQ measurements was 10.4 μg m⁻³, which corresponds with a median ratio of 0.98. As
 456 mentioned in the previous section, a source of error when comparing the TES retrieval with
 457 the CMAQ data is the fact that the two data sources were not entirely aligned in both space
 458 and time.

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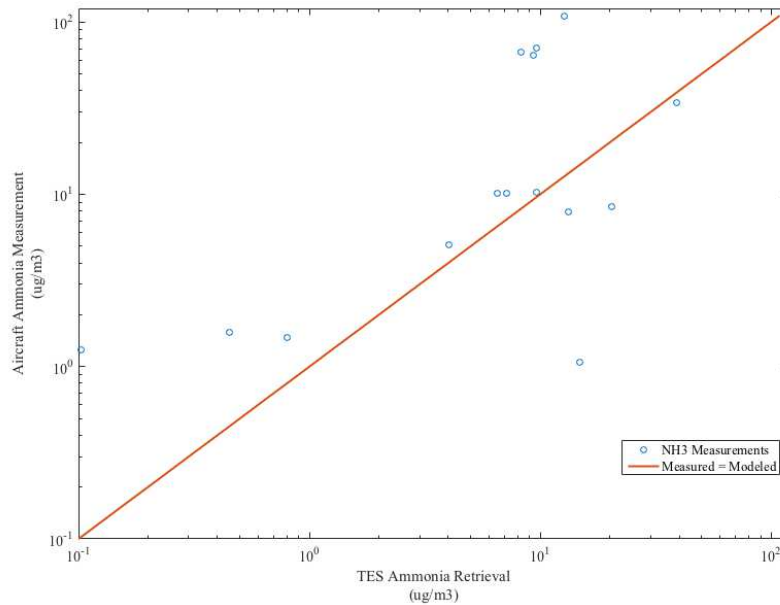
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Table 3. Comparison of *in-situ* aircraft measurements with TES retrievals for NH₃.

	NH ₃ ($\mu\text{g m}^{-3}$)
<i>In situ</i> aircraft measurements	
Average	33.1
Standard deviation	35.5
Maximum	108.8
Median	10.2
TES retrievals	
Average	14.8
Standard deviation	11.8
Maximum	40.5
Median	10.4
Comparison statistics	
Normalized mean bias of TES retrieval	-55%
Ratio of average measured value to average TES retrieval	2.2
Ratio of median measured value to median modeled value	0.98
Correlation coefficient (r)	0.25
Coefficient of determination (r ²)	0.06
Number of observations	12

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Figure 9. Compares the *in-situ* aircraft measurements with the satellite observations obtained from TES on a log-log scale plot. The solid red line shows where the measured points should have fallen if the model predictions were exactly correct. The TES NH₃ retrievals were much closer to the observed aircraft measurements when compared with the model output.

3.4. Analysis of Model Bias in Relation to Previous Studies and the NH₃ Emissions Inventory

Several studies have been conducted to determine how well the CMAQ model can predict NH₃ concentration. In general, it has been found that the CMAQ model has a tendency to under predict NH₃ concentrations, particularly over large source regions (Table 4) (Gilliland *et al.*, 2006; Kelly *et al.*, 2014; Butler *et al.*, 2015; Schifer *et al.*, 2016; Battye *et al.*, 2016). Gilliland *et al.* (2006) used an inverse modeling technique with CMAQ v4.4 to predict NH₃ emissions for the continental United States (CONUS). The results of this study indicated that the emissions inventory is too high for the winter months and too low for the summer months. Similar results were found by Butler *et al.* (2015), who used CMAQ v4.7.1 to predict NH₃ concentrations in Susquehanna River Watershed of New York and Pennsylvania. When comparing ambient concentration measurements of NH₃ with the model predictions, it was found that the model under estimated concentration by 8-60%. In addition to this, it was also found that the NH₃ under estimations were particularly high over the agricultural regions. Kelly *et al.* (2014) found similar results when comparing NH₃ measurements obtained from the California Research at the Nexus of Air Quality and Climate Change (CalNex) field campaign that occurred May-June, 2010, with model predictions from CMAQ v5.0.2. In addition to this, it was also found that the CMAQ model also predicted lower concentrations of NH₃ in some urban regions as well. Battye *et al.* (2016) found comparable results to Kelly *et al.* (2014) when comparing NH₃ measurements from the Deriving Information on Surface conditions from Column and Vertically Resolved Observations Relevant to Air Quality (DISCOVER AQ) field campaign (July-August, 2014) with NOAA's NAQFC CMAQ model (v5.0.2) over the agricultural regions of northeastern Colorado. Schifer *et al.* (2016) used GEOS-Chem (v9-02, driven by GEOS-5 assimilated meteorology) to simulate concentrations of atmospheric ammonia across the United States from 2008 to 2012 and found that the model tended to under predict ammonia concentrations near large source regions, under predicting concentrations by 26% when compared with surface sites. The current study found similar results to Gilliland *et al.* (2006), Kelly *et al.* (2014), Butler *et al.* (2015), Schifer *et al.* (2016) and Battye *et al.* (2016), where the CMAQ model (v4.6) under estimates NH₃ concentration, with the results being most comparable to Kelly *et al.* (2014) and Battye *et al.* (2016).

Meteorological factors can have a major impact on the emission and removal processes of ammonia. Both wind speed and relative humidity have been found to be inversely proportional to ammonia concentrations (Flechard and Fowler, 1998; Kapoor *et al.*, 1992; Parmar *et al.*, 2001; Phillips *et al.*, 2004; Sharma *et al.*, 2010; Sharma *et al.*, 2011), while temperature was primarily found to be directly proportional to ammonia concentrations (Flechard and Fowler, 1998). Ammonia concentrations tend to be lower during clear skies and when precipitation occurred (Kapoor *et al.*, 1992). Schifer *et al.* (2016) also notes the importance of meteorology in atmospheric air quality. They found that meteorology contributed to 64% of the changes in the surface concentration of ammonia when compared with reductions air pollution. Therefore, errors in the meteorology used within the CMAQ model could have contributed to the large under estimations observed in this study.

Another potential source of error in NH₃ predictions is the absence of the bidirectional flux model within the CMAQ version used. Cooter *et al.* (2012) and Bash *et al.* (2013) found that the use of the bidirectional flux in the model increases NH₃ concentrations on average by about 10% over the continental US. It is important to note that the flux will likely be higher over agricultural regions due to the abundance of NH₃ in the agricultural

523 cropping system. Error in the model processes used to handle NH₃ emissions may also
524 contribute to the observed bias. For example, the 12 km spatial resolution may lead to the
525 model overlooking high concentrations that are smaller than the grid size. However, the
526 similar results found by Kelly *et al.* (2014), who used an updated version (5.0.1) of the
527 CMAQ model with a 4km vertical resolution and 34 vertical layers, suggest that this is not
528 entirely the cause of the under estimation. Another likely source of error is the use of older
529 National Emissions Inventories (NEI). For example, Gilliland *et al.* (2006) used the 2001
530 NEI while Battye *et al.* (2016) and this current study used the 2005 NEI. Recall from the
531 above discussion that the NH₃ emissions increased 25% between the 2005 and 2011 (EPA,
532 2005; EPA, 2011). In contrast to this, NH₃ emissions increased 85% from the 2005 NEI to
533 the 2014 NEI after major changes were made to the methodologies used to calculate the
534 agricultural emissions of ammonia. In addition, it is possible that an increase in animal
535 activity could have contributed to the increase in NH₃ emission. Because the majority of the
536 under estimations, both in this study and the literature, tend to occur over agricultural
537 regions, it is likely that a major contributor to this under estimation of NH₃ concentrations is
538 due to an under estimation of NH₃ emissions from agricultural sources in the US National
539 Emissions Inventories used in most research (i.e. inventories that are older than the current
540 2014 NEI).

541 542 543 **4. Conclusion**

544
545 The NOAA NAQFC CMAQ model under predicted NH₃ concentrations in California
546 measured during the CalNex2010 field campaign by a factor of 2.4 (NMB = -58%), with a
547 median ratio of 0.8. Similarly, the NOAA NAQFC CMAQ model under predicted NH₃
548 concentrations in California by a factor of 4.5 (NMB = -77%), with a median ratio of 3.1,
549 when compared with measurements obtained from TES. In contrast to this, the CMAQ
550 model had a fairly good handle on NH₄⁺ concentrations, only over predicting by a factor of
551 0.7 (NMB = 43%), with a median ratio of 0.7. When comparing the median values for NH₃
552 and NH₄⁺, it was found that the median values were fairly similar for both modeled and
553 measured values. Despite the model under estimating NH₃ concentrations, particularly at
554 high concentrations the NH₄⁺ projections were fairly accurate, which suggests that the issue
555 lies within the prediction of gaseous NH₃. These results indicate that while the NOAA
556 CMAQ model represents the partitioning of NH₃, there is still uncertainty in predicting
557 concentrations of gaseous NH₃. This also suggests that the NH₃ levels in California exceed
558 the levels of the acidic species necessary for the gas-to-particle conversion. Therefore, this
559 will have major implications for PM_{2.5} reduction strategies. In addition to this, it is important
560 to note that a portion of the error in comparing the TES retrieval measurements with the
561 CMAQ model prediction is likely due to the fact that the retrieval and the predicted
562 concentration did not line up completely in space and time.

563 Recall that there are at least four potential sources of error within the CMAQ model: the
564 lack of the inclusion of the bidirectional flux model, potential errors in the model processes
565 used, errors within the NH₃ emissions used in the model and errors in the meteorology used
566 within the models. The addition of bidirectional flux to the model would increase NH₃
567 emissions, particularly over the agricultural regions, and therefore likely contributes to the
568 model bias observed in this study. Similarly, ammonia emissions in the 2014 NEI increased

569 85% from the emissions calculated by the 2005 NEI, primarily due to the update in the
570 methodologies used to calculate agricultural emissions of ammonia. Because the majority of
571 the highest model bias occur in areas of agriculture, it is likely that a major part of the
572 problem lies within the agricultural emissions of NH₃ in the 2005 NEI. Other potential
573 sources of the observed bias include the changes made to the diurnal and temporal
574 representation of ammonia emissions within the NEI as well as errors with the meteorology
575 used within the models. Based on the results of this study, it seems that the two major
576 sources of error within the model lies with the 2005 NEI NH₃ emissions and the lack of the
577 bidirectional flux model used in the model.

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581

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