Fluorescent dissolved organic matter as a multivariate biogeochemical tracer of submarine groundwater discharge in coral reef ecosystems.

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1 Abstract

2 In Hawai'i and other Pacific high islands submarine groundwater discharge (SGD) 3 can be a significant and continuous source of solutes to nearshore reefs and may 4 play a key role in the structure and function of benthic coral and algal communities. 5 Identifying SGD sources and linking them to reef biogeochemistry is technically 6 challenging. Here we analyzed spectra of fluorescent dissolved organic matter 7 (fDOM) in coral reefs in the context of a suite of biogeochemical parameters along 8 gradients of SGD to characterize fDOM composition and evaluate the utility of fDOM 9 signatures in tracking groundwater dispersal and transformation. We spatially 10 mapped water column chemistry in Maunalua Bay, O'ahu, Hawai'i by collecting 24 water samples in grids at each of two ~ 0.15 km² regions during both high and low 11 12 tides over a two-day period. We observed clear horizontal gradients in the majority 13 of 15 measured parameters, including inorganic and organic solutes and organic 14 particles that tracked concentrations of conservative SGD tracers (radon, salinity 15 and silicate). Multivariate scanning excitation-emission fluorometry successfully 16 differentiated two distinct groundwater sources and delineated regions of SGD 17 dispersion in each reef from the surrounding water column samples without 18 detectable groundwater. Groundwater was consistently depleted in DOC and 19 enriched in nutrients; although the two SGD sources varied widely in fDOM quantity and fluorophore proportions, indices of humification were consistently elevated in 20 21 SGD at both sites. Our results provide a robust spectral characterization of fDOM in 22 SGD-influenced coral reefs and indicate the potential for this rapid and cost-effective 23 measurement technique to be useful in tracking SGD dispersal in nearshore 24 ecosystems.

1 Introduction

2 Coastal ecosystems experience dynamic inputs of material from benthic, fluvial, 3 groundwater and offshore habitats. Groundwater can be a significant and 4 continuous source of solutes to nearshore reefs and may play a key role in the 5 structure and function of benthic coral and macroalgal communities, as well as 6 influencing local coastal oceanography and planktonic communities. Groundwater 7 nutrient and organic matter pollution, whether through agricultural fertilization, on-8 site sewage disposal or runoff from industrial/urban land uses, is a major 9 eutrophication concern for coral reefs because of their adaptation to relatively low 10 nutrient conditions (Fabricius, 2005; Lapointe, 1997). However, identifying 11 groundwater sources and linking them to reef biogeochemistry is technically 12 challenging. 13

14 Coral reefs are highly productive ecosystems adapted to oligotrophic oceans, and it 15 remains an open question how they acquire sufficient macro- and micro-nutrients 16 to maintain high productivity in low-nutrient waters (e.g., Alldredge et al., 2013). 17 Submarine groundwater discharge (SGD) is a phenomenon common to the Hawaiian Islands (Dollar and Atkinson, 1992; Johnson et al., 2008; Street et al., 2008; 18 19 Swarzenski et al., 2013) and other Pacific high islands (Kim et al., 2011) wherein 20 groundwater is continuously and directly discharged into shallow coastal reef 21 ecosystems. SGD is assumed to be a fundamental feature of reef ecosystems where 22 fluxes are significant (Cyronak et al., 2014; Paytan et al., 2006), and tracking the rate 23 and extent of groundwater dispersion in coastal regions has been an area of 24 significant active research in Hawai'i and elsewhere (Johnson et al., 2008; Knee et 25 al., 2010; Moore, 2010; Street et al., 2008). Current techniques to understand where 26 and when SGD is diffused throughout the nearshore habitat include thermal imaging 27 (Johnson et al., 2008), dve tracer studies (Burnett et al., 2006), geophysical 28 exploration (Dimova et al., 2012), radioisotopic tracers (Charette et al., 2008), and 29 mapping of conservative solute concentrations (Street et al., 2008).

30

31 The influence of SGD on the structure and function of coral reefs is poorly 32 understood. The elevated levels of nitrate and phosphate found in SGD in many 33 regions of Hawai'i (Johnson et al., 2008; Knee et al., 2010; Street et al., 2008) have 34 been implicated as a key factor in coastal eutrophication (Dailer et al., 2010), 35 changes in benthic algal composition (Smith et al., 2010; Stimson and Larned, 2000) 36 and alteration of nearshore plankton biomass and community structure (Fabricius, 37 2005; McCook, 1999; Parsons et al., 2008). Despite our conceptualization of SGD as 38 driving eutrophication, we have few studies mapping the distribution of organic 39 matter in the water column of reefs experiencing significant SGD inputs (Tedetti et 40 al., 2011). A key question for the role of SGD in coral ecosystems is how SGD may 41 influence the organic composition of coral reefs, both through allochthonous 42 subsidies and through stimulation of autochthonous productivity.

43

44 Dissolved organic matter (DOM) in aquatic ecosystems is a significant component of 45 the total organic content of marine ecosystems. The pool of DOM in the oceans is 46 vast, containing carbon equivalent to the CO₂ in the Earth's atmosphere, and 47 compositionally complex, with degradation time scales that vary greatly from hours to many years (Hansell and Carlson, 2014). A portion of DOM fuels food webs 48 49 through metabolism by single-celled osmotrophs such as heterotrophic Bacteria and 50 Archaea that are subsequently grazed by microbial eukarvotes, a process known as 51 the "microbial loop" (Azam et al., 1983). The organic matter content of groundwater 52 can vary widely depending on the geological and hydrological factors defining 53 groundwater catchments and biogeochemical processes altering solutes within the 54 subterranean estuary (STE) (Kim et al., 2012). SGD in island systems can be sourced 55 from a variety of different ages and levels of human impact (Knee et al., 2010; 56 Wolanski et al., 2009), and little is known about the characteristics of groundwater 57 organic matter in Pacific islands (Tedetti et al., 2011). SGD passes through the STE, 58 a biogeochemical reactor that is analogous in metabolic complexity to surface 59 estuaries where terrestrial freshwater and recirculated seawater mix, differing 60 markedly with regards to sunlight exposure, residence time and redox conditions. 61 The sources of DOM in the STE can be diverse and include terrestrial inputs (Tedetti

62 et al., 2011), locally produced DOM within the STE (Santos et al., 2009), and marine 63 DOM, which enters the STE via seawater recirculating through the coastal aquifer 64 (Beck et al., 2007; Goñi and Gardner, 2003; Kim et al., 2012). Direct allochthonous 65 DOM subsidies from SGD may have varying degrees of lability relative to ambient 66 DOM depending on age and composition (Burdige et al., 2004; Kim et al., 2012); if 67 allochthonous DOM in SGD is labile, it could stimulate the microbial loop in reefs 68 thereby supporting a portion of the reef food web. Autochthonous production 69 stimulated by SGD nutrient subsidies may also produce labile DOM that supports 70 higher trophic levels (Johnson and Wiegner, 2013; Lee et al., 2010). Both subsidies 71 of DOM may have significant impacts on reef ecosystem function, and understanding 72 the relationship between SGD and DOM in reefs is an important step toward 73 understanding how SGD influences reef ecosystems and how groundwater 74 contamination may alter ecosystems processes.

75

76 The composition of DOM in aquatic environments is known to be highly complex, 77 comprising a diverse suite of thousands of molecules ranging in molecular weight 78 across many orders of magnitude (Hansell and Carlson, 2014). One method of 79 characterizing DOM is through spectral analysis of a subset of DOM that exhibits 80 autofluorescence, typically stimulated by ultraviolet and blue light (Coble, 1996). 81 This fluorescent DOM (fDOM) can exhibit variable fluorescence across a range of 82 excitation and emission wavelengths, and scanning fluorescence spectroscopy can 83 produce a three dimensional map of the fDOM in a sample that varies through space 84 and time according to subtle shifts in chemical composition of the complex molecular assemblage (Nelson and Coble, 2009). Analysis of the multivariate 85 spectral characteristics of fDOM by generating an excitation-emission matrix (EEM) 86 87 from a sample is a cost-effective analysis that requires minimal laboratory training 88 and equipment and can produce a suite of informative data about the organic matter 89 chemistry of the water. In marine ecosystems, the variation in fDOM characteristics 90 has been used to differentiate between a variety of DOM sources including 91 terrestrial (Coble, 1996), algal (Determann et al., 1998), microbial (Stedmon and 92 Markager, 2005) and anthropogenic (Dabestani and Ivanov, 1999; Ferretto et al.,

93 2014). Additionally, fDOM characterization has proven useful in differentiating

94 contributions from rivers, groundwater, coastal margins and reefs, and the open

ocean (Chen et al., 2003; Helms et al., 2013; Osburn et al., 2013; Tedetti et al., 2011).

96 If fDOM exhibits clearly defined characteristics across gradients of SGD influence it

97 has the potential to serve as a promising tool for understanding the role of

98 groundwater in reef ecosystems.

99

100 The present study sought to examine the relationship between SGD inputs and the 101 field of particulate, dissolved and fluorescent organics in coral reef ecosystems. We 102 identified two regions of a single contiguous reef system with relatively predictable 103 nearshore inputs of SGD (Maunalua Bay, O'ahu, Hawai'i), which range from 12,000 104 to 16,000 m³ d⁻¹ (Holleman, 2011). SGD here is composed of brackish groundwater 105 (salinity 2-5) discharging through channelized groundwater conduits (Dimova et al., 106 2012) thus bypassing STE processes typical for tidal flats. Since the water contains 107 minimal recirculated seawater and an extensive STE is absent, its terrestrial DOM 108 signature is preserved making it a unique SGD tracer. We first mapped the extent of 109 SGD by collecting water samples in a grid centered on an identified spring discharge 110 site at low and high tide and measuring a suite of inorganic solute concentrations, 111 including relatively conservative groundwater tracers (salinity, radon and silicate) 112 and various chemical species of N and P. From these same water samples we 113 collected a suite of dissolved and particulate organic matter samples, including bulk 114 measurements of DOM, particulate organic C and N, chlorophyll and cytometric 115 counts of picoplankton and bacterioplankton. We tested whether these parameters 116 correlated spatially and by site with the inorganic solutes measured, examining if 117 samples from different regions of the reef clustered together in patterns consistent 118 with SGD influence on the organic matter field. Finally, we conducted spectral 119 analyses on fDOM to understand the characteristics of fDOM in different 120 groundwater sources and evaluate the potential for tracking SGD dispersal and 121 alteration across the reef ecosystem. Our results demonstrate that SGD in coral reef 122 habitats alters not only the composition of inorganic solutes such as salinity, silicate 123 and nutrients, but also bulk concentrations of dissolved and particulate organics and

- 124 the spectral characteristics of fDOM. We discuss the potential for fDOM
- 125 measurements to be developed into a cost-effective tool for tracking SGD in similar
- 126 coral reefs dominated by spring groundwater inputs.
- 127 Methods
- 128
- 129 Water collection

130 We collected water samples at each of two nearshore fringing coral reef sites within 131 Maunalua Bay, Oʻahu, Hawaiʻi (Figure 1a) over a two day period; 24 samples were 132 collected near Wailupe Beach Park on 28 May 2014 and 24 samples were collected 133 at Black Point on 29 May 2014 (Figure 1b,c). The majority of samples (32 of 48) 134 were collected synoptically during morning low tides (-6 to -9 cm) with an 135 additional subset (16) collected during afternoon high tides (+67 cm; Fig. S1). 136 Depths at these sites are generally < 2 m, and the majority of samples (38) were 137 collected from the surface (0.2m below sea level); a subset of samples were 138 additionally collected from bottom waters (roughly 0.2 m above the benthos). Using 139 kayaks, surface water samples were hand-collected into 5 L high-density 140 polyethylene carboys (73062, US Plastics, Lima, OH, USA); bottom water samples 141 were collected in 5 L horizontal Niskin samplers (101005H, General Oceanics, 142 Miami, FL, USA) and immediately siphoned to carboys for processing. Carboys and 143 Niskin samplers were previously conditioned with seawater, soaked overnight in 144 10% HCl. thoroughly rinsed with low-organic deionized water (DIW: Barnstead 145 Nanopure Diamond, Thermo Fisher Scientific, Asheville, NC, USA) and stored dry 146 before sampling. All sample containers were also triple-rinsed with sample water 147 before filling. All filtration and subsampling of water was done on site within 2 h of 148 collection.

149

150 Sample collection and storage

All storage vials were acid soaked, thoroughly rinsed with DIW, air dried, and then
triple rinsed with sample water before collection. Samples for total alkalinity (TA)

- 153 were collected first directly from the carboy spigot. Duplicate TA samples (125 mL
- each) were transferred into polypropylene sample bottles (Huang et al., 2012) and

155 amended with 50 μ L of half-saturated HgCl₂. All subsequent samples were 156 transferred to long-term storage vials via gentle (1 mL s⁻¹) peristaltic pumping 157 directly from the carboy through platinum-cured silicone tubing over a period of 158 roughly 1 h. For dissolved nutrient analyses, the filtrate from a 0.2 µm 159 polyethersulfone filter (Sterivex, Millipore, Billerica, MA, USA) was collected in acid-160 washed and sample rinsed 50 mL polypropylene tubes and frozen to -20 °C. 161 Samples for fDOM analysis were collected from the 0.2µm polyethersulfone filtrate 162 (after a minimum of 250 mL sample flushing to avoid filter DOM leaching) in amber 163 glass vials with teflon-lined septate caps (acid-washed and DIW rinsed) and stored 164 in a dark refrigerator free of volatile organics. For dissolved organic carbon (DOC) 165 analyses the filtrate from glass fiber filters (Whatman GF/F, GE Life Sciences, 166 Pittsburgh, PA, USA) was collected in glass vials with teflon-lined septate caps (acid-167 washed and DIW rinsed) and frozen to -20 °C in an organic-free freezer. All glass 168 vials and filters were pre-combusted within days of sampling (2 h at 400 $^{\circ}$ C) and 169 stored in a laboratory free of volatile organics. For analysis of chlorophyll *a* and 170 particulate organic carbon/nitrogen, 600 to 1000 mL of sample water for each 171 sample was filtered onto a 25 mm GF/F filter, folded in half, wrapped in Al foil and 172 frozen to -20 °C. For flow cytometry (FCM), 1.5 mL of unfiltered water was fixed at 173 0.5% paraformaldehyde (amended with 100 µL 8% ampulated paraformaldehyde, 174 Electron Microscopy Sciences, Hatfield, PA, USA) in a 2 mL polypropylene cryovial, 175 mixed briefly and then frozen to -80 °C. All samples were immediately refrigerated 176 in the field and frozen or refrigerated within 6 hours of collection for long-term 177 storage.

178

179 Inorganic nutrient and organic matter concentration measurements

180 Nutrient samples were thawed to room temperature, mixed thoroughly and

181 analyzed on a Seal Analytical Segmented Flow Injection AutoAnalyzer AA3HR for

182 simultaneous determination of soluble reactive phosphate (PO₄³⁻), ammonium

183 (NH₄+), nitrate+nitrite (N+N; NO₃- + NO₂-), silicate (SiO₄) and total dissolved

184 nitrogen and phosphorus (TDN, TDP; via in-line persulfate/UV oxidation). DOC and

185 TDN samples (GF/F filtrate frozen in glass vials) were measured as non-purgeable 186 organic carbon and nitrogen via acidification, sparging and high-temperature 187 catalytic oxidation on a Shimadzu TOC-L with TMN-L attachment, ensuring that 188 Deep Seawater Reference waters from the University of Miami Consensus Reference 189 Materials Project measured within specifications in each run to facilitate 190 comparison of results to those obtained by the international DOM community. 191 Chlorophyll *a* (Chla) concentrations were measured by acetone extraction and 192 fluorescence spectroscopy on a modified Turner 10-AU fluorometer following 193 Welschmever (1994). Particulate organic carbon (POC) and nitrogen (PON) 194 concentrations were determined via filter combustion on an Exeter Analytical CE 195 440 Elemental Analyzer after acid fumigation to remove particulate inorganic 196 carbon, drying, weighing and packing into tin capsules. TA samples were analyzed 197 using open cell potentiometric titrations on a 166 Mettler T50 autotitrator and 198 calibrated against a Certified Reference Material (Dickson et al., 2007). Salinity was 199 measured as electrical conductivity with a combination platinum ring electrode – 200 thermistor (Metrohm 6.0451.100) on a Metrohm conductivity module using Tiamo 201 software (v2.4). All solute and particulate samples were analyzed in the SOEST 202 Analytical Laboratory (http://www.soest.hawaii.edu/S-LAB/). 203

204 To assess field and technical replicability, a separate nutrient sample was collected 205 in situ at a subset of 15 locations and filtered immediately through a 206 polyethersulfone 0.45 μm groundwater cartridge filter (AquaPrep 600, Pall Life 207 Sciences, Ann Arbor, MI, USA). Samples were stored refrigerated in 250 mL 208 polyethylene bottles for 1 month and analyzed for salinity and by flow injection 209 autoanalyzer in parallel with the primary samples. The replicated samples were 210 representative of the range of biogeochemical zones, spanning 75% of the 211 lognormal data range for each parameter. Linear models of log₁₀-transformed data 212 collected by the two methods demonstrated strong congruency. The two methods 213 were highly correlated (r > 0.98 for TDN, TDP, PO₄³⁻, N+N, SiO₄, Salinity and r = 0.85214 for NH₄⁺). In addition, both least-squares and orthogonal (Model II or reduced major 215 axis) regression model slopes for all 7 parameters (p < 0.001) were not significantly 216 different from 1 (95% confidence intervals bracketed 1). Intercepts were nearly all 217 non-significant (intercept p > 0.25, except TDP p = 0.0027), indicating no offset 218 between the two sample sets; when intercepts were constrained to zero, slopes 219 remained not significantly different from 1. This analysis indicates that 220 measurements of standard solutes were robust to minor variation in sample 221 collection and storage as well as field variation in sampling; the primary samples 222 were used in all subsequent analyses.

223

224 Dissolved organic N and P (DON, DOP) were calculated as the difference between 225 TDN, TDP and inorganic species of N and P: DOP = TDP – PO_4^{3-} and DON = TDN – 226 N+N - NH₄⁺. TDN was measured using two separate methods: via high temperature 227 catalytic oxidation (HTCO) and subsequent ozonation chemiluminescence of 0.7 μm 228 filtrate (GF/F) on a Shimadzu TMN-L and via persulfate alkaline oxidation and 229 subsequent colorimetric cadmium reduction of 0.2 μ m filtrate (Sterivex) on a Seal 230 Analytical AA3HR (described above). The two methods yielded highly correlated 231 measures of TDN (r = 0.97, n = 46) but the HTCO method yielded consistently lower 232 estimates of TDN (Model II lognormal regression slope of 0.85 with 95 % confidence 233 interval of 0.79 to 0.91). Estimates of DON derived from the two TDN measurements 234 did not covary and were not significantly related to N+N or DOC (p > 0.05). When 235 N+N concentrations exceeded 40 µmol L⁻¹ estimates of DON from HTCO were 236 negative; for all subsequent analyses, Seal autoanalyzer TDN measurements were 237 used to maintain methodological consistency with other N and P measures.

238

239 *Radon activity measurements*

240 Coastal radon activities were measured using a RAD AQUA closed air loop

241 continuous equilibrium exchanger accessory for the RAD7 radon detector

242 (Durridge Inc., Billerica, MA). The system was mounted on a small boat hand-pulled

along the shoreline and in perpendicular transects. The system was mounted on a

small boat and hand-pulled along pre-determined GPS transects. Coastal springs and

245 diffuse seepage was identified by moving along the shoreline and shore-246 perpendicular transects were used to determine the extent of significant 247 groundwater plumes at the two focus areas. The air-water exchanger of the RAD-248 AQUA was fed by water using a submersible bilge pump submersed 0.2 m below the 249 water surface. The instrument recorded radon in 5-minute integrated intervals 250 providing a spatial resolution of 50-100 meters. Radon in air values were converted 251 to radon in water activities using temperature and salinity recorded by a YSI (V2-2) 252 multiparameter probe (Schubert et al., 2012). It has been shown previously that the 253 nearshore water residence time at the sampling sites is one tidal cycle so the 254 reported radon values are not corrected for radon decay and evasion to the 255 atmosphere (Holleman, 2011). The radon survey covered the whole bay area but 256 only results relevant to Black Point and Wailupe are included in this analysis.

257

258 Flow cytometry

259 Flow cytometry was used to measure both autofluorescent and total nucleic-acid 260 stained cell concentrations in fixed unfiltered water samples (Nelson et al., 2011). 261 Samples were thawed and placed in 250 µL aliquots in 96-well autosampler plates 262 in duplicate; one of the two wells was mixed with SYBR Green I stain (1X final 263 concentration) within 2.5 h of analysis. Samples were analyzed on an Attune 264 Acoustic Focusing Cytometer with Autosampler Attachment (Life Technologies, 265 Eugene, OR, USA). Samples were run at flow speeds of 100 μ L min⁻¹ on standard 266 sensitivity; 150 μ L of sample was aspirated, 75 μ L was counted and data was 267 collected only from the last 50 μ L (event rates were empirically determined to be 268 steady only after 25 µL of continuous sample injection). For SYBR-stained cells a 269 blue laser (488 nm, threshold 10,000 rfu, voltage 2300mV) was used to excite the 270 dye and cell counts obtained by increasing the voltage to maintain event counts of 271 blank controls (SYBR-stained 0.2 µm filtered DIW) below 100 events s⁻¹ and event 272 counts of environmental samples below 1500 events s⁻¹. This allowed for clear 273 gating of plankton cells as populations distinct from instrument noise in bivariate 274 plots of sidescatter and green fluorescence (530/30 nm bandpass fluorescence; BL1 275 channel). For autofluorescent cells a combination threshold on Violet (405 nm) OR 276 Blue (488 nm) laser excitation and red emission (600 nm and 640 nm longpass 277 filters, respectively VL3 and BL3 channels, 1000 rfu, 2500 mV) was used, and size-278 based sidescatter gating was applied to differentiate autofluorescent photosynthetic 279 bacterioplankton (PBact) from photosynthetic autofluorescent picoeukaryotes 280 (PEuks). Concentrations were corrected for stain and paraformaldehyde dilution 281 factors, and heterotrophic bacterioplankton counts (HBact) were calculated as the 282 difference of SYBR and total autofluorescent counts. These settings were empirically 283 tested for streamwater, coastal and open ocean heterotrophic and autofluorescent 284 bacterioplankton from the North Pacific Subtropical Gyre down to depths of 4000 m 285 with densities ranging from 100 to 2000 cells μ L⁻¹ for SYBR-stained cells and 1 to 286 500 cells µL⁻¹ for autofluorescent cells; counts matched those derived from 287 epifluorescent microscopy within 10% in all cases.

288

289 Fluorescent dissolved organic matter (fDOM) measurement

290 Analysis of fDOM was conducted on a Horiba Aqualog scanning fluorometer with 291 150W Xe excitation lamp, Peltier-cooled CCD emission detector and simultaneous 292 absorbance spectrometer. Samples were warmed to room temperature (22 °C) for 2 293 hours while the Xe bulb warmed. Excitation-emission matrices (EEMs) were 294 measured from each of 48 samples in a 1 cm DIW-leached and rinsed quartz cuvette 295 (3-Q-10, Starna Cells, Atascadero, CA, USA) with 4 DIW blanks run at the start and 296 end of the contiguous 3 h analysis period. Water was excited through a 5nm 297 bandpass subtractive double monochromator in declining 5nm sequence intervals 298 from 500 to 240 nm and emission was integrated 4s at each step and binned in 4.65 299 nm intervals (8-pixel bins) from 250 to 800 nm. Scans were processed using custom 300 scripts in Matlab (v2007b) as follows: 1) first inner filter effect correction was 301 applied to account for the quenching of fluorescence by absorbance following the 302 recommendations of Kothawala, et al. (2013) by multiplying by the antilog of the 303 average of absorbances at the wavelengths of excitation and emission for each 304 fluorescence data point, 2) next EEMs were scaled to Raman units (RU) by dividing

by the integrated emission range of 381 to 426 nm at an excitation of 350 nm in

averaged DIW blanks (Lawaetz and Stedmon, 2009; Murphy et al., 2010) and 3)

307 average DIW blank EEMs were subtracted from each sample.

308

309 fDOM modelling and indices

310 We used Parallel Factor Analysis (PARAFAC) to derive four modelled fDOM 311 components with the DOMFluor toolbox (v1.7; Stedmon and Bro, 2008), trimming 312 Rayleigh and Raman scatter, testing for outliers (none were identified), deriving up 313 to 6 PARAFAC components then using split-half validation and random initialization 314 to determine the appropriate number of modeled components (in this case only the 315 first 4 could be validated). We also calculated a suite of derived indices from each 316 EEM that are commonly used to differentiate aspects of fDOM character and help 317 interpret DOM sources. The ratio of marine-derived to terrigenous fDOM (e.g. M:C) 318 was calculated as the ratio of fluorescence at Ex310/Em410 divided by fluorescence 319 at Ex345/Em445 (Burdige et al., 2004). The M:C has had utility in differentiating 320 between marine- and terrestrial-derived fDOM (Burdige et al., 2004; Helms et al., 321 2013). The Fluorescence index (FI; McKnight et al., 2001), calculated as the ratio of 322 fluorescence at 470 nm to 520 nm under 370 nm excitation (Cory et al., 2010; Maie 323 et al., 2006), expresses the ratio of terrigenous vs. autochthonous-produced humic 324 DOM. Similarly, the fluorescent biological index (BIX), which is associated with 325 microbially-derived and autochthonous DOM, was calculated as the ratio of 326 fluorescence at 380 nm to 430 nm under 308 nm excitation (Huguet et al., 2009). 327 BIX >1 can indicate a strong signal of recent autochthonous DOM production, 328 whereas those <0.7 reflect older authorhthonous DOM (Huguet et al., 2009). Lastly, 329 the fluorescent humification index (HIX), often used to estimate the extent of DOM 330 diagenesis or maturation in soils, was calculated as the integrated fluorescence from 331 434 to 480 nm divided by the integrated fluorescence from 300 to 346 nm under 332 254nm excitation (Zsolnay et al., 1999). High HIX values (>10) indicate aromatic 333 DOM (potentially from terrestrial or marine humic acids) whereas low values (<4) 334 reflect more autochthonous origin (Birdwell and Engel, 2010). Lastly, we used the

- absorbance spectra to calculate specific ultraviolet absorbance (SUVA₂₅₄) by
- dividing the linear absorbance (m⁻¹) by DOC (mg L⁻¹) (Weishaar et al., 2003).
- 337

338 Statistical analyses

339 All nutrient, organic, carbonate, fDOM indices and flow cytometry parameters were 340 log₁₀-transformed to better approximate a gaussian (normal) distribution before 341 statistical analysis; raw fDOM values were normally distributed and were not 342 transformed for statistical analysis. Hierarchical clustering (Ward's minimum 343 variance method) was used to group samples according to similarity in multiple 344 biogeochemical parameters as a way to define clusters of samples with similar 345 properties. Each parameter was first standardized (by subtracting the column mean 346 and dividing by the column standard deviation) to avoid weighting clusters by 347 absolute measurement values. To conservatively define groups of samples 348 according to relative proportion of SGD influence based on inorganic chemical 349 composition, samples were initially clustered by the full suite of 7 standard 350 inorganic solute measurements made (PO₄³⁻, N+N, NH₄⁺, SiO₄, Salinity, TA and Rn). 351 This clustering approach differentiated samples into spatially-distributed 352 "biogeochemical provinces" interpreted as SGD Springs, Transition SGD mixing 353 Zones, Diffuse SGD Zones and Ambient Reef waters (detailed further in results). 354 Analysis of variance (ANOVA) was then used to test if mean values of organic 355 parameters differed among the inorganic biogeochemical provinces and sites, with 356 Tukey and Dunnet's *post hoc* tests used to assess pairwise differences among groups 357 at α = 0.05. Chi-square tests were used to assess similarity in cluster assignment of 358 samples using different suites of organic variables. Pearson correlation and linear 359 regression models (least squares and orthogonal/reduced major axis/Model II 360 approaches) were used to assess covariance among variables. 361

362 363 **Results:**

364 Distributions of dissolved inorganic solutes and delineation of groundwater influence

365 At each site SGD sources and dispersal patterns were clearly visualized by contour 366 mapping of conservative inorganic solute tracer concentrations in surface samples 367 at low tide (Figure 2a-f). Concentration gradients were consistent with rapid 368 dilution within 200m of the source springs at each site. Contours of the fDOM 369 humification index (HIX) closely tracked these conservative solute gradients across 370 the reef platform (Figure 2g,h), and HIX was highly correlated with salinity and 371 silicate consistently at both sites (r > 0.75; Figure S2), demonstrating that fDOM 372 parameters tracked salinity.

373

374 Hierarchical clustering of samples according to the suite of 7 standard inorganic 375 solute measurements (SiO₄, Salinity, Rn, PO₄³⁻, N+N, NH₄⁺ and TA) separated 376 samples into six distinct groups, which we refer to subsequently as "biogeochemical 377 provinces" because of their spatial differentiation (Figure 3a). Groundwater springs 378 at Wailupe and Black Point were distinct (BP Spring and WL Spring Provinces), 379 areas of significant SGD mixing at Wailupe and Black Point were distinct (BP 380 Transition Zone and WL Transition Zone Provinces), while Diffuse SGD Zones and 381 Ambient Reef provinces did not differ between sites (Figure 3b,c). Figure 3d,e 382 provides a conceptual spatial schematic of the biogeochemical provinces defined in 383 Figure 3a that are referenced throughout this study (e.g. Spring, Transition Zone, 384 Diffuse Zone and Ambient Reef).

385

386 At the Springs, silicate concentrations were > 500 μ mol L⁻¹, salinities < 10 and radon 387 activities > 150 dpm L⁻¹ while Ambient Reef waters had silicate concentrations < 5 388 µmol L⁻¹, salinities near 30 and radon activities < 20 dpm L⁻¹; Transition and Diffuse 389 Zones exhibited characteristic intermediate silicate concentrations and did not 390 differ significantly from Ambient Reef sources in salinity or radon (Table 1, Figure 391 S3). Sites did not differ significantly in any of the inorganic tracer solutes except that 392 Transition Zone waters had more radon at Black Point (mean 148 dpm L⁻¹) than at 393 Wailupe (mean 43 dpm L⁻¹; Figure S3). Springs at both sites were significantly 394 higher in N+N (> 50 μ mol L⁻¹) and PO₄³⁻ (> 1.5 μ mol L⁻¹) than any other samples:

395 nearby Transition Zone samples remained significantly higher (> 5X) than adjacent 396 Diffuse Zone and Ambient Reef waters that did not differ significantly from each 397 other (< 1.5 μ mol L⁻¹ N+N and < 0.15 μ mol L⁻¹ PO₄³⁻; Figure S4). In contrast, NH₄+ 398 concentrations were depleted in Springs (near limits of detection) relative to the 399 adjacent Transition Zone and Diffuse Zone waters, both of which were strikingly 400 enriched in NH₄⁺ (mean 0.9 μmol L⁻¹) above Ambient Reef samples (mean 0.3 μmol 401 L^{-1} ; Table 1 and Figure S4). TA in the groundwater Springs differed markedly 402 between the two sites, being significantly elevated at Black Point (mean 2826 µmol 403 kg⁻¹) and significantly depleted at Wailupe (mean 1616 μmol kg⁻¹) relative to all 404 other biogeochemical provinces at both sites (which did not differ significantly; 405 mean concentrations 2250 µmol kg⁻¹; Table 1 and Figure S4).

406

407 Distributions of particulate and dissolved organics:

408 Hierarchical clustering of samples according to a suite of 9 measured dissolved and 409 particulate organic matter concentrations and ratios (Chl a, Picoeukaryotic 410 phytoplankton, autotrophic bacterioplankton, heterotrophic bacterioplankton, DOC, 411 DON, DOP, DOC:N and TN:TP) yielded 6 distinct groups (Figure 4a) with spatial 412 distributions of sample types consistent with SGD gradients (Figure 4b,c). Group 413 assignment of samples by clustering on inorganic solutes (Fig. 3a) and organic 414 matter (Fig. 4a) was highly congruent (Contingency $R^2 = 0.63$, Pearson Chi-square 415 0.96 and p < 0.0001 for Low Tide samples), with 75% of the samples assigned to 416 identical groups (Figure 4a). Spring samples were all assigned perfectly, but Black 417 Point samples were more homogenous spatially in terms of organic matter than 418 observed with inorganic solutes and did not separate clearly among Transition 419 Zone, Diffuse Zone and Ambient Reef types, potentially indicating a more extensive 420 influence of SGD on the reef organic field (Figure 4b). Wailupe spatial patterning of 421 organic matter was consistent with inorganic solutes (Figure 4c). 422

423 DOC was significantly depleted in both Springs relative to the surrounding waters
 424 (mean 85 μmol L⁻¹); concentrations in Wailupe Springs (mean 20 μmol L⁻¹) were

425 more than twice as low as those in Black Point Springs (mean 47 μmol L-1; Table 1 426 and Figure S5). The two Springs had very different DON concentrations, both 427 significantly different from the surrounding waters (mean 6.5 µmol L⁻¹), with Black 428 Point highly enriched (mean 34 μ mol L⁻¹) and Wailupe significantly depleted (mean 429 1.4 µmol L⁻¹). Dissolved organic phosphorus was unresolvable in Springs and near 430 detection limits in Transition Zone regions, and did not differ among Diffuse Zones 431 and Ambient Reef waters (mean 0.3 µmol L⁻¹). Particulate organic concentrations 432 (POC, PON and chl *a*; Fig. S6) and flow cytometry (Picoeukaryotic phytoplankton, 433 Autotrophic and Heterotrophic bacterioplankton; Fig S7) pairwise differences 434 among biogeochemical provinces were mostly non-significant due to high variance, 435 but overall exhibited trends of particulate depletion in Springs and plankton

436 enrichment in the surrounding Transition Zone waters (Figs. S6, S7).

437

438 *fDOM characteristics and distributions:*

439 The PARAFAC modelling validated 4 fluorescence components that covaried with fluorescence regions widely identified from marine systems (Regions: A - humic-like 440 441 UV excitation; M – visible, blue-shifted, marine humic-like; C – visible excitation, 442 humic-like; and T – aromatic amino protein-like; Coble 1996) both in terms of 443 spectral characteristics (Fig. S8) and in terms of standardized distributions among 444 variables within this dataset (Figure S8e). In addition, the spectral loadings of our 445 PARAFAC components (Fig. S8a-d) matched PARAFAC components found in marine 446 systems in various recent reviews: Component 1 $(\text{Em}(2^{\circ})/\text{Ex}: 260(375) \text{ nm}/375)$ 447 nm) is consistent with component C1 from Jørgensen et al. (2011) and component 448 C2 from Ishii and Boyer (2012). Our Component 2 $(\text{Em}(2^{\circ})/\text{Ex:} < 250(325) \text{ nm} / 10^{\circ})$ 449 400-480 nm) corresponds to component C4 from Jørgensen et al. (2011) and 450 component C3 from Ishii and Boyer (2012). Component 3 (Em/Ex(2°): 300-451 380nm/510(480nm) corresponds to component C4 in Kowalczuk et al. (2009). 452 Component 4 (Ex/Em(2°): 260nm/330(510nm) corresponds to component C2 in 453 Jørgensen et al. (2011).

454

455 Each of the four PARAFAC components differed significantly among the 456 biogeochemical provinces (ANOVA p < 0.0001). At Black Point both Spring and 457 Transition Zone provinces were enriched relative to Diffuse Zone and Ambient Reef 458 waters for all components, indicating that total fDOM was elevated in the 459 groundwater. In contrast, Wailupe Transition and Diffuse Zones were enriched 460 relative to both Spring and Ambient Reef waters (Fig. 5), consistent with the idea of 461 production of fDOM in the SGD-influenced reef waters of Wailupe. Notably, at Black 462 Point, although fDOM decreased from Spring to Ambient Reef waters, DOC exhibited 463 some enrichment in the transition and diffuse zones, suggesting that autochthonous 464 production of non-fluorescent DOM may have occurred in the diffuse zones. 465 Within any of the four biogeochemical provinces there were clear site differences in 466 fDOM quantity: Black Point Spring samples were enriched in all four components 467 relative to Wailupe Springs while the reverse was true in Diffuse Zone samples, with 468 Wailupe enriched for components A, M and C (Fig. 5), again consistent with the idea 469 of production of fDOM in the SGD-influenced reef waters of Wailupe.

470

471 The four ratio-based fDOM indices exhibited very different patterns within sites, emphasizing that each index is assessing a different aspect of the character of FDOM 472 473 (Figure 6). In SGD Springs, the humification index, HIX, was more than double (> 7)Ambient Reef values (< 3) at both sites (Table 1). HIX was highly correlated with 474 475 salinity and silicate consistently between sites (r > 0.75; Figure S2) and declined 476 continuously with distance from the Springs (Figure 2). The M:C index covaried with 477 HIX (r = 0.83) and was also significantly enriched in Springs and in Transition Zone 478 waters relative to the Diffuse Zone and Ambient Reef waters. Relative to ambient 479 waters, the Wailupe Spring had a significantly higher fluorescence index (FI) and 480 SUVA254 was significantly greater in the Black Point Spring. BIX ranged from 0.76-481 0.86 across both sites, and was generally elevated in Black Point relative to Wailupe, 482 but did not differ significantly among water types. Indices did not differ between 483 sites within a given biogeochemical province (Figure 6).

484

485 Hierarchical clustering of samples according to a suite of 9 fDOM-derived 486 parameters (4 PARAFAC components, 4 fDOM indices and SUVA₂₅₄) separated 487 samples into 6 groups (Figure 7a) with spatial distributions of sample types 488 consistent with SGD gradients (Figure 7b,c). Group assignment of samples by fDOM 489 characteristics was generally congruent with clustering by inorganic solutes (Fig. 3) 490 and organic matter (Fig. 7; Contingency $R^2 = 0.52$ and 0.50, respectively, Pearson 491 Chi-square 93 and 78, respectively and p < 0.0001 for Low Tide samples), with 70% 492 of the samples assigned to identical groups. As with the organic matter clustering, 493 Spring samples were all assigned perfectly, but at both sites the other 494 biogeochemical provinces were less clearly differentiated spatially in terms of fDOM 495 than observed with inorganic solutes. Notably, both Transition Zone and Diffuse 496 Zone samples appeared to be different in fDOM parameters between Wailupe and 497 Black Point (Figure 7a), potentially indicating a more extensive influence of SGD on 498 the reef fDOM field. The two SGD springs (and the two diffuse zones) may be 499 differentiated according to their fDOM amount (i.e. fluorescence intensities of 500 fluorophores) (Fig. 5) but cannot be clearly discriminated by fluorophore ratio 501 indices (Fig. 6).

502

503 **Discussion**

504

505 Biogeochemical characteristics of groundwater entering Maunalua Bay

506 Groundwater discharging from springs in Maunalua Bay showed some consistencies

and differences between the Black Point and Wailupe sites. Both sites released

508 groundwater with biogeochemical profiles consistent with previous studies of SGD

in coral reefs (Swarzenski et al., 2013), including low salinities and elevated

- 510 concentrations of radon and silicate, elevated PO₄³⁻ and N+N, depleted NH₄⁺
- 511 concentrations and depleted DOC, POC, chl *a* and phytoplankton cells (Table 1, Figs.
- 512 2-4 and S3-S7). However, some measurements were strongly and significantly
- 513 different between the two Springs. First, the Wailupe Springs were significantly
- 514 depleted in TA and DON relative to the adjacent waters whereas the Black Point TA
- 515 and DON were significantly greater than the adjacent waters, suggesting a

516 fundamentally different hydrological origin (Figs. S4, S5). Second, while both 517 Springs were depleted in DOC relative to the adjacent waters, the Wailupe site had 518 nearly half the DOC concentrations of the Black Point site (Fig. S5), yet Black Point 519 Springs were strongly depleted in bacterioplankton, nearly 4 times less than 520 Wailupe Springs or the surrounding ocean (Fig. S7). Aside from the unexpectedly 521 high concentrations of bacterioplankton in the Wailupe Springs, these patterns 522 suggest that groundwater flowpaths at these two sites are very different, potentially 523 capturing differences in land-use and geology in these two parts of the watershed. 524 Black Point has a higher density of on-site sewage disposal (septic and cesspool) 525 systems (Whittier and El-Kadi, 2009), but we do not have evidence that the high 526 nutrient or fDOM levels are due to these potential sources. The difference in radon 527 concentrations between the springs (Table 1) suggest that groundwater at Black 528 Point flows through rocks generating more radon (i.e., more enriched in U and Ra) 529 than the rocks and sediments at Wailupe, indicating differences in the geologic 530 make-up of the aquifer. Land-use, including density of septic systems, presence of 531 historic agricultural sites, etc. is a likely cause for differences in nutrient and organic 532 matter levels.

533

534 fDOM characteristics of groundwater entering Maunalua Bay

535 Across sites, SGD was significantly enriched in aromatic/humic components (e.g., 536 Regions A and M. HIX. M:C) and had higher specific ultraviolet absorbance (Table 1). 537 The Black Point Springs had significantly more of all 4 fDOM components than the 538 Wailupe Springs and were enriched in all components relative to the Ambient Reef 539 waters (Fig. 5); the Wailupe Springs had fDOM quantities identical to Ambient Reef 540 waters for all four components. The lack of significant differences in fDOM indices 541 between these two sites within any given biogeochemical province (Fig. 6) suggests 542 that SGD fDOM molecular composition was similar. However, groundwater DOC 543 concentrations overall were significantly lower than the overlying reef, and when 544 combined with elevated fDOM (at least at Black Point) and significantly higher ratios 545 of humic compounds (HIX Index, Figs 2 and 6) our data indicate that a greater 546 proportion of the DOC in the groundwater is fluorescent. Indeed, ratios between any

547 of the four PARAFAC components and DOC concentrations, a proxy for specific 548 fDOM fluorescence, were significantly enriched in both Springs relative to the 549 adjacent waters (ANOVA with Tukey post hoc p < 0.0001), indicating that 550 groundwater DOC has a much higher fluorescence than marine DOC. Although both 551 overall fDOM and DOC concentrations were significantly higher in the Black Point 552 site (Figs 5 and S5), ratios of fDOM components to DOC did not differ between sites, 553 only between biogeochemical provinces, again emphasizing that the DOC in SGD is 554 consistently highly fluorescent with a strong humic component (Figs 2, 6, S2).

555

556 Dispersal and biogeochemical influence of submarine groundwater discharge 557 In an observational study it is difficult to separate cause from effect, and in the case 558 of the current dataset we cannot definitively determine whether parameters that 559 differ significantly in the Transition and Diffuse Zones from the Ambient Reef waters 560 are driven by dilution of SGD or stimulation of biogeochemical processes that 561 subsequently alter the characteristics of the water. However, the vast majority of the 562 measured parameters exhibited statistically robust gradients spatially concordant 563 with SGD dispersal across the reef. A handful of parameters did not follow this 564 trend, and it is likely that these represent reef-specific production or consumption 565 processes. Ammonium was the only inorganic solute that exhibited enrichment in 566 the Transition Zone and Diffuse Zone provinces above both Spring and Ambient 567 Reef endmembers, suggesting a production process (Figure S4). Moreover, this 568 parameter was only significantly enriched in the Transition Zone province, not in 569 the adjacent Reef waters, possibly indicating rapid consumption and 570 ammonification of DON, dissimilatory reduction of nitrate to ammonia (DNRA), or 571 recycling of nitrate to ammonia through organic assimilation and remineralization. 572 A similar pattern was observed in particulate organic matter, with the Wailupe site 573 exhibiting enriched Chl a and eukaryotic phytoplankton counts suggestive of some 574 stimulation of water column productivity by the groundwater nutrient delivery. 575 Finally, there was consistent enrichment of fDOM components A,C and T in the 576 Transition Zone and Diffuse Zone provinces above levels found in the Springs and Ambient Reef zones (Fig. 5). Region C has been associated with microbial 577

578 production processes in marine systems and Region T is generally associated with 579 proteinaceous material because of the similarities to pure tryptophan fluorescence 580 (Coble et al., 2014). One interpretation of these patterns is that these components 581 represent fDOM being produced by the reef habitat; whether that production is 582 influenced directly or indirectly by SGD inputs or is simply a characteristic of reefs 583 generally is not currently known. Certainly there are potential physicochemical 584 changes in DOM across these gradients due to photodegradation, metal-ligand 585 bonding, pH, and salinity shifts that impart changes to fDOM character (Helms et al., 586 2013; Osburn et al., 2013).

587

588 Recommendations for the use of fDOM to track groundwater discharge in reefs 589 This study shows clearly how fDOM spectral analyses can be used to differentiate 590 water masses according to the degree of influence of SGD in two sites with very 591 different SGD organic matter profiles, consistent with previous work conducted in 592 coral reef environments with allochthonous DOM inputs (Tedetti et al., 2011). Based 593 on our observations, it is clear that groundwater entering Maunalua Bay contains a 594 significant quantity of fDOM with more than double the humification index (HIX) of 595 the receiving Diffuse Zone and Ambient Reef waters (Figure 6, Table 1), consistent 596 with fDOM from sedimentary and volcanic sources in reef ecosystems (Tedetti et al., 597 2011). This single feature of the fDOM spectra (i.e., elevated HIX) was strongly 598 correlated with both salinity and silicate in both sites with identical slopes, and from 599 this simple index there is the potential to model these inorganic solute 600 concentrations from the HIX value (Figure S2). In addition, the groundwater sources 601 to Black Point stood out clearly from the surrounding reef in humic fluorescence 602 regions A and M (Figure 5), though not at Wailupe. These differences suggest that 603 fDOM characteristics may be able to differentiate groundwater according to land 604 use, hydrology, or other factors, allowing the development of fDOM as a 605 groundwater source-tracking tool in concert with other biogeochemical parameters. 606 Future studies should consider the use of continuous sensor monitoring of coastal 607 fDOM (our PARAFAC component C1 exhibits a secondary emission peak 608 corresponding to the excitation-emission maxima of commercial DOM sensors) and

609 examination of the interacting roles of photobleaching and residence time in 610 defining the extent of fDOM distributions in coastal waters. Because fDOM samples 611 are relatively easy to collect (filtering a few mL of water into glass vials and dark 612 refrigerated storage), are unaffected by gas exchange and quick to analyze (the 613 scans took less than 5 minutes each), fDOM may prove a cost-effective and efficient 614 monitoring tool for mapping groundwater dispersal in reefs. Analyzing a sample 615 EEM spectra also provides a wealth of additional ratio-based indices and values of 616 the literature-derived identified spectral regions and our results hint that with 617 larger datasets from more reefs fDOM may be used as cost-effective monitoring tool 618 to identify new and promising indices to differentiate SGD within coastal waters. 619 620 **ACKNOWLEDGEMENTS** 621 622 The project described in this publication was supported in part by a 623 grant/cooperative agreement from the National Oceanic and Atmospheric 624 Administration, Projects R/SB-14PD (CEN), R/SB-13 (FIMT), R/SB-12 (MJD) and 625 R/SB-11 (HD) sponsored by the University of Hawai'i Sea Grant College Program, 626 School of Ocean and Earth Science and Technology, under Institutional Grant No. 627 NA14OAR4170071 from the NOAA National Sea Grant Office, Department of 628 Commerce (UNIHI-SEAGRANT-JC-14-29). SJG was supported in part by Cooperative 629 Agreement Number G12AC00003 from the United States Geological Survey (USGS). 630 NJS was supported by a NOAA Dr. Nancy Foster Scholarship. CR and KL are 631 supported by the National Science Foundation Graduate Research Fellowship. This 632 is HIMB contribution 1618 and SOEST contribution TBA. The funders had no role in 633 the study design, data collection and analysis, decision to publish, or preparation of 634 the manuscript. The contents of this publication are solely the responsibility of the 635 authors and do not necessarily represent the official views of the USGS, NOAA, or 636 any of their respective subagencies.

FIGURE LEGENDS

Figure 1. Maps of the sampling locations. Panel (a) shows the location of the two sampling sites in Maunalua Bay, O'ahu, with the inset showing the location within the Main Hawaiian Islands. Water samples were collected at each site, Black Point to the West (b) and Wailupe to the East (c), with red markers at each location where water was collected.

Figure 2. Fluorescent dissolved organic matter (fDOM) in the spatial context of Submarine Groundwater Discharge (SGD) in Maunalua Bay. Contour plots of conservative solutes and fDOM solutes at Black Point (a,c,e,g) and Wailupe (b,d,f,h) 28-29 May 2014, including salinity (a,b), silicate concentrations (c,d) (log₁₀ µmol L⁻ ¹), radon concentrations (e,f) (dpm L⁻¹) and the fDOM humification index (g,h) (HIX). Contour gridding and interpolation was done with the kriging function (spherical semivariogram model) in ArcGIS 10.3 Spatial Analyst. All contour plots were generated from samples collected at low tide.

Figure 3. Hierarchical clustering of samples into biogeochemical provinces according to inorganic solute concentrations. Panel (a) is a hierarchical clustering dendrogram grouping samples according to similarity in log₁₀ concentrations of 7 inorganic solutes (shown in heat map with legends at right; X indicates no data). The tips of the dendrogram are colored to match the dendrogram clusters and define biogeochemical provinces by sets of samples with similar chemistry. WL refers to Wailupe and BP refers to Black Point. Panels (b) and (c) illustrate the spatial extent of each biogeochemical province by mapping the sample points color-coded by dendrogram clusters - defined with large colored boxes in panel (a) - at Black Point and Wailupe, respectively. Panels (d) and (e) provide a conceptual illustration of the spatial arrangement of biogeochemical provinces.

Figure 4. Hierarchical clustering of samples according to particulate and dissolved organic matter concentrations. Panel (a) is a hierarchical clustering dendrogram grouping samples (tips colored according to biogeochemical provinces delineated in Fig. 3) according to similarity in log10 concentrations of 8 organic measurements (shown in heat map with legends at right; X indicates no data). Panels (b) and (c) define the spatial extent of organic provinces by mapping the sample points color-coded by dendrogram clusters - defined with large colored boxes in panel (a) - at Black Point and Wailupe, respectively.

Figure 5. Comparison of fDOM PARAFAC components among the biogeochemical provinces in each reef site. Boxes depict standard interquartile ranges with medians and are labeled at top with letters for ANOVA Tukey *post hoc* tests; all ANOVA models p < 0.0001 and samples with different letters are significantly different at α = 0.05.

Figure 6. Comparison of fDOM indices among the biogeochemical provinces in each reef site. Boxes depict standard interquartile ranges with medians and are labeled at top with letters for ANOVA Tukey *post hoc* tests; all ANOVA models p < 0.01 and samples with different letters are significantly different at $\alpha = 0.05$.

Figure 7. Hierarchical clustering of samples according to fDOM spectral

characteristics. Panel (a) is a hierarchical clustering dendrogram grouping samples (tips colored according to biogeochemical provinces delineated in Fig. 3) according to similarity in fDOM PARAFAC components and ratio-based indices (shown in heat map with legends at right; X indicates no data). Panels (b) and (c) define the spatial extent of fDOM character by mapping the sample points color-coded by dendrogram clusters – defined with large colored boxes in panel (a) – at Black Point and Wailupe, respectively.

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		Wailupe				Black Point			
		Spring	Transition	Diffuse	Reef*	Reef	Diffuse	Transition	Spring
Practical Salinity		4.2	24.4	30.4	28.8	30.5	32.6	25.5	7.1
SiO ₄ (µmol L-1)		668.5	225.2	32.2	2.8	3.2	16.6	106.8	626.9
Rn (dpm L-1)		160.4	30.4	23.7	5.6	1.7	14.2	127.2	269.7
N+N (µmol L-1)		61 63	8.50	0.24	0.14	0.23	1.16	21.15	114 16
PO43- (µmol L-1)		1.89	0.49	0.05	0.04	0.10	0.15	0,68	3.09
NH4* (µmol L-1)		0.05	0.89	0.43	0.26	0.34	0.52	0.94	0.22
TA (µmol kg ⁻¹)		1614	2143	2234	2280	2238	2263	2362	2824
DOC (µmol L-1)		20.0	70.8	91.4	83.7	77.3	88.3	88.5	45.6
DON (µmol L-1)		1.4	5.0	6.8	5.7	6.1	6.6	6.6	33.6
Chl a (µg L-1)		0.06	0.29	0.12	0.08	0.07	0.09	0.14	0.03
HBact (cells mL-1)		4.5×10 ⁵	2.3×10 ⁵	5.0×10 ⁵	3.2×10 ⁵	3.7×10 ⁵	2.3×10 ⁵	2.2×10 ⁵	1.0×10 ⁵
PBact (cells mL-1)		1.6×10 ³	7.2×10 ³	1.8×10 ³	4.0×103	1.8×10 ³	1.9×10 ³	1.4×10 ³	7.2×10 ²
PEuks (cells mL-1)		7.7×10 ³	9.2×10 ⁴	6.6×10 ³	6.0×10 ³	4.9×103	5.2×10 ³	6.8×10 ³	5.4×10 ³
	fDOM: A	0.028	0.061	0.056	0.025	0.029	0.037	0.052	0.075
log ratio to Ambient -1 -0.6 -0.2 0.2 0.6 1 1.4 1.4 1.8 2.2 2.6 3	fDOM: M	0.036	0.073	0.065	0.029	0.034	0.043	0.062	0.091
	fDOM: C	0.006	0.018	0.018	0.007	0.009	0.011	0.015	0.017
	fDOM: T	0.008	0.026	0.026	0.016	0.017	0.021	0.027	0.023
	SUVA ₂₅₄	0.95	0.87	0.82	0.65	0.98	0.72	0.88	1.44
	M:C	1.25	1.11	1.05	1.02	1.01	1.03	1.07	1.20
	FI	1.82	1.72	1.73	1.73	1.79	1.75	1.75	1.76
	BIX	0.83	0.79	0.82	0.80	0.84	0.83	0.82	0.85
	HIX	7.78	4.28	3.60	2.35	2.74	3.00	3.47	6.98

Table 1. Mean values of each parameter in each of the biogeochemical provinces shaded according to magnitude of significant differences among regions. Rows are parameters, columns are biogeochemical provinces, and values are geometric means. Shaded cells are significantly different from the Ambient Reef (WL) province (denoted at top by *; Dunnet's *post hoc* test), with color and intensity scaled by mean log-ratio relative to Ambient Reef waters (legend at left). POC, PON and DOP are excluded due to a lack of data to properly test each province.















