

1 **Chemical and Biological Impacts of Ocean Acidification Along the West Coast**
2 **of North America**

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25 **Anthropogenic CO₂, Upwelling, Pteropod Dissolution**

26 **Abstract**

27 The continental shelf region off the west coast of North America is seasonally exposed to
28 water with a low aragonite saturation state by coastal upwelling of CO₂-rich waters. To
29 date, the spatial and temporal distribution of anthropogenic CO₂ (C_{anth}) within the CO₂-
30 rich waters is largely unknown. Here we adapt the multiple linear regression approach to
31 utilize the GO-SHIP Repeat Hydrography data from the northeast Pacific to establish an
32 annually updated relationship between C_{anth} and potential density. This relationship was
33 then used with the NOAA Ocean Acidification Program West Coast Ocean Acidification
34 (WCOA) cruise data sets from 2007, 2011, 2012, and 2013 to determine the spatial
35 variations of C_{anth} in the upwelled water. Our results show large spatial differences in
36 C_{anth} in surface waters along the coast, with the lowest values (37–55 μmol kg⁻¹) in strong
37 upwelling regions off southern Oregon and northern California and higher values (51–63
38 μmol kg⁻¹) to the north and south of this region. Coastal dissolved inorganic carbon
39 concentrations are also elevated due to a natural remineralized component (C_{bio}), which
40 represents carbon accumulated through net respiration in the seawater that has not yet
41 degassed to the atmosphere. Average surface C_{anth} is almost twice the surface
42 remineralized component. In contrast, C_{anth} is only about one third and one fifth of the
43 remineralized component at 50 m and 100 m depth, respectively. Uptake of C_{anth} has
44 caused the aragonite saturation horizon to shoal by approximately 30–50 m since the
45 preindustrial period so that undersaturated waters are well within the regions of the
46 continental shelf that affect the shell dissolution of living pteropods. Our data show that

47 the most severe biological impacts occur in the nearshore waters, where corrosive waters
48 are closest to the surface. Since the pre-industrial times, pteropod shell dissolution has, on
49 average, increased approximately 20–25% in both nearshore and offshore waters.

50

51 **1. Introduction**

52 Since the beginning of the Industrial Revolution, the global oceans have absorbed about
53 28% (~550 billion tons) of the total anthropogenic carbon dioxide (CO₂) emissions
54 (Canadell et al., 2007; IPCC, 2013). This absorption of atmospheric CO₂ has increased
55 ocean acidity in a process referred to as “anthropogenic” ocean acidification (OA). Over
56 the past 250 years, the pH of open-ocean surface waters has decreased by approximately
57 0.11 units, equivalent to an increase of about 28% in hydrogen ion concentration (Gattuso
58 et al., 2015). When CO₂ enters the ocean, it reacts with water to form carbonic acid,
59 which consumes carbonate ions (CO₃²⁻) via the release of protons. In direct
60 correspondence with these changes, the CO₃²⁻ concentration has declined about 16%
61 from preindustrial values through the year 2000. By the end of this century, surface ocean
62 pH is expected to decline by another 0.3–0.4 units, and CO₃²⁻ concentration is expected to
63 decline by ~50% (Feely et al., 2004, 2009; Orr et al., 2005; Doney et al., 2009a,b;
64 Gattuso et al., 2015).

65

66 Organisms that produce calcium carbonate (CaCO₃) shells or skeletons made of aragonite
67 or calcite are expected to encounter increasing physiological challenges as the saturation
68 state of aragonite and calcite decreases due to OA (Fabry et al., 2008; Guinotte and
69 Fabry, 2008; Hofmann and Todgham, 2010; Gaylord et al., 2011; Barton et al., 2012;

70 Bednaršek et al., 2012a, 2014a,b; Hettinger et al., 2012; Frieder et al., 2014; Gattuso et
71 al., 2015; Waldbusser et al., 2015; Somero et al., 2016). The saturation state of aragonite
72 (Ω_{ar}) and calcite (Ω_{cal}) is a function of the concentrations of calcium (Ca^{2+}) and CO_3^{2-} ,
73 and pressure-dependent stoichiometric solubility product, K_{sp}^* : ($\Omega = [\text{Ca}^{2+}][\text{CO}_3^{2-}] / K_{\text{sp}}^*$)
74 (Mucci, 1983), such that Ω_{ar} and Ω_{cal} will decline as more CO_2 is taken up by the oceans.
75 At $\Omega=1$, carbonate minerals are in equilibrium with surrounding seawater; at $\Omega>1$,
76 precipitation or preservation of carbonate minerals is thermodynamically favored; and at
77 $\Omega<1$, dissolution is favored.

78

79 Recent models suggest that the shallower waters along the California Current Large
80 Marine System (CCLME) will become undersaturated more often and for longer
81 durations over the next several decades to a century (Gruber et al. 2012; Hauri et al.,
82 2013; Turi et al., 2016). Persistence of acidified water in the coastal waters of the west
83 coast of North America could have profound consequences for marine organisms,
84 ecosystems, and the ecosystem services of this region (Doney et al., 2009a; Gattuso and
85 Hansson, 2011; Feely et al., 2012a; Ekstrom et al., 2015; Gaylord et al., 2015; Somero et
86 al., 2016). Increasing CO_2 may have significant biological and ecological effects, with
87 potential feedbacks to biogeochemical cycles. Declines in CaCO_3 saturation state,
88 particularly Ω_{ar} , will pose increasing physiological challenges to calcifying invertebrates
89 such as pteropods, bivalves, and echinoderms (Wootton et al., 2008; Hettinger et al.,
90 2012; Kroeker et al., 2013; Frieder et al., 2014; Bednaršek et al., 2012a; 2014a,b;
91 Waldbusser et al., 2015; Barton et al., 2015; Somero et al., 2016).

92 Pteropods are an important food source for organisms across lower
93 (e.g. macrozooplankton) and higher trophic levels in the oceans. In the North Pacific
94 Ocean, pteropods are seasonally substantial portion of the diets of pink and chum salmon
95 (Groot and Margolis, 1991), sablefish and rock sole (Armstrong et al., 2005; Aydin et al.,
96 2005). Moreover, they are among the species most affected by ocean acidification, with
97 shell dissolution already occurring in the natural environment (Bednaršek et al., 2014a).
98 Consequently, pteropods are ideal sentinel organisms to study how the changes since the
99 pre-industrial times are affecting aragonite dissolution in the CCLME, and help to
100 identify which of the regions are the most vulnerable to the anthropogenic changes. In
101 this paper we estimate the amount of anthropogenic CO₂ (C_{anth}) in the CCLME region
102 and determine its impact on pteropod shell dissolution comparatively for cruises in 2011
103 and 2013.

104

105 **1.1. Physical and biogeochemical setting**

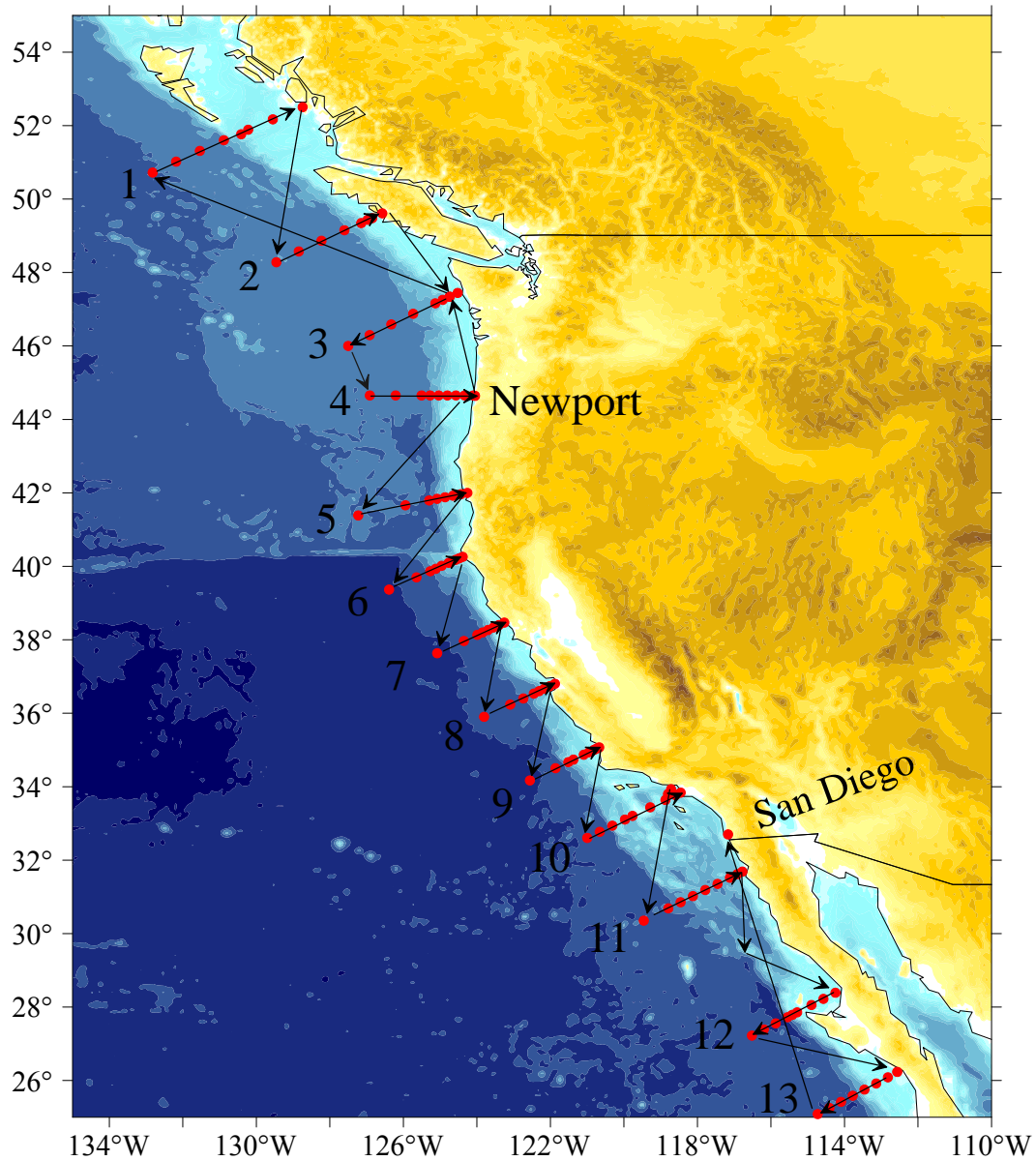
106 The CCLME is a large-scale oceanographic feature along the west coast of North
107 America, an eastern boundary current extending from northern Vancouver Island in
108 Canada to Punta Eugenia in Mexico, and landward into large estuarine systems such as
109 the San Francisco Bay and the Salish Sea (Fig. 1). The coastal waters off the west coast
110 of North America are strongly affected by seasonal upwelling, which typically begins in
111 early spring when the Pacific High moves into the subarctic North Pacific, resulting in a
112 strengthening of the northwesterly winds. These winds drive net surface waters offshore
113 via Ekman transport, which induces the upwelling of low pH, nutrient- and CO₂-rich,
114 intermediate depth (100 to 300 m) offshore waters onto the continental shelf (Hales et al.,

115 2006; Feely et al., 2008; Gruber et al., 2012; Harris et al., 2013; Hauri et al., 2013; Turi et
116 al., 2016). The upwelling lasts from spring to early or late fall, when winter storms return.
117 Within the CCLME, the upwelling supports highly productive communities and fisheries
118 on the continental shelf and slope, and in the estuaries (Hickey, 1979; Thomson et al.,
119 1989; Thomson and Krassovski, 2010). Thus, while upwelling plays a defining role in
120 CCLME biogeochemistry, productivity, and ecology, it also contributes to the impacts of
121 local and regional oceanographic processes that exacerbate the effects of anthropogenic
122 OA. Here we use the term “corrosive” to refer to waters that are undersaturated with
123 respect to aragonite ($\Omega_{ar}<1$), a condition that results from some combination of: 1)
124 oceanic uptake of anthropogenic CO₂, and 2) build-up of CO₂ from the natural respiration
125 processes in the ocean interior (C_{bio}) that occur in offshore waters prior to upwelling or
126 on the continental shelf after those interior waters have upwelled. These processes are
127 already affecting coastal regions such that corrosive waters have previously been
128 observed in large coastal regions including Arctic and Alaskan coastal waters, as well as
129 the CCLME (Feely et al., 2008; Bates et al., 2013; Mathis et al., 2014a,b, 2015).

130

131 Many of the ecosystems within the CCLME are particularly vulnerable because of the
132 combined effects of acidification, warming, upwelling, and hypoxia, which are expected
133 to increase under anthropogenic climate change (Rykaczewski and Dunne, 2010; Somero
134 et al., 2016). The term “hypoxia” implies diminished levels of oxygenation under which
135 many species of fish and invertebrates are negatively impacted. Conditions ranging from
136 hypoxic ($<65 \mu\text{mol kg}^{-1}$) to anoxic ($0 \mu\text{mol kg}^{-1}$) have been observed in near-bottom
137 waters on the inner continental shelf within the CCLME, particularly in the late summer

138 and early fall months when respiration-induced oxygen depletions are at their maximum
139 extent (Grantham et al., 2004; Hales et al., 2006; Chan et al., 2008; Booth et al., 2012;
140 Siedlecki et al., 2016). High CO₂ concentrations and hypoxia are linked mechanistically
141 because aerobic respiration of organic matter consumes oxygen and produces CO₂ in
142 approximate stoichiometric equivalence (170:117) (Anderson and Sarmiento, 1994).
143 Thus, processes that create aquatic oxygen deficits can also exacerbate corrosive
144 conditions for calcareous organisms.



145

146 Figure 1. Map of the station locations for the 2007 West Coast cruise. The black line

147 shows the cruise track. The 2011, 2012, and 2013 cruises included subsets of these

148 stations and, in some cases, a few additional stations.

149

150 **2. Analytical methods**

151 **2.1. Chemical methods**

152 In the late spring of 2007 and late summers of 2011, 2012, and 2013 we conducted
153 detailed observations of carbonate system chemistry and other physical, chemical, and
154 biological parameters along the western North American continental shelf, both via ship-
155 based cruises and shore-based sampling (Fig.1). Water samples from the cruises were
156 collected in modified Niskin-type bottles and analyzed under laboratory conditions for
157 dissolved inorganic carbon (DIC), total alkalinity (TA), oxygen, and nutrients. During the
158 cruises in 2011 and 2013, samples were also measured directly for pH_T . DIC was
159 analyzed using coulometric titration (Johnson et al., 1987; DOE, 1994; Ono et al., 1998).
160 TA was measured by the potentiometric titration method (Millero et al., 1993; DOE,
161 1994; Ono et al., 1998). Certified Reference Materials were analyzed with both the DIC
162 and TA samples as an independent verification of instrument calibrations (Dickson et al.,
163 2007). The ship-based DIC and TA data are both precise and accurate to within $2 \mu\text{mol}$
164 kg^{-1} . The spectrophotometric method described in Byrne et al. (2010) and Liu et al.
165 (2011) was used to measure pH on the total scale (pH_T) for the 2011 and 2013 cruises.
166 Shore-based measurements of pH_T from *in-situ* sensors and DIC and TA from discrete
167 samples were also provided through the OMEGAS (8 sites) and UC Davis Coastal
168 Transect (47 sites) projects, respectively. *In-situ* records were collected using Durafet®-
169 based sensors that were calibrated against seawater and/or TRIS-based Certified
170 Reference Materials. Bottle samples were analyzed for DIC (via infrared CO_2 ; Monterey
171 Bay Aquarium Research Institute) and TA (Metrohm 855 autotitrator), and were cross-

172 verified with pH determined spectrophotometrically for pH, using the total pH scale. The
173 saturation state of seawater with respect to aragonite was calculated from the DIC and TA
174 data using the program CO2SYS developed by Lewis and Wallace (1998), using the
175 Lueker et al. (2000) carbonate constants, Dickson (1990) for the KSO₄, and Lee et al.
176 (2010) for total boron. The pressure effect on the solubility, for samples collected at
177 depth, is estimated from the equation of Mucci (1983), incorporating adjustments to the
178 constants recommended by Millero (1995). Based on the uncertainties in the DIC and TA
179 measurements and the thermodynamic constants, the uncertainty in the calculated Ω_{ar} is
180 approximately 0.02. Oxygen analysis was conducted by modified Winkler titration
181 (Carpenter, 1965), and nutrients (nitrate, nitrite, ammonium, phosphate, silicate) were
182 frozen at sea and analyzed using a Technicon AutoAnalyzer II (UNESCO, 1994) at
183 Oregon State University.

184

185 **2.2. Pteropod shell dissolution**

186 Pteropod shell dissolution was determined on shells collected from 16 stations for the
187 2011 cruise and 20 stations during the 2013 cruise. The samples were stored in 90%
188 buffered ethanol. Between 15 to 30 pteropods of *Limacina helicina* were blindly picked
189 from samples selected randomly with no prior knowledge of station location or carbonate
190 chemistry conditions. Following the methods described in Bednaršek et al. (2012c), the
191 shells were repeatedly washed with distilled water before being subjected to chemical
192 shell dehydration, followed by a plasma etching procedure for periostracum removal. All
193 treated shells were analyzed for shell dissolution using a scanning electron microscope
194 (SEM) and identified for the presence of dissolution patterns and the proportion of more

195 severe types of shell dissolution (Type II and Type III). Following the categorization
196 scheme outlined in Bednaršek et al. (2012c), Type II dissolution indicates deeper
197 penetrating dissolution that precedes Type III, which affects large parts of shell
198 crystalline structure, making shells less compact and more fragile.

199

200 **2.3. Estimating coastal C_{anth} and C_{bio}**

201 Seawater upwelling along the continental shelf of the west coast of North America comes
202 from the thermocline waters of the North Pacific subtropical and subarctic gyres. We
203 therefore used the gyre thermocline C_{anth} , estimated by Carter et al. (submitted)
204 employing the methods outlined in Supplementary Materials section SM1.1 (this paper) -
205 to estimate upwelling water C_{anth} for the years 2007, 2011, 2012, and 2013. This method
206 is similar to the approach for used by Feely et al. (2008). Our approach for estimating
207 coastal C_{anth} and biological remineralization C (C_{bio}) involves the following steps:

- 208 1. Open ocean C_{anth} estimates are used to derive polynomials relating thermocline C_{anth}
209 to potential density σ_{θ} for both 2004 and 2013 along P02 (two polynomials) and in
210 2006 and 2015 along P16N (two additional polynomials). See Supplementary
211 Materials section SM1.2 for details on this step.
- 212 2. A grid of seawater properties shoreward of the 200 m depth isobath is determined
213 along the West Coast from our hydrographic surveys in 2007, 2011, 2012, and 2013
214 using the procedure detailed in the Supplementary Materials section SM1.3.
- 215 3. The four polynomials determined in step 1 are used with the σ_{θ} estimates determined
216 in step 2 to estimate C_{anth} for all gridded locations.

217 4. We interpolate among the four estimates from step 3 to obtain sets of estimates
218 specific to the 4 years of interest (2007, 2011, 2012, and 2013) at each location. We
219 interpolate among the four polynomials both by date to select between the earlier and
220 later polynomials for each section, and by gridded seawater spiciness to select
221 between the P02 and P16 polynomials.

222 5. We directly estimate C_{bio} , or the amount of additional DIC present as a result of
223 organic matter remineralization, from seawater properties using methods described in
224 detail in Supplementary Materials SM1.2.

225 These gridded properties are used for volume-weighted seawater average properties.
226 Also in SM1.2, the uncertainties in these quantities are estimated to be of order $\pm (1\sigma) 10$
227 $\mu\text{mol kg}^{-1}$, yielding a 95% confidence interval of $\sim 20 \mu\text{mol kg}^{-1}$. We refer to the sum of
228 C_{anth} and C_{bio} as “enriched DIC.”

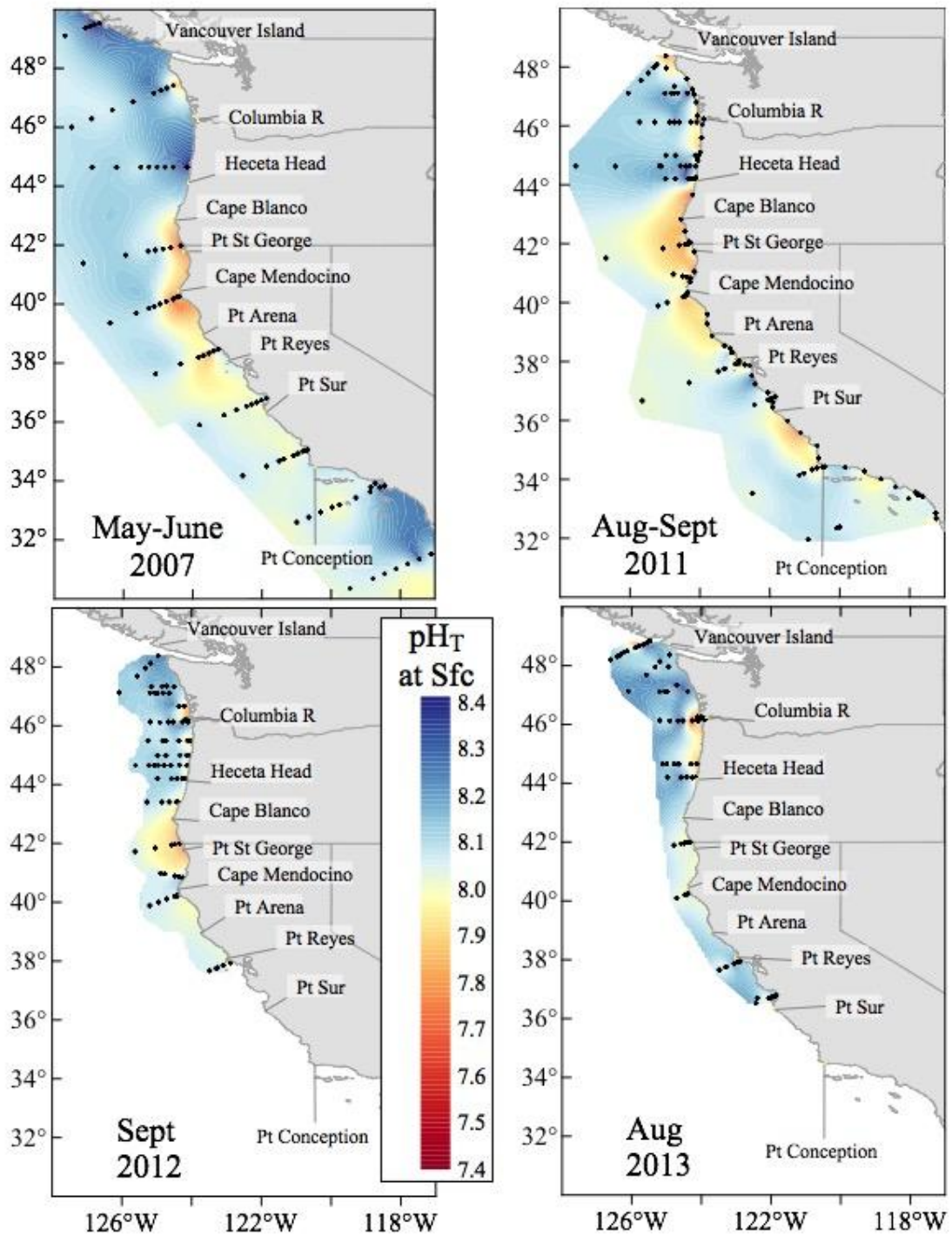
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230 **3. Results**

231 **3.1. Coastal distributions of acidified water**

232 During the four cruises, various stages and strengths of upwelling were observed from
233 central Vancouver Island, Canada to Baja California, Mexico. The observations revealed
234 that, on average, acidified, corrosive CO_2 -rich waters (*insitu* $\text{pH}_T < 7.75$; $\Omega_{\text{ar}} < 1.0$; $\text{DIC} >$
235 $2190 \mu\text{mol kg}^{-1}$) were upwelled from depths of 150–250 m to depths as shallow as 20–
236 200 m in most areas and close to the surface in the region between northern California
237 near Cape Mendocino to Heceta Head, Oregon (Figs. 2–4). Maps of surface ocean pH_T
238 and DIC during the four West Coast survey cruises show that *insitu* pH_T values ranged
239 from 7.7 to 8.3, with the lowest pH_T values and highest DIC concentrations occurring in

240 the upwelled water near the coast (Figs. 2 and 3). Moving offshore, pH_T values quickly
241 increase to open-ocean values ranging from 8.0 to 8.3. The 2011 pH_T map includes
242 complementary shore-based nearshore and intertidal pH_T data from the same period,
243 collected using Durafet-style autonomous sensors (Fig. 2), which reinforces the notion
244 that the greatest spatial variability of pH_T appears in closest proximity to the shore (Chan
245 et al., submitted). The excellent consistency among the intertidal, nearshore, and offshore
246 data suggests that the uptake of anthropogenic CO_2 , upwelling/mixing, and respiration
247 processes are the primary drivers of pH_T distributions along the coast. Our results for the
248 four cruises follow the seasonal patterns described by Chan et al. (submitted) from field
249 data. Consistent with those results, Turi et al (2016) found similar patterns in their
250 hindcast biogeochemical model outputs, with higher pH values in the spring and lower
251 pH values in the late summer. One exception is the low pH_T , high O_2 , low Ω_{ar} values in
252 surface waters immediately seaward of the Columbia River Estuary in 2011, 2012, and
253 2013, which were dominated by the outflow of low salinity, low alkalinity, and high DIC
254 riverine water in the surface layer (Evans et al., 2013).



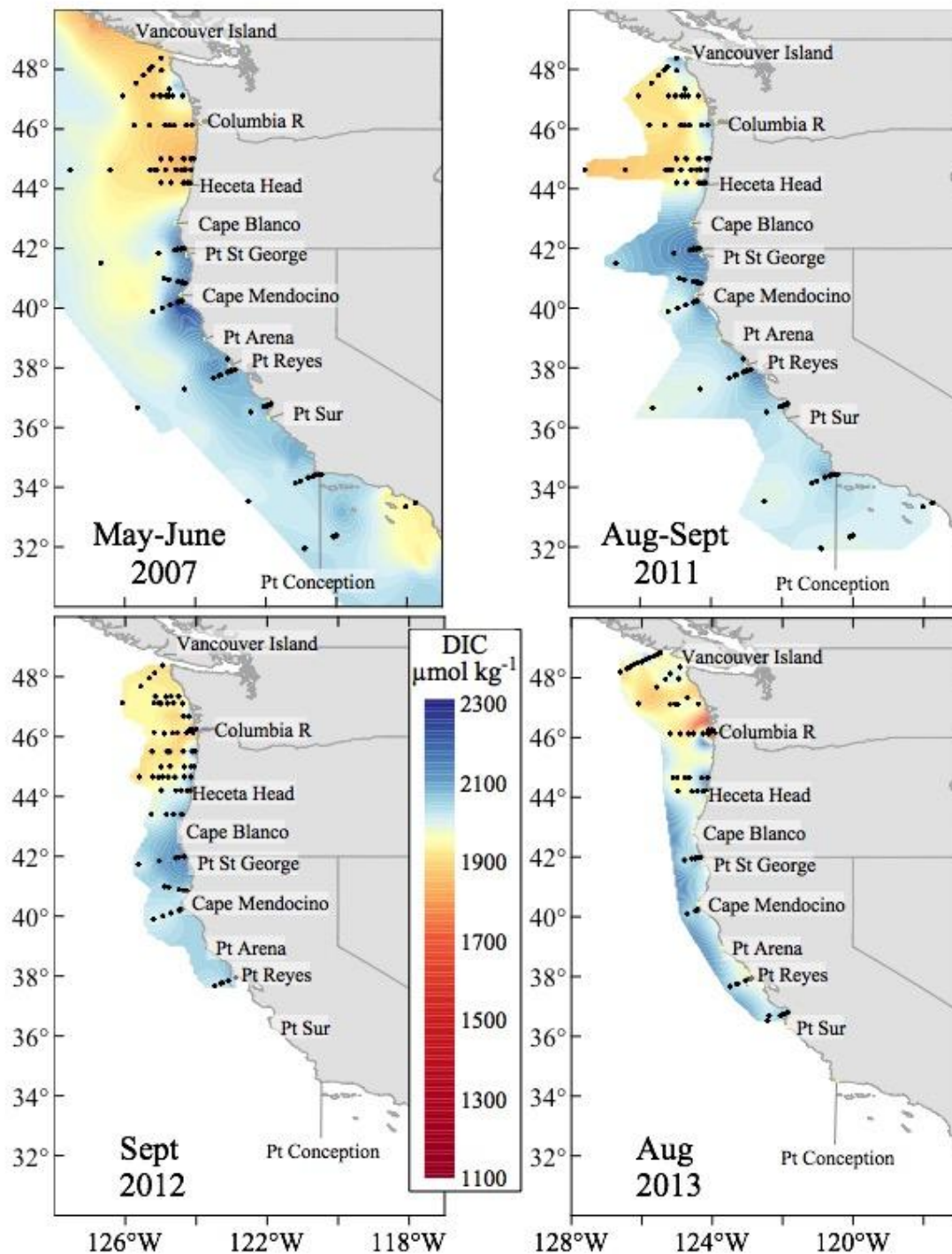
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256 Figure 2. Maps of surface ocean pH_T values for the 2007, 2011, 2012, and 2013 cruises.

257 The 2011 map includes the shore-based intertidal data.

258

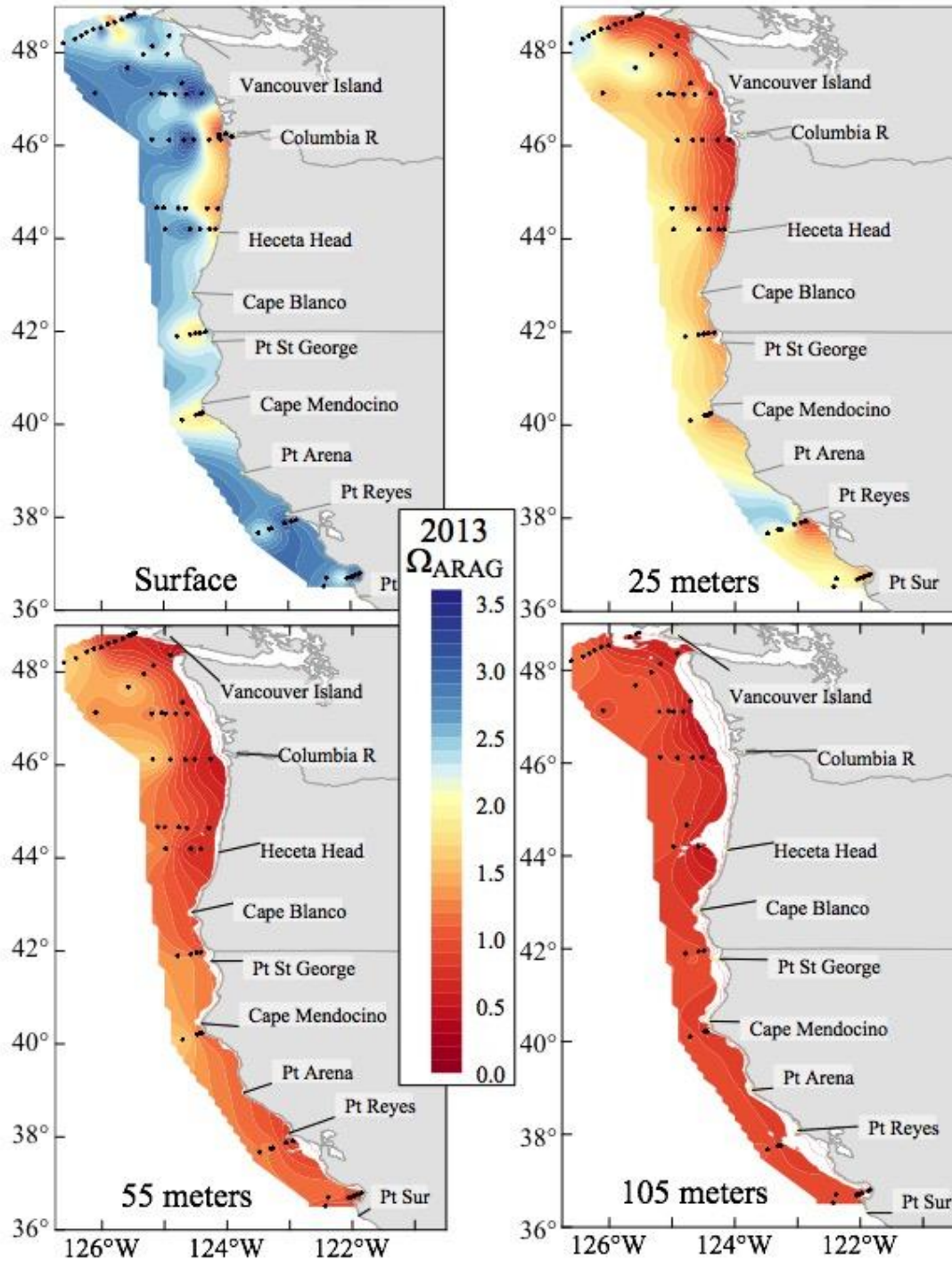
259 The corrosive waters along the inner- and mid-shelf regions were due to the combined
260 impacts of anthropogenic CO₂ uptake and upwelling of respiration-enriched CO₂ waters
261 along the coast (Figs. 4 and 5). Nearshore upwelled waters were characterized by low-pH
262 seawater (pH <7.75) with Ω_{ar} values near or below 1.0 and potential density >26.0 kg m³.
263 In 2013, for example, along Line 6 offshore of Newport, Oregon, the 26.1 kg m³ potential
264 density surface shoaled from a depth range of 150–200 m offshore to the surface near the
265 coast (Fig. 5). This density surface was co-located with isolines of $\Omega_{\text{ar}} = 1.0$, DIC = 2190
266 $\mu\text{mol kg}^{-1}$, and pH = 7.75. However, pH decreased, and DIC and the partial pressure of
267 CO₂ (pCO₂) increased shoreward in the region surrounding this isopycnal due to CO₂
268 release from local remineralization of organic matter. Upwelling of CO₂-enriched
269 seawater caused the entire water column shoreward of the 50 m isobath along Line 6 to
270 become undersaturated with respect to aragonite (Fig 5D). The lowest Ω_{ar} values (<0.70)
271 found shoreward of the 200 m isobath were observed in the near-bottom waters of the
272 mid-shelf region where respiration provides an additional CO₂ contribution that decreases
273 Ω_{ar} . The uptake of anthropogenic CO₂ has caused the corrosive ($\Omega_{\text{ar}} < 1$) waters to shoal
274 by about 30–50 m since preindustrial times such that they are within the density layers
275 that are currently being upwelled along the west coast of North America (Feely et al.,
276 2012b).
277



278

279 Figure 3. Maps of surface DIC concentrations in $\mu\text{mol kg}^{-1}$ for the 2007, 2011, 2012, and
 280 2013 cruises. The nearshore upwelling regions are delineated by DIC concentrations in
 281 excess of $2050 \mu\text{mol kg}^{-1}$. Black dots indicate measurement locations. Open circles on the

282 2011 and 2013 cruises indicate stations where both chemical and biological samples were
283 taken.

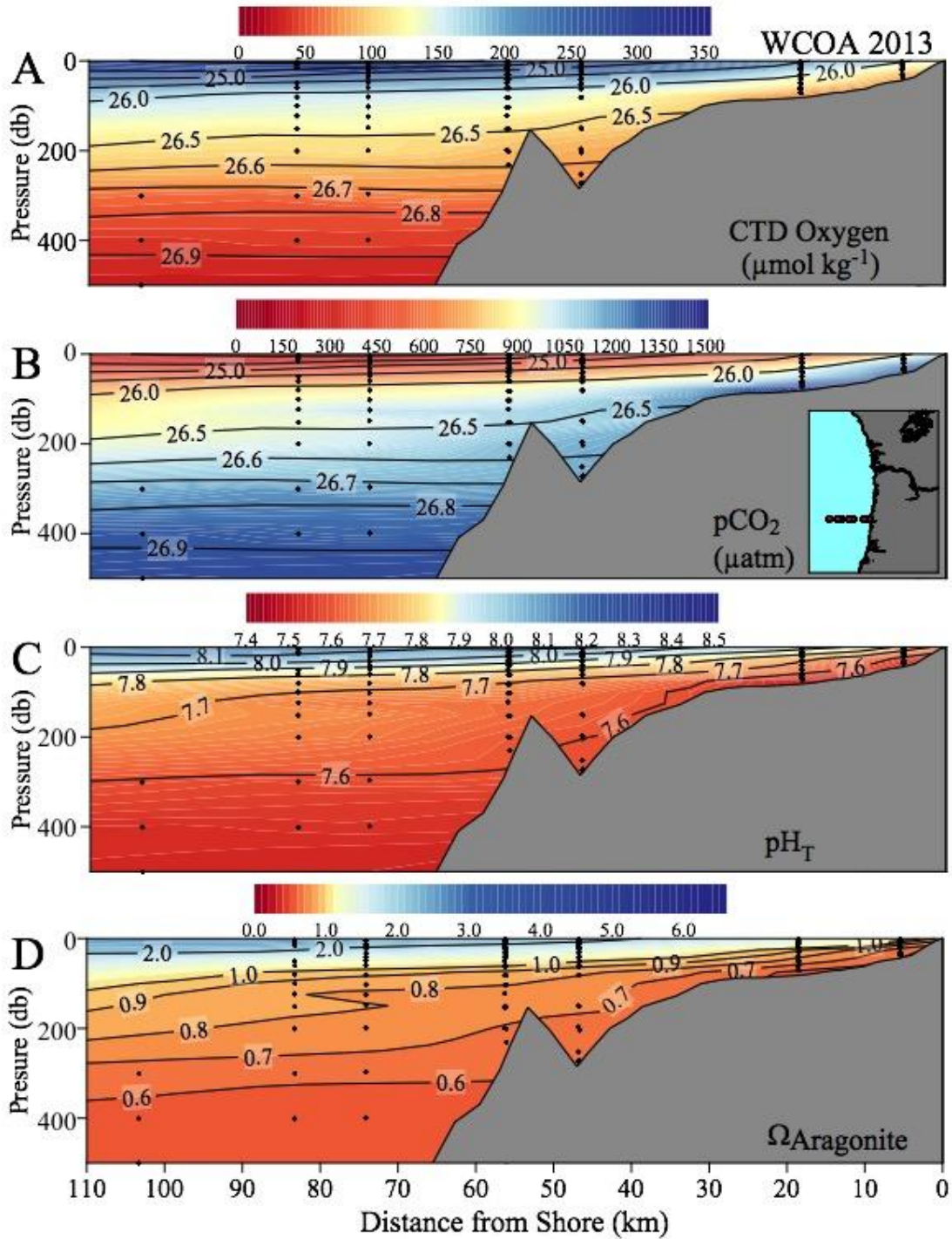


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285 Figure 4. Aragonite saturation state at the surface, 25 m, 55 m, and 105 m depth during

286 the 2013 West Coast survey.

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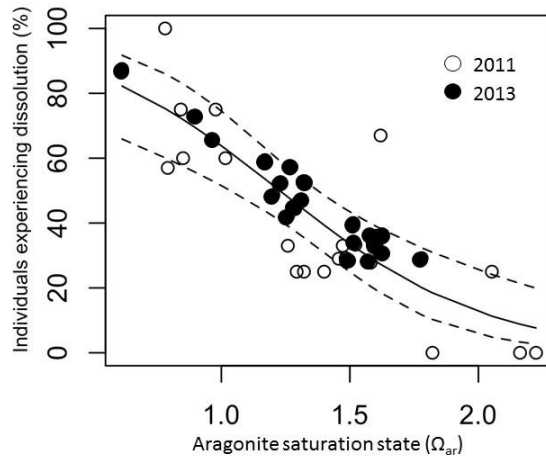
289 Figure 5. Vertical sections of: (A) dissolved oxygen, (B) pCO_2 , (C) pH_T , and (D) Ω_{ar}

290 along the 2013 Line 6 stations off Newport, OR. Black dots indicate measurement
291 locations and the isolines lines in (A) and (B) show the potential density in kg m^{-3} .

292

293 **3.2. Pteropod dissolution and water chemistry**

294 The water column hydrographic data were combined with the chemical data for the
295 nearshore and offshore regions and the aragonite saturation state (Ω_{ar}) was calculated for
296 the upper 55 m or 100 m of the water column in the nearshore and offshore, respectively.
297 Diel vertical migration of *L. helicina* is within this depth range. The values from the
298 region off Southern California were not taken into account, as we did not have pteropod
299 dissolution data for that region. There was a strong negative linear correlation between
300 the percentage of pteropods with Type II and Type III dissolution shell impacts and Ω_{ar} in
301 2011 and 2013 (Fig. 6, $R^2=0.74$, $p< 0.001$). We have fitted the combined data (2011 and
302 2013) to a logarithmic function and generated the equation: $y = -66.29 \ln x + 61.21$ ($R^2 =$
303 0.74). This relationship was used for estimating the percentage of individuals with severe
304 dissolution for both pre-industrial and current Ω_{ar} values (Table 2).



305

306 Figure 6. Percentage of individuals affected by severe dissolution as a function of
 307 aragonite saturation state (integrated over the upper 100 m) for the 2011 (open circles)
 308 and 2013 (closed circles) data. The dashed lines show the 95% confidence interval for the
 309 logarithmic function.

310

311 During both 2011 and 2013 cruises, pteropod shell dissolution was observed to be
 312 significantly higher in the nearshore region of the CCLME. Currently, on average 57%
 313 of individuals are affected by dissolution in the nearshore regions, but only 36% in the
 314 offshore region (Table 2). This greater incidence of dissolution-affected individuals is
 315 consistent with lower aragonite saturation state in the nearshore region (average $\Omega_{ar} =$
 316 1.07) compared to the offshore region (average $\Omega_{ar} = 1.47$).

317

318 4. Discussion

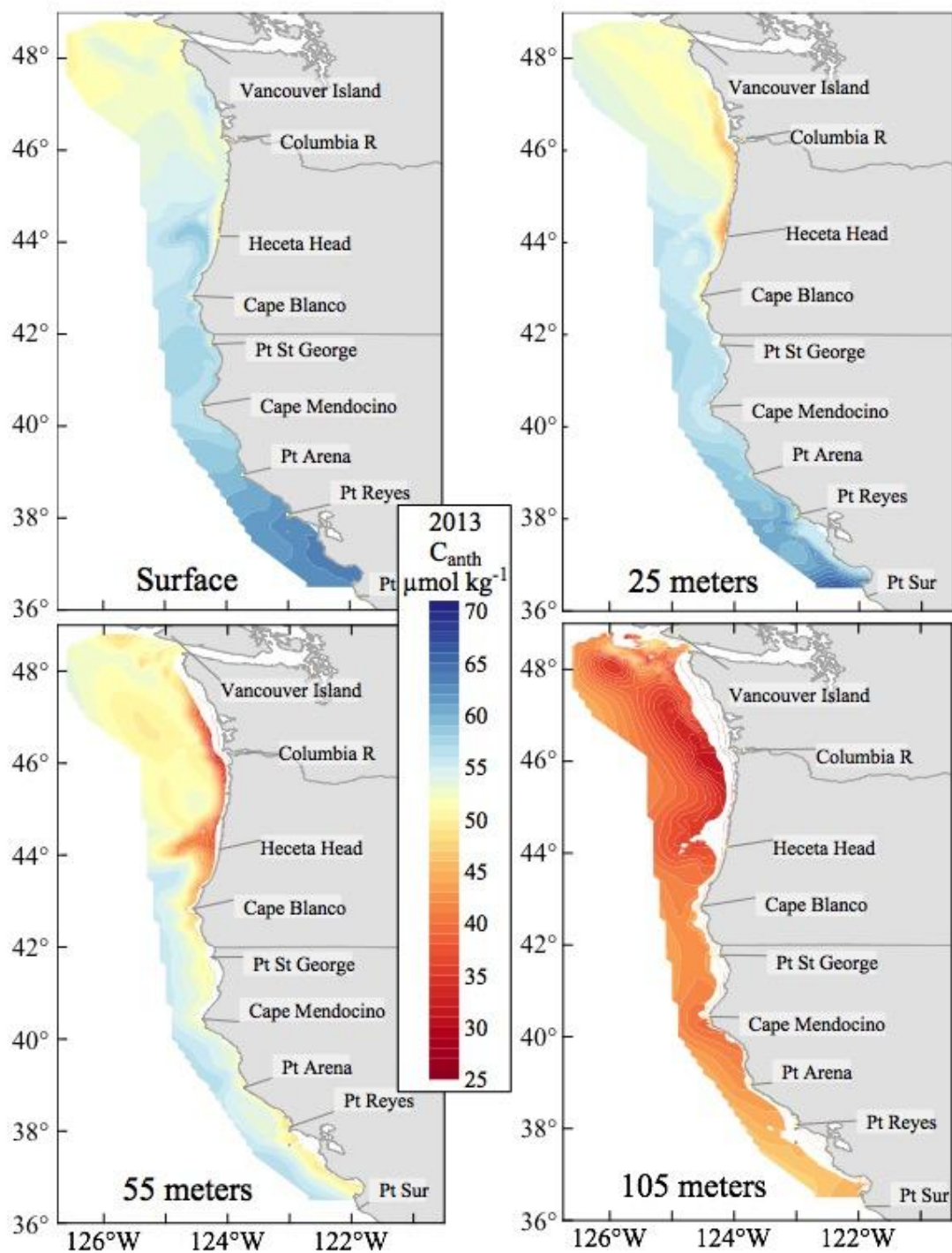
319 4.1. Estimates of C_{anth} and C_{bio} in the CCLME

320 Because the increased DIC concentrations along the coast are the result of uptake of C_{anth}
321 and upwelling of CO_2 -rich respired CO_2 (C_{bio}) waters from below we have estimated the
322 contributions of both C_{anth} and C_{bio} throughout the water column. Our estimates of the
323 distribution of C_{anth} from the coast out to the open-ocean for 2013 are presented as maps
324 for surface, 25, 55, and 105 m (Fig. 7), and a summary of the regional averages are given
325 in Table 1. In nearshore surface waters, C_{anth} ranges from about 37 to 60 $\mu\text{mol kg}^{-1}$, with
326 increasing concentrations north and south of the region near Cape Blanco. The lowest
327 C_{anth} concentrations (ranging from 37 to 55 $\mu\text{mol kg}^{-1}$) are centered near the strong
328 upwelling center between the region south of the Columbia River to Cape Mendocino. To
329 the north and south of this region nearshore C_{anth} concentrations are somewhat higher,
330 indicating mixing of the upwelled water with water that has been in recent contact with
331 the atmosphere. The highest C_{anth} concentrations (ranging from 44 to 63 $\mu\text{mol kg}^{-1}$) are
332 located in the offshore surface waters. At 25 m in the nearshore region, the influence of
333 the upwelled water is more pronounced, with C_{anth} concentrations ranging from 33 to 55
334 $\mu\text{mol kg}^{-1}$ along most of the coastline. At deeper nearshore depths, C_{anth} ranges from 33
335 to 56 $\mu\text{mol kg}^{-1}$ at 55 m and from 29 to 53 $\mu\text{mol kg}^{-1}$ at 105 m.

336

337 Average C_{anth} and C_{bio} concentrations are shown in Figure 8 and a summary of the
338 regional and depth averages for C_{anth} and C_{bio} is given in Table 1. For comparison, Table
339 1 also shows enriched carbon contributions found at the surface and at 200 m depth at the
340 most northwestern station (i.e., most offshore) within each study region. In offshore
341 surface waters, nearly all of the enriched DIC ($C_{\text{anth}} + C_{\text{bio}}$) is from C_{anth} , whereas at 200 m
342 only about 19% of the enriched DIC is from C_{anth} and the remainder is from C_{bio} .

343 In the nearshore region of the CCLME, enriched DIC in surface waters ranged from 41 to
344 $148 \mu\text{mol kg}^{-1}$, with an average of about 65% of enriched DIC in the surface waters due
345 to C_{anth} and the remainder due to C_{bio} . Enriched DIC at 50 m is larger than at the surface
346 (range: $129 - 172 \mu\text{mol kg}^{-1}$), but the percentage due to C_{anth} is lower ($\sim 32\%$). Finally, at
347 100 m, only about 22% of the enriched DIC is due to C_{anth} . There is some year-to-year
348 variability within the regions but the highest contributions of C_{anth} and total enriched-DIC
349 generally occur in the later years. While the percentage of C_{anth} in the nearshore upwelled
350 water is lower than the surrounding water, the total amount of enriched DIC is highest in
351 the nearshore upwelled water and, consequently, those nearshore upwelled waters are the
352 most corrosive to calcifying organisms. In subsurface waters, the most corrosive
353 conditions occur in the onshore bottom waters within 20 km of the coast. The uptake of
354 C_{anth} has caused the aragonite saturation horizon to shoal by approximately 30–50 m
355 since the preindustrial period so that undersaturated waters are well within the regions of
356 the continental shelf that affect the shell dissolution of living pteropods (Feely et al.,
357 2008).
358
359



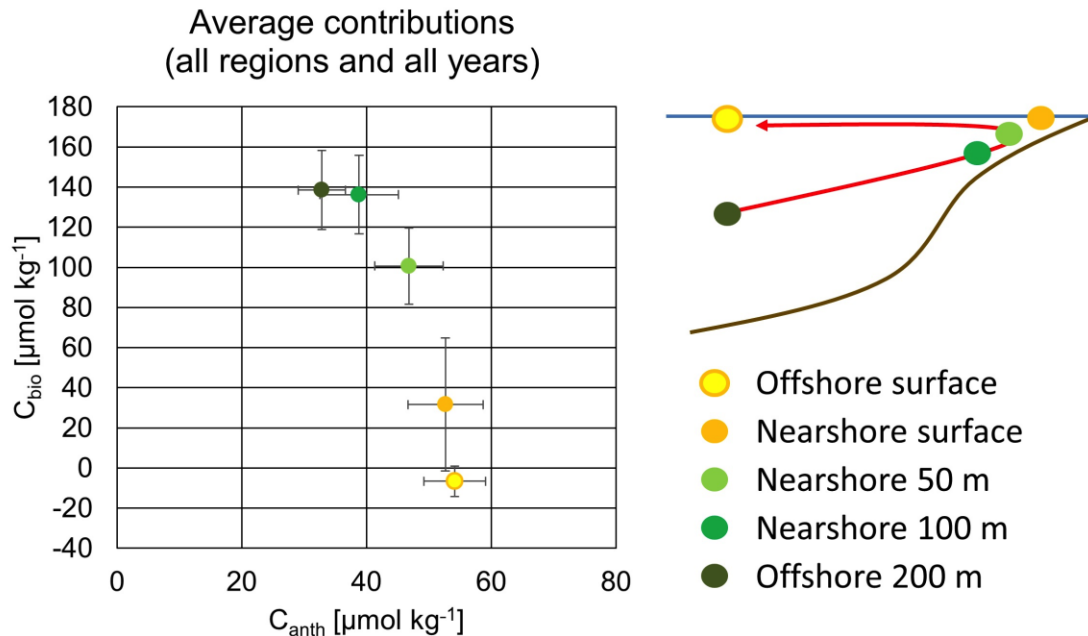
360

361 Figure 7. Distribution of C_{anth} in $\mu\text{mol kg}^{-1}$ at the surface, 25 m, 55 m, and 105 m depth

362 for the 2013 West Coast survey.

Table 1. Anthropogenic carbon (C_{anth}), remineralized carbon (C_{bio}) and anthropogenic percentage of total enriched carbon ($\%C_{\text{anth}}$) by region (W: Washington, O: Oregon, NC: Northern California, SC: Southern California), cruise year, and depth for averages of gridded coastal properties shoreward of the 200 m isobath (left columns) and for the northwestern stations occupied offshore of each region (right columns). All values are expressed in $\mu\text{mol kg}^{-1}$. Estimated average uncertainty is approximately $\pm 10 \mu\text{mol kg}^{-1}$ (1σ). Column averages are calculated weighting all regions and years equally. Negative C_{bio} values suggest either net autotrophy or physically derived oxygen supersaturation.

		<i>Grid average shoreward of 200 m isobath</i>									<i>Northwestern station in region</i>					
Depths		0-10 m			50-60 m			100-110 m			Surface			200 m		
State	Year	C_{anth}	C_{bio}	$\%C_{\text{anth}}$	C_{anth}	C_{bio}	$\%C_{\text{anth}}$	C_{anth}	C_{bio}	$\%C_{\text{anth}}$	C_{anth}	C_{bio}	$\%C_{\text{anth}}$	C_{anth}	C_{bio}	$\%C_{\text{anth}}$
W	2007	47	-27	232	47	85	35	34	131	21	47	-5	113	30	140	18
W	2011	51	97	34	47	125	27	36	149	20	54	0	101	33	133	20
W	2012	52	83	39	49	117	29	36	150	19	55	-7	114	30	154	16
W	2013	53	48	53	48	111	30	39	153	20	55	-6	113	31	159	16
O	2007	47	24	66	44	93	32	34	147	19	48	-9	122	32	117	21
O	2011	52	31	62	42	114	27	34	158	18	52	-5	112	35	99	26
O	2012	54	18	75	48	88	35	39	131	23	53	-1	102	33	143	19
O	2013	55	37	60	46	121	27	37	149	20	56	-7	115	33	141	19
NC	2007	37	32	54	33	96	26	29	135	18	44	-10	128	27	134	17
NC	2011	57	49	54	47	108	30	42	138	23	56	-5	110	27	157	15
NC	2012	58	28	68	53	82	39	49	112	30	56	4	94	33	140	19
NC	2013	60	17	78	52	97	35	43	137	24	58	-8	115	33	139	19
SC	2007	55	-22	167	43	116	27	38	135	22	51	-6	114	32	153	17
SC	2011	59	28	68	56	56	50	53	82	39	58	0	99	42	95	31
SC	2012	-	-	-	-	-	-	-	-	-	60	-9	117	37	155	19
SC	2013	-	-	-	-	-	-	-	-	-	63	-31	196	37	157	19
Averages		53	28	65	47	102	32	39	136	22	54	-7	115	33	139	19



363

364 Figure 8. Plot of C_{bio} vs C_{anth} in offshore and nearshore waters in the California Current

365 Large Marine Ecosystem. The simple schematic in the upper right is a cross section of the

366 coast with offshore being to the left, and with the mean path of upwelling water indicated

367 as a red arrow. Error bars express standard deviations for various estimates from each

368 region and depth (Table 1) rather than uncertainty, which is approximately $\pm 10 \mu\text{mol}$

369 kg^{-1} (1σ) for C_{bio} and C_{anth} .

370

371 4.2. Biological impacts evaluated as pteropod shell dissolution

372 Co-locating biological responses and chemical observations allows for direct

373 comparison of results in 2011 and 2013. Pteropod dissolution has been found to be

374 highly correlated with aragonite saturation conditions in 2011 (Bednaršek et al., 2014a).

375 Consequently, we have used the same procedure to also correlate the extent of dissolution

376 also for 2013. Pteropod shell dissolution significantly increased from offshore to

377 nearshore in the CCLME. Pteropods were ~22% more likely to be affected by severe

378 shell dissolution in nearshore waters compared with offshore waters. Consistent with
379 these results, nearshore Ω_{ar} values were approximately 40% lower than offshore values,
380 indicating a strong negative correlation between the percentage of pteropod individuals
381 with severe shell dissolution and Ω_{ar} (Fig. 6).

382

383 In 2011 and 2013, C_{anth} contributed approximately 22 - 65% of the enriched DIC in the
384 coastal areas over the period of the spring and summer measurements through the top
385 100 m (Table 1). This contribution lowered average seawater Ω_{ar} values from
386 approximately 1.39 to 1.05 in the nearshore region in 2011, and from 1.46 to 1.08 in
387 2013. Offshore, the contribution of C_{anth} reduced Ω_{ar} from an average of 2.21 to 1.51 in
388 2011, and from 2.09 to 1.43 in 2013 since the pre-industrial times. Consequently, based
389 on the newly developed relationships in Figure 6, we estimate that the percentage of
390 pteropods affected with severe dissolution due to the C_{anth} contribution in 2011 increased
391 19% in the nearshore waters and 26% in the offshore waters (Table 2). In 2013, we
392 estimate C_{anth} had increased the percentage of individuals affected by dissolution by
393 20% and 25% in nearshore and offshore waters, respectively (Table 2). The 2013 results
394 are comparable to the results for 2011, providing further evidence for increasing
395 incidence of severe dissolution with increasing C_{anth} and decreasing Ω_{ar} . The estimate of
396 pteropod dissolution from C_{anth} is comparable to that reported previously (Bednaršek et
397 al., 2014a), where dissolution was estimated based on the difference between pre-
398 industrial and current DIC values.

399

400 The observed relationship between Ω_{ar} and severe shell dissolution suggests that changes

401 in the carbonate chemistry due to C_{anth} are already having an impact on *L. helicina*.
 402 Although the percentage of individuals affected by dissolution in the nearshore region is
 403 ~22% greater than in the offshore region, the increase due to anthropogenic CO_2 of
 404 approximately 20–25% is comparable in both regions. Surprisingly, the relative change
 405 in the extent of pteropod dissolution in the offshore regions suggest that they are at least
 406 as vulnerable, or perhaps even more vulnerable, to the changes imposed by the C_{anth}
 407 uptake over the last several decades. This may be related to the much lower natural
 408 variability in offshore waters as compared with the nearshore waters.

409

410 Table 2. Average pre-industrial and current aragonite saturation states (calculated for
 411 years 2011 and 2013) and average percentage of individuals affected by severe
 412 dissolution in the pre-industrial times and currently for the nearshore and offshore regions
 413 of CCLME.

414

Year	Location	Ω_{ar} , preind.	Ω_{ar} , current	% ind. with severe dissolution, preind.	% ind. with severe dissolution, current
2011	nearshore	1.39	1.05	39	58
2013	nearshore	1.46	1.08	36	56
2011	offshore	2.21	1.51	8	34
2013	offshore	2.09	1.43	12	37

415

416 Shell dissolution as observed in pteropods along the west coast of North America affects
 417 their swimming abilities (Bednaršek et al, unpublished results), and can potentially
 418 enhance predation pressure and increase energetic costs of vital biological processes
 419 (Lischka et al., 2011; Wood et al., 2008; Manno et al., 2012). This chronic exposure to
 420 undersaturated conditions results in sub-lethal effects of compromised physiological
 421 state that may, over longer time periods, affect the overall pteropod population in the
 422 CCLME (Weisberg et al., 2016). Given that pteropods are equally abundant nearshore

423 and offshore (Mackas and Gailbraith, 2012; Bednaršek et al., 2012b), changes due to OA
424 intensification might have ecological implications in both regions. Additionally, the role
425 of pteropods as potentially important prey species requires better understanding of
426 trophic interactions with their predators on the regional level in the CCLME. Integrating
427 pteropods as an independent functional group in end-to-end modeling efforts can help
428 reveal the impacts of potential pteropod biomass decreases on higher trophic levels.
429 Introducing pteropods in such models would require incorporating information on
430 pteropod diet, life-history stages, and physiological and feeding responses, which has
431 recently been reviewed by Bednaršek et al. (2016).

432

433 **5. Conclusions**

434 By combining chemical and biological studies in the field we are able to provide a clearer
435 picture of the extent of C_{anth} distributions and its likely impact on pteropod shell
436 dissolution. Our results suggest that large-scale declines in the aragonite saturation states
437 of the CCLME resulting from the uptake of C_{anth} in open-ocean and coastal waters are
438 leading to increased incidence of pteropod shell dissolution and potentially creating
439 significant challenges for these organisms. Since the pre-industrial times, pteropod shell
440 dissolution has, on average, increased approximately 20–25% in both nearshore and
441 offshore waters in the CCLME. The capacity of these organisms to acclimatize and adapt
442 to OA, amid concurrent changes in temperature, dissolved oxygen, and other drivers
443 remains largely unknown. Nevertheless, the results shown here clearly indicate that
444 humankind may already be having a significant impact on a species that may play a vital
445 | role in this large and important marine ecosystem.

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447

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455

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736 **Figure and table captions**

737 **Figure 1.** Map of the station locations for the 2007 West Coast cruise. The black line
738 shows the cruise track. The 2011, 2012, and 2013 cruises included subsets of
739 these stations and, in some cases, a few additional stations.

740 **Figure 2.** Maps of surface ocean pH_T values for the 2007, 2011, 2012, and 2013 cruises.
741 The 2011 map includes the shore-based intertidal data.

742 **Figure 3.** Maps of surface DIC concentrations in $\mu\text{mol kg}^{-1}$ for the 2007, 2011, 2012, and
743 2013 cruises. The nearshore upwelling regions are delineated by DIC
744 concentrations in excess of $2050 \mu\text{mol kg}^{-1}$. Black dots indicate measurement
745 locations. Open circles on the 2011 and 2013 cruises indicate stations where both
746 chemical and biological samples were taken.

747 **Figure 4.** Aragonite saturation state at the surface, 25 m, 55 m, and 105 m during the
748 2013 West Coast survey.

749 **Figure 5.** Vertical sections of: (A) dissolved oxygen, (B) pCO_2 , (C) pH_T , and (D) Ω_{ar}
750 along the 2013 Line 6 stations off Newport, OR. Black dots indicate measurement
751 locations and the isolines lines in (A) and (B) show the potential density in kg m^{-3} .

752 **Figure 6.** Percentage of individuals affected by severe dissolution as a function of
753 aragonite saturation state (integrated over the upper 100 m) for the 2011 (open
754 circles) and 2013 (closed circles) data. The dashed lines show the 95% confidence
755 interval for the logarithmic function.

756 **Figure 7.** Distribution of C_{anth} in $\mu\text{mol kg}^{-1}$ at the surface, 25 m, 55 m, and 105 m depth
757 for the 2013 West Coast survey.

758 **Figure 8.** Plot of C_{bio} vs C_{anth} in offshore and nearshore waters in the California Current

759 Large Marine Ecosystem. The simple schematic in the upper right is a zonal
760 section of the coast with offshore being to the left, and with the mean path of
761 upwelling water indicated as a red arrow. Error bars express standard deviations
762 for various estimates from each region and depth (Table 1) rather than uncertainty,
763 which is 1σ of $\pm 10 \mu\text{mol kg}^{-1}$ for C_{bio} and C_{anth} .

764 **Table 1.** Anthropogenic carbon (C_{anth}), remineralized carbon (C_{bio}) and anthropogenic
765 percentage of total enriched carbon ($\%C_{\text{anth}}$) by region (W: Washington, O:
766 Oregon, NC: Northern California, SC: Southern California), cruise year, and
767 depth for averages of gridded coastal properties shoreward of the 200 m isobath
768 (left columns) and for the northwestern stations occupied offshore of each region
769 (right columns). All values are expressed in $\mu\text{mol kg}^{-1}$. Estimated average
770 uncertainty is approximately $\pm 10 \mu\text{mol kg}^{-1}$ (1σ). Column averages are calculated
771 weighting all regions and years equally. Negative C_{bio} values suggest either net
772 autotrophy or physically derived oxygen supersaturation.

773 **Table 2.** Average pre-industrial and current aragonite saturation states (calculated for
774 years 2011 and 2013) and average percentage of individuals affected by severe
775 dissolution in the pre-industrial times and currently for the nearshore and offshore
776 regions of CCLME.

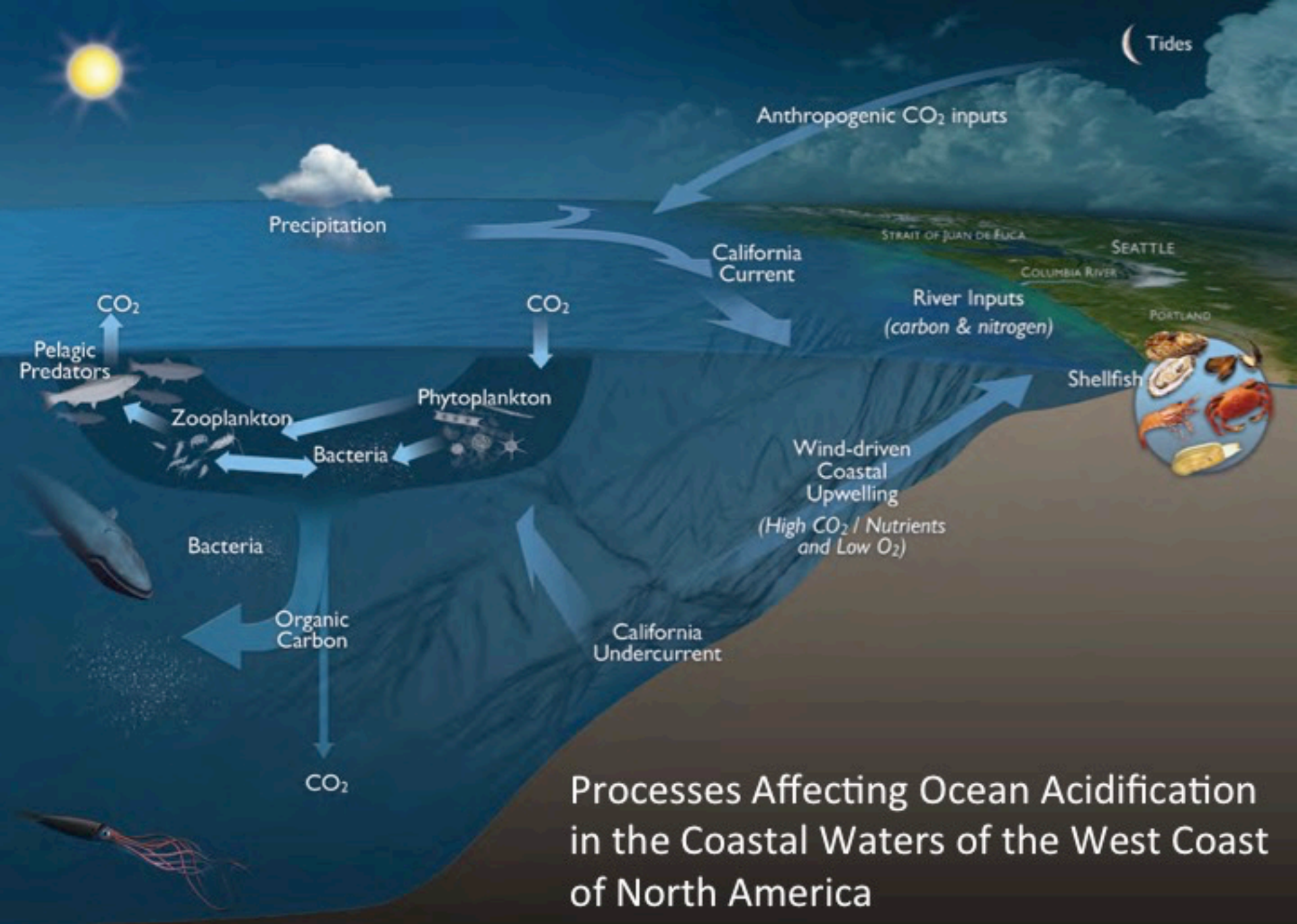
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Processes Affecting Ocean Acidification in the Coastal Waters of the West Coast of North America