A quantification method for peroxyacetyl nitrate (PAN)

2 using gas chromatography (GC) with a non-radioactive

3 pulsed discharge detector (PDD)

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10

11 Abstract

12 In this study, we developed a method for continuous PAN measurements by gas

13 chromatography (GC) with a non-radioactive pulsed discharge detector (PDD).

14 Operational parameters were optimized based on the ratio of peak height over

baseline noise (P/N ratio). The GC/PDD system was compared with a traditional

16 radioactive electron-capture detector (ECD). In the lab, the method detection limit

17 (MDL) of the new GC/PDD method (9 pptv) was lower than the radioactive GC/ECD

18 method (15 pptv), demonstrating its excellent potential. The MDL of GC/PDD in the

19 field campaign at the Mt. Bachelor Observatory (MBO) was 23 pptv, higher than in

20 the lab. This was caused in part by the decreased slope of the calibration curve

resulting from the low air pressure level at MBO. However, the MDL level of

22 GC/PDD at MBO is still low enough for accurate PAN measurements, although

special attention should be paid to its application at high-elevation sites. Observations

of PAN were conducted at MBO in the summer of 2016 with the GC/PDD system,

and provided more evidence of the performance of the system. PAN was found to be

26 highly correlated with CO. The promising performance of GC/PDD which does not

27 require a radioactive source makes it a useful approach for accurate PAN

28 measurements in the field.

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29 **1** Introduction

Peroxyacetyl nitrate (PAN) is a useful indicator of photochemical smog and acts as a 30 potentially significant reservoir of nitrogen oxides (NO_x) in the troposphere (Nielsen 31 et al., 1981; Singh and Hanst, 1981; Singh and Salas, 1983). PAN is highly stable in 32 the cold regions of the middle and upper troposphere providing NO_x storage and 33 hemispheric-scale transport (Singh et al., 1986), and has been recognized as a 34 dominant form of reactive nitrogen (NO_{ν}) in the free troposphere (Koike et al., 2003). 35 36 The lifetime of PAN, due to thermal decomposition, ranges from a few hours to several months, depending on temperature (Singh, 1987). Photolysis is the main loss 37 process of PAN in the upper troposphere (Talukdar et al., 1995). The thermal 38 decomposition of PAN redistributes NO_x far from its source region, enhancing ozone 39 (O₃) production in descending Asian plumes observed in the western U.S. (Heald et 40 al., 2003; Kotchenruther et al., 2001a,b; Moxim et al., 1996; Zhang et al., 2008). 41 PAN decomposition was found to contribute 11%–30% toward NO_x production and 42 enhance O_3 photochemical tendency by 0.13–0.41 ppbv d⁻¹ in the springtime 43 44 northeastern Pacific troposphere (Kotchenruther et al., 2001a). Ford et al. (2002) measured PAN in ambient and snowpack interstitial air at Summit, Greenland, and 45 found that PAN represented 30%–60% of NO_v and snowpack acted as a source of 46 PAN in summer. PAN was found to contribute about 20% to NO_{y} on average at the 47 remote Jungfraujoch research station in the Swiss Alps (Whalley et al., 2004). Fischer 48 et al. (2011) reported the springtime PAN at the Mt. Bachelor Observatory (MBO) 49 50 from 2008 to 2010 and linked the interannual variability to biomass burning, transport 51 efficiency over the central and eastern Pacific, and transport temperature. PAN 52 elevation in urban areas in East Asia suggested a more local impact of NO_x pollution in summer and fall (Lee et al., 2012; Xu et al., 2015; Zhang et al., 2015). The study of 53 54 Ungermann et al. (2016) indicated that eddy shedding provides a rapid horizontal transport pathway of Asian pollution into the extratropical lowermost stratosphere 55 with a timescale of only a few days. In regional wildfire plumes, $\Delta PAN/\Delta CO$ ranged 56 from 1.46–6.25 pptv ppbv⁻¹ and PAN represented 25%–57% of the observed NO_y in 57 aged plumes (Briggs et al., 2016). Therefore, PAN chemistry is a crucial part of the 58

photochemical processes and the long-range transport of NO_x and O_3 . Accurate PAN

60 measurement is thus of great importance to improve the understanding of NO_x and O_3 61 formation and transformation.

62 The gas chromatography (GC) with electron capture detection (ECD) method is

63 hitherto the most common method to measure PAN and its homologues (Fischer et al.,

64 2010; Fischer et al., 2011; Flocke et al., 2005; Gao et al., 2014; Lee et al., 2013;

65 Moravek et al., 2014; Schrimpf et al., 1995; Williams et al., 2000; Xu et al., 2015;

66 Zellweger et al., 2000; Zellweger et al., 2003; Zhang et al., 2009; Zhang et al., 2015).

67 Other methods have also been used, such as GC with luminol chemiluminescence

detection (LCD) (Gaffney et al., 1998; Lee et al., 2012; Marley et al., 2004), GC with

69 negative ion chemical ionization mass spectrometry (NICI-MS) (Tanimoto et al.,

1999), thermal dissociation laser-induced fluorescence (TD-LIF) (Day et al., 2002),

proton-transfer-reaction mass spectrometry (PTR-MS) (Hansel and Wisthaler, 2000),

and thermal dissociation chemical ionization mass spectrometry (TD-CIMS) (Slusher

et al., 2004). The GC/ECD method is widely adopted for PAN measurement due to its
accuracy and low method detection limit (MDL).

75 However, a pivotal weakness of the GC/ECD method is the radioisotope-based

ECD, which uses nickel-63 (63 Ni) as a stable source of electrons (beta particles) that

are accelerated towards a positively charged anode, generating a steady current

78 (Fischer et al., 2010; Flocke et al., 2005; Moravek et al., 2014; Schrimpf et al., 1995;

79 Williams et al., 2000). Because of their potentially hazardous properties, the use of

radioactive materials must be closely regulated to protect the health and safety of the

81 public and the environment. Transportation licensing and routine leak check are

required for the radioactive sources, making the use of the GC/ECD method

83 inconvenient, especially for field work. Ionization-based gas chromatographic

84 detectors mainly include flame ionization detector (FID), thermionic ionization

detector (TID), photo-ionization detector (PID), electron-capture detector (ECD),

helium ionization detector (HID), and pulsed discharge detector (PDD) (Poole, 2015).

87 PDD is actually a special type of ECD (also known as PD-ECD) using an additional

dopant gas (Cai et al., 1998; Forsyth, 2004). The dopant gas is first ionized by the

photons from the discharge, generating electrons which constitute the detector 89 standing current. When electron capturing compounds enter the detector, a decrease in 90 the detector standing current occurs, which is the PDD response. Most manufacturers 91 will only sell a full GC, and there is no commercially available GC that would be 92 appropriate to use for PAN. The advantage of the PDD is that it is non-radioactive and 93 94 can be purchased stand-alone with a compatible electrometer without having to dismantle a GC. Ford et al. (2002) described the use of a GC/PD-ECD system to 95 96 measure PAN, but provide no information on the operating characteristics or 97 performance.

In this study we developed a GC/PDD method for PAN measurement. As an 98 alternative to the traditional radioactive GC/ECD method, the GC/PDD combination 99 provides high accuracy and low MDL. The GC/PDD method has enhanced 100 deployability due to its non-radioactive source. We compared the two methods in the 101 lab using a state-of-the-art calibration system, and employed the GC/PDD method in 102 the 2016 summer campaign at the Mt. Bachelor Observatory (MBO, 2.8 km asl) to 103 104 evaluate its on-site performance. Overall the GC/PDD method has similar or better performance as the radioactive GC/ECD method. 105

106 2

2 Materials and methods

107 The GC/PDD PAN detection system consisted of a capillary GC column, a Valco 108 Instruments Co. Inc. (VICI) PDD Model D-2, an 8-port valve with a sample loop, a pump, a humidifier, a helium gas cylinder with a helium purifier, a dopant gas 109 cylinder, a mass flow controller (MFC), and two restrictors (Fig. 1). Similar to the 110 system described by Fischer et al. (2010), the system incorporated a photochemical 111 112 PAN calibration source using a calibrated nitric oxide (NO) standard and excess acetone to generate PAN under illumination of an ultraviolet (UV) lamp (Fischer et al., 113 2010). 114

115 2.1 PAN detection system

116 An 8-port Valco valve was utilized to control the operational modes of the system.

117 Under the loading mode, air samples were collected through the inlet to a 1.5 mL

sample loop by a pump. The sample loop is made from 1/8'' polyetheretherketone 118 (PEEK) tubing. PEEK tubing was reported to have no loss of PAN and a better 119 performance than perfluoroalkoxy (PFA) and stainless steel tubing (Fischer et al., 120 2010; Flocke et al., 2005). Oxygen (O₂) diffuses through PFA tubing, causing a noisy 121 background, and PAN loss has been observed on stainless steel. Ultra-high purity 122 (UHP) helium (He) was used as a carrier/discharge gas. UHP He was further purified 123 with an Agilent Gas Clean Filter System to remove O₂ and moisture. The carrier gas 124 flow rate was set to be 7 mL min⁻¹ using a Valco restrictor. Prior to entering the valve 125 and column the carrier gas was humidified by a cartridge filled with 200 g of copper 126 sulfate pentahydrate (CuSO₄·5H₂O, 99.995% purity), temperature controlled to 35°C 127 (Flocke et al., 2005). The addition of moisture minimizes loss of PAN in the column, 128 valve and tubing. The filling in the cartridge was changed every two months. 129 When the 8-port valve switches to the injection mode, the purified and humidified 130 carrier gas passes through the valve and a 1/16" PEEK connecting tubing, and enters 131 the column. We used a 15 m Restek Rtx-200MS (1 mm film thickness, 0.53 mm inner 132 133 diameter) capillary column. The column was controlled to 20°C by a bidirectional temperature controller (TE Technology TC 36-25 RS232) and a thermoelectric device 134 (TE Technology AC-073). The sample flow enters one end of the VICI PDD which 135 was controlled to 60° C for detection. The PDD used 30 mL min⁻¹ UHP He as a 136 discharge gas and 1.8 mL min⁻¹ of 5% methane (CH₄) in He as a dopant gas. The flow 137 rates of the discharge gas and the dopant gas were controlled with an MFC and a 138 Valco restrictor, respectively. 139 140 The PDD was set to the electron capture mode which requires the addition of a 141 dopant gas. The dopant gas used in this study is first ionized by the photons from the 142 discharge gas generating electrons. When compounds that can capture electrons enter

the detector the standing current in the PDD is decreased. Before tests the detector

144 was baked at 300°C for 12 h to stabilize its background. The output voltage signal

145 from the detector was converted to a digital signal by the PeakSimple

146 Chromatography Data System, and the PeakSimple software was used to collect the147 output from the detector. With the above mentioned conditions, the PAN peak occurs

at about 3.6 min (within the range of 3.3–3.9 min). Samples were collected every 5
min.

150 2.2 PAN calibration source

The PAN calibration source was also used in our previous study (Fischer et al., 2010). We used NO and excess acetone in the presence of O_2 and UV to generate PAN. NO is converted to NO₂ and then to PAN following Reactions (1) to (4) (Fischer et al., 2010; Roberts et al., 2004):

¹⁵⁵
$$CH_3C(O)CH_3 + 2O_2 \xrightarrow{h\nu} CH_3C(O)OO + CH_3OO$$
 (1)

¹⁵⁶
$$CH_3C(0)OO + NO \rightarrow CH_3C(0)O + NO_2$$
 (2)

¹⁵⁷
$$CH_3OO + NO \rightarrow CH_3O + NO_2$$
 (3)

¹⁵⁸
$$CH_3C(0)OO + NO_2 \rightarrow CH_3C(0)OONO_2$$
 (4)

The conversion efficiency of the calibrator from NO to PAN is 93±3% based on 159 previous work (Fischer et al., 2010; Flocke et al., 2005). We used Scott-Marrin 160 standard cylinders with a NO mixing ratio of 0.996 ppmv and an acetone mixing ratio 161 162 of 20 ppmv in ultrapure air. Both gas cylinders were National Institute of Standards and Technology (NIST)-traceable (±2% tolerance). The flow rates of NO and acetone 163 were 1.2 and 38 mL min⁻¹, respectively, controlled by MFCs. PAN from the calibrator 164 was diluted with hydrocarbon free (HCF) air from a pressurized cylinder. The flow 165 rate of HCF air ranged from 660 to 5000 mL min⁻¹ to generate a range of PAN mixing 166 ratios from 221 to 1590 pptv. The system requires approximately 30 min to produce a 167 steady PAN calibration source. 168

169 **2.3** Comparison with a GC/ECD system

To evaluate the performance of the GC/PDD system, we compared it to the GC/ECD system used in our previous study (Fischer et al., 2010). The same GC with the same configuration was used, but the PDD was replaced with a Shimadzu Mini-2 ECD. The temperature of the ECD was controlled to 40°C. UHP He was used as a carrier gas at a flow rate of 27–28 mL min⁻¹, and UHP nitrogen (N₂) was used as a make-up gas at a flow rate of 2–3 mL min⁻¹. Uncertainties, sensitivities, and MDLs of these two

methods were compared using the same PAN calibration source. Three sets of

177 calibrations using the GC/PDD method were conducted to ensure its stability. Either

two sets were at least one week away from each other, including a 6-point calibrationeach time.

180 2.4 Application in the 2016 summer campaign at MBO

The GC/PDD method was deployed in the 2016 summer campaign at the Mt. 181 Bachelor Observatory (MBO). MBO is located on the summit of a dormant volcano in 182 central Oregon (43°58'39" N 121°41'10" W, 2763 m asl). The site has been used for 183 atmospheric chemistry research for over 13 years (Jaffe et al., 2005). The PAN 184 measurement in the campaign took place from July 29 to September 27, 2016. The 185 campaign lasted for two months and two sets of calibrations were performed within 186 187 the campaign. Sub-micron aerosol scattering coefficient (σ_{sp}) and carbon monoxide (CO) were also measured during the campaign. We used a multi-wavelength 188 nephelometer Model 3563 manufactured by TSI Inc. to measure sub-micron aerosol 189 scattering at 450, 550 and 700 nm (blue, green and red) (Laing et al., 2016). The σ_{sp} at 190 green wavelength are adjusted to standard temperature and pressure condition (STP, 191 273 K and 1 atm) for analysis in this study. CO was measured using a Picarro G2302 192 193 cavity ring-down spectrometer (Gratz et al., 2015). All data are reported every 5 min in the Coordinated Universal Time (UTC). 194

195 2.5 Backward trajectories and wildfire locations

We calculated 10-day air mass backward trajectories from MBO for every hour of the 196 197 two identified plume events using the Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT) model version 4 (Draxler and Hess, 1998). Global Data 198 Assimilation System (GDAS) $1^{\circ} \times 1^{\circ}$ gridded meteorological data were used. The 199 starting height was set to be 1800 m above ground level based on terrain height in the 200 gridded meteorological data (Gratz et al., 2015). Daily MODIS fire detection data for 201 North America was obtained from the United States Department of Agriculture 202 (USDA) Forest Service (http://activefiremaps.fs.fed.us/gisdata.php). Daily MODIS 203 fire detection data for Eurasia was downloaded from the Fire Information for 204

Resource Management System (FIRMS) of the US National Aeronautics and Space
Administration (NASA) (https://firms.modaps.eosdis.nasa.gov/download).

207 **3 Results and discussion**

208 **3.1** Chromatogram integration and operational parameters

Fig. 2 shows the chromatograms of two samples from the calibration source with PAN
mixing ratios of 138 and 1070 pptv. The PAN signals occurred at about 3.7 min. The
chromatograms from the GC/PDD method indicate good performance and low MDL
of the system. We developed an integration method for the chromatogram using
MATLAB. Peak fitting was performed within a 1-min range in the vicinity of the peak
using a Modified Gaussian Equation (MGE):

215
$$f(x) = \lambda_1 \exp\left[-\left(\frac{x-\lambda_2}{\lambda_3}\right)^2\right] + \lambda_4 + \lambda_5 x$$
(5)

In Equation (5), λ_1 is a scale factor of the peak; λ_2 is the center of the peak; λ_3 is a width factor of the peak; λ_4 is the baseline offset; and λ_5 is the baseline slope. We then integrated the fitted line to obtain the peak area for the PAN signal. Fig. 3 shows the chromatograms and their fitted lines of three samples from the PAN source in a calibration case with a fluctuating baseline. The MGE method exhibits an excellent fit for both high and low signal cases (Fig. 3).

222 The flow rates of carrier gas and dopant gas are crucial operational parameters 223 affecting the performance of the GC/PDD system. Baseline noise is represented by the standard deviation of the baseline within a 1-min range in the vicinity of the peak. 224 The ratio of peak height over baseline noise (P/N ratio) is an important indicator to 225 depict the performance of the system. Therefore, we fixed the PAN mixing ratio to 226 227 862 pptv, and changed the flow rates of carrier gas and dopant gas. Peak heights, baseline noises and the P/N ratios under different conditions are shown in Fig. 4. Peak 228 height and baseline noise both increase when the dopant gas flow rate increases, while 229 the P/N ratio climbs up first and then declines. High P/N ratio implies high 230 detectability. Therefore, there is an optimal dopant gas flow rate. Two sets of tests 231 were conducted when carrier gas flow rates were set to be 8 and 9 mL min⁻¹, 232

respectively. Under high dopant gas condition (>1.5 mL min⁻¹), reducing the carrier 233 gas also has an increasing effect for both peak height and baseline noise, and there is 234 also an optimal carrier gas flow rate. The tests showed in Fig. 4 were taken place in 235 the lab (1013 hPa and 293 K). For MBO (730 hPa and 293 K), the optimal operational 236 parameters were a bit different from the lab due to the air pressure discrepancy. At 237 MBO, slightly lower carrier gas flow rate (7 mL min⁻¹) and slightly higher dopant gas 238 flow rate (1.8 mL min⁻¹) were used, although we did not do the same extensive set of 239 test conditions as in the lab. 240

241 3.2 Comparisons between GC/ECD and GC/PDD

Comparisons between GC/ECD and GC/PDD were conducted both in the lab at the 242 University of Washington Bothell (UWB) and at MBO. The same calibration source 243 244 was used for the two detection systems at 6-7 PAN mixing ratio levels. Parameters for calibrations are shown in Table 1. The calibrations by GC/ECD and GC/PDD at UWB, 245 have high correlations ($R^2 = 0.997$) and show good linearity. The most important 246 parameter to evaluate the performance of the two methods is the method detection 247 248 limit (MDL), which is defined as three times the standard deviation of the replicates of a low calibration standard over the slope of the calibration curve. It is often used 249 250 when the noise for the blank is not available, which is true in chromatography. MDLs were calculated using the following equation: 251

252 MDL =
$$\frac{3\sigma}{s}$$
 (6)

In Equation (6), s is the slope of the calibration curve (mV s $pptv^{-1}$), and σ is the 253 standard deviation of the peak area (mV s) at the lowest PAN mixing ratio (~200 pptv) 254 during calibration. The standard deviations for GC/ECD and GC/PDD at ~200 pptv in 255 the lab were 0.34 and 0.031 mV s, respectively. Therefore, with the obtained slopes, 256 the corresponding MDLs were 15 and 9 pptv, respectively. The precisions of the two 257 systems at ~1000 pptv were estimated to be 0.4% and 2.5% from replicate 258 259 observations. The accuracy of the PAN mixing ratio produced by the calibrator was 260 estimated to be 7.7% by our previous study (Fischer et al., 2010). The overall uncertainty was calculated as the root sum of the squares of the precision and the 261

262 accuracy. Therefore, the overall uncertainties for GC/ECD and GC/PDD at UWB were found to be 7.7% and 8.1% (at ~1000 pptv), respectively (see Table 1). 263 With the same method, the precision of the system at MBO at ~1000 pptv was 264 estimated to be 6.1%. With the precision (6.1%) and the accuracy (7.7%), the overall 265 uncertainty of the GC/PDD method at MBO was reported to be 9.8% at ~1000 pptv 266 (see Table 1), compared to 8.2% using GC/ECD method at MBO (Fischer et al., 2010). 267 In the calibration at MBO during the 2016 summer campaign, the standard deviation 268 269 was 0.031 mV s. Therefore, with Equation (6), the MDL for the campaign was estimated to be 23 pptv (see Table 1), higher than the MDL at UWB (9 pptv). This 270 was caused by the decreased slope of the calibration curve (4.1 mV s $pptv^{-1}$) resulting 271 from the low air pressure level (~730 mbar) at MBO, compared to the slope at UWB 272 $(10.0 \text{ mV s pptv}^{-1})$. The slopes were lower at MBO for both methods. This was due to 273 the lower air pressure at the high elevation MBO site. The air pressure difference 274 between UWB and MBO causes the change of the total sample volume resulting in a 275 ~30% loss in the injection mass. Furthermore, the pressure discrepancy has a more 276 277 significant impact on the slope for GC/PDD than that for GC/ECD. The larger sensitivity reduction suggests that the PDD detector also has a pressure dependence, 278 but we don't have any further evidence on how it works. However, the MDL level of 279 GC/PDD at MBO is still low enough for accurate PAN measurements, although 280 special attention should be paid to its application at high-elevation sites. 281

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2 **3.3** Plume events in the 2016 summer campaign at MBO

Measurements of PAN were made at MBO from July 29 to September 27, 2016. 283 Totally 7622 valid PAN data (5-min) were obtained from the 2016 summer campaign 284 at MBO. The mean PAN mixing ratio was 221 pptv, and the median value was 194 285 pptv. The 5th–95th percentile range was 87–434 pptv. Observations of PAN at MBO 286 in summer 2016 fit a lognormal distribution, reflecting the background PAN level in 287 the western U.S. and influence from regional and long-range transport wildfires. To 288 289 evaluate the performance of the GC/PDD method in pollution episodes, we chose two typical plume events in the 2016 summer campaign at MBO for multi-pollutant 290

analysis. Fig. 5 shows the 10-day backward trajectories for the two plume events with
wildfire locations in North America during the events and in Eurasia one week before
the events which is the approximate transport time of air masses. Event 1 was under
significant influence of Siberian wildfires, especially at lower elevation in eastern
Russia, and the backward trajectories of Event 1 had no overlap with regional
wildfires in North America. On the contrary, Event 2 was transported from large fires
in northern California.

Fig. 6 and Fig. 7 exhibit the observations of PAN, CO and aerosol scattering (σ_{sn}) 298 during Event 1 and Event 2, respectively. Event 1 experienced two peaks at 12:00 and 299 15:00 UTC on July 31, and Event 2 had a longer pollution episode from 12:00–19:00 300 on August 30 and two separate peaks at 5:00 and 19:00 on August 31. CO and PAN 301 had significant positive correlations in both events ($R^2 > 0.8$). The correlation between 302 CO and σ_{sp} was even more significant, especially for Event 2 (R² = 0.96). The 303 enhancement ratios (ERs) of $\Delta PAN/\Delta CO$ for Event 1 and Event 2 were 1.56 and 1.91 304 pptv ppbv⁻¹, respectively, close to the lower end of the range (1.46–6.25 pptv ppbv⁻¹) 305 306 reported by Briggs et al. (2016) for regional wildfire plumes. Much more significant difference occurred on the ERs of $\Delta \sigma_{sp} / \Delta CO$ for Event 1 (0.08 Mm⁻¹ ppbv⁻¹) and 307 Event 2 ($1.00 \text{ Mm}^{-1} \text{ ppbv}^{-1}$). The comparison between the Siberian wildfire event 308 (Event 1) and the regional wildfire event (Event 2) indicates that aerosols generated 309 from wildfires could be scavenged during long-range transport (Zhang and Jaffe, 2017) 310 while PAN is not significantly scavenged by cloud processing. The PAN level in 311 wildfire plumes is more likely related to the air temperature. In 2 Siberian wildfire 312 plume events in spring 2008 (Fischer et al., 2010), the $\Delta PAN/\Delta CO$ ratios were 4.0 and 313 6.3 pptv ppbv⁻¹ under an average air temperature of -3° C observed at MBO, much 314 higher than in the 2 wildfire plume events identified in the 2016 summer campaign at 315 an average air temperature of 11°C at MBO. This could be resulted from the shift of 316 NO_x-PAN balance towards the PAN side at lower temperature (Singh and Hanst, 317 1981). However, temperature is not the only factor. Fire emission ratios of NO_x/CO 318 and photochemical conditions are also very important to PAN formation (Fischer et al., 319 2014). The high correlation between CO and PAN in different types of plume events 320

and the relationship between temperature and $\Delta PAN/\Delta CO$ provided more evidence on the reliability of the new GC/PDD method.

323 **4** Conclusion

We developed a new method for PAN measurements using GC with a non-radioactive 324 325 PDD. The system was optimized and calibrated using PAN generated from a calibrated source of NO and excess acetone in the presence of O₂ and UV. A Modified 326 Gaussian Equation (MGE) method was used for peak fitting to smooth the signal at 327 328 low PAN mixing ratio levels considering a sloping baseline. The new GC/PDD method was compared to the traditional GC/ECD method in the lab. The MDL of 329 GC/PDD operated in the lab was 9 pptv, lower than that of GC/ECD (15 pptv), 330 331 showing the excellent performance of the GC/PDD system. The MDL of GC/PDD at MBO was 23 pptv while the MDL of GC/ECD at MBO stayed the same (15 pptv). 332 The sensitivity reduction for GC/PDD was due to the lower air pressure at the high 333 elevation MBO site. Besides the sample volume loss, the PDD detector also has a 334 335 pressure dependence. However, the MDL level at MBO is still low enough for accurate PAN measurements. The overall uncertainties for GC/ECD and GC/PDD in 336 337 the lab at ~1000 pptv were 7.7% and 8.1%, respectively, while that for GC/PDD at MBO was found to be 9.8%. The new method was used in the 2016 summer 338 campaign at MBO. Based on the analyses of wildfire plume events, PAN was found to 339 340 be highly correlated with CO. Results of enhancement ratios show that aerosols could be largely scavenged during long-range transport while the PAN level is more likely 341 related to the air temperature. The similar levels of significance for the CO-PAN 342 correlations between GC/ECD and GC/PDD and the similar $\Delta PAN/\Delta CO$ ratios 343 344 obtained from these two methods for the same types of events provided more evidence on the reliability of the GC/PDD method. Results in this study indicate that 345 the performance of the new GC/PDD method is almost as good as GC/ECD. With its 346 non-radioactive essence, GC/PDD could be widely applied to continuous PAN 347 measurements in the ambient air. 348

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- 354 University of Washington data repository (<u>https://digital.lib.washington.edu/researchworks</u>).

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516 **Tables**

517 **Table 1** Calibration parameters, method detection limits (MDLs) and overall

Parameters	GC/PDD		GC/ECD	
	UWB	MBO	UWB	MBO
Slope for calibration (mV s $pptv^{-1}$)	10.0	4.1	65.9	45.2
R^2 for calibration	0.997	0.996	0.997	0.994
MDL (pptv)	9	23	15	15 ^a
Uncertainty at ~1000 pptv (%)	8.1	9.8	7.7	8.2 ^a

518 uncertainties of GC/PDD and GC/ECD at UWB and MBO

519 Note:

^a These estimations were adapted from Fischer et al. (2010).

522 Figure captions

Fig. 1. Schematic diagram of the custom gas chromatograph pulsed discharge detector(GC/PDD) system.

Fig. 2. Examples of PAN chromatograms: (a) response to a PAN mixing ratio of 138
pptv; (b) response to a PAN mixing ratio of 1070 pptv.

- **Fig. 3.** Examples of PAN chromatogram fittings using MATLAB: (a) response to a PAN mixing ratio of 1070 ppty; (b) response to a PAN mixing ratio of 438 ppty; (c)
- response to a PAN mixing ratio of 138 pptv.
- **Fig. 4.** Peak heights, baseline noises and the ratios between them (P/N ratio) under different carrier gas and dopant gas flow rates. Baseline noise is represented by the standard deviation of the baseline within a 1-min range in the vicinity of the peak.
- **Fig. 5.** Backward trajectories for (a) Event 1 (2016/7/31 6:00–2016/8/1 5:00) with
- Asian fire locations on July 24, 2016 and North American fire locations on July 31,
- 535
 2016, and (b) Event 2 (2016/8/30 8:00–2016/9/1 4:00) with North American fire
- 536 locations on August 30–31, 2016.
- 537 Fig. 6. Observations of PAN, CO and aerosol scattering coefficient at MBO during
- Event 1 from July 31 to August 1, 2016, a period with significant influence from
- 539 Siberian wildfire smokes. The scatter plots show the data fit with linear regression.
- **Fig. 7.** Observations of PAN, CO and aerosol scattering coefficient at MBO during
- 541 Event 2 from August 30 to September 1, 2016, a period with significant influence
- 542 from regional (northern California) wildfire smokes. The scatter plots show the data
- 543 fit with linear regression.

Figures



Fig. 1. Schematic diagram of the custom gas chromatograph pulsed discharge detector (GC/PDD) system.



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Fig. 5. Backward trajectories for (a) Event 1 (2016/7/31 6:00–2016/8/1 5:00) with Asian fire locations on July 24, 2016 and North American fire locations on July 31, 2016, and (b) Event 2 (2016/8/30 8:00–2016/9/1 4:00) with North American fire locations on August 30–31, 2016.



Fig. 6. Observations of PAN, CO and aerosol scattering coefficient at MBO during Event 1 from July 31 to August 1, 2016, a period with significant influence from Siberian wildfire smokes. The scatter plots show the data fit with linear regression.



Fig. 7. Observations of PAN, CO and aerosol scattering coefficient at MBO during Event 2 from August 30 to September 1, 2016, a period with significant influence from regional (northern California) wildfire smokes. The scatter plots show the data fit with linear regression.