1	Spatiotemporal variations in the abundance and composition of bulk and chromophoric dissolved
2	organic matter in seasonally hypoxia-influenced Green Bay, Lake Michigan USA
3	
4	Stephen E. DeVilbiss, Zhengzhen Zhou, J. Val Klump, and Laodong Guo*
5	
6	School of Freshwater Sciences, University of Wisconsin-Milwaukee, 600 E Greenfield Ave.,
7	Milwaukee, WI 53204, USA.
8	
9	* Corresponding author. E-mail: guol@uwm.edu
10	
11	

# 12 Abstract

13 Green Bay, Lake Michigan USA, is the largest freshwater estuary in the Laurentian Great 14 Lakes and receives disproportional terrestrial inputs as a result of a high watershed to bay surface 15 area ratio. While seasonal hypoxia and the formation of "dead zones" in Green Bay have 16 received increasing attention, there are no systematic studies on the dynamics of dissolved 17 organic matter (DOM) and its linkage to the development of hypoxia. During summer 2014, 18 bulk dissolved organic carbon (DOC) analysis, UV-vis spectroscopy, and fluorescence 19 excitation-emission matrices (EEMs) coupled with PARAFAC analysis were used to quantify 20 the abundance, composition and source of DOM and their spatiotemporal variations in Green 21 Bay, Lake Michigan. Concentrations of DOC ranged from 202 to 571  $\mu$ M-C (average = 361±73) 22  $\mu$ M-C) in June and from 279 to 610  $\mu$ M-C (average = 349±64  $\mu$ M-C) in August. In both months, absorption coefficient at 254 nm (a254) was strongly correlated to bulk DOC and was most 23 24 abundant in the Fox River, attesting a dominant terrestrial input. Non-chromophoric DOC 25 comprised, on average,  $\sim 32\%$  of bulk DOC in June with higher terrestrial DOM and  $\sim 47\%$  in 26 August with higher aquagenic DOM, indicating that autochthonous and more degraded DOM is 27 of lower optical activity. PARAFAC modeling on EEM data resulted in four major fluorescent 28 DOM components, including two terrestrial humic-like, one aquagenic humic-like, and one 29 protein-like component. Variations in the abundance of DOM components further supported 30 changes in DOM sources. Mixing behavior of DOM components also indicated that while bulk 31 DOM behaved quasi-conservatively, significant compositional changes occurred during transport 32 from the Fox River to the open bay.

33

34 Keywords: Dissolved organic matter, CDOM, hypoxia, Green Bay, Lake Michigan

#### 36 **1. Introduction**

37 Freshwater bays and estuaries comprise some of the most biologically productive, yet 38 anthropogenically stressed ecosystems in the Laurentian Great Lakes (Smith et al. 1988; 39 Herdendorf 1990). Terrestrial inputs of nutrients and organic carbon are major components 40 driving estuary biogeochemical cycling and have implications ranging from degraded ecosystem 41 health to food web perturbations (Klump et al. 1997). Southern Green Bay, Lake Michigan USA 42 has historically been classified as a hypereutrophic system as a result of high nutrient and 43 organic carbon loading from the Lower Fox River, which drains some of the most intensely 44 farmed land in Wisconsin (Ankley et al. 1992; Smith et al. 1988). Despite decades of nutrient 45 regulation, Green Bay has continued to experience reeutrophication, recurring bouts of hypoxic 46 conditions, and the formation of dead zones over the past years (Klump et al. 2009; Egen 2014).

47 Carbon dynamics can be closely related to dissolved oxygen (DO) and the development of 48 hypoxia (Bianchi et al. 2010), as photochemical and heterotrophic oxidation of dissolved organic 49 matter (DOM) consumes DO in both surface and bottom waters (Diaz and Rosenberg 2008; 50 Green et al. 2006). Additionally, DOM plays an important role in the fate and transport of many 51 organic contaminants and heavy metals (McCarthy and Zachara 1989; Santschi et al., 1999). 52 Therefore, knowledge of sources, abundance, chemical composition, and cycling pathways of 53 DOM is important to better understanding of the overall ecosystem health in aquatic 54 environments. There have been several studies on dissolved and particulate organic matter in 55 Green Bay since the 1980s (Eadie et al. 1992; Achman 1993; O'loughlin and Chin 2009). 56 However, the Green Bay ecosystem and Lake Michigan as a whole have experienced significant 57 ecological and environmental change over the last two decades including increasing 58 anthropogenic influence, the introduction of invasive species (Qualls et al. 2007; De Stasio et al.

59 2014), and increases in the extent of hypoxia (Egen 2014). Baseline data on the abundance, 60 distribution, and composition of DOM in Green Bay are still lacking, but are indispensable for 61 trend analysis and understanding the response of carbon dynamics to climate and environmental 62 changes, especially the formation of hypoxia.

63 Dissolved organic matter is a heterogeneous mixture of reduced carbon compounds and 64 comprises the largest pool of active organic matter in aquatic environments (Leenheer and Croué 65 2003; Guo et al. 2003). Aquatic systems receive DOM inputs from allochthonous and 66 autochthonous sources such as riverine discharge and in situ production, respectively (Bauer and 67 Bianchi 2011). Chromophoric dissolved organic matter (CDOM) is operationally defined as the 68 fraction of bulk DOM that absorbs light over a broad range of visible and UV wavelengths 69 (Coble 2007; Rochelle-Newall and Fisher 2002; Zhang et al. 2009). As such, CDOM plays many 70 important roles in aquatic systems such as influencing biological production by regulating the 71 penetration of photosynthetically active radiation, protecting organisms from UV radiation, and 72 contributing to the organic carbon cycle due to high photoreactivity (Belzile et al. 2002; 73 Williamson et al. 1999). In addition to providing many important roles in aquatic systems, 74 CDOM is also useful in tracing riverine inputs and biogeochemical cycling of DOM (Helms et 75 al. 2008; Matsuoka et al. 2012). A sub-fraction of CDOM with fluorescent properties 76 (fluorescent-DOM, or FDOM) has received increasing interest in recent years due to its 77 usefulness as a proxy for the bulk DOC pool (Chen et al., 2003; Cory and McKnight, 2011). 78 Fluorescence excitation-emission matrices (EEMs) combined with parallel factor analysis 79 (PARAFAC) have been successfully used to identify allochthonous and autochthonous 80 fluorescent DOM components in a diverse range of natural environments (Stedmon and 81 Markager 2005; Wang et al. 2007; Chari et al. 2012; Xu et al. 2013; Zhou et al. 2016). So far,

there are no systematical studies on bulk DOC and chromophoric DOM in Green Bay using fluorescence EEMs coupled with PARAFAC modeling to track changes in DOM composition across the trophic gradient along the terrestrial-aquatic continuum in Green Bay.

85 The objectives of this study were to 1) examine the abundance, composition, spatial distribution, and temporal variation of DOC, CDOM, and FDOM in Green Bay; 2) evaluate the 86 87 relative importance of non-chromophoric DOM and allochthonous and autochthonous DOM 88 sources in different sampling months; 3) identify cycling processes and transformation pathways 89 of DOM in the water column under different thermal stratification conditions and the linkage 90 between DOM dynamics and hypoxic conditions; and 4) establish baseline data for trend 91 analysis and to provide new insights into better understanding the impact of organic carbon 92 loading to the health of Green Bay.

### 93 2. Materials and methods

#### 94 **2.1. Study Site**

95 Green Bay, located in northwestern Lake Michigan, USA is the largest freshwater estuary in 96 the Laurentian Great Lakes (Fig. 1). The average depth of the bay is 20 m, ranging from ~3 m in 97 the far southern end to 53 m in the north. Extending  $\sim$ 190 km from south to north and  $\sim$ 22 km from west to east, Green Bay has a volume of  $\sim 67 \text{ km}^3$ . Green Bay's watershed drains about 98 40,000 km<sup>2</sup>, which encompasses almost half of the drainage basin of Lake Michigan and 99 100 resultantly is highly impacted by surrounding landscapes (Harris and Christie 1987; Klump et al. 101 1997). Of the 11 rivers and streams that drain into Green Bay, the Fox River is the largest input with a mean discharge of 118 m<sup>3</sup> s<sup>-1</sup> (Mortimer 1978). The Lower Fox River watershed is 50.2% 102 103 agriculture, 34.6% urban, 14.7% natural forests and wetlands, and the remaining 0.6% under 104 construction (WDNR, 2012). Additionally, the lower Fox River used to contain the highest

density of pulp and paper mills in the world resulting in large amounts of nutrients and
contaminants being flushed into the southern portion of the bay (Harris and Christie 1987),
consequently causing high productivity, eutrophic conditions and the formation of dead zones,
especially in southern Green Bay (Valenta, 2013; Lin et al., 2016).

## 109 **2.2. Sampling**

Water samples were collected from Green Bay on board the RV Neeskay on the 4<sup>th</sup> and 5<sup>th</sup> 110 111 of June 2014 and on the 24th, 25th, and 26th of August 2014 at 30 and 43 sampling locations, 112 respectively (Fig. 1). Surface samples were taken with a submersible pump and stored in acid 113 washed, triple-rinsed, 2 L HPDE bottles (Nalgene). Samples were kept on ice in the dark until 114 further processing (within 2 days). Hydrographic data, including dissolved oxygen (DO), specific 115 conductivity, pH, chlorophyll-a fluorescence (Chl-a) and water temperature, were measured 116 using a CTD (Seabird) and two YSI sondes. Water samples were filtered through pre-combusted 117 0.7 µm GF/F (Whatman) and stored in pre-combusted (550°C) glass vials for DOC analysis and 118 in acid washed, triple-rinsed, HPDE bottles (Nalgene) for CDOM and EEM measurements. DOC 119 samples were acidified with concentrated HCl to a  $pH \le 2$ . Both DOC and CDOM samples were 120 stored at 4°C until analysis (within two weeks).

# 121 2.3. Measurements of DOC and UV-vis absorbance

122 Concentrations of DOC were measured with a Shimadzu TOC-L analyzer equipped with an 123 ASI-L autosampler using the high temperature combustion method (Guo and Santschi 1997). 124 Prior to analysis, samples were extensively sparged with zero air for >5 minutes to remove 125 inorganic carbon. Three to five replicate measurements of 150  $\mu$ L each were made for each 126 sample. Blanks, including water blanks and instrument blanks, were usually <2  $\mu$ M-C and 127 compensated from the original measurements accordingly. Detection limit was  $\leq 1 \mu$ M and precision was ≤2% in terms of coefficient of variance. Calibration curves were generated before
analysis and ultra-pure water, internal standards, and certified DOC standards (University of
Miami) were measured every eight samples for quality assurance (Zhou and Guo 2012).

131 UV-visible absorption spectra were measured with a spectrophotometer (Agilent 8453) 132 using a 1 cm path-length quartz cuvette over a wavelength range of 190-1,100 nm with 1 nm 133 increments. Samples were diluted with ultrapure water to an absorbance value of  $\leq 0.02$  at 260 134 nm to minimize inner-filtering effects (Ohno, 2002). Ultrapure water was scanned as a blank 135 before samples analysis daily. The water blank was subtracted and the refractive index effect was 136 corrected by subtracting the averaged absorbance between 650 and 800 nm (Stedmon et al. 2000; Zhou and Guo 2012). Absorption coefficients at a specific wavelength  $\lambda$  ( $a_{(\lambda)}$ , in m<sup>-1</sup>) were 137 138 calculated as  $a_{(\lambda)} = 2.303 A(\lambda)/L$ , where A( $\lambda$ ) is the absorbance at wavelength  $\lambda$  (nm) and L is the 139 cuvette path-length (in m). Specific UV absorbance at 254 nm (SUVA<sub>254</sub>) was calculated as A<sub>254</sub>/DOC concentration resulting in a dimension of m<sup>-1</sup>/mg-C/L or L/mg-C/m. Spectral slope 140 141 through linear fit of the logarithm of absorption coefficients over the wavelength interval of 275-142 295 nm (S<sub>275-295</sub>) was calculated to provide information on DOM molecular weight and the 143 influence of terrigenous DOM inputs (Helms et al. 2008). Spectral slope has been found to 144 negatively correlate with DOM molecular weight, or the higher molecular weight DOM has a 145 lower spectral slope value.

## 146 2.4. Measurements of fluorescence EEMs and PARAFAC Modeling

Fluorescence excitation emission matrices (EEMs) were measured using a Horiba Fluoromax-4 spectrofluorometer. Before analysis, samples were diluted with ultra-pure water to an absorbance value of  $\leq 0.02$  at 260 nm to minimize inner-filtering effects (Ohno 2002). Each sample was scanned from excitation wavelength 220-480 nm with 5 nm increments and emission 151 wavelength 240-600 nm with 2 nm increments in a 1 cm path-length quartz cuvette. The 152 bandpass width was 5 nm for both excitation and emission. A water blank was subtracted from 153 each sample and areas affected by Rayleigh and Raman scattering peaks were eliminated by 154 setting data to zero. The fluorescence index (FIX), an indicator of DOM quality, was calculated 155 as the ratio of fluorescence intensities between emission wavelengths 450 and 500 nm at 156 excitation wavelength 370 nm (McKnight et al. 2001). The biological index (BIX), a DOM 157 source indicator, was calculated as the ratio of emission intensity at 380 nm divided by the 158 emission intensity maximum observed between 420 and 435 nm, obtained at excitation 159 wavelength 310 nm (Parlanti et al. 2000). The humification index (HIX), which provides insight 160 into the degree of DOM humification, was calculated as the area under the emission spectra 435-161 480 nm divided by the peak area under the emission spectra 300-345 + 435-480 nm, at excitation 162 wavelength 254 nm (Ohno 2002).

PARAFAC modeling was done with MATLAB software using the DOMFlour Toolbox (Stedmon and Bro 2008). Sample matrices were calibrated, corrected, and normalized to maximum fluorescence intensity (excluding water scattering peaks) before analysis. A nonnegativity outlier test was performed and no samples were identified and subsequently removed as outliers. A split-half analysis was performed for model validation. PARAFAC analysis was also run separately for June and August to examine differences in DOM components between months.

170 **2.5. Statistical and spatial analyses** 

Statistical analyses (e.g. ANOVA, T-Test, significance) were performed with Sigmaplot
software (version 12.5). Contour maps were generated using Surfer 12 (Golden Software) using
the kriging method.

# 174 3. Results and discussion

## 175 **3.1. Hydrological features of Green Bay**

176 During June 2014, Green Bay had yet to fully thermally stratify and water temperature 177 ranged from 11.1 to 22.1°C with an average of 15.7±2.7°C. In contrast, by August the bay was 178 fully stratified and the average water temperature was 22.0±0.7°C, which was significantly 179 warmer than June (p < 0.0001, Table 1). As a result, DO concentrations were highly variable 180 between the two sampling months. In June, DO was regulated predominantly by temperature in both surface and subsurface water, as indicated by a significant negative correlation ( $r^2 = 0.7532$ ; 181 182 p < 0.0001, Table 1). Conversely, a much less significant, positive relationship existed between water temperature and DO in August ( $r^2 = 0.0829$ ; p = 0.0314, Table 1). Further, bottom water 183 DO level was as low as 2.6 mg  $L^{-1}$  and an apparent hypoxic zone had developed in the colder 184 185 bottom waters of the bay (Fig. 2), accompanied by an increase in phosphate concentration in 186 deeper waters in central Green Bay (Lin et al., 2016). This is likely related to increased sediment 187 oxygen demand and reduced mixing of aerated surface water resulting from thermal stratification 188 of the water column as well as the degradation of DOM in the water column (Klump et al. 2009; also see discussion below). In contrast, bottom water DO in June averaged  $12.6\pm1.5$  mg L<sup>-1</sup>, and 189 the lowest concentration was 8.6 mg  $L^{-1}$ , which is higher than criteria for hypoxia. 190

Surface distributions of specific conductivity (Figs 3a and 3e), which can be used as a tracer
of river water inputs due to its conservative mixing behavior, highlight significant differences in
hydrology between June and August (Modlin and Beeton, 1970). Surface distributions of specific

194 conductivity in June were consistent with previous observations indicating the presence of the 195 Fox River plume (Lathrop et al. 1990). The Fox River discharges into Green Bay at the 196 southernmost end and is the predominant river input into the bay (Mortimer 1978). The river 197 flow deflects east due to the Coriolis force as it moves north and mixes with open bay water. 198 Additionally, the combination of the Coriolis force and prevailing winds have also been shown 199 to induce a compensating southward flow of open Lake Michigan water along the west coast of 200 the bay, which was evident due to a strong specific conductivity gradient along both the north-201 south and east-west axes of the bay (Martin et al. 1995). Conversely, specific conductivity was 202 elevated along the western coast of the bay in August, suggesting temporal variations in Green 203 Bay's surface hydrology.

Although Chl-*a* concentrations were not significantly different between the two sampling months (p = 0.261), spatial distributions were somewhat variant (Figs. 3d and 3h). While Chl-*a* was highest at the southernmost sample stations for both months, Chl-*a* was elevated west of Sturgeon Bay in June, but was highly concentrated in the southeastern portion of the bay in August in the same region as the surface water DO minimum (Fig. 3e), suggesting DO consumption from organic matter degradation. However, no such trend was observed in June (Fig. 3b).

211 **3.2. DOC abundance and distribution** 

DOC concentrations ranged from 202 to 561  $\mu$ M-C with an average of 361±73  $\mu$ M-C in June, and from 279 to 610  $\mu$ M-C with an average of 349±64  $\mu$ M-C in August. No significant difference of DOC abundance between June and August was observed (p = 0.343) (Table 2). Throughout the summer, the highest DOC concentrations were observed in the lower Fox River (546 - 610  $\mu$ M) and the lowest concentrations were found at the northernmost sampling stations 217 influenced with waters from Lake Michigan (Figs. 4a and 4e). Although DOC concentration was 218 slightly higher in the Fox River in August than June (610 vs. 546 µM), potentially due to dilution 219 by high river discharge in June, DOC concentrations were elevated farther north in the bay in 220 June than August, likely also a result of  $\sim 2.5$  times higher Fox River discharge in June than August (251 m<sup>3</sup> sec<sup>-1</sup> vs. 92 m<sup>3</sup> sec<sup>-1</sup>; Data from USGS gauging station 040851385). This 221 222 highlights the importance of river inputs on temporal DOC dynamics in Green Bay. Also 223 apparent in Green Bay were low DOC concentrations along the northwestern coast of the bay, 224 concurrent with low specific conductivity suggesting the influence of open Lake Michigan water, 225 which is significantly lower in DOC (average of 154 µM-C in June and July, 2013; Zhou et al., 2013). This supports previous observations and indicates the importance of the surface water 226 227 hydrology and water mixing with open Lake Michigan water on DOC dynamics in Green Bay 228 (Martin et al. 1995; Mortimer 1978; Modlin and Beeton, 1970).

#### 229

#### **3.3 Variations in CDOM characteristics**

The absorption coefficient at 254 nm ( $a_{254}$ ) ranged from 12.4 to 58.5 m<sup>-1</sup> (average = 27.9±9.3 230 m<sup>-1</sup>) in June and from 16.8 to 59.8 m<sup>-1</sup> (average = 24.7 $\pm$ 8.1 m<sup>-1</sup>) in August. Values of a<sub>254</sub> were 231 232 significantly greater in June than those in August (p = 0.028), showing higher chromophoric DOM components early in the summer. Significant correlation was observed between a254 and 233 DOC, both in June ( $r^2 = 0.9408$ ; p < 0.0001) and August ( $r^2 = 0.8968$ ; p < 0.0001), indicting 234 235 similar chromophoric DOM sources in Green Bay throughout the summer. In addition, spatial 236 distributions of a254 resembled those of bulk DOC in both months (Figs. 4b and 4f). The 237 correlation coefficient between a<sub>254</sub> and DOC was slightly higher in June than August, further 238 attesting to the dominance of riverine inputs to DOC dynamics in June and suggesting alternative 239 DOM sources, for example, from *in situ* production in August (Table 2).

Throughout the summer, the Fox River was a source of highly aromatic DOM, as indicated by elevated SUVA<sub>254</sub> values ranging from 2.22 to 3.77 L mg-C<sup>-1</sup> m<sup>-1</sup> (average =  $3.01\pm0.0.29$  L mg-C<sup>-1</sup> m<sup>-1</sup>) in June, and from 2.02 to 3.55 L mg-C<sup>-1</sup> m<sup>-1</sup> (average =  $2.58\pm0.34$  L mg-C<sup>-1</sup> m<sup>-1</sup>) in August. Higher SUVA<sub>254</sub> values were found in June possibly associated with a higher influx of allochthonous DOM from terrestrial and/or anthropogenic sources. A distinct south to north gradient was also apparent in both months (Figs. 4c and 4g) suggesting rapid mixing of river and bay waters.

247 Bulk DOM molecular weight in southern Green Bay in August appeared to be affected by 248 processes other than river-bay mixing, as no distinct south-to-north gradient was observed for S<sub>275-295</sub> as was seen in June (Figs 4d and 4h, respectively). Elevated S<sub>275-295</sub> values in August 249 250 were located in the same region where Chl-a was high and DO was lowest suggesting DO 251 consumption from organic matter degradation during an algal bloom decline (Figs 3h and 3e, 252 respectively). Given that DOC concentration does not follow this trend, it is likely that biological 253 activity plays a more important role in altering the composition, and therefore DOM molecular 254 weight in southeastern Green Bay later in the summer (Moran et al., 2000; Amon and Benner, 255 1996). In both sampling months, however,  $S_{275-295}$  showed irregular distributions, possibly as a 256 result of spatial heterogeneity and in situ processes such as production and microbial and 257 photochemical degradation that were non-uniformly affecting DOM molecular weight (Miller and Moran 1997). Values of  $S_{275-295}$  ranged from 0.0173 to 0.0222 nm<sup>-1</sup> (average = 258  $0.0197\pm0.0012$  nm<sup>-1</sup>) in June and from 0.0202 to 0.0292 nm<sup>-1</sup> (average =  $0.0233\pm0.002$  nm<sup>-1</sup>) in 259 260 August. Higher S<sub>275-295</sub> values in August indicate an overall lower molecular weight DOM pool. 261 This could result from a variety of factors including decreased terrestrial DOM inputs, increased photodegradation, and additional in situ DOM processing. In both months, lower S275-295 values 262

indicate the Fox River was also a source of higher molecular weight (HMW) DOM (Fig. 4d). On
the other hand, it seems that both autochthonous and degraded DOM gave rise to an overall
higher spectral slope value and thus lower molecular weight DOM pool.

266 Although DOC concentrations were not significantly different between June and August, all 267 derived optical properties, including a254, SUVA254, and S275-295, showed great differences. This 268 indicates that, while the abundance or quantity of DOM did not vary significantly, its quality and 269 composition did change over the summer. In general, the bulk DOC pool was less 270 chromophoric, less aromatic, and lower in molecular weight in August than in June. Nonchromophoric DOC comprised, on average, ~33% of bulk DOC in June while it comprised up to 271 272 47% in August. Lower non-chromophoric DOC abundance in the early summer is consistent 273 with higher riverine or terrestrial DOM inputs which in general have a higher aromaticity compared to open lake waters (Table 2) and higher average SUVA254 values in June compared to 274 August (3.01±0.29 L mg-C<sup>-1</sup> m<sup>-1</sup> vs. 2.59±0.34 L mg-C<sup>-1</sup> m<sup>-1</sup>). This is in agreement with higher 275 a<sub>254</sub> values in June and may be potentially attributed to increased photochemical degradation and 276 277 *in situ* DOM sources as indicated by higher Chl-*a* abundance in August than in June  $(2.96 \pm 1.40)$  $\mu$ g L<sup>-1</sup> vs. 2.38±0.89  $\mu$ g L<sup>-1</sup>, and Table 1). 278

# 279 **3.4. Sources of dissolved organic matter**

By using specific conductivity as a mixing index between Fox River and bay waters and Chl-*a* as a proxy for primary production, it was possible to infer the relative importance of each source on DOC dynamics throughout the summer. In June, DOC was more significantly correlated to specific conductivity ( $r^2 = 0.6443$ , *p* < 0.0001) than Chl-*a* ( $r^2 = 0.3030$ , *p* = 0.0097). In contrast, DOC in August was more significantly correlated to Chl-*a* ( $r^2 = 0.4098$ , *p* < 0.0001) than specific conductivity ( $r^2 = 0.1260$ , *p* = 0.0246), indicating a shift from predominant terrestrial DOM sources in June to more *in situ* DOM sources and/or photochemically degradedand microbially modified DOM in August.

288 There were, however, two distinct relationships between specific conductivity and bulk DOC in August (Fig. 5a). A highly linear trend ( $r^2 = 0.6443$ , p < 0.0001) existed and spatially 289 290 corresponded to what has previously been identified as the Fox River plume along the eastern coast of the bay (Fig. 5b). A separate cluster with no significant linear relationship ( $r^2 = 0.1313$ , 291 292 p = 0.2237) also existed and was spatially located along the western coast of the bay as well as 293 near Sturgeon Bay. Given that specific conductivity is elevated along the western coast of the 294 bay, one possible explanation is that river discharge along the western shore more strongly 295 influenced DOM dynamics in August. For example, in June the Oconto, Peshtigo, and 296 Menominee rivers along Green Bay's west shore collectively comprised ~58% of Fox River 297 discharge. However, in August the same rivers comprised upwards of 80% of Fox River 298 discharge (Data from USGS gauging stations). Therefore, not only was primary production more 299 influential in August, discharge from rivers other than the Fox River also played a significant 300 role in regulating DOM dynamics in August.

## 301 **3.5.** Variations in excitation-emission matrix characteristics and fluorescence indices

Fluorescence EEMs in Green Bay waters resembled typical spectra for natural DOM samples with distinct terrestrial humic-like peaks A and C and protein-like peaks T and B (Coble 2007) (Fig. 6). In June, peak A was significantly more abundant than August (p = 0.049) while peak T was significantly less abundant (p = 0.0161), indicating the increased importance of *in situ* produced DOM as summer progresses, which is in agreement with increased Chl-*a* concentrations in August (Fig. 6). However, peaks C and B were not significantly different month to month suggesting the moieties producing these peaks did not vary temporally (Table 309 3). Further, in June both terrestrial and *in situ* DOM peaks were most abundant in the Fox River.
310 In August only terrestrial humic-like peaks were most abundant in the Fox River, suggesting a
311 temporal change in riverine DOM composition. Protein-like peaks were most abundant at station
312 GB6E, located in the southeastern portion of the bay in close proximity to the Chl-*a* maximum.

313 Indices derived from fluorescence EEMs all indicate that although DOM from the Fox River 314 predominated in the bay, as the summer progressed freshly produced DOM became more 315 dominant in the bulk DOM pool. The HIX, an indicator of the degree of DOM humification 316 (Ohno 2002), was significantly higher in June (Tables 3 and 4) suggesting that terrestrial DOM 317 was more abundant early in the summer before hypoxia development, which is in agreement 318 with higher a<sub>254</sub> and discharge from the Fox River discussed above. The BIX, which indicates the 319 contribution of autochthonous or freshly produced DOM, was significantly higher in August 320 (Tables 3 and 4) indicating increased importance of *in situ* produced DOM as the summer 321 progressed and hypoxia began to develop. The FIX is also an indicator of DOM source, with 322 higher values (~1.8) indicating DOM from the extracellular release and leachate from bacteria 323 and algae and lower values ( $\sim$ 1.2) related to terrestrial DOM sources (McKnight et al. 2001). In 324 both months, the average FIX value was low (Tables 3 and 4) indicating a dominant terrestrial 325 source in Green Bay regardless of month. Yet, the FIX was still significantly higher in August 326 than June (p < 0.0001) suggesting that although terrestrial DOM was predominant, autochthonous 327 DOM was more influential to the bulk DOM pool in August and may affect DO dynamics more 328 than allochthonous DOM.

In comparison, the contribution of freshly produced DOM in Green Bay was lower than those reported for the Yangtze River Estuary, another anthropogenically stressed system, but DOM humification was greater (Sun et al., 2014 and Table 4). In relation to the rest of the

332 Laurentian Great Lakes, DOM humification in Green Bay was consistently higher than open 333 Lakes Superior, Lake Michigan, and Lake Huron (Zhou et al., 2016 and Table 4). However, it 334 was less than those in the lower Great Lakes, including Lake Erie and Lake Ontario in August 335 (Table 4). On the other hand, freshly produced DOM in Green Bay, indicated by BIX values, 336 was consistently lower than the Great Lakes as a whole (Table 4). Average FIX values were also 337 lower in Green Bay than all other Great Lakes (Table 4). Overall, DOM in Green Bay seemed to 338 contain more degraded components than open lake waters where in situ production could be 339 more dominant.

## 340 **3.6.** Variations in fluorescent DOM components derived from PARAFAC analysis

341 Four PARAFAC-derived fluorescent-DOM components were identified each sampling 342 month, with slight differences in Ex/Em wavelength (Table 5 and Fig. 7). It should be noted that, 343 because PARAFAC was run separately for June and August, the assignment of components was 344 not the same. In both sampling months, C1 and C2 were identified as terrestrial humic-like and 345 aquagenic humic-like, respectively (Guéguen et al. 2011; Coble 2007; Stedmon and Markager 346 2005; Yamashita et al. 2008; Cory and McKnight 2005). In June, C3 was identified as terrestrial-347 humic like while C4 was identified as protein-like. However, in August, the protein-like 348 component was identified as C3 while C4 was identified as terrestrial humic-like (C3 and C4 are 349 reversed in June and August) (Yamashita et al. 2008; Cory and McKnight 2005). These 350 differences are outlined in Table 5, and Fig. 7 represents the average PARAFAC components for both months, which is why C3 and C4 in June and August are reversed in Table 5. 351

Humic like DOM was more abundant early in the summer while protein –like DOM became more dominant in August indicating a shift from a predominant terrestrial source in June to a more biologically controlled system in August. In June, both terrestrial humic-like DOM

components were highly correlated with each other ( $r^2 = 964$ , p < 0.0001) (Figs. 8a and 8c) and 355 with bulk DOC ( $r^2 = 0.772$ , p < 0.0001 and  $r^2 = 0.764$ , p < 0.0001) suggesting their source was 356 357 from Fox River discharge. Elevated abundance of aquagenic humic-like DOM north of Sturgeon 358 Bay along the eastern shore suggests the presence of an alternative source other than the Fox 359 River (Fig. 8b). Conversely, the aquagenic humic-like component was not elevated north of 360 sturgeon bay in August, suggesting temporal variations in its source and composition (Figs. 8e, 361 8f, and 8h). Differences in the spatial distribution of protein-like DOM were evident between 362 June and August (Figs. 8d and 8g) and may be linked to nutrient availability regulated by Fox 363 River discharge. When Fox River discharge was high early in the summer and the plume 364 migrated farther north, protein-like DOM was abundant farther north. Later in the summer when 365 discharge decreased, protein-like DOM was most abundant in the southern bay in close 366 proximity to the Chl-a maxima attesting its biological origin.

### 367 **3.7. Behavior of DOM during river-bay water mixing**

Previous studies have reported rapid mixing of Fox River water with open Green Bay water 368 369 in the southern portion of the bay (Klump et al. 2009). As a result, high sedimentation rates occur 370 in the southern bay, removing riverine constituents from the water column at an elevated rate 371 (Klump et al. 1997). Similar patterns were observed in Green Bay in June and August, where 372 two distinct mixing zones were evident (Fig. 9). In June, DOC decreased 27% from ~545 to 373 ~400  $\mu$ M-C within zone I, which extended 29 km from the Fox River mouth (Fig. 9a). In 374 August, however, DOC decreased 47% from ~610 to ~325 µM-C within zone I which only 375 extended 19 km from the Fox River mouth, likely as a result of lower discharge indicating 376 temporal variations in mixing behavior (Fig. 9c), stemming from reduced river discharge as 377 summer progresses. In both months, DOC concentration remained fairly constant in zone II.

However, greater fluctuation was seen in zone II in June indicating a highly dynamic systemwhile DOC appeared more conservative in zone II in August.

Chromophoric DOM ( $a_{254}$ ) displayed similar trends with DOC during the summer with two distinct mixing zones, but decreases in zone I were much greater. In June,  $a_{254}$  decreased 47% from ~57 m<sup>-1</sup> to ~30 m<sup>-1</sup> in zone I (Fig. 9b). In August,  $a_{254}$  decreased almost 75% from ~60 m<sup>-1</sup> to ~19 m<sup>-1</sup> in Zone I suggesting a highly dynamic and reactive nature of CDOM during early mixing of riverine and bay waters (Fig. 9d). Similar to DOC,  $a_{254}$  values remained fairly constant in zone II in both months with higher variations in CDOM abundance in the early summer.

386 Mixing plots based on the relationship between DOC or  $a_{254}$  and specific conductivity were 387 used to estimate the apparent removal/addition of DOM and CDOM in Green Bay (Loder and 388 Reichard 1981). The hydrological cycle of Green Bay is complex and is influenced by multiple 389 riverine sources, mixing with Lake Michigan waters, and variable wind patterns (Mortimer 1978; 390 Martin et al. 1995; Waples and Klump 2002). As a result, simple two end-member mixing of 391 constituents may be difficult to model. With an average addition of less than 10%, bulk DOM 392 and a<sub>254</sub> mixing plots suggest a conservative mixing behavior of DOM although there were 393 evident DOM additions at most stations in June, likely due to light shielding by high CDOM and 394 thus lower photodegradation and *in situ* production (Figs. 10a and 10b). In August, DOM as a 395 whole appeared less conservative than in June with a removal of  $\sim 13\%$ . However, when sample 396 stations were divided into the Fox River plume and the western coast as determined by specific 397 conductivity in Fig. 5, a quasi-conservative mixing behavior of bulk DOC and CDOM was 398 observed with both addition and apparent removal at some individual stations (Figs. 10c and 399 10d). In August, CDOM removal appears to be much higher than DOC in both the Fox River 400 plume and along the western coast indicating again a higher reactivity of optically active DOM

401 components, which is further buttressed by higher non-chromophoric DOC relative to June.
402 However, when an averaged endpoint which was more representative of other open Green Bay
403 stations was used to generate the theoretical mixing line, CDOM appeared highly conservative
404 (Fig. 10).

In June, SUVA<sub>254</sub> values decreased sharply while  $S_{275-295}$  increased sharply with distance from the Fox River mouth indicating a rapid decrease in both aromaticity and mean molecular weight of the bulk DOM pool (Figs. 4c and 4d). While SUVA<sub>254</sub> showed a similar trend in August, no such trend was observed for  $S_{275-295}$  indicating other controlling factors besides mixing/dilution were affecting DOM molecular weight such as biological and photochemical degradation of DOM (Figs. 4g and 4h). Regardless, DOM composition appears to be altered the most significantly during early mixing in the southern portion of the bay.

412 The mixing behavior of individual DOM components relative to specific conductivity also 413 suggests a highly dynamic nature of the DOM pool. In June, the terrestrial humic-like DOM 414 components behaved conservatively with an average removal of only 7%, the aquagenic humic-415 like component displayed an average removal of 17%, while the protein-like component 416 displayed an addition of 15%. Bulk FDOM, however, behaved conservatively with a calculated 417 removal of <4% between all four components. This, in conjunction with the mixing behavior of 418 DOC, suggests that although bulk DOC appeared to behave conservatively, DOM composition 419 was significantly altered with different extents highlighting the dynamic nature of DOM mixing 420 and cycling in Green Bay in June. In August, all humic-like DOM components underwent 421 significant removal with an average removal of 24%, consistent with higher CDOM removal and 422 higher abundance of non-chromophoric DOC. Contrary to June, the protein-like component C3

behaved conservatively with addition of only 5% suggesting the rate of microbial degradation
was greater than the rate of *in situ* production of DOM.

425 The conservative nature of terrestrial humic-like components in June may be explained by 426 their refractory nature. The percent contribution of both C1 and C3 increased in the southern bay 427 with distance from the Fox River (Figs. 11a and 11c). Because the Fox River appears to be their 428 source, this observation can potentially be explained by the removal of other DOM components 429 relative to C1 and C3. Further, C1 and C3 normalized to DOC concentration (C1/DOC and 430 C3/DOC, respectively) both remained relatively constant until as far north as Sturgeon Bay 431 (Figs. 12a and 12c). This suggests that although DOC concentration decreased rapidly with 432 distance from the Fox River, the relative abundance of C1 and C3 did not, further buttressing 433 their similar source and refractory nature (Ishii and Boyer, 2012; Diffey, 2002). No such trend 434 was observed for the other DOM components (Figs. 11 and 12).

## 435 *4. Conclusions*

436 Dissolved organic carbon dynamics in Green Bay displayed both spatial and temporal 437 variability. Although bulk DOC concentration was not significantly different between June and 438 August, optical characteristics (i.e. a<sub>254</sub>, SUVA<sub>254</sub>, and S<sub>275-295</sub>) were, indicating the chemical 439 composition or quality of DOM was different. A significant difference in the percent of non-440 chromophoric DOC (33% in June and 47% in August) further supports this observation and 441 indicates a less aromatic and lower molecular weight DOM pool in August. Four similar 442 fluorescent DOM components were identified with PARAFAC analysis in June and August: two 443 terrestrial humic-like, one aquagenic humic-like, and one protein-like DOM. While Green Bay 444 was dominated by allochthonous DOM, in situ production was more influential in late summer 445 with more protein-like DOM presence, resulting in changes in chemical composition of the bulk 446 DOM pool. Further, *in situ* production may have influenced dissolved oxygen dynamics and 447 hypoxia development in the southern portion of the bay as a result of heterotrophic oxidation of 448 DOM in the water column. The humification index was significantly higher in June while the 449 biological and fluorescence indices were significantly higher in August. This indicates that the 450 bulk DOM pool was more humified in June and comprised of fresher DOM in August, 451 concurrent with higher concentration of Chl-*a* in August.

452 Hydrology was also a significant factor influencing both the spatial and temporal variation 453 of DOM. The dominant circulation in the bay (Hamidi et al. 2015) significantly influenced the 454 spatial distribution of DOM, deflecting DOC-rich Fox River water along the eastern coast with 455 DOC-poor open Lake Michigan waters being transported south along the western coast of the 456 bay. While bulk DOM was generally conservative during estuarine mixing, August displayed 457 unique mixing behaviors between the southern portion of the bay and the western coast, 458 influenced by a significant increase in the discharge of small rivers along the western coast 459 relative to the Fox River. Specifically, each fluorescent DOM component displayed two separate, 460 linear relationships with specific conductivity indicating the increased complexity of DOM 461 sources and sinks in August relative to June. DOM dynamics in Green Bay have unique temporal 462 variation controlled by both allochthonous organic matter loading as well as in situ production, 463 and are linked to riverine discharge, biological activity, hydrological condition, and dissolved 464 oxygen dynamics in the bay.

# 466 Acknowledgements

467 We thank Jianyu Ni, Dirk Koopmans, Shelby LaBuhn, Don Szmania, Kim Weckerly, R/V 468 Neeskay Captain Gregory Stamatelakys, Geoffrey Anderson, and Crew Members for their 469 assistance during sample collection, Peng Lin for technical assistance during sample processing 470 and analysis, and two anonymous reviewers for their constructive comments, which improved the manuscript. This work was supported in part by NOAA Sea Grant (R/HCE-16 to LG and 471 472 R/HCE-12 to JVK), the NOAA CSCOR Coastal Hypoxia Research Program (Grant NA10NOS4780139 to JVK), the Michigan Water Center, and the University of Wisconsin-473 474 Milwaukee (RGI).

# 476 **References**

- 477 Achman, D. 1993. Volatilization of polychlorinated biphenyls from Green Bay, Lake Michigan.
  478 Environ. Sci. Technol. 27: 75-87.
- 479 Amon, R, and Benner, R. 1996. Photochemical and microbial consumption of dissolved organic
- 480 carbon and dissolved oxygen in the Amazon River system. Geochim. Cosmochin. Acta. 60:481 1783-1792.
- Amon, R. and Benner, R. 1996. Bacterial utilization of different size classes of dissolved organic
   matter. Limnol. Oceanogr. 41: 41–51.
- Ankley, G. T., Lodge, K., Call, D. J., Balcer, M. D., Brooke, L. T., Cook, P.M., Kreis Jr., R. G.,
  Carlson, A. R., Johnson, R. D., Niemi, G. J., Hoke, R. A., West, C. W., Giesy, J. P., Jones,
  P.D., and Fuying, Z.C. 1992. Integrated assessment of contaminated sediments in the lower
  fox river and green bay, Wisconsin. Ecotoxicol. Environ. Saf. 23: 46–63.
- Bauer, J. E., and Bianchi, T. S. 2011. Dissolved organic carbon cycling and transformation.
  Treatise on Estuarine and Coastal Science. 5: 7–68, doi: 10.1016/B978-0-12-374711-2.005027.
- Belzile, C., Vincent, W. F., and Kumagai, M. 2002. Contribution of absorption and scattering to
  the attenuation of UV and photosynthetically available radiation in Lake Biwa. Limnol.
  Oceanogr. 47: 95–107.
- Bianchi, T.S., DiMarco, S.F., Cowan, J.H., Hetland, R.D., Chapman, P., Day, J.W. and Allison,
  M.A., 2010. The science of hypoxia in the Northern Gulf of Mexico: a review. *Science of the Total Environment*, 408(7): pp.1471-1484.
- Chari, N.V.H.K., Rao, P., and Sarma, N. 2013. Fluorescent dissolved organic matter in the
  continental shelf waters of western Bay of Bengal. J. Earth Syst. Sci. 122: 1325–1334.
- Chari, N.V.H.K., Sarma, N. S., Pandi, R. S., and Murthy, N. K. 2012. Seasonal and spatial
  constraints of fluorophores in the midwestern Bay of Bengal by PARAFAC analysis of
  excitation emission matrix spectra. Estuar. Coast. Shelf. S. 100: 162–171.
- 502 Coble, P. G. 2007. Marine optical biogeochemistry: the chemistry of ocean color. Chem. Rev.
   503 107: 402–18.
- Cory, R. M., and McKnight, D. M. 2005. Fluorescence spectroscopy reveals ubiquitous presence
   of oxidized and reduced quinones in dissolved organic matter. Environ. Sci. Technol. 39:
   8142–8149.
- 507 De Stasio, B., Schrimpf, M., and Cornwell, B. 2014. Phytoplankton communities in Green Bay,
  508 Lake Michigan after invasion by Dreissenid Mussels: increased dominance by cyanobacteria.
  509 Diversity. 6: 681–704.
- 510 Diaz, R. J., and Rosenberg, R. 2008. Spreading dead zones and consequences for marine 511 ecosystems. Science 321: 926–929.
- 512 Diffey, B. L. 2002. Sources and measurement of ultraviolet radiation. Methods: 4–13.
- Eadie, B. J., Klump, J. V., Landrum, P. F. 1992. Distribution of hydrophobic organic compounds
  between dissolved and particulate organic matter in Green Bay waters. J. Great Lakes Res. 18:
  91–97.
- 516 Egan, D. 2014. Dead zones haunt Green Bay as manure fuels algae blooms. Milwaukee J.
   517 Sentin., Sept 30. http://www.jsonline.com/news/wisconsin/dead-zones-haunt-green-bay-as 518 manure fuels algae blooms die offe b00244002r1 274684741 html
- 518 manure-fuels-algae-blooms-die-offs-b99344902z1-274684741.html
- Green, R. E., Bianchi, T. S., Dagg, M. J., Walker, N. D., and Breed, G. A. 2006. An organic
  carbon budget for the Mississippi River turbidity plume and plume contributions to air-sea
  CO2 fluxes and bottom water hypoxia. Estuar. Coast. 29: 579–597.

- Guéguen, C., Granskog, M. A., McCullough, G., Barber, D. G. 2011. Characterisation of colored
   dissolved organic matter in Hudson Bay and Hudson Strait using parallel factor analysis. J.
   Marine. Syst. 88: 423–433.
- 525 Guo, L., and Santschi, P. H. 1997. Measurements of dissolved organic carbon (DOC) in sea 526 water by high temperature combustion method. Atca. Oceanol. Sin. 16: 339–353.
- 527 Guo, L., Santschi, P. H., and Warnken, K. W., 1995. Dynamics of dissolved organic carbon 528 (DOC) in oceanic environments. Limnol. Oceanogr. 40: 1392–1403.
- Gustafsson, Ö. and Gschwend, P. M., 1997. Aquatic colloids: Concepts, definitions, and current
   challenges. Limnol. Oceanogr. 42(3), pp.519–528.
- Hamidi, S.A., H. Bravo, J. V. Klump, and J.T. Waples. 2015. The role of circulation and heat
  fluxes in the formation of stratification leading to hypoxia in Green Bay, Lake Michigan, J.
  Great Lakes Res., dx.doi.org/10.1016/j.jglr.2015.08.007
- Harris, V. A., and Christie, J. 1987. The lower Green Bay remedial action plan: nutrient and
  eutrophication management, Technical Advirsory Committee report. Publication no. WR-16787. Wisconsin Department of Natural Resources, Madison, Wis.
- Helms, J. R., Stubbins, A., Ritchie, J. D., Minor, E. C. 2008. Absorption spectral slopes and
  slope ratios as indicators of molecular weight, source, and photobleaching of chromophoric
  dissolved organic matter. Limnol. Oceanogr. 53: 955–969.
- 540 Herdendorf, C. E. 1990. Great Lakes Estuaries. Estuaries. 13: 493–503.
- Ishii, S., and Boyer, T. 2012. Behavior of reoccurring PARAFAC components in fluorescent
  dissolved organic matter in natural and engineered systems: a critical review. Environ. Sci.
  Technol. 46: 2006-2017.
- Klump, J. V., Edgington, D. N., Sager, P. E., and Robertson, D.M. 1997. Sedimentary
  phosphorus cycling and a phosphorus mass balance for the Green Bay (Lake Michigan)
  ecosystem. Can. J. Fish. Aquat. Sci. 54: 10–26.
- 547 Klump, J. V., Fitzgerald, S. A., and Waples, J. T. 2009. Benthic biogeochemical cycling, nutrient
  548 stoichiometry, and carbon and nitrogen mass balances in a eutrophic freshwater bay. Limnol.
  549 Oceanogr. 54: 692–712.
- Lathrop, R. G., Vande Castle, J. R., and Lillesand, T. M. 1990. Monitoring river plume transport
  and mesoscale circulation in Green Bay, Lake Michigan, through satellite remote sensing. J.
  Great. Lakes. Res. 16: 471–484.
- Leenheer, J. A., and Croué, J. 2003. Characterizing aquatic dissolved organic matter. *Environ. Sci. Technol.*, *37*(1): 18A-26A.
- Lin, P., Klump, J.V. and Guo, L. 2016. Dynamics of dissolved and particulate phosphorus
   influenced by seasonal hypoxia in Green Bay, Lake Michigan. Science of The Total
   *Environment*, 541: 1070-1082.
- Loder, T. C., and Reichard, R. P. 1981. The dynamics of conservative mixing in estuaries.
   Estuaries. 4: 64-69.
- Martin, S. C., Hinz, S. C., Rodgers, P. W., Bierman Jr., V. J., DePinto, J. V., and Young, T. C.
  1995. Calibration of a hydraulic transport model for Green Bay, Lake Michigan. J. Great.
  Lakes. Res. 21: 599–609.
- Matsuoka, A., Bricaud, A., Benner, R., Para, J., Sempére, R., Prieur, L., Bélanger, S., and Babin,
   M. 2012. Tracing the transport of colored dissolved organic matter in water masses of the
- 565 Southern Beaufort Sea: relationship with hydrographic characteristics. Biogeosciences, 9: 566 925–940.
- 567 McCarthy, J., and Zachara, J. 1989. Subsurface transport of contaminants. Environ. Sci. Technol.

- 568 23: 496–502.
- McKnight, D. M., Boyer, E. W., Westerhoff, P. K., Doran, P. T., Kulbe, T., and Anderson, D. T.
   2001. Spectrofluorometric characterization of dissolved organic matter for indication of
   precursor organic material and aromaticity. Limnol. Oceanogr. 46: 38–48.
- Miller, W. L., and Moran, M. A. 1997. Interaction of photochemical and microbial processes in
  the degradation of refractory dissolved organic matter from a coastal marine environment.
  Limnol. Oceanogr. 42: 1317–1324.
- Modlin, R.F., and Beeton, A.M. 1970. Dispersal of Fox River water in Green Bay, Lake
   Michigan. Proceedings of the 13<sup>th</sup> Conference of Great Lakes Research.
- Moran, M. A., Sheldon, W. M., and Zepp, R. G. 2000. Carbon loss and optical property changes
   during long-term photochemical and biological degradation of estuarine dissolved organic
   matter. Limnol. Oceanogr. 45: 1254–1264.
- Mortimer, C. H. 1978. Water movement, mixing, and transport in Green Bay, Lake Michigan.
   Univ. Wisconsin Sea Grant Institute No. WI-SG-78-234.
- O'loughlin, E. J., and Chin, Y. 2009. Quantification and characterization of dissolved organic
  carbon and iron in sedimentary porewater from Green Bay, WI, USA. Biogeochemistry. 71:
  371–386.
- Ohno, T. 2002. Fluorescence inner-filtering correction for determining the humification index of
   dissolved organic matter. Environ. Sci. Technol. 36: 742–746.
- Parlanti, E., Wörz, K., and Lamotte, G. M. 2000. Dissolved organic matter fluorescence
  spectroscopy as a tool to estimate biological activity in a coastal zone submitted to
  anthropogenic inputs. Org. Geochem. 31: 1765–1781.
- Qualls, T. M., Dolan, D. M., Reed. T., Zorn, M. E., and Kennedy, J. 2007. Analysis of the
  impacts of the zebra mussel, Dreissena polymorpha, on nutrients, water clarity, and the
  chlorophyll-phosphorus relationship in Lower Green Bay. J. Great Lakes. Res. 33: 617–626.
- Rochelle-Newall, E., and Fisher, T. 2002. Chromophoric dissolved organic matter and dissolved
   organic carbon in Chesapeake Bay. Mar. Chem. 77: 23–41.
- Santschi, P.H., Guo, L., Means, J.C. and Ravichandran, M. 1999. Natural organic matter binding
   of trace metals and trace organic contaminants in estuaries. *Biogeochemistry of Gulf of Mexico Estuaries*, pp.347-380, Wiley.
- 598 Smith, P. L., Ragotzkie, R. A., Andren, A. W., Harris, H. J. 1988. Estuary rehabilitation: the 599 Green Bay story. Oceanus. 31: 12–20.
- Stedmon, C. A., Markager, S. and Kaas, H., 2000. Optical properties and signatures of
  chromophoric dissolved organic matter (CDOM) in Danish coastal waters. Estuar. Coast.
  Shelf. S. 51: 267–278.
- 603 Stedmon, C. A., Markager, S., and Bro. R. 2003. Tracing dissolved organic matter in aquatic 604 envrionments using a new approach to fluorescence spectroscopy. Mar. Chem. 82: 239-254.
- 605 Stedmon, C., and Bro, R. 2008. Characterizing dissolved organic matter fluorescence with 606 parallel factor analysis: a tutorial. Limnol. Oceanogr. Methods. 6: 572-579.
- Stedmon, C. A., and Markager, S. 2005. Resolving the variability in dissolved organic matter
   fluorescence in a temperate estuary and its catchment using PARAFAC analysis. Limnol.
   Oceanogr. 50: 686–697.
- 610 Sun, Q., Wang, C., Wang, P., Hou, J. and Ao, Y. 2014. Absorption and fluorescence
- 611 characteristics of chromophoric dissolved organic matter in the Yangtze Estuary. Environ.
  612 Sci. Poll. Res. 21: 3460-3473.
- 613 Wisconsin Department of Natural Resources. 2012. Total maximum daily load and watershed

- 614 management plan for total phosphorus and total suspended solids in the Lower Fox River615 basin and lower Green Bay.
- Wang, Z., Liu, W., Zhao, N., Li, H, Zhang, Y., Si-Ma, W., Liu, J. 2007. Composition analysis of
  colored dissolved organic matter in Taihu Lake based on three dimension excitation-emission
  fluorescence matrix and PARAFAC model, and the potential application in water quality
  monitoring. J. Environ. Sci. 19: 787–791.
- Waples, J., and Klump, J. V. 2002. Biophysical effects of a decadal shift in summer wind
   direction over the Laurentian Great Lakes. Geophys. Res. Lett. 29: 1–4.
- Williamson, C. E., Hargreaves, B.R., Orr, P.S., Lovera, P.A. 1999. Does UV play a role in changes in predation and zooplankton community structure in acidified lakes? Limnol.
  Oceanogr. 44: 774–783.
- Ku, H., Cai, H., Yu, G. and Jiang, H. 2013. Insights into extracellular polymeric substances of
   cyanobacterium Microcystis aeruginosa using fractionation procedure and parallel factor
   analysis. *Water Res.* 47(6): 2005-2014.
- Yamashita, Y., Jaffé, R., Maie, N., Tanoue, E. 2008. Assessing the dynamics of dissolved
  organic matter (DOM) in coastal environments by excitation emission matrix fluorescence
  and parallel factor analysis (EEM-PARAFAC). Limnol. Oceanogr. 53: 1900–1908.
- Yang, L., Choi, J. H., Hur, J. 2014. Benthic flux of dissolved organic matter from lake sediment
  at different redox conditions and the possible effects of biogeochemical processes. Water Res,
  61, 97-107.
- Zhang, Y., van Dijk, M.A., Liu, M., Zhu, G., Qin, B. 2009. The contribution of phytoplankton
  degradation to chromophoric dissolved organic matter (CDOM) in eutrophic shallow lakes:
  Field and experimental evidence. Water Res. 43: 4685–4697.
- Zhou, Z., and Guo, L. 2012. Evolution of the optical properties of seawater influenced by the
  Deepwater Horizon oil spill in the Gulf of Mexico. Environ. Res. Lett, 7: 025301, doi:
  doi:10.1088/1748-9326/7/2/025301.
- Karal Zhou, Z., Guo, L., and Minor, E. C 2016. Characterization of chromophoric dissolved organic
  matter in the Laurentian Great Lakes using fluorescence EEMs and PARAFAC techniques. J.
  Great Lake Res. doi: 10.1016/j.jglr.2016.04.006.
- Zigah, P. K., Minor, E. C., and Werne, J. P. 2012. Radiocarbon and stable isotope geochemistry
   of organic and inorganic carbon in Lake Superior. Global Biogeochem. Cy. 26. GB1023,
- 648 doi:10.1029/2011GB004132.
- 649

650 Figure Captions

- Fig. 1. A map of the study area and sample locations in Green Bay, Lake Michigan in June andAugust 2014.
- **Fig. 2.** Distributions of bottom water dissolved oxygen (DO, mg L<sup>-1</sup>) in June (a) and August
- 654 2014 (b) highlighting the hypoxic zone in Green Bay that developed in August but was not655 present in June
- **Fig. 3.** Spatial distributions of dissolved oxygen (DO, mg  $L^{-1}$ ), specific conductivity ( $\mu$ S cm<sup>-1</sup>),

surface water temperature (°C), and chlorophyll-*a* (Chl-*a*,  $\mu$ g L<sup>-1</sup>) in Green Bay during June

658 (upper panels, a-d) and August (lower panels, e-h) 2014.

659 Fig. 4. Spatial distributions of dissolved organic carbon (DOC, μM-C), absorption coefficient at

660 254 nm ( $a_{254}$ , m<sup>-1</sup>), Specific UV absorbance at 254 nm (SUVA<sub>254</sub>, L mg-C<sup>-1</sup> m<sup>-1</sup>), and spectral

661 slope between 275 and 295 nm  $(S_{275-295}, nm^{-1})$  in Green Bay during June (upper panels) and 662 August (lower panels) 2014.

663 Fig 5. Linear regression between dissolved organic carbon (DOC) and specific conductivity in

August (a). White circles represent sample locations in the Fox River plume while red circles

- represent sample locations along the western coast and near Sturgeon Bay and are spatially
- 666 depicted in (b). Fox River Plume y=2.0956x-281.1078,  $r^2=0.6443$ , p<0.0001. Western

667 Coast - y=1.6178x-223.6927,  $r^2=0.1313$ , p=0.2237.

668 Fig. 6. Examples of excitation-emission matrices (EEMs) from a) the Fox River mouth, b) the

669 Chl-*a* maxima, c) bottom water (30 m) at Station GB-48, and d) surface water (1 m) at670 Station GB-17.

671 Fig. 7. Excitation-emission plots of fluorescent-DOM components identified by PARAFAC

analysis in both June and August (See Table 5 for specific Ex/Em data of each component).

In June, components 1 and 3 were terrestrial humic-like, component 2 was aquagenic humic-

674	like, and component 4 was protein-like. In August, components 1 and 4 were terrestrial
675	humic-like, component 2 was aquagenic humic-like, and component 3 was protein-like.
676	Fig. 8. Spatial distributions of PARAFAC-derived fluorescent-DOM components (in ppb-QSE)
677	in June (a-d) and August 2014 (e-h). A more detailed description of each component can be
678	found from Table 5.
679	Fig. 9. Relationship between distance from the Fox River mouth and DOC concentration in June
680	(a), distance from the Fox River mouth and $a_{254}$ in June (b), distance from the Fox River
681	mouth and DOC concentration in August (c), and distance from the Fox River mouth and $a_{254}$
682	in August (d) in Green Bay showing two distinct mixing zones. Black ovals represent Zone 1
683	and red ovals represent Zone 2.
684	Fig. 10. Relationship between dissolved organic carbon (DOC) concentration ( $\mu$ M-C) and
685	specific conductivity ( $\mu$ S cm <sup>-1</sup> ) in Green Bay in June (a), absorption coefficient at 254 nm
686	$(a_{254}, m^{-1})$ and specific conductivity in June (b) DOC and specific conductivity in August (c),
687	and $a_{254}$ and specific conductivity in August (d). F.R. = Fox River, W.C. = west coast.
688	Mixing lines were derived from the equation of line between the points of lowest and highest
689	specific conductivity. In August, the station with the lowest specific conductivity had higher
690	than usual concentrations of DOC, so an averaged endpoint was also plotted for comparison.
691	Fig. 11. Spatial distributions of the percent contribution of each fluorescent-DOM component to
692	bulk fluorescent-DOM (FDOM) in Green Bay during June (a-d) and August (e-h).
693	Fig. 12. Spatial distributions of PARAFAC-derived fluorescent-DOM components normalized to
694	DOC concentration (C <sub>i</sub> /DOC, ppb-QSE/ $\mu$ M-C) in Green Bay during June (a-d) and August
695	(e-h).
696	

**Table 1.** Sampling locations and their hydrographic parameters, including water depth, water
temperature (Temp), specific conductivity (Sp. Cond), dissolved oxygen (DO), and chlorophyll-a
(Chl-*a*) in surface waters of Green Bay during June and August 2014.

Station ID Latitude Longitude		Longitude (W)	Temp	Sp. Cond	DO	Chl-a	Water Depth
	(N)	0 ()	(°C)	$(\mu S \text{ cm}^{-1})$	$(mg L^{-1})$	$(\mu g L^{-1})$	(m)
		Juna 2014	~ /				
DoDoro Dom		Julie 2014					67
Dereie Dalli Ear Mouth	-	- 00000,00 20"	22.1	410	-	-	0.7
CD (	44 32 13.72	00 00 00.20 07%52'00 74''	22.1 10.2	410	1.23	-	7.8
CD 12 (Sto E)	44 39 20.01	8/ 33 00./4 87845'02 00''	10.5	227	11.04	3.4 2.2	5.9
OD-12 (Sta-r) Sto D	44 43 28.20	8/ 43 03.00 87°47'22 40''	16.0	327	10.11	3.3 2.2	9.9
Sla-D	44 4/ 15.55	0/4/32.49 07072,40 02"	10.9	302 205	10.37	2.2	11.0
GD-30 D CD 17	44 37 38.21	0/ 23 40.03 07015'26 71''	16.9	303 205	-	20	19.4
GD-17 CD 42	44 4/ 5/.11	8/ 43 30./4 87822,58 00.	10.3	293	10.30	2.0	15.2
OD-42	43 01 14.65	0/ 22 30.99 97941'20 00''	10./	208	11.25	5.1 2.2	50.7 15.2
GB-20 (Sta-G)	44°50 20.04	8/41 39.00	17.9	298	10.14	5.2 1.0	15.5
Sla-D	44 40 18.23	8/ 32 40.04 97922,04 27,2	1/.0	209	10.55	1.9	0.5
GB-25 (Sta-H)	44 54 25.00	8/*32 04.3/	18.3	308	10.57	4.4	23.0
GB-30	44°55 39.98	8/°26 40.98	1/.5	298	10.72	3.7	18.0
Sta-C	44*43 42.95	8/ 31 13.39	10.9	324	10.35	-	ð.ð 14.0
Sta-A	44°49 48.72	8/°40 51.95	18.1	307	10.02	1.9	14.0
GB-9 (Sta-E)	44°42°22.03°	8/°49'11.65''	15.9	298	10.56	2.4	8.1
GB-26	44°53 01.22	8/*3/ 5/.88	14.0	293	11.32	2.1	21.4
Condos	44°56°30.28°	8/°50°09.26°	12.1	300	11.14	1.2	10.2
GB-SB	44°53°42.32°	8/°24°50.33°	1/.3	309	-	-	13.0
GB-CI	45°10'01.60"	8/°16°56.14	14.9	300	11.07	-	18.6
GB-39	44°58°39.76°	8/°30°40.16	12.4	295	12.07	-	27.5
GB-47	45°04′00.24″	8/°23'00.26''	13.8	296	12.45	3.5	29.7
GB-44 (Sta-1)	45°01°32.44°	8/°28'41.06''	12./	287	12.00	2.5	32.8
GB-64 B	45°14′21.4/	8/°19′26.94′′	14.6	295	11.38	-	11.3
GB-52	45°06′3′/.96′	8/°23'02.65	13.2	286	11.70	1.8	25.1
GB-59 (Sta-J)	45°10°17.77°	8/°26°48./1	14.0	290	10.56	1.3	29.9
GB-67	45°11′58.05′	8/°26'50.6/	13.7	296	11.81	1.7	29.3
GB-48	45°04′16.12′′	87°26′47.12″	12.2	281	12.40	2.3	31.2
GB-64B (Sta K)	45°12′00.91′′	8/°15′2/./6′	14.7	-	-	-	23.8
GB-53	45°06′39.4′/′′	87°26′42.43′′	12.3	284	12.31	1.1	30.0
GB-73	45°14′38.36′′	87°23′02.61	11.1	289	13.60	1.9	31.2
Average			15.7	304	11.0	2.5	18.8
±Stdev			±2.72	±24	±1.2	±1.1	±9.1
		August 201	4				
Fox River	44°32'15.72''	88°00'08.28''		_			
GB-6E	44°39'27 53''	87°49'49 44''	22.1	318	10.8	39	46
GB-SC	44°34'29 03''	87°59'06 75''	23.4	361	14.2	5.6	7.6
GB-2E	44°36'14 58''	87°56'05 11''	22.8	332	5.6	61	43
GB-2S	44°36'10 51''	87°56'49 88''	22.0	340	11.4	3.7	5.0
GB-2W	44°35'57 63''	87°57'56 29''		-	- I I.T	-	33
GB-5	44°39'27 53''	87°49'49 44''	22.6	311	71	53	62
GB-8	44°42'25 68''	87°45'28 09''	21.0	322	10.4	3 2	73
GB-1B	44°38'31 81''	87°53'17 70''	22.9	316	65	6.6	53
GB-9	44°47'19 91''	87°49'05 62''	21.9	315	12.1	3 2	8 2
GB-10R	<u>44°41'32 00''</u>	87°50'20 96''	21.7 22.0	312	11.1	37	75
GB-16B	44°47'44 72''	87°41'44 87''	21.0	312	11.2	2.8	11.2

GB-13	44°45'02.52''	87°48'28.75''	21.7	314	11.0	3.4	-
MR-Mouth	45°05'15.34''	87°32'49.95''	22.1	336	9.3	1.2	17.2
Condos	44°46'28.25''	87°50'09.79''	21.9	308	12.8	2.4	10.1
GB-18	44°47'40.25''	87°49'14.56''	22.4	340	9.6	2.7	11.2
GB-30B	44°54'59.19''	87°27'28.51''	20.7	-	9.1	-	-
GB-12	44°44'58.52''	87°45'49.63''	21.6	309	11.5	3.4	9.6
GB-18B	44°48'26.76''	87°48'06.56''	22.3	340	10.0	5.4	11.1
GB-7	44°39'30.29''	87°56'46.46''	23.4	342	9.3	2.8	4.6
GB-17	44°47'40.13''	87°45'46.16''	21.5	308	11.6	2.2	12.2
GB-11	44°42'11.06''	87°56'15.27''	23.3	339	9.3	3.3	-
GB-25	44°52'30.46''	87°33'35.36''	20.8	296	9.3	3.1	-
GB-20	44°50'50.68''	87°37'51.79''	21.3	300	9.4	2.7	14.4
GB-31	44°55'17.99''	87°30'23.36''	21.0	339	9.5	2.5	23.7
GB-14B	44°45'25.88''	87°52'44.90''	22.7	340	9.6	4.0	6.1
GB-19	44°47'24.49''	87°52'15.44''	23.0	330	9.6	4.0	6.1
GB-32	44°55'39.63''	87°34'14.43''	21.1	304	10.8	1.7	-
GB-26S	44°49'24.32''	87°40'14.24''	21.0	305	11.3	2.4	13.2
GB-45	45°01'21.51''	87°34'20.81''	21.7	337	8.7	1.5	14.8
GB-22	44°50'18.35''	87°45'33.47''	22.0	339	9.7	3.8	13.3
GB-39	44°58'30.38''	87°30'27.45''	21.3	301	11.8	1.6	27.4
GB-44	45°01'12.50''	87°30'29.90''	21.5	300	11.4	1.5	28.0
GB-48	45°04'02.23''	87°26'38.18''	22.0	335	9.5	1.8	30.7
GB-48W	45°06'14.15''	87°31'12.02''	22.2	318	9.4	1.4	21.0
GB-30E	44°52'58.85''	87°24'52.17''	22.3	339	10.5	3.8	7.1
GB-30	44°55'37.91''	87°26'45.72''	21.4	301	13.4	2.5	17.9
GB-39-Pump	44°58'30.39''	87°30'27.46''	21.3	301	11.8	1.6	27.4
GB-38	44°58'28.02''	87°26'43.57''	23.3	318	7.3	1.3	27.4
GB-54	45°06'40.74''	87°30'18.73''	22.0	334	9.5	1.7	22.5
GB-43	45°01'14.88''	87°26'51.71''	21.2	299	11.6	1.5	31.2
GB-47	45°04'00.81''	87°22'59.11''	21.3	298	11.8	1.5	29.4
GB-42	45°01'09.16''	87°23'09.90''	21.6	300	11.7	1.6	30.7
			<b>22</b> ^	200	10.2	<b>a</b> ^	1.1.5
Average			22.0	320	10.3	3.0	14.6
±Stdev			$\pm 0.7$	±17	±1.7	±1.4	±9.2

702 Table 2. Dissolved organic carbon (DOC) and optical properties of dissolved organic matter,

703 including absorption coefficient at 254 nm (a<sub>25</sub>), specific UV absorbance at 254 nm (SUVA<sub>254</sub>),

704	and spectral slope at 275-295 nm	$(S_{275-295})$ in (	Green Bay	in June and Au	igust 2014.
		( 1, 5, 1, 5)	2		0

Station ID	DOC	a <sub>254</sub>	SUVA <sub>254</sub>	S <sub>275-295</sub>
	(µM-C)	$(m^{-1})$	$(L mg-C^{-1} m^{-1})$	$(nm^{-1})$
	J	une 2014		
DePere Dam	561	58.4	3.77	0.0173
Fox Mouth	546	56.5	3.74	0.0173
GB-6	442	35.0	2.86	0.0202
GB-12 (Sta-F)	414	33.4	2.92	0.0213
Sta-B	396	31.7	2.90	0.0194
GB-38 B	396	33.9	3.10	0.0186
GB-17	394	32.6	2.99	0.0201
GB-42	385	34.1	3.21	0.0190
GB-20 (Sta-G)	384	31.7	2.99	0.0203
Sta-D	383	31.1	2.94	0.0199
GB-25 (Sta-H)	378	32.0	3.06	0.0191
GB-30	375	32.3	3.12	0.0190
Sta-C	375	32.7	3.15	0.0192
Sta-A	374	31.1	3.01	0.0197
GB-9 (Sta-E)	374	30.4	2.94	0.0201
GB-26	371	32.1	3.13	0.0193
Condos	360	29.1	2.93	0.0197
GB-SB	354	31.2	3.19	0.0195
GB-CI	354	29.4	3.01	0.0203
GB-39	336	27.7	2.98	0.0194
GB-47	321	25.6	2.88	0.0216
GB-44 (Sta-I)	320	27.2	3.07	0.0188
GB-64 B	310	22.1	2.58	0.0213
GB-52	309	26.2	3.06	0.0185
GB-59 (Sta-J)	304	27.9	3.33	0.0188
GB-67	291	24.5	3.05	0.0207
GB-48	291	23.6	2.94	0.0189
GB-64B (Sta K)	288	22.1	2.78	0.0203
GB-53	250	17.8	2.58	0.0214
GB-73	202	12.4	2.22	0.0222
Average	361	30.5	3.01	0.0197
±Stdev	73	9.3	0.40	0.0012
	Aı	1911st 2014		
Fox River	610	59.8	3.55	0.0202
GB-6E	495	40.3	2.95	0.0237
GB-SC	488	41.5	3.08	0.0247
GB-2E	458	29.8	2.35	0.0257

GB-2S	436	37.8	3.14	0.0224
GB-2W	426	38.2	3.25	0.0215
GB-5	390	27.9	2.59	0.0220
GB-8	385	28.5	2.68	0.0212
GB-1B	365	31.0	3.08	0.0212
GB-9	358	25.7	2.59	0.0217
GB-10B	356	24.2	2.46	0.0226
GB-16B	350	26.8	2.77	0.0213
GB-13	349	24.7	2.57	0.0217
MR-Mouth	349	21.2	2.20	0.0225
Condos	345	23.8	2.49	0.0225
GB-18	344	25.9	2.73	0.0211
GB-30B	342	21.5	2.28	0.0215
GB-12	339	27.4	2.92	0.0215
GB-18B	336	23.4	2.52	0.0282
GB-7	335	26.9	2.91	0.0220
GB-17	333	24.6	2.67	0.0210
GB-11	325	24.5	2.73	0.0224
GB-25	324	21.5	2.41	0.0238
GB-20	322	25.6	2.88	0.0214
GB-31	321	18.7	2.11	0.0266
GB-14B	321	25.0	2.82	0.0226
GB-19	319	24.4	2.76	0.0228
GB-32	317	21.7	2.48	0.0226
GB-26S	316	20.9	2.40	0.0257
GB-45	313	17.9	2.08	0.0261
GB-22	312	24.5	2.84	0.0222
GB-39	312	20.8	2.41	0.0230
GB-44	311	17.6	2.05	0.0266
GB-48	311	21.4	2.49	0.0221
GB-48W	310	21.1	2.46	0.0263
GB-30E	310	21.3	2.49	0.0221
GB-30	308	22.2	2.60	0.0211
GB-39-Pump	306	20.3	2.41	0.0223
GB-38	302	16.8	2.02	0.0292
GB-54	300	18.3	2.21	0.0256
GB-43	293	19.7	2.43	0.0230
GB-47	289	16.9	2.11	0.0268
GB-42	279	17.3	2.24	0.0254
Average	349	25.3	2.59	0.0233
±Stdev	64	8.1	0.34	0.0020

Table 3. Fluorescence indices and fluorescent-DOM component abundance derived from
PARAFAC modeling based on fluorescence excitation-emission matrices for samples collected
from Green Bay during June and August 2014. FIX=fluorescence index, BIX=biological index,
and HIX=humification index. C1, C2, C3 and C4 are fluorescent-DOM component-1, 2, 3, and

711 4, respectively.

Station ID	FIX	BIX	HIX	C1	C2	C3	C4
			-		(ppb-0	QSE)	
			June 20	14			
DePere Dam	1.22	0.66	2.58	9.49	6.49	3.79	4.37
Fox Mouth	1.23	0.65	2.17	8.66	6.07	3.32	4.69
GB-6	1.19	0.69	2.08	5.95	3.61	2.31	4.05
GB-12 (Sta-F)	1.17	0.67	2.39	6.10	3.60	2.43	2.86
Sta-B	1.15	0.71	2.59	3.64	3.74	1.72	2.06
GB-38 B	1.11	0.67	2.94	4.57	3.37	2.10	2.11
GB-17	1.15	0.63	1.71	5.26	3.02	2.15	3.78
GB-42	1.15	0.69	2.96	4.48	3.61	1.82	1.98
GB-20 (Sta-G)	1.16	0.65	1.76	5.42	3.19	2.27	3.98
Sta-D	1.13	0.67	2.85	4.90	3.04	2.20	2.38
GB-25 (Sta-H)	1.14	0.68	3.11	5.46	2.68	2.20	2.01
GB-30	1.15	0.64	0.93	3.07	2.18	1.31	3.95
Sta-C	1.16	0.77	2.23	3.48	4.14	1.54	2.32
Sta-A	1.14	0.69	2.79	5.62	2.81	2.36	2.81
GB-9 (Sta-E)	1.13	0.66	2.76	5.44	2.87	2.20	2.56
GB-26	1.23	0.65	2.31	5.46	2.94	2.23	2.88
Condos	1.13	0.67	3.13	4.49	2.94	2.09	2.05
GB-SB	1.12	0.67	2.71	4.91	3.12	2.28	2.62
GB-CI	1.16	0.72	2.73	3.51	3.59	1.68	1.91
GB-39	1.13	0.65	1.30	5.19	2.85	2.06	4.57
GB-47	1.14	0.65	1.47	4.41	2.20	1.81	3.22
GB-44 (Sta-I)	1.15	0.68	1.55	4.20	2.50	1.73	3.59
GB-64 B	1.14	0.69	1.09	3.64	3.74	1.72	2.06
GB-52	1.11	0.65	3.05	4.07	3.30	1.93	1.95
GB-59 (Sta-J)	1.10	0.64	2.56	3.81	2.69	1.84	2.19
GB-67	1.12	0.72	2.31	2.46	2.58	1.26	1.55
GB-48	1.13	0.70	2.76	4.02	3.10	1.78	1.89
GB-64B (Sta K)	1.12	0.70	2.97	3.26	2.90	1.53	1.57
GB-53	1.09	0.66	1.38	3.03	1.87	1.33	2.68
GB-73	1.11	0.67	2.85	1.85	1.24	0.77	0.90
Average	1.14	0.68	2.33	4.67	3.20	2.00	2.71
±Stdev	0.04	0.03	0.65	1.60	1.03	0.58	0.99
			August 2	014			
Fox River	1.24	0.66	2.78	9.39	7.87	4.28	4.49
GB-6E	1.23	0.69	1.12	5.72	5.19	7.81	2.70

GB-SC	1.21	0.68	2.41	6.29	5.23	4.22	2.97
GB-2E	1.22	0.69	1.54	5.11	4.59	4.90	2.45
GB-2S	1.22	0.70	1.65	5.77	4.40	5.74	2.70
GB-2W	1.22	0.69	1.57	5.94	5.32	5.47	2.80
GB-5	1.18	0.73	0.97	3.47	3.28	5.78	1.67
GB-8	1.19	0.71	1.41	3.83	3.57	4.18	1.87
GB-1B	1.16	0.74	2.00	4.02	4.16	3.56	1.99
GB-9	1.17	0.75	2.25	3.39	3.44	2.59	1.69
GB-10B	1.18	0.73	1.93	4.53	3.95	3.22	2.19
GB-16B	1.17	0.72	1.44	3.51	3.32	3.78	1.75
GB-13	1.19	0.74	1.59	4.53	3.88	4.57	2.18
MR-Mouth	1.13	0.72	0.85	2.26	1.93	3.99	1.06
Condos	1.16	0.75	1.56	3.26	3.01	3.61	1.57
GB-18	1.17	0.75	1.71	3.31	3.29	3.56	1.63
GB-30B	1.16	0.73	2.58	3.05	2.66	2.07	1.49
GB-12	1.18	0.75	2.02	3.31	3.43	3.03	1.63
GB-18B	1.19	0.75	1.29	3.41	3.22	4.00	1.65
GB-7	1.18	0.74	1.77	3.46	3.20	2.77	1.68
GB-17	1.18	0.73	2.28	3.46	3.08	2.75	1.66
GB-11	1.20	0.76	3.26	3.49	3.30	1.74	1.71
GB-25	1.16	0.72	1.14	2.94	2.68	4.03	1.43
GB-20	1.18	0.73	2.61	2.54	2.84	1.63	1.28
GB-31	1.17	0.72	1.54	3.28	3.39	3.58	1.63
GB-14B	1.18	0.73	2.35	3.48	3.17	2.42	1.65
GB-19	1.17	0.74	1.70	3.60	3.25	3.01	1.76
GB-32	1.15	0.72	1.45	3.55	3.03	3.82	1.69
GB-26S	1.18	0.75	2.45	3.55	3.40	2.63	1.79
GB-45	1.16	0.73	1.53	2.55	2.75	3.07	1.32
GB-22	1.17	0.72	1.39	3.52	3.26	3.52	1.70
GB-39	1.17	0.76	1.06	1.73	1.93	3.24	0.86
GB-44	1.14	0.74	1.64	2.71	2.73	2.83	1.37
GB-48	1.17	0.70	2.04	2.87	2.66	2.61	1.44
GB-48W	1.15	0.72	1.15	2.81	2.54	3.70	1.36
GB-30E	1.16	0.77	1.58	2.70	2.84	3.14	1.36
GB-30	1.15	0.77	1.54	2.69	3.04	3.33	1.41
GB-39-Pump	1.15	0.73	1.42	2.84	2.61	3.29	1.36
GB-38	1.15	0.72	0.67	1.56	1.40	3.75	0.73
GB-54	1.16	0.72	1.05	2.91	2.75	4.23	1.46
GB-43	1.15	0.72	0.42	0.71	0.89	2.95	0.37
GB-47	1.14	0.72	1.39	2.67	2.74	3.38	1.33
GB-42	1.13	0.73	2.10	2.40	2.56	1.95	1.23
Average	1.17	0.73	1.68	3.54	3.30	3.57	1.72
±Stdev	0.03	0.02	0.59	1.45	1.14	1.16	0.67

- **Table 4.** Comparison of the values of the humification index (HIX), the biological index (BIX),
- 715 and the fluorescence index (FIX) between Green Bay, the Laurentian Great Lakes, Lake Taihu,
- and the Yangtze River estuary. Values are averages  $\pm 1$  standard deviation.

Site	HIX	BIX	FIX	Reference
Green Bay (June)	$2.33\pm0.65$	$0.68\pm0.03$	$1.14\pm0.03$	This study
Green Bay (August)	$1.68\pm0.59$	$0.73\pm0.02$	$1.17\pm0.02$	۲۵
Lake Superior	$1.13\pm0.02$	$0.91\pm0.06$	$1.14\pm0.02$	Zhou et al. (2016)
Lake Michigan	$1.28\pm0.25$	$0.89\pm0.03$	$1.19\pm0.01$	۲۵
Lake Huron	$1.23\pm0.28$	$0.90\pm0.06$	$1.21\pm0.02$	۲۵
Lake Erie	$1.98 \pm 0.59$	$0.91\pm0.02$	$1.22 \pm 0.02$	۲۵
Lake Ontario	$1.91 \pm 0.68$	$0.92\pm0.03$	$1.22 \pm 0.02$	۲۵
Lake Taihu	1.7-3.3	0.71-0.89	1.55-1.79	Yang et al. (2014)
Yangtze River Estuary	$1.10 \pm 0.39$	$1.16 \pm 0.23$	No data	Sun et al. (2014)

**Table 5**. Description of PARAFAC-derived fluorescent-DOM components in Green Bay during

June and August 2014. Note that the order of C3 and C4 are reversed between sampling months.

The assignment of component plots in Figure 7 corresponds to the order in August in this table.

June Ex/Em (nm)	August Ex/Em (nm)	Component Description
C1 – 255 (310)/440	C1 – 250 (315)/458	Similar to Peak A - UVC humic like. Dominates export from natural catchments and exported from agricultural catchments. Allochthonous.
C2 – 240 (290)/354	C2 – 240 (300)/392	Similar to Peak M – marine humic-like. May be allochthonous, autochthonous, or a result of microbial activity.
C3 – 285/508	C4 – 280/508	UVA humic like. High molecular weight and aromatic. Fluorescence resembles fulvic acid. Widespread.
C4 – 265/312	C3 – 270/316	Similar to Peak B – Tyrosine-like. May indicate amino acids, free or bound proteins, and may indicate more degraded peptide material.