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# Long-Range Transport Influence on Key Chemical Components of $PM_{2.5}$ in the Seoul Metropolitan Area, South Korea, during the Years 2012–2016

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**Abstract:** This study identified the key chemical components based on an analysis of the seasonal variations of ground level PM<sub>2.5</sub> concentrations and its major chemical constituents (sulfate, nitrate, ammonium, organic carbon, and elemental carbon) in the Seoul Metropolitan Area (SMA), over a period of five years, ranging from 2012 to 2016. It was found that the mean PM<sub>2.5</sub> concentration in the SMA was  $33.7 \ \mu g/m^3$ , while inorganic ions accounted for 53% of the total mass concentration. The component ratio of inorganic ions increased by up to 61%–63% as the daily mean PM<sub>2.5</sub> concentration increased. In spring, nitrate was the dominant component of PM<sub>2.5</sub>, accounting for 17%–32% of the monthly mean PM<sub>2.5</sub> concentrations. In order to quantify the impact of long-range transport on the SMA PM<sub>2.5</sub>, a set of sensitivity simulations with the community multiscale air-quality model was performed. Results show that the annual averaged impact of Chinese emissions on SMA PM<sub>2.5</sub> concentrations ranged from 41% to 44% during the five years. Chinese emissions' impact on SMA nitrate ranged from 50% (winter) to 67% (spring). This result exhibits that reductions in SO<sub>2</sub> and NO<sub>X</sub> emissions are crucial to alleviate the PM<sub>2.5</sub> concentration. It is expected that NO<sub>X</sub> emission reduction efforts in China will help decrease PM<sub>2.5</sub> concentrations in the SMA.

Keywords: PM<sub>2.5</sub>; SMA; chemical composition; sulfate; nitrate; long-range transport

# 1. Introduction

Ambient particulate matter with a diameter less than 2.5  $\mu$ m (PM<sub>2.5</sub>) is composed of inorganic ions, carbonaceous materials, crustal substances, metallic components, sea salt, water, and so on. Globally, carbonaceous materials are considered to be a major component of PM<sub>2.5</sub> [1,2]. However, in Northeast Asia, the concentration of inorganic ions in PM<sub>2.5</sub> is reported to be higher than that of carbonaceous materials [3–5]. The concentration of inorganic ions is affected by atmospheric physiochemical processes and its primary and precursor emissions (NO<sub>X</sub>, SO<sub>2</sub>, and NH<sub>3</sub>) [6–10]. In Northeast Asia, in particular, precursor emissions and meteorological conditions change markedly depending on the season. Thus, the concentration and composition of PM<sub>2.5</sub> show clear seasonal variations [11,12].



Exposure to ambient PM<sub>2.5</sub> is a probable cause of several serious diseases (chronic obstructive pulmonary disease, cardiovascular disease, and some cancers). This is more evident in the vulnerable sectors of the sensitive population, such as children and elders, than in the general population [13–15]. In Northeast Asia, including China, days with high PM<sub>2.5</sub> concentration occur frequently during winter and spring [16–18]. To address this, China has developed and implemented comprehensive emission reduction policies, such as the phasing out of outdated industrial sites and strengthening the industrial emission standards [19,20]. In particular, more intensive emission reduction policies were enforced during autumn and winter [21]. For example, production restrictions have been implemented for high emitting industries (i.e., steel) and emission limits have been placed on thermal power plants for the periods between October and March [21]. Furthermore, the Korean Ministry of Environment and Seoul city recently began considering stronger emissions-reduction policies during the cold season.

The Korea–United States Air Quality study (KORUS-AQ) interim report explains that secondary particle formation and growth is important to determine  $PM_{2.5}$  concentrations in South Korea [22]. Concentrations of secondary  $PM_{2.5}$  components are determined through complex processes, including emission, advection, diffusion, chemical reaction, and removal. Studies have shown that the  $PM_{2.5}$  concentrations of countries in Northeast Asia are influenced by long-range transport [23–27]. Therefore, to implement an effective control strategy for decreasing  $PM_{2.5}$  concentrations, it is necessary to quantify the major chemical components of  $PM_{2.5}$ , its seasonal variability, and its major emission sources.

The results of previous studies that analyze haze episodes in South Korea are as follows. Shin et al. [17] analyzed  $PM_{2.5}$  and its chemical composition during high  $PM_{2.5}$  events in spring 2014 for Seoul, South Korea. They found that inorganic ions in the Seoul metropolitan area (SMA) accounted for approximately 60% of  $PM_{2.5}$  by mass. According to Kim et al. [28], the sum of the inorganic ion mass fraction is 50% of the total  $PM_{2.5}$  mass when smog events in Seoul occurred from 2003 to 2004. Many other case studies on haze have reported that inorganic ions are major components of  $PM_{2.5}$  [29–31]. In a meta-analysis of  $PM_{2.5}$  concentrations in Seoul, Han and Kim [32] showed that, while the overall  $PM_{2.5}$  concentration decreased, concentrations of nitrate and ammonium tended to increase during the years 1986–2013. Trends in  $PM_{2.5}$  and nitrate concentrations may be related to recent Chinese anthropogenic emission reductions [19,33]; however, this assumption should be supported by continuous and long-term measurement and analysis of component concentrations. Han et al. [34,35] reported on the characteristics of inorganic ions in South Korea over two years; however, their analysis was conducted prior to the Chinese emission reductions. Hence, analysis using recent observational data is required.

Three-dimensional photochemical models can be used to quantify the long-range transport of air pollutants [36–39]. For example, some of the most frequently used air-quality modeling systems is the Weather Research and Forecasting (WRF) [40], Sparse Matrix Operator Kernel Emissions (SMOKE; [41]), and the Community Multiscale Air Quality (CMAQ; [42]) models [22,26]. Previous studies utilizing this system have reported that the impact of foreign emissions on South Korean PM<sub>2.5</sub> concentrations during days with high PM<sub>2.5</sub> concentrations was 60%–80% [43,44]. Furthermore, long-term studies that analyzed long-range transport of PM<sub>2.5</sub> seasonally or annually in South Korea reported that foreign emissions ranged from 40% to 70%, depending on the season [45–48]. However, few studies have considered long-term seasonal variations of PM<sub>2.5</sub> chemical components.

This study aims to examine the seasonal concentration and composition of surface (ground layer of the atmosphere)  $PM_{2.5}$  observed in the SMA and to identify the dominant chemical components during high-concentration  $PM_{2.5}$  events. To achieve this, the seasonal chemical composition of  $PM_{2.5}$  was analyzed to identify the key  $PM_{2.5}$  components on days with high  $PM_{2.5}$  concentrations based on five years of the recently collected data (2012–2016) from an SMA super site. We distinguished the influence of foreign impacts on  $PM_{2.5}$  concentrations and identified the individual major chemical components in the SMA by using a three-dimensional photochemical model. This paper is structured as follows: Section 2 describes the methods used in this study; Section 3 describes the seasonal changes

in individual PM<sub>2.5</sub> major chemical components and quantifies foreign impacts, using air-quality modeling; and, finally, Section 4 summarizes our findings.

## 2. Data and Methods

## 2.1. Surface Observation Data

Hourly data, including PM<sub>2.5</sub> composition and concentration, from the SMA super site (latitude: 37.6098, longitude: 126.9348) provided by the National Institute of Environmental Research, were used in this study. PM<sub>2.5</sub> mass concentrations were measured through the beta-ray absorption method, using a BAM1020 (MetOne Instrument Inc., Grants Pass, OR, USA). Ionic components were monitored by ion chromatography, using a URG-9000D Aerosol Ion Monitor (URG Corporation, Chapel Hill, NC, USA), and carbonaceous components were measured by using a 4F-semi-continuous carbon field analyzer (Sunset Laboratory Inc., Portland, OR, USA). Elemental carbon (EC) and organic carbon (OC) were determined by using the National Institute for Occupational Safety and Health's thermal optical transmittance method. Observation data were automatically recorded, using a semi-continuous instrument at 1-hour intervals. These automatic measurements used heated inlets to remove particle-bound water, so water content was excluded in this study [49]. Further details on the measurement techniques and the reliability of evaluation methods can be seen in Park et al. [50,51].

Because each component was measured by using a different instrument, the concentrations of some components were sometimes not measured. To analyze the relative ratios of the individual major chemical components, we only used data that contained concentrations of all five major constituents of  $PM_{2.5}$ , namely sulfate, nitrate, ammonium, OC, and EC. Our analysis shows that the composition of measured  $PM_{2.5}$  was sulfate (20%), nitrate (20%), ammonium (13%), OC (11%), and EC (5%) (Figure S1). We present a detailed description in Section 3.1. The 'unidentified substances' classification represents the components, excluding the sum of the concentrations of these five major components. Between 2012 and 2016, 58% of hourly data samples contained all of these components. Daily mean concentration was discarded unless at least 75% of the day's hourly average concentrations were available, leaving 51% (933 days) of the data for analysis. Seasonally, 45%, 42%, 52%, and 63% of winter, spring, summer, and fall data, respectively, were used (Table 1). The monthly average concentration was determined only when five or more daily mean concentrations were available. Using the selected observation data, annual and seasonal changes in  $PM_{2.5}$  ratios and chemical composition were analyzed.

	Jan	Feb	Mar	Apr	May	Jun	Jul	Aug	Sep	Oct	Nov	Dec
# of daily samples (Daily Mean)	69	60	71	67	65	80	81	78	83	107	97	75

Table 1. Available daily concentration samples for each month from 2012 to 2016.

### 2.2. Air-Quality Modeling

A three-dimensional photochemical modeling system was used for the PM<sub>2.5</sub> sensitivity simulations. WRF was used to prepare meteorological input data to be used in a chemical transport model. SMOKE and the Model of Emissions of Gases and Aerosols from Nature (MEGAN) [52] were used to process anthropogenic and biogenic emissions for air-quality modeling, respectively. Finally, CMAQ was used for chemical-transport modeling. The WRF model was configured with 36 vertical layers and used initial fields from the Final Operational Global Analysis (FNL) reanalysis provided by the National Centers for Environmental Prediction (NCEP; Table 2). WRF simulation results were converted for use in the air-quality model, using the Meteorology–Chemistry Interface Processor (MCIP) version 3.6, and the vertical layers were interpolated to 22 layers.

WRF	Description
Version	3.4.1
Initial field	FNL
Planetary boundary layer scheme	YSU
Microphysics	WSM6
Land surface model scheme	NOAH
Short wave radiation	NASA Goddard

Table 2. Details of the weather research and forecasting (WRF) model configuration used in this study.

Emission input data were divided into South Korea and Northeast Asia. For Northeast Asia, the MICS-Asia 2010 [53] emissions inventory was used. These data were applied in many air-pollutant-behavior studies in the region [26,27,54]. For South Korea, 2010 Clean Air Policy Support System (CAPSS) [55] data were used. Temporal allocation, spatial allocation, and chemical speciation were performed for each source classification code (SCC) through SMOKE, to incorporate the emission inventory into the air-quality simulation [56]. Biogenic emissions were estimated by using MEGAN. Meteorological data were used for the calculation of biogenic emissions and for vertical allocation of point sources. Air-quality simulation was performed by using CMAQ version 4.7.1. For the chemical mechanism, Statewide Air Pollution Research Center, Version 99 (SAPRC99) [57] was used, and for the aerosol module, Aerosol module version 5 (AERO5) [58] was used (Table 3). The model domain for Northeast Asia included the Korean Peninsula, China, and Japan, and used a 27 km horizontal resolution to account for the influence of long-range transport (Figure 1).

Table 3. Details of the community multiscale air-quality model (CMAQ) configuration used in this study.

CMAQ	Description		
Version	4.7.1		
Chemical Mechanism	SAPRC99		
Chemical Solver	EBI		
Aerosol Module	AERO5		
Boundary Condition	Default profile		
Advection Scheme	YAMO		
Horizontal Diffusion	Multiscale		
Vertical Diffusion	Eddy		



**Figure 1.** Community Multiscale Air Quality Model (CMAQ) domain used in this study. The gray area shows the Seoul Metropolitan Area (SMA), and the black dot represents the Bulkwang super site.

### 2.3. Model Performance Evaluation

The performance of the base simulation was evaluated by a comparison of its results with surface measured concentrations. Table S1 shows that the statistical evaluation of CMAQ model performance for PM<sub>2.5</sub> was performed by comparing the results of the air-quality modeling system (described in Section 2.2) with the observed data from the SMA super site. The performance statistics proposed by Emery et al. [59] were used as the evaluation criteria. For PM<sub>2.5</sub>, the normalized mean bias (NMB) was -11%, the normalized mean error (NME) was 30%, and the correlation coefficient (R) was 0.72, indicating that model performance is consistent with the goal level suggested by Emery et al. [59]. Sulfate and OC components were underestimated by 28% and 8%, respectively, while nitrate and EC components were overestimated by 33% and 121%, respectively. Simulations of all components satisfied the performance criteria proposed by Emery et al. [59]. At the same time, the model shows interannual variability of performance. PM<sub>2.5</sub> was underestimated for the years 2013 and 2014 (MB:  $-8.1 \,\mu$ g/m<sup>3</sup>, NMB: -22%), while the simulated PM<sub>2.5</sub> concentrations for the year 2015 were similar to observed PM<sub>2.5</sub> concentrations (MB:  $-0.6 \mu g/m^3$ , NMB: 2%). We noticed that the simulated nitrate bias had increased from 2.4  $\mu$ g/m<sup>3</sup> (2012) to 5.0  $\mu$ g/m<sup>3</sup> (2015) (Figure S2). We compared 1 h average observed values of meteorological variables with the WRF-simulated values in Northeast Asia during the years 2012 to 2016 (Table S2). Meteorological observation data were obtained from the United States National Centers for Environmental Prediction (NCEP) and Meteorological Assimilation Data Ingest System (MADIS). Overall, the simulated 2 m temperature and 10 m wind speed satisfied the statistical benchmarks that were proposed by Emery et al. [60]. According to Zheng et al. [19], Chinese  $NO_X$ and SO<sub>2</sub> emissions decreased by 23% and 53%, respectively, during the years 2012 to 2016. However, this study used a fixed emission inventory. This may affect nitrate overestimation and simulated sulfate bias reduction in recent years. Model bias changes warrant further analysis in the future, but a detailed analysis is outside the scope of this study. Figures S3 and S4 show a comparison of the simulated and observed PM<sub>2.5</sub> concentrations in China from 2015 to 2016. Surface observation data for China were obtained only in 2015 and 2016. For PM<sub>2.5</sub>, NMB, NME, and R were -4%, 13%, and 0.91, respectively. These values meet the goal level suggested by Emery et al. [59].

## 2.4. Estimation of Chinese Emission Impacts

The impact of Chinese emissions on surface  $PM_{2.5}$  concentrations in the SMA were analyzed, using the brute-force method (BFM), which measures the sensitivity of resultant concentrations by using a perturbed emission input. In addition to the air-quality simulation that used base emissions (the base simulation), an impact analysis simulation was conducted that changed the target emissions (simulation with perturbed emissions). The sensitivity factor, which represents the airborne pollutant concentration change relative to the emissions change, was then estimated, using the following equation [61]:

$$S_{p,i} = \frac{C_p - C_{p,i}}{\Delta e_i} \tag{1}$$

where  $S_{p,i}$  indicates the sensitivity factor for the emission area, *i*, and the concentration of pollutant, *p*, derived through BFM;  $C_p$  is the atmospheric concentration of nitrate, sulfate, ammonium, OC, and EC in SMA;  $C_{p,i}$  is the re-simulated concentration of emission area, *i*, after changing the emissions by e%; and  $\Delta e_i$  is the emission-reduction rate of emission area, *i*. In this study, *i* indicates Chinese sources and the emissions of all precursors were reduced simultaneously. A BFM emission-reduction rate of 50% was used to derive the sensitivity factor. This rate was determined after reviewing previous studies that used BFM to analyze the impacts of PM emissions in Northeast Asia [46,62,63]. A high reduction rate is too small [64].

The zero-out contribution (ZOC) [65] represents the change in concentration when the emission reduction rate of the target area is assumed to be 100%, using the derived sensitivity factor. ZOC was

used as the concentration impact of the target emissions in this study. The impact of Chinese emissions on the  $PM_{2.5}$  concentration in SMA was calculated by adding the ZOCs of each component, in the following manner:

$$ZOC_{p,i} = S_{p,i} \times 100 \tag{2}$$

$$ZOC_i = \sum_{p=1}^{n} (ZOC_{p,i})$$
(3)

For the air-quality simulation, it was difficult to perfectly represent the observed concentrations during the target period. When utilizing simulation results, the US Environmental Protection Agency recommends using the observed and simulated concentrations and considering the relative changes of the simulated concentrations according to emission changes [66]. Taking this into account, this study corrected the impact-analysis results, using the relative ratio of the simulated and observed concentrations. This correction is similar to the relative response factor (RRF) of the United States EPA, and it is described as the contribution correction factor (CCF) in this study [67].  $CCF_{p,d}$  is the ratio of the observed and simulated concentrations for each component (p) and was calculated by day (d), in the following way:

$$CCF_{p,d} = \frac{C(OBS)_{p,d}}{C(MOD)_{n,d}}$$
(4)

The process of correcting the emission impact using CCF is shown in Equation 5. The impact of emission area, *i*, on component *p* in daily units (*d*;  $ZOC_{i,p,d}$ ) is multiplied by  $CCF_{p,d}$ , which is determined by component and day. The corrected ZOC (Adjusted  $ZOC_{i,d}$ ) is considered to be the impact of Chinese emissions.

Adjusted 
$$ZOC_{i,d} = \sum_{p=1}^{n} (ZOC_{i,p,d} \times CCF_{p,d})$$
 (5)

This methodology to estimate the Chinese emission impact used a fixed emissions inventory and perturbed Chinese emissions. Therefore, the methodology has a limitation, in that recent changes in Chinese emissions cannot be considered.

# 3. Results and Discussion

## 3.1. Concentration and Chemical Composition of PM<sub>2.5</sub>

The average PM<sub>2.5</sub> concentration observed at the SMA super site during 2012–2016 was  $33.7 \,\mu g/m^3$ . On average, the five major constituents account for 69% of the PM<sub>2.5</sub> concentrations. Unidentified substances make up 31% of  $PM_{2.5}$  and a significant part of  $PM_{2.5}$  mass. However, the analysis of unidentified substances is challenging because previous studies have reported that the unidentified substances possibly consisted of various ion and trace metal species (e.g., Na<sup>+</sup>, K<sup>+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>, Cl<sup>-</sup>, and Fe<sup>-</sup> [5,31]), and their measurements have large uncertainties [50]. Unidentified substances are discussed further in the last part of this section. To identify the major chemical components during high PM<sub>2.5</sub> events in the SMA, daily mean PM<sub>2.5</sub> concentrations were divided into 10  $\mu$ g/m<sup>3</sup> concentration bins. The average chemical component ratios for each bin are shown in Figure 2. The number of days in the  $0-10 \ \mu g/m^3$  concentration bin was 56 days (6%), and the composition ratios of unidentified substances, inorganic ions, OC, and EC were 45%, 29%, 19%, and 7%, respectively. When the PM<sub>2.5</sub> concentration was lower than 10  $\mu$ g/m<sup>3</sup>, the unidentified substances component was high, while inorganic ions and carbonaceous components (OC + EC) were similar. When applying the South Korean OM/OC ratio (1.6–1.8) proposed in previous studies [68,69], the ratio of carbonaceous components became higher than that of inorganic ions. Previous studies that have measured the chemical composition of PM<sub>2.5</sub> globally reported that the mass concentration of OM was approximately

40% [1]. Cheng et al. [2] reported that the mass concentration of carbonaceous components was approximately 39%. Globally,  $PM_{2.5}$  contains more carbonaceous components than inorganic ions; however, in China and South Korea, when the  $PM_{2.5}$  concentration is higher than 30 µg/m<sup>3</sup>,  $PM_{2.5}$  contains more inorganic ions than carbonaceous components [2]. This indicates that the importance of carbonaceous components and inorganic ions can differ depending on  $PM_{2.5}$  concentration.

The 10–20 (210 days) and 20–30  $\mu$ g/m<sup>3</sup> (231 days) PM<sub>2.5</sub> concentration bins had the highest observation frequencies and together accounted for 47% of observation days. At these concentrations, the ratio of inorganic ions increased from 40% to 47%, whereas the ratio of carbonaceous components decreased from 20% to 18%. This change in component ratios continued until the 50–60  $\mu$ g/m<sup>3</sup> bin, and no significant differences were observed in higher concentration bins. From the 30–40  $\mu$ g/m<sup>3</sup> bin, which is higher than the national ambient air-quality standard of 35  $\mu$ g/m<sup>3</sup> for daily mean PM<sub>2.5</sub> in South Korea, the ratio of inorganic ions exceeded 50% of the total PM<sub>2.5</sub> mass, with the nitrate ratio exceeding that of sulfate. Although observations were limited, when the PM<sub>2.5</sub> concentration was 100  $\mu$ g/m<sup>3</sup>, the ratio of inorganic ions was high (63%). In summary, inorganic ions were the largest components of PM<sub>2.5</sub> in the SMA. Nitrate became predominant for the bins of PM<sub>2.5</sub> concentrations exceeding 30  $\mu$ g/m<sup>3</sup>.



(B) Relative chemical composition of PM2.5

**Figure 2.** (**A**) Frequency of observed daily mean  $PM_{2.5}$  and the mass fraction of inorganic ions for each concentration bin at the SMA super site, from 2012 to 2016. (**B**) Relative chemical composition of  $PM_{2.5}$  for each concentration bin at the same place and period as (A).

### 3.2. Seasonal Changes

Figure 3A shows the frequency distribution of PM<sub>2.5</sub> concentrations during the cold (winter and spring) and warm seasons (summer and autumn). Frequencies are shown as probabilities to account

for the varying amounts of data available in the cold and warm seasons. The mean and median  $PM_{2.5}$  concentrations in the cold season were 41.9 and 36.5 µg/m<sup>3</sup>, respectively, making it approximately 13 µg/m<sup>3</sup> higher when compared to the warm season. The 20–30 µg/m<sup>3</sup> concentration bin was the highest, at 22.9%, and the probability of a concentration over 70 µg/m<sup>3</sup> was 10% during the cold season. In contrast, a concentration of 10–20 µg/m<sup>3</sup> PM<sub>2.5</sub> was most likely (31%) during the warm season, and more than 90% of the observed concentrations were under 50 µg/m<sup>3</sup>.



Figure 3. Occurrence probability of (A)  $PM_{2.5}$ , (B) nitrate, and (C) sulfate for each air pollutant concentration bin at the SMA super site from 2012 to 2016. Blue and red lines represent the cold (winter and spring) and warm seasons (summer and autumn), respectively.

Figure 3B,C shows the frequency distribution of nitrate and sulfate concentrations during the cold and warm seasons. The mean concentrations of nitrate and sulfate were 9.4  $\mu$ g/m<sup>3</sup> (23%) and 7.6  $\mu$ g/m<sup>3</sup> (19%) respectively, in the cold season. Notably, the nitrate concentration in the cold season was twice that of the warm season. Conversely, the sulfate concentration was 6.3  $\mu$ g/m<sup>3</sup> (23%) during the warm season, higher by 1.6  $\mu$ g/m<sup>3</sup> as compared to nitrate. Figure 4 shows the correlation of nitrate and sulfate concentrations, with the daily mean  $PM_{2.5}$  concentration. It was found that, in the cold season, nitrate concentrations tend to be higher than sulfate concentrations for the same  $PM_{2.5}$  concentration, while in the warm season, this tendency was reversed. This clearly represents the seasonality in SMA of sulfate and nitrate, the major PM<sub>2.5</sub> components. It is possible that, during the warm season, sulfate formation was enhanced due to temperature and relative humidity increase [70,71], while nitrate could be dissociated and evaporated at high temperatures [17,70,72]. Conversely, in the cold season, nitrate formation could be enhanced, and the nitrate concentration would remain high due to meteorological conditions facilitating the accumulation and conversion of air pollutants. Previous studies for measurements of PM<sub>2.5</sub> and its chemical components in the SMA also showed the highest nitrate concentrations in spring and winter [4,16,35]. At the same time, nitrate concentrations were generally highest in winter and autumn in observational studies conducted in China [73–76]. The seasonal variances of nitrate concentrations in SMA seem to be influenced by regional characteristics of SMA.

Overall, the results suggest that managing inorganic ions is an important factor in controlling SMA  $PM_{2.5}$  concentrations. From an emission-control standpoint, this means that ionic aerosol precursors such as  $NO_X$ ,  $SO_2$ , and  $NH_3$  should be reduced. In particular, Figure 3A shows that the nitrate ratio is high during the cold season, when high  $PM_{2.5}$  concentrations are frequently observed in the SMA. Hence,  $NO_X$  emission control should be given a high priority for more effective control of annual  $PM_{2.5}$  concentrations and occurrence of days with high  $PM_{2.5}$  concentration over the region.



**Figure 4.** Scatter plots of (**A**) seasonal observed daily mean nitrate versus  $PM_{2.5}$  and (**B**) seasonal sulfate versus  $PM_{2.5}$  for each season at the SMA super site, from 2012 to 2016. Blue and red lines represent the cold (winter and spring) and warm seasons (summer and autumn), respectively.

In this section, the seasonal concentration and chemical composition of  $PM_{2.5}$  are analyzed to determine how the composition of the particulates changed with concentration. However, there is uncertainty with the observed data with respect to mass closure, where the total mass of  $PM_{2.5}$  differs by approximately 30% from the mass of the major chemical components (nitrate, sulfate, ammonium, EC, and OC). Possible reasons for the discrepancy would be the missing measurements for some chemical components and uncertainties in automatic measurements. This shows that the quality of the observational data can affect the veracity of the research results. Park et al. [50] evaluated the reliability of automated hourly mean measurements by comparing them with 24 h filter-measurement results at a different super site, using the measurement methods employed in this study. They found underestimations in automated hourly mean measurements of 10%, 24%, 17%, and 17% for sulfate, nitrate, OC, and EC, respectively. If we assume that the observed data used in this study have a similar level of uncertainty, then the composition ratio of inorganic ions could be further increased, especially considering that Park et al. [50] found that the underestimation of nitrate was the most significant. Consequently, this uncertainty is unlikely to impact the results of this study, and because the underestimation of nitrate is the highest, the importance of NO<sub>X</sub> emission control is still valid.

## 3.3. Seasonal Chinese Emission Impacts on PM<sub>2.5</sub>

Previous studies have shown that the long-range transport of air pollutants in Northeast Asia affects  $PM_{2.5}$  concentrations [23–26,54]. When determining the major sources and impact estimation for the seasonal management of  $PM_{2.5}$  in South Korea, long-range transport impacts need to be quantified according to the season and chemical components. Simulation results in Figure 5A show that the period (2012–2016) mean impact of Chinese emissions on SMA  $PM_{2.5}$  concentrations was approximately 43% (14.5 µg/m<sup>3</sup>) and ranged between 41% (12.1 µg/m<sup>3</sup>; 2012) and 44% (17.9 µg/m<sup>3</sup>; 2014). The concentrations and impact were adjusted by CCF, as described in Section 2.4. In 2014, the annual mean SMA  $PM_{2.5}$  concentration was 40.7 µg/m<sup>3</sup>, the highest in the analysis period. Annual data for 2015 were excluded because of lack of data.

The estimated monthly mean impact of Chinese emissions on SMA  $PM_{2.5}$  ranged from 3 to 43  $\mu$ g/m<sup>3</sup>. Chinese emission impacts increased in winter and spring possibly due to the prevailing

winds of the northwest monsoon transporting air pollutants from upwind areas [46]. Figure 5B shows that the impact of Chinese emissions on daily mean  $PM_{2.5}$  concentrations in winter and spring were 16.3 µg/m<sup>3</sup> (38%) and 20.2 µg/m<sup>3</sup> (50%), respectively, decreasing to 13.6 µg/m<sup>3</sup> (47%) and 10.1 µg/m<sup>3</sup> (39%) in summer and autumn, respectively. In summer, the mass concentration impact of Chinese emissions is low; however, the relative impact (47%) is higher than in winter. Concentrations of the major chemical components changed seasonally (Figures 2 and 3). As described above, inorganic ions are the dominant components of PM<sub>2.5</sub> in SMA. Moreover, the relative importance of the carbonaceous components decreased as the PM<sub>2.5</sub> concentration increased. Given this, we focused on the analysis of the Chinese emission impact on inorganic ions.



**Figure 5.** (A) Time series of monthly average  $PM_{2.5}$  concentration and the Chinese emission impact (adjusted by CCF). (B) Scatter plots of daily  $PM_{2.5}$  concentration and Chinese emission impact (adjusted by CCF) for each season.

Figure 6 shows that the seasonal nitrate concentration ranged from 4.9 (summer) to 9.6  $\mu$ g/m<sup>3</sup> (winter) and that the Chinese impact on nitrate concentrations ranged from 50% (winter) to 67% (spring). During the spring, when the nitrate concentration was the highest, the impact of Chinese emissions on nitrate was 55%–74%, corresponding to 11%–24% of the PM<sub>2.5</sub> mass concentration. Meanwhile, seasonal SMA sulfate concentrations ranged from 4.4 (autumn) and 8.7  $\mu$ g/m<sup>3</sup> (summer), and the impact of Chinese emissions was the highest in summer (58%) and lowest in winter (30%). It has been reported that winter and spring seasons have conditions favorable to the transport of air pollutants

from China to South Korea [77,78]. Nevertheless, the Chinese impact on sulfate concentrations in winter was lower than that in summer. Figure S5 shows that the Chinese impact on a total S (defined as a sum of S mass in SO<sub>2</sub> and S mass in sulfate) was higher in winter than in summer. Chinese impact on a total S and S mass in sulfate show different seasonal variations: higher for SO<sub>2</sub> than for sulfate in the winter season. It may be caused by the (S in sulfate)/(total S) ratio being lower in winter than in summer (Figure 7). Consequently, low Chinese impact on sulfate concentration is evident during the cold months. Previous studies have reported SO<sub>2</sub>-to-sulfate conversion is slow due to limited liquid-and gas-phase oxidation during winter [79,80].



**Figure 6.** Seasonal Chinese emission impact on daily mean (**A**) nitrate and (**B**) sulfate concentrations in the SMA, from 2012 to 2016. Chinese emission impact was adjusted by CCF, as described in Section 2.4.



**Figure 7.** Spatial distribution of ratios of vertical column densities (VCDs) of S in sulfate to total S VCDs for Northeast Asia during (**A**) summer (June 2016–August 2016) and (**B**) winter (December 2015–February 2016).

Figure 8 shows the relative impacts of Chinese emissions based on daily mean  $PM_{2.5}$  concentration in the SMA. Regardless of seasonality, as daily mean  $PM_{2.5}$  concentration increased, the impact of Chinese emissions increased, as well. A similar pattern is apparent for nitrate and sulfate. Below concentrations of 10 µg/m<sup>3</sup>, the Chinese emission impact is 22% of  $PM_{2.5}$  mass concentration and the impact on inorganic ions is 13% of  $PM_{2.5}$  mass concentration. In contrast, at concentrations of 60–70  $\mu$ g/m<sup>3</sup>, the Chinese emission impact on SMA PM<sub>2.5</sub> concentration increases to 49%, and the impact on inorganic ion concentration is 21.9  $\mu$ g/m<sup>3</sup>, corresponding to 34% of the PM<sub>2.5</sub> mass concentration. During high PM<sub>2.5</sub> events, the impact of Chinese emissions on nitrate concentrations is the most significant, suggesting that Chinese NO<sub>X</sub> emissions can be a major factor driving high concentrations of PM<sub>2.5</sub> in the SMA. NO<sub>X</sub> in the atmosphere has short residence time and is converted to HNO<sub>3</sub>, PAN (Peroxyl Acetyl Nitrate), HONO (nitrous acid), and so on. It is known that nitrate or PAN could be transported over long distances [81,82]. In order to quantify the influence of Chinese emission for SMA, we calculated Chinese impact of total N (defined as a sum of N mass concentrations in NO<sub>X</sub>, NO<sub>Z</sub>, and nitrate). Previous studies for PM<sub>2.5</sub> transboundary influence have reported that PM<sub>2.5</sub> is often transported total N inflow, we present the total N vertical column densities (VCDs) in this study. VCDs were calculated by integrating the number of total N molecules for all the vertical layers. Figure 9 shows the Chinese emission impact on total N VCDs in the SMA. Approximately 14 % of total N VCDs was influenced by Chinese emissions. In winter and spring, Chinese emission impacts on SMA total N VCDs ranged from 15% to 20%.



**Figure 8.** Mean impact of Chinese inorganic ions and the relative impact from Chinese emission for each PM<sub>2.5</sub> concentration bin.



**Figure 9.** Spatial distribution of total N VCDs and Chinese impact on total N VCDs for Northeast Asia from 2012 to 2016 (**top**). Further, time series of monthly mean total N VCDs and Chinese impact on total N VCDs in the SMA during this period (**bottom**).

The composition ratio of nitrate in  $PM_{2.5}$  is currently increasing due to reduced  $SO_2$  emissions in Northeast Asia [85,86]; hence,  $NO_X$  emission controls in Northeast Asia are expected to become important for  $PM_{2.5}$  management in the SMA.

## 4. Study Limitations

This study used a CCF, which corrects simulation results based on observed data, to estimate Chinese emission impacts. While a CCF has the advantage of correcting simulation results and analyzing observed concentrations through relatively simple post-processing, it is challenging to improve fundamental errors in the simulation results. When Chinese emission impacts for sulfate and nitrate concentrations were compared before and after applying the CCF, the ZOCs for sulfate and nitrate increased 1.4 times and 0.8 times, respectively (Table 4). In contrast, the relative impact of each component differed by 1%–2% before and after applying the CCF. Care must be taken when applying the CCF, because it can play an important role in prioritizing the precursors that need to be reduced first to decrease PM<sub>2.5</sub> concentrations. However, it does not have a significant influence on estimating the relative impact of foreign emissions for single components. Given this, it is more advisable to interpret our results in terms of relative ratios rather than the mass concentration.

Recent changes in Chinese emissions can cause uncertainty in air-quality simulation. In this study, fixed emission input data were used for air-quality simulation. However, according to a recent report, Chinese SO<sub>2</sub> and NO<sub>X</sub> emissions have decreased by 53% and 23%, respectively, during the period of 2012 to 2016 [19], and this decrease was marked in Eastern China which, is close to South Korea [87]. Therefore, it is possible that the impact of Chinese emissions estimated in this study could be overestimated. An additional emission impact analysis that reflects current Chinese emission reduction trends will be necessary in the future.

	Chinese Emission Impact	Sulfate	Nitrate
(a) Base case	Concentration (µg/m <sup>3</sup> )	2.3	5.1
	Relative (%)	46.2	57.8
(b) CCF case	Concentration (µg/m <sup>3</sup> )	3.2	3.9
	Relative (%)	48.7	58.4

**Table 4.** Chinese emission impact to Seoul Metropolitan Area (SMA) sulfate and nitrate concentrations (a: not adjusted; b: adjusted using the CCF).

# 5. Conclusions

This study analyzed seasonal changes in the concentration and composition ratio of  $PM_{2.5}$  by using observational data from the SMA super site during the years 2012 to 2016. The impact of Chinese emissions on  $PM_{2.5}$  concentration and composition in SMA through long-range transport were quantified by using a three-dimensional chemistry transport model (CMAQ) and sensitivity simulations.

The mean  $PM_{2.5}$  mass concentration was 33.7 µg/m<sup>3</sup>, with inorganic ions comprising 53%. When  $PM_{2.5}$  concentration increased above 60 µg/m<sup>3</sup>, the ratio of inorganic ions increased up to 63%, suggesting that the role of inorganic ions is important during high  $PM_{2.5}$  concentration events. The highest  $PM_{2.5}$  concentrations were observed during winter and spring.  $PM_{2.5}$  concentrations ranged from 30.0 to 59.8 µg/m<sup>3</sup> for the study period. During winter and spring, nitrate became the dominant chemical component, and its concentrations were 16%–26% (winter) and 17%–32% (spring) of the total  $PM_{2.5}$  mass concentrations. Consequently, NO<sub>X</sub> emission controls will be crucial for reducing  $PM_{2.5}$  concentration during the cold season (winter and spring).

The impact of Chinese emissions on  $PM_{2.5}$  concentrations were estimated to identify the sources of  $PM_{2.5}$  in the SMA. According to CMAQ results, Chinese emissions impact was 42% (40%–44%) of  $PM_{2.5}$  concentrations in the SMA. The impact of Chinese emissions on  $PM_{2.5}$  concentration and the composition varied seasonally. Chinese emissions impacts were 16.2 µg/m<sup>3</sup> (38%) and 19.7 µg/m<sup>3</sup>

(48%) of SMA PM<sub>2.5</sub> concentrations during winter and spring, respectively, and 13.0  $\mu$ g/m<sup>3</sup> (45%) and 9.5  $\mu$ g/m<sup>3</sup> (36%) during summer and autumn, respectively. The relative impact of Chinese emissions to PM<sub>2.5</sub> concentration in summer was approximately 7% higher as compared to winter. The Chinese influence on nitrate and sulfate concentrations, the major PM<sub>2.5</sub> chemical components, was also seasonally affected. Nitrate concentrations due to Chinese emissions were high in winter (2.7–7.6  $\mu$ g/m<sup>3</sup>; 44%–63%) and spring (4.3–10.3  $\mu$ g/m<sup>3</sup>; 55%–74%), while sulfate concentrations were high in spring (1.2–6.0  $\mu$ g/m<sup>3</sup>; 25%–56%) and summer (1.1–12.5  $\mu$ g/m<sup>3</sup>; 26%–70%).

The results of this study can be used to establish effective emission control strategies for  $PM_{2.5}$  concentration reduction. For high  $PM_{2.5}$  cases (exceeding a daily average concentration of 10  $\mu$ g/m<sup>3</sup>), emission control of inorganic aerosol precursors over Northeast Asia is necessary. Considering recent SO<sub>2</sub> emission reduction in Northeast Asia, reduction of NO<sub>X</sub> emissions in the region is a crucial factor driving  $PM_{2.5}$  concentrations in the SMA. A future work is warranted to verify the effect of changes in anthropogenic emissions over Northeast Asia on long-range transport and air-quality improvement.

**Supplementary Materials:** The following are available online at http://www.mdpi.com/2073-4433/11/1/48/s1: Table S1: Statistics of CMAQ model performance evaluation for daily mean PM<sub>2.5</sub> concentration and chemical composition at the SMA super site for the period of 2012-2016.; Table S2: Statistics of WRF model performance evaluation for 1 hr average temperature and wind speed over Northeast Asia in modeling domain for the period of 2012-2016. Figure S1: PM<sub>2.5</sub> chemical composition observed at the SMA super site for the period of 2012-2016.; Figure S2: Time series (left) and scatter (right) of monthly mean PM<sub>2.5</sub> and its components at the SMA super site from 2012 to 2016. Black circles and red lines present the observed and modeled concentrations, respectively.; Figure S3: Spatial tile plot of simulated period mean PM<sub>2.5</sub> concentrations overlaid by the observations at surface air quality monitoring sites in China during the period of 2015-2016. Circles represent the locations of air quality monitoring sites in China for the period of 2015 to 2016. Black and red line presents the observed and modeled concentrations, respectively.; Figure S5: Seasonal Chinese impacts to Total S, SO<sub>2</sub>, and sulfate concentration estimated over the SMA for the period of 2015–2016. The percentages represent SO<sub>2</sub> /Total S ratio and sulfate /Total S ratio.

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**Conflicts of Interest:** The scientific results and conclusions, as well as any views or opinions expressed herein, are those of the author(s) and do not necessarily reflect the views of NOAA or the Department of Commerce.

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