RESEARCH ARTICLE



Seasonal and geographical variability of nitryl chloride and its precursors in Northern Europe

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Measurements of nitryl chloride (ClNO₂) and its precursors (O₃, NO₂, particulate chloride) were made in 2014-2016 at three contrasting locations in the United Kingdom: Leicester, Penlee Point and Weybourne. ClNO₂ was observed at all sites and in every season, with the highest concentrations between 00:00 and 04:00 GMT. The median nocturnal concentration of ClNO₂ ranged between the detection limit (4.2 ppt) and 139 ppt. A clear seasonal cycle, with maxima in spring and winter, and significant differences between locations in the same season were observed. The main source of particulate chloride was sea salt aerosol (including at Leicester, ~ 200 km from the coast). In general, ClNO₂ levels were controlled by the concentrations of O₃ and NO₂, rather than by the uptake and reaction of N₂O₅ with particulate chloride. Under these conditions, the seasonality and geographical distribution of ClNO₂ can be explained in terms of O₃-limited and NO₂-limited regimes affecting the formation of the N₂O₅ precursor. A global version of the GEOS-Chem model at medium resolution ($2^{\circ} \times 2.5^{\circ}$) was not able to fully capture the observed seasonality of ClNO₂, mostly because the model overestimated the concentrations of the precursors, particularly of nocturnal O₃. A higher-resolution $(0.25^{\circ} \times 0.3125^{\circ})$ version of GEOS-Chem showed better agreement with the observations, although it still overestimated ClNO₂ concentrations during summer.

KEYWORDS

chlorine, ClNO₂, nitryl chloride, ozone, seasonality, variability

1 | INTRODUCTION

Chlorine atoms (Cl) are highly reactive in the atmosphere and affect several atmospheric chemical processes: the oxidation of organic compounds (including methane, a key climate forcer), the formation of tropospheric ozone and the cycling of sulphur and nitrogen (Saiz-Lopez and von Glasow, 2012).

One of the mechanisms by which unreactive particulate chloride (Cl⁻) can be converted into reactive gas-phase

chlorine is via the nocturnal formation of nitryl chloride (ClNO₂), followed by its photolysis at sunrise to yield Cl and NO₂ (1). ClNO₂ is formed from the reaction of N₂O₅ on chloride-containing particles (5) (Finlayson-Pitts *et al.*, 1989; Behnke *et al.*, 1994; Finlayson-Pitts, 2003; Roberts *et al.*, 2008; 2009):

$$ClNO_2 \xrightarrow{hv} Cl + NO_2 \tag{1}$$

$$O_3 + NO_2 \rightarrow NO_3 + O_2 \tag{2}$$

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$$NO_3 + NO_2 \rightleftharpoons N_2O_5 \tag{3}$$

$$N_2O_5 + H_2O_{(aq)} \rightarrow 2 HNO_3$$
 (4)

$$N_2O_5 + H_2O_{(aq)} + Cl_{(aq)}^- \rightarrow HNO_3 + ClNO_2 + OH_{(aq)}^-$$
 (5)

The chemical equilibrium between NO₃ and N₂O₅ (Brown et al., 2003) and the short lifetime of NO₃ during the day, due to its rapid photolysis rate and reaction with NO, dictate that reactions 3-5 are important only during the night. Except for slow dry deposition, CINO₂ losses are thought to be insignificant at night (Roberts et al., 2008; Kim et al., 2014). During the day, ClNO₂ photolyses, with a lifetime of \sim 40 min (at midday, 52°N), resulting in a diel pattern with night-time maxima and daytime minima. The conversion of N_2O_5 to $ClNO_2$ (5) has been shown to occur even at low concentrations of particulate chloride (Roberts et al., 2009). The main source of chloride in the troposphere is sea salt aerosol, although sub-micron particles can also contain Cl following, for example, uptake of HCl from anthropogenic sources (Thornton et al., 2010; Mielke et al., 2011). In addition, saline soils and road salt can be locally important sources of chloride (Jordan et al., 2015).

The formation of ClNO₂ reduces the formation of HNO₃ (due to the competition between 4 and 5) and therefore slows the overall loss of nitrogen oxides via HNO₃ deposition (Brown *et al.*, 2006), leaving more NO₂ available for ozone formation in the morning; the photolysis of ClNO₂ releases Cl atoms (1), enhancing the oxidation of organic compounds and ozone production (Osthoff *et al.*, 2008; Simon *et al.*, 2009; Sarwar *et al.*, 2012; Riedel *et al.*, 2014; Young *et al.*, 2014; Wang *et al.*, 2016).

Observations of CINO₂ in the troposphere have been reported over the last decade by several research groups, mostly in North America (Osthoff *et al.*, 2008; Kercher *et al.*, 2009; Thornton *et al.*, 2010; Mielke *et al.*, 2011; 2013; Riedel *et al.*, 2012; 2013; Young *et al.*, 2012; Kim *et al.*, 2014; Mielke *et al.*, 2016), with only a few in Europe (Phillips *et al.*, 2012; Bannan *et al.*, 2015; 2017) and in Asia (Tham *et al.*, 2014; Tham *et al.*, 2016; Wang *et al.*, 2016). These studies have shown that ClNO₂ is widely distributed in the troposphere, even away from the coast, at concentrations ranging from a few hundred ppt to several ppb.

This paper presents the first seasonally complete set of CINO_2 measurements in Europe, drawn from three sites across the United Kingdom. The observations were used, together with modelling tools, to investigate the spatial and temporal variability of CINO_2 in the United Kingdom and, by extension, Northern Europe.

2 | METHODS

2.1 | Fieldwork

Measurements of ClNO₂, O₃, NO₂ and aerosol composition were taken during seven periods between 2014 and 2016 at three locations in the United Kingdom:

- 1. Leicester (52°38′N, 01°08′W) is a middle-sized city (pop: ~330,000) in central England, ~200 km from the ocean. The measurements were made at two sites on the University campus: the Department of Chemistry and the local AURN station (Automatic Urban and Rural Network, http://uk-air.defra.gov.uk/networks/site-info? uka_id=UKA00573), which is classified as an urban background site (Hama *et al.*, 2017). The two sites are ~400 m apart and, for this work, were analysed together.
- 2. Penlee Point (50°19′N, 04°11′W) is on the southwestern coast of the United Kingdom, ∼6 km southwest of the city of Plymouth (pop: ~250,000). The Penlee Point Atmospheric Observatory (PPAO, http://www.westernchannelobservatory.org.uk/penlee/) is located ~11 m above mean sea level and ~30 m from the shoreline. The site receives unpolluted air from the Atlantic Ocean during prevailing southwesterly conditions (~50% of the time, Yang *et al.*, 2016).
- 3. Weybourne (52°57′N, 01°07′E) is on the eastern coast of the United Kingdom, ~180 km northeast of London. The Weybourne Atmospheric Observatory (WAO, http://weybourne.uea.ac.uk/) is located ~150 m from the shoreline and receives clean air masses from the North Sea, as well as continental outflow from the United Kingdom and/or Northern Europe (Penkett et al., 1999).

The locations and dates of the measurements periods cover a variety of chemical conditions and geographical areas, from unpolluted coastal to urban environments. A map of the study region is given in the Supplementary Information (SI. 1, Supporting Information), and a summary of the dataset is shown in Table 1.

2.2 | Instruments

ClNO₂ was measured by Chemical Ionisation Mass Spectrometry (CIMS). The Leicester CIMS instrument (THS Instruments LLC, GA) was operated in negative ion mode, with a configuration similar to Liao et al. (2011), using iodide (I⁻) as the reagent ion. ClNO₂ was detected as the cluster ion $[I.CINO_2]^-$ at m/z = 208 and 210 amu, as in Osthoff et al. (2008) and Thaler et al. (2011). The instrument was calibrated by generating ClNO₂ with a flow of Cl₂ (5 ppm in N₂) over a bed of sodium nitrite (NaNO₂) and NaCl, similar to Thaler et al. (2011). Quantitative determination of the amount of ClNO₂ produced was obtained by thermally decomposing ClNO₂ to NO₂ at \sim 350 °C and measuring NO₂ by broadband cavity-enhanced spectroscopy (Thalman et al., 2015). During fieldwork, the CIMS instrument's background signal was determined every hour by diverting the sample flow through a heated (~175 °C) stainless steel coil, which destroys all ClNO₂ in the sample. The

TABLE 1 Date, season and location of the measurement periods. Median, interquartile range and maximum measured concentrations of ClNO₂, NO₂, O₃ (15-min average between 00:00–04:00 GMT) and particulate Cl⁻ (24 h average) during each measurement period

| Date Year Season | 3–11 Mar 2014 Spring | 4–28 Aug 2014 Summer | 11–19 Dec 2014 Winter | 20 Apr-8 May 2015 Spring | 26 Jun-3 Aug 2015 Summer | 22–29 Sep 2015 Autumn | 1–26 Feb 2016 Winter |
|------------------------|----------------------------|--|-----------------------------|--------------------------------|--------------------------------|-----------------------------|----------------------------|
| Location | Leicester | Leicester | Leicester | Penlee Point | Weybourne | Leicester | Leicester |
| CINO ₂ | 110 | 15.4 | 50.5 | <dl<sup>a</dl<sup> | 21.5 | 27.7 | 139 |
| (ppt) | 50.1-150 | <dl-31.2a< td=""><td>23.4–100</td><td><dl-6.3<sup>a</dl-6.3<sup></td><td>9.1-67.4</td><td>18.7-41.5</td><td>5.5-281</td></dl-31.2a<> | 23.4–100 | <dl-6.3<sup>a</dl-6.3<sup> | 9.1-67.4 | 18.7-41.5 | 5.5-281 |
| | 274 | 74.2 | 248 | 922 | 1,100 | 75.6 | 733 |
| NO_2 | 9.0 | 8.6 | 6.8 | 0.9 | 2.7 | 13.1 | 7.9 |
| (ppb) | 5.8-21.6 | 4.4–14.1 | 5.3-11.4 | 0.4-1.9 | 1.6-3.8 | 6.0-19.6 | 5.1-20.5 |
| | 33.5 | 26.4 | 38.3 | 6.3 | 9.4 | 32.9 | 43.4 |
| O_3 | 24.2 | 17.0 | 23.2 | 39.0 | 26.7 | 8.4 | 22.1 |
| (ppb) | 15.7–28.9 | 11.4–21.3 | 18.6-29.0 | 32.5-42.8 | 22.3-32.3 | 3.3-15.7 | 11.3-29.6 |
| | 35.7 | 30.3 | 34.9 | 50.3 | 82.7 | 26.7 | 39.8 |
| Cl ^{-c} | 0.7 | <dl<sup>b</dl<sup> | 1.3 | _ | 2.3 | _ | _ |
| $(\mu g/m^3)$ | 0.4-1.2 | <dl-0.6<sup>b</dl-0.6<sup> | 1.1-1.5 | _ | 1.5-4.8 | _ | _ |
| | 2.6 | 2.0 | 1.6 | _ | 22.2 | _ | _ |

^a Detection limit (DL) = 4.2 ppt.

1-min detection limit for ClNO₂ was 4.2 ppt at 2σ (5.1 ppt at 3σ), with 18% accuracy and 14% precision.

O₃ was measured by UV absorption in Leicester (Model T400, Teledyne Technologies Inc., City of Industry, CA, Hama et al., 2017), in Penlee Point (Model 205, 2B Technologies, Boulder, CO, Yang et al., 2016) and in Weybourne (Model 49i, Thermo Fisher Scientific Inc., Franklin, MA). NO₂ was measured at Leicester and in Penlee Point with a Teledyne Model T200 NO/NO₂/NO_x analyser, using chemiluminescence detection with a molybdenum converter (Hama et al., 2017). This type of instrument is known to be subject to interference from PAN and other NO_v species (Reed et al., 2016): although this is unlikely to be an issue for Leicester, where fresh NO_x emissions dominate, it may be significant for Penlee Point. Concentrations of PAN in clean oceanic air masses are typically lower than 300-400 ppt (Roberts et al., 1996; Heard et al., 2006); with a conversion factor of PAN on a molybdenum converter of 80-95% (Williams et al., 1998; Steinbacher et al., 2007), it can be estimated that the interference was of the order of \sim 300 ppt. In Weybourne, NO₂ was measured with a dualchannel chemiluminescent instrument using a UV-LED photolytic converter (Air Quality Design Inc., Golden, CO, Reed et al., 2016).

Aerosol composition was measured in 2014 in Leicester and in 2015 at Weybourne but not during the other periods. A PM10 Leckel SEQ47/50 sequential sampler was used in Leicester (Hama *et al.*, 2017), and a multistage Sierra-type cascade impactor was used in Weybourne. Soluble ionic components of the filters from both sites were extracted and analysed by ion chromatography, as described in Baker *et al.* (2007). Details of the aerosol samplers and analytical procedures are given in the Supplementary Information

(SI. 2). All measurements were made at \sim 5 m above ground; the data were processed and analysed using R and the open air package (Carslaw and Ropkins, 2012).

3 | RESULTS

3.1 | Overview

The highest concentrations of ClNO₂ were typically observed between 00:00 and 04:00 GMT. The median ClNO₂ concentration at the different sites during this period varied between the detection limit (4.2 ppt) and 139 ppt (15-min averages, Table 1); the maximum concentration was 1,100 ppt, measured in Weybourne in July 2015. These observations are consistent with previous reports (e.g., Osthoff et al., 2008; Mielke et al., 2011; Riedel et al., 2012). In Europe, peak concentrations between 50 and 800 ppt were observed in Kleiner Feldberg (Germany) and in London (United Kingdom) by Phillips et al. (2012) and Bannan et al. (2015). More recently, Bannan et al. (2017) reported peak concentrations of up to 65 ppt in Weybourne (United Kingdom).

Median particulate chloride concentrations of up to $1.3 \, \mu g/m^3$ and $2.3 \, \mu g/m^3$ were measured in Leicester and Weybourne, respectively. Chloride was well correlated ($r^2 = 0.81$ –0.99) with Na⁺ at both sites (Figure 1, left). In Weybourne, the measured Cl⁻/Na⁺ mass ratio was 1.87, very close to the average seawater ratio of 1.8 (Stumm and Morgan, 1995). The size-resolved measurements of aerosol composition at Weybourne showed that the percentage of total Cl⁻ in sub-micron aerosol was 5–19% (median = 12%), indicating that most Cl⁻ was in the coarse fraction (e.g., sea

^b Detection limit (DL) = $0.2 \,\mu\text{g/m}^3$.

^c Particle diameter <10 μm (PM10) for the Leicester data and >0.01 μm for the Weybourne data.

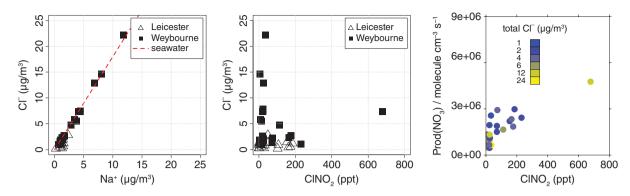


FIGURE 1 Measured particulate Cl^- versus Na^+ (left) and $ClNO_2$ (middle); the red dashed line in the left panel is the average chloride/sodium mass ratio in seawater (1.8). NO_3 production rate (from measured O_3 and NO_2) versus $ClNO_2$, coloured by measured particulate chloride (right). Data from Leicester and Weybourne, averaged to the sampling intervals of the aerosol impactors

salt). In Leicester, the measured Cl⁻/Na⁺ mass ratio was 1.32, consistent with the dechlorination of aerosol during transport from the ocean (Yang *et al.*, 2011; Saiz-Lopez and von Glasow, 2012). In both Weybourne and Leicester, the measured Mg²⁺/Na⁺ mass ratio was 0.12–0.13, equal to the average seawater ratio (Jordan *et al.*, 2015). Aerosol composition was not measured in Penlee Point, but the observatory is on a headland ~30 m from the water's edge, and sea spray is always abundant. These data strongly suggest that sea salt aerosol was the main source of Cl⁻ not only at the two coastal sites but also in Leicester (~200 km from the ocean), providing enough surface area for reaction 5 to occur.

The mean ClNO₂ nocturnal maxima in Leicester and Weybourne were not correlated with particulate Cl⁻ (Figure 1, middle and right), which suggests that the concentration of chloride was not, in general, a limiting factor for the formation of ClNO₂.

The complete time series of the measured species are given in the Supplementary Information (SI. 3). In this paper, the observational dataset was aggregated and analysed in two ways: (a) data collected in the same place but in different seasons and (b) data collected in different places and in the same season. The aim is to investigate the seasonal and geographical patterns of CINO₂ and of its precursors, rather than the short term (e.g., day-to-day) variability.

3.2 | Seasonal variability

Observations in Leicester were made in all seasons and provide information on the seasonal variability of CINO₂. A seasonal cycle is apparent (Figure 2), consistent with the observations made by Mielke *et al.* (2016) in Calgary (Canada). The highest median CINO₂ concentrations were observed in spring (110 ppt) and in winter (50.5–139 ppt) and the lowest in summer (15.4 ppt) (Table 1).

The nocturnal levels of NO₂ were similar in all seasons, with median concentrations of 6.8–9.0 ppb in spring/winter and 8.6–13.1 ppb in summer/autumn. The median nocturnal O₃ concentrations showed a larger seasonal range and were higher during spring/winter (22.1–24.2 ppb), compared to

summer/autumn (8.4–17.0 ppb). This suggests that, on average, the variability of O_3 was more influential than that of NO_2 in driving the seasonal changes of $CINO_2$ in Leicester. Furthermore, the NO_3 production rate ($k_1[NO_2][O_3]$) showed a weak but positive correlation with $CINO_2$ (Figure 1, right), which points to reaction 2 as the rate-limiting step for $CINO_2$ formation.

The duration of the daylight period, which changes with the season, can have an effect on the concentration of $CINO_2$ as its formation occurs in the dark. Autumn and spring have similar daylight periods but different levels of $CINO_2$, O_3 and NO_2 (Figure 2), suggesting that changes in the chemical precursors are more important. Several processes involved in the production of $CINO_2$ have temperature dependencies. The mean temperature in Leicester was 6 °C in winter and 15 °C in summer: simulations with a simple box model (section 3.4 and SI. 4) showed that the temperature difference impacts $CINO_2$ concentrations by 12% or less, which suggests a limited impact of temperature on the seasonal variability of $CINO_2$.

3.3 | Geographical variability

Observations were made in two different locations in spring (Leicester and Penlee Point) and in summer (Leicester and Weybourne), thus providing information on the geographical variability of ClNO₂ in the same season.

The interquartile range of ClNO₂ concentrations during spring was 50.1--150 ppt in Leicester and <DL-6.3 ppt in Penlee Point. It must be noted that the observations made in Penlee Point were highly skewed: the mean concentration was 66.1 ppt (Figure 2), and the median was 1.7 ppt (i.e., below the detection limit). This was mostly due to a single night (April 30, 2015) characterized by northeasterly winds, when the concentration of ClNO₂ reached 922 ppt. For most of the measurement period, southwesterly winds transported clean air masses from the Atlantic Ocean to Penlee Point, and the concentration of ClNO₂ was below 6.3 ppt. Although the median O_3 concentrations were $\sim 40\%$ lower in Leicester than in Penlee Point, the median NO_2

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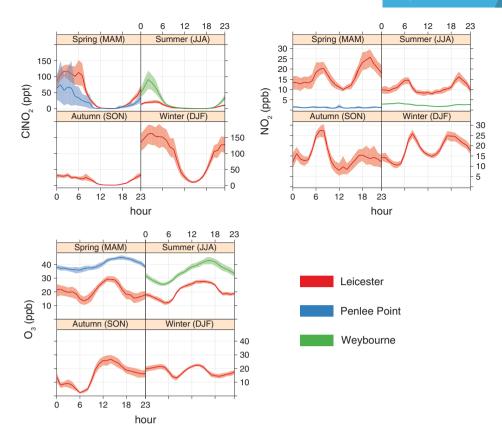


FIGURE 2 Average diel profiles of $CINO_2$, NO_2 and O_3 in different seasons and locations. The shaded areas are the 95% confidence intervals of the mean. The median and quartile concentrations are given in Table 1

concentrations were a factor of 10 higher, leading to higher ClNO $_2$ concentrations in Leicester (Table 1). This result is consistent with the spring observations previously made by Bannan *et al.* (2017) in Weybourne: although O $_3$ concentrations were higher (45–50 ppb) than in Leicester, the mean ClNO $_2$ concentration was about a factor of 6 lower (\sim 15 ppt) due to lower NO $_x$ levels in Weybourne.

During summer, the median O_3 concentrations were \sim 40% lower and the median NO_2 concentrations a factor of 3 higher in Leicester, compared to Weybourne. Contrary to springtime, lower O_3 and higher NO_2 resulted in lower $CINO_2$ concentrations in Leicester (15.4 ppt) than in Weybourne (21.5 ppt). This suggests that the concentration of $CINO_2$ in the two locations was, on average, controlled by different mechanisms, with NO_2 being the more influential parameter in Weybourne and O_3 in Leicester.

The Leicester and Weybourne data can be compared to the other European measurements, which were obtained during summer (Phillips *et al.*, 2012; Bannan *et al.*, 2015). In London, O₃ concentrations (~20 ppb, Bohnenstengel *et al.*, 2015) were similar to those observed in Leicester and lower than those observed in Weybourne, while NO₂ concentrations were generally higher (10–15 ppb), which resulted in higher mean concentrations of ClNO₂ (~150 ppt, Bannan *et al.*, 2015). The pattern was less clear for Kleiner Feldberg, likely because air masses of continental origin were also

sampled (Phillips *et al.*, 2012), and under those conditions, it cannot be assumed that particulate chloride was not a limiting factor.

3.4 | Discussion

The measurements presented here, and those by Bannan *et al.* (2015; 2017), suggest that $CINO_2$ levels in the United Kingdom are controlled by the gas-phase precursors (O_3 and NO_2), rather than by the uptake and reaction of N_2O_5 with particulate chloride. To investigate the relationship between $CINO_2$, O_3 and NO_2 in more detail, a simple box model was developed. The model includes reactions 2–5, plus:

$$NO + O_3 \rightarrow NO_2 + O_2 \tag{6}$$

$$NO + NO_3 \rightarrow 2 NO_2 \tag{7}$$

$$RH + NO_3 \rightarrow RO_2$$
 (8)

where RH represent a generic hydrocarbon (with the rate coefficient of NO_3 + propene and initial concentration of 1 ppb) and RO_2 the corresponding peroxy radical. The model was run 96 times, each time with a different initial concentration of NO_2 (0.1–50 ppb) or O_3 (1–100 ppb). The uptake coefficient of N_2O_5 was set to 0.01 (Bertram and Thornton, 2009) and the $CINO_2$ yield to 0.5 (Roberts *et al.*, 2009). The model was run for 4 hr at constant temperature

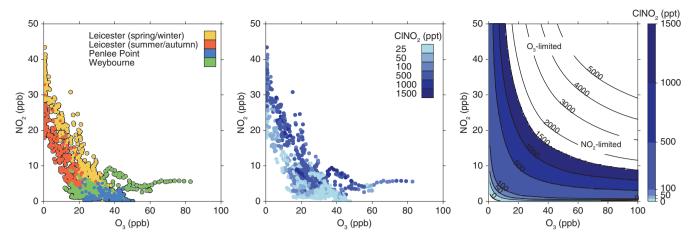


FIGURE 3 Measured O₃ and NO₂ coloured by measurement period and site (left). Measured O₃ and NO₂ coloured by measured ClNO₂ (middle). Modelled ClNO₂ versus initial O₃ and NO₂ concentrations, as calculated with the simple box model (right)

(285 K, mean of all measurement periods) to simulate the formation of ClNO₂ during the night.

The measured and modelled concentrations of ClNO2 are plotted in Figure 3 (middle and right) as a function of O₃ and NO₂. Although the model is very simple, it is able to capture the main features of the observations. Running the model with different temperature (278 K), N₂O₅ uptake coefficient (0.005) or ClNO₂ yield (0.2) does not substantially alter the shape of the envelope of model results (see Supplementary Information, SI. 4). Two distinct regions with lower ClNO₂ concentrations can be identified (Figure 3, right): one is NO₂-limited, and the other is O₃-limited. The observations at Penlee Point and Weybourne fall predominantly in the regime where ClNO₂ concentrations are constrained by the availability of NO2, while those in Leicester fall mainly in the O₃-limited regime (Figure 3, left). The highest ClNO₂ concentrations are obtained when the chemical conditions fall between these two regimes, which is the case for some of the observations from Leicester and Weybourne.

This is consistent with the analysis presented above (sections 3.2 and 3.3). As Leicester is largely O₃-limited, the seasonal variability of ClNO₂ is controlled by the variability of O₃ concentrations, and the variability of NO₂ has less impact. Lower O₃ concentrations also explain why the ClNO₂ concentrations observed in Leicester during summer were, on average, lower than in Weybourne, even though NO_2 levels were a factor of 3 higher (Figure 2). On the other hand, ClNO₂ concentrations in Penlee Point and Weybourne were mostly constrained by the availability of NO2, rather than of O₃. Although the two locations cannot be directly compared as the measurements were made in different seasons, Weybourne usually experienced higher NO2 concentrations than Penlee Point, resulting in higher ClNO₂ concentrations even under similar O₃ levels (Figure 3).

As shown previously (Figure 1, right), both the Leicester and the Weybourne data suggest that NO₃ production (2) was the rate-limiting step in the sequence of reactions leading to ClNO₂ formation (2–5), which is consistent with the existence of O₃-limited and NO₂-limited chemical regimes. Extrapolating the results of this simple box model, it can be concluded that, under conditions of abundant sea salt aerosol, the formation of ClNO₂ can be either O₃-limited or NO₂-limited. These conditions are likely to occur in most of the United Kingdom and Northern Europe, as well as in other regions across the world.

CHEMICAL TRANSPORT MODELLING

The chemical transport model GEOS-Chem (v10-01, http:// www.geos-chem.org, Bey et al., 2001) was run for the period of the observations driven by offline GEOS-FP meteorology with 72 levels, extending to 0.01 hPa. The model includes HO_x-NO_x-VOC-O₃-BrO_x chemistry and has been recently updated to include Cl, Br, I chemistry (Schmidt et al., 2016; Sherwen et al., 2016a; 2016b), as described in detail by Sherwen et al. (2017). The chlorine chemical mechanism in the model includes further reactions of chlorine and bromine with organics compounds, ClNO2 formation following N₂O₅ uptake on sea salt, and heterogeneous iodine cycling to produce ICl, IBr. Reaction probabilities for N₂O₅ uptake follow Evans and Jacob (2005) for sea salt aerosol (producing ClNO₂) and non-sea salt aerosol (producing HNO₃).

The model was run at a medium global resolution $(2^{\circ} \times 2.5^{\circ})$ for 2 years for the seasonal comparisons, following a 1-year "spin up". A higher-resolution "nested" simulation $(0.25^{\circ} \times 0.3125^{\circ})$, Sherwen et al., 2017) was also run for two contrasting observational periods—spring and summer—following a 2-week "spin-up" for each analysis period. The average surface concentrations for the grid box closest to the measurements sites are compared with the measured average diel profiles of ClNO2, NO2 and O3 in

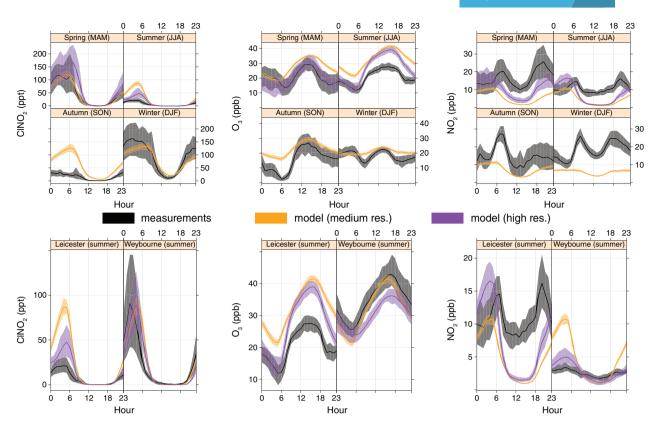


FIGURE 4 Modelled and measured average diel profiles of ClNO₂, NO₂ and O₃, divided by season (in Leicester, top) and by location (in summer, bottom). The shaded areas are the 95% confidence intervals of the mean

Figure 4. Additional plots are given in the Supplementary Information (SI. 5).

The medium-resolution version GEOS-Chem of $(2^{\circ} \times 2.5^{\circ})$ was able to reproduce the diel ClNO₂ cycle but failed to reproduce the seasonal cycle as observed in Leicester. The model showed good agreement with the measurements during spring and winter but overestimated CINO2 during summer and autumn (Figure 4, top). However, the model showed good agreement with the summertime measurements of ClNO₂ at Weybourne (Figure 4, bottom), as previously reported by (Sherwen et al., 2017). The highresolution version of GEOS-Chem $(0.25^{\circ} \times 0.3125^{\circ})$ calculated roughly the same concentrations of ClNO2 as the medium-resolution version for spring in Leicester and summer in Weybourne, thereby agreeing reasonably well with the observations. Although the high-resolution model showed improved agreement for summer in Leicester, it still overestimated the measurements of ClNO₂ by a factor of \sim 2 (Figure 4). The comparison between modelled and measured O₃ and NO₂ suggests that the improved performance of the higher-resolution model was due to a better representation of nocturnal O₃ during summer (ClNO₂ being mostly O₃limited in Leicester, section 3.4).

The GEOS-Chem results complement the work by (Sherwen *et al.*, 2017), which focused on the observations at Weybourne in summer 2015 and explored the regional effects of halogen chemistry. The expanded set of model results and measurements discussed here, covering a larger

temporal and geographical range, highlight the challenges and possible future directions for chemical transport models. A key issue is the high bias in modelled surface ozone, a known problem for GEOS-Chem (e.g., Travis *et al.*, 2016). The simulation of the seasonal cycle of ClNO₂ is a significant challenge for chemical transport models. The models need to be able to accurately simulate the nocturnal concentrations of O₃, NO₂, NO₃, N₂O₅ and effective Cl⁻ surface area. It is likely that significant uncertainties in nocturnal chemical and physical processes lead to resolution dependencies in current chemical transport models. The results presented here suggest that there is a tendency for lower resolution models to overestimate ClNO₂ concentrations, which may not be apparent in higher-resolution models.

5 | CONCLUSIONS

Measurements of ClNO₂ and its precursors (O₃, NO₂, particulate chloride) were taken in 2014–2016 at Leicester, Penlee Point and Weybourne, three sites in different parts of the United Kingdom. The measurements covered all seasons, and ClNO₂ was observed at all locations and in all seasons. The median nocturnal concentrations of ClNO₂ ranged from below the detection limit (4.2 ppt) to 139 ppt. The highest concentration (1,100 ppt) was observed in Weybourne in July 2015.

ClNO $_2$ displayed a clear seasonal cycle, with maxima in spring and winter. Significant differences were observed between different locations during the same season. Analysis of the dataset indicates that sea salt aerosol was the main source of particulate chloride (Cl $^-$). Observational evidence suggests that production of NO $_3$ (2) was the rate-limiting step in the formation of ClNO $_2$, rather than the aerosol uptake of N $_2$ O $_5$ and its reaction with Cl $^-$. In general, the data indicate that the observed variability of ClNO $_2$ was driven by the availability of either O $_3$ or NO $_2$, depending on the season and the location. The seasonal and geographical variability of ClNO $_2$ can be explained in terms of O $_3$ -limited and NO $_2$ -limited chemical regimes, broadly reproducible with a simple chemical box model.

The GEOS-Chem chemical transport model at medium resolution $(2^{\circ} \times 2.5^{\circ})$ could reproduce the diel cycle of CINO₂, but not the observed seasonal cycle and geographical distribution. The likely reason is that GEOS-Chem overestimated the concentrations of CINO₂ precursors, particularly of O₃. Simulations at higher resolution $(0.25^{\circ} \times 0.3125^{\circ})$ improved the fidelity of the simulation, although CINO₂ was still overestimated in Leicester during summer. Future work is clearly necessary to improve the representation of night-time chemistry and dynamics in models.

This work, as well as Phillips *et al.* (2012) and Bannan *et al.* (2015; 2017), shows that the presence of ClNO₂ is widespread across Northern Europe: where there is sufficient particulate chloride and the total aerosol surface area is large enough (e.g., where sea salt aerosol is abundant), ClNO₂ chemistry is expected to occur. These conditions are typical of the United Kingdom and large areas of Northern Europe; they are also likely to be encountered in other parts of the world and are particularly relevant for coastal urbanized regions and megacities.

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Conflict of interests

The authors declare no conflicts of interest.

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SUPPORTING INFORMATION

Additional supporting information may be found online in the Supporting Information section at the end of the article.

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